

Ferromagnetism in oxide and dichalcogenide based diluted magnetic semiconductors

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Ferromagnetism in Oxide and Dichalcogenide Based Diluted Magnetic Semiconductors

Xiang Ding

A thesis in fulfilment of the requirements for the degree of

Doctor of Philosophy



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Faculty of Science

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Over the past decades, the world has witnessed the rapid development of semiconductor industry according to Moore's law. However, the pace of advancement has slowed in recent years and is already nearing its end due to the limits of miniaturization. The industry is in urgent need for new materials and device designs, for example, spintronics. Spintronics is a multifunction al device that utilizes spin degrees of freedom of electrons. The key issue in spintronics is the manipulation of spins. Diluted magnetic semiconductors are able to combine both semiconducting and magnetic properties into spintronics. Room-temperature ferromagnetism has been predicted to achieve in transition metal oxides like TiO_2 and transition metal dichalcogenides such as MoS_2 and WS_2 , which are the subjects of this dissertation.

In the first part of this dissertation, 5 at% Co-doped TiO₂ thin films were grown on LaAlO₃ and SrTiO₃ substrates by pulsed laser deposition. The growth mechanism and magnetic properties of the thin films were investigated in detail. It was found that although only anatase phase was observed in X-ray diffraction, the epitaxial films can be either fully strained or fully relaxed based on the choice of deposition substrates. The Co dopants tend to segregate near the interface, but the size and distribution of the nanoclusters depend on the deposition parameters. Higher temperatures and slow growth rate lead to bigger cluster size and greater distance to the interface. The highest magnetic moments reaches 4.73 μ_B /Co at room temperature, supported by the first-principles calculations considering spin-orbit coupling. Polarised neutron reflectometry measurement reveals that Co clusters and possible surrounding defects are the main magnetic sources.

The second part of this dissertation focuses on defect engineering in WS_2 and MoS_2 . Both WS_2 annealed in reducing gas and Nd-doped MoS_2 show enhanced room-temperature ferromagnetism. The magnetic properties of the annealed WS_2 powders are governed by the sulfur vacancies, which significantly increase in density with increased annealing temperature. In the Nd-doped MoS_2 crystals, the exchange coupling of Nd^{3+} ions and sulfur vacancies introduced by ion implantation can form magnetic polarons that determine the magnetic properties.

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LIST OF PUBLICATIONS

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2. <u>X. Ding</u>, and J. B. Yi. "Synthesis and Applications of 2d Transition Metal Dichalcogenides." In Functional Materials and Electronics, *Apple Academic Press*, 2018, pp. 1-33.

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R. Zheng, and X. Wang. "Intrinsic and spatially nonuniform ferromagnetism in Co-doped ZnO films." *Physical Review B*, 96(10) (2017): 104423.

5. X. Luo, L.T. Tseng, Y. Wang, N. Bao, Z. Lu, <u>X. Ding</u>, R. Zheng, Y. Du, K. Huang, L. Shu, and A. Suter. "Intrinsic or Interface Clustering-Induced Ferromagnetism in Fe-Doped In₂O₃-Diluted Magnetic Semiconductors." *ACS applied materials & interfaces*, 10(26) (2018): 22372.

6. S. Ahmed, P. Viboon, <u>X. Ding</u>, N. Bao, Y. H. Du, T. S. Herng, J. Ding, and J. B. Yi. "Annealing effect on the ferromagnetism of MoS₂ nanoparticles." *Journal of Alloys and Compounds*, 746 (2018): 399.

7. S. Ahmed, <u>X. Ding</u>, P.P. Murmu, N. Bao, R. Liu, J. Kennedy, J. Ding, and J. B. Yi. "Magnetic properties of Co doped WSe₂ by implantation." *Journal of Alloys and Compounds*, 731 (2018): 25.

ABSTRACT

Over the past decades, the world has witnessed the rapid development of semiconductor industry according to Moore's law. However, the pace of advancement has slowed in recent years and is already nearing its end due to the limits of miniaturization. The industry is in urgent need for new materials and device designs, for example, spintronics. Spintronics is a multifunctional device that utilizes spin degrees of freedom of electrons. The key issue in spintronics is the manipulation of spins. Diluted magnetic semiconductors are able to combine both semiconducting and magnetic properties into spintronics. Room-temperature ferromagnetism is expected in transition metal oxides like TiO_2 and transition metal dichalcogenides such as MoS_2 and WS_2 , which are the subjects of this dissertation.

In the first part of this dissertation, 5 at% Co-doped TiO₂ thin films were grown on LaAlO₃ and SrTiO₃ substrates by pulsed laser deposition. The growth mechanism and magnetic properties of the thin films were investigated in detail. It was found that although only anatase phase was observed in X-ray diffraction, the epitaxial films can be either fully strained or fully relaxed based on the choice of substrates. The Co dopants tend to segregate near the interface, but the size and distribution of the nanoclusters depend on the deposition parameters. Higher temperatures and slow growth rate lead to bigger cluster size and greater distance to the interface. The magnetic moments can reach $3.5 \mu_{\rm B}/\text{Co}$ at room temperature, supported by the first-principles calculations considering spin-orbit coupling. Polarised neutron reflectometry measurement reveals that Co clusters and possible surrounding defects are the main magnetic sources.

The second part of this dissertation focuses on defect engineering in WS₂ and MoS₂. Both WS₂ annealed in reducing gas and Nd-doped MoS₂ show enhanced room-temperature ferromagnetism. The magnetic properties of the annealed WS₂ powders are governed by the sulfur vacancies, which significantly increase in density with increased annealing temperature. In the Nd-doped MoS₂ crystals, the exchange coupling of Nd³⁺ ions and sulfur vacancies introduced by ion implantation can form magnetic polarons that determine the magnetic properties.

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Chapter 1 Introduction

1.1 Background

In semiconductor industry, the number of transistors in a dense integrated circuit is expected to increase twice in every two years. This is the famous Moore's law predicted by Gordon Moore. Moore's prediction has been proved accurate for several decades and served as a long-term guide in the semiconductor industry. However, the doubling has already started to falter in recent years due to unavoidable heat problem and other fundamental limits. The International Technology Roadmap for Semiconductors (ITRS) in 2016, also known as the final roadmap, has officially ended the long pursuit of Moore's law. Next-generation semiconductor chips demand the development of novel semiconductor materials, especially in low dimensions. Diluted magnetic semiconductors (DMSs), which can combine semiconducting and magnetic properties in a single device, are of great importance and attract massive research interests.

In general, DMSs are semiconducting matrix materials doped with low-concentration transition or rare-earth metals. The dopants, such as Mn, Co, Fe and Gd, have uncompensated electron spins in partially occupied 3d or 4f orbitals and can induce longrange ferromagnetism into the matrix materials. The combination of semiconducting and magnetic properties leads to potential integration of logic semiconductors and magnetic information-storage capability in a single electric component. DMSs can be employed as critical component in spintronics, which manipulates the spin degrees of freedom of electrons, rather than mere charge degrees of freedom in traditional electronics. These devices possess potential advantages of higher integration density, higher processing speed, and lower power consumption. To date, the most mature DMS system is Mn doped semiconductors, especially (Ga,Mn)As. (Ga,Mn)As has been extensively studied and is considered as the prototype DMS, providing textbook magnetic and thermodynamic properties. The methods used to fabricate (Ga,Mn)As and other DMS materials are usually in conditions far beyond thermal equilibrium, such as pulsed-laser deposition (PLD), molecular beam epitaxy (MBE), and sputtering. The induced ferromagnetism in (Ga,Mn)As is mediated by carriers (holes) and can be explained by mean-field Zener model. As the hole density in (Ga,Mn)As can be tuned via extrinsic means, such as electric field, light and strain, this provides researchers possible methods to manipulate the magnetic properties, including magnetization, Curie temperature and magnetic anisotropy, without changing temperature. Upon the successful modulation of the magnetic properties, spintronic devices like (Ga,Mn)As based memory cell have been designed and constructed. However, the low Curie temperature (~ 200 K) severely limits its potential applications.

Practical applications require DMSs materials with ferromagnetism persisted to above room temperature. Theoretical studies have predicted that room-temperature ferromagnetism is expected in some nitride and oxide systems, such as GaN and ZnO. This leads to massive research exploration in the following decade in the field of diluted magnetic semiconducting oxides (DMSO) and most research has been carried out on TiO₂ and ZnO. Co doped TiO₂ has demonstrated to be a remarkable DMS system with roomtemperature ferromagnetism in both rutile and anatase forms in a variety of morphologies, such as thin films, nanoparticles, and nanorods. However, the origin of the ferromagnetism in the system has not been fully understood. In the meantime, there has been renewed interest in 2D transition metal dichalcogenides (TMDCs) due to their unique band structure, for example, spin-orbit splitting and inherently coupled spin and valley carriers.

1.2 Research objectives

A critical issue in Co doped TiO₂ is the puzzle relationship between growth conditions and Co distribution in TiO₂. Using similar techniques, segregation of Co, usually in the form of metallic nanoclusters, are observed in epitaxial thin films as often as homogeneously distributed Co atoms. The uncertainty of growth mechanism also leads to strong fluctuations in the magnetic properties. For instance, the magnetic moments observed in Co doped TiO₂ significantly from 2 to 0.01 μ _B/Co reported by different research groups. In addition, the relatively high magnetic moment is only achieved by magnetometer measurement and there is no evidence to directly observe the high magnetic moment of the magnetic element and therefore, the mechanism for the high magnetic moment is still unclear. In this dissertation, 5% Co doped TiO_2 thin films were deposited on LaAlO₃ (001) and SrTiO₃ (001) substrate using a pulsed laser deposition system (PLD). With the delicate control of deposition parameters, we achieved the microstructure of metallic Co nanoclusters embedded in TiO₂ matrix with extremely high magnetic moment over 3.5 μ_B/Co at room temperature. This result is striking since it is well-known that bulk Co has a magnetic moment of only 1.72 μ_B/Co . We proposed the theory of confinement effect induced high magnetization of clusters. Moreover, from the fundamental science, it is challenging to achieve a large magnetization at high temperature. However, the thin film we prepared has a Curie temperature close to 1000 K. The embedded Co clusters are structurally, chemically and magnetically protected by the oxide matrix, enabling more stable magnetic properties. In addition, the structure and magnetic properties of the thin films can be controlled by tuning PLD growth parameters.

2D transition metal dichalcogenides (TMDs) have attracted wide interest due to many extraordinary properties and potential applications. MoS_2 and WS_2 are two common

TMD materials that have been extensively investigated. Especially, MoS₂ is almost the second most studied 2D material next to Graphene. Hence, in this thesis, the research is focused on MoS₂ and WS₂. Although pure low-dimensional MoS₂ and WS₂ have exhibited room temperature ferromagnetism due to vacancies or edge states, the magnetic study of TMDCs are still in its infantry and the manipulation of magnetic properties through doping and other defect engineering methods remain rarely explored. This dissertation aims to employ defects engineering to manipulate magnetic properties of TMD materials via heat treatment and ion implantation. It is revealed that defects can lead to the enhancement of room-temperature ferromagnetism and ferromagnetic properties are strongly associated with defect concentrations. Furthermore, during the investigation, we first discovered negative irreversible magnetization, usually observed in random field systems, in TMD materials.

The research outlines are shown below:

- To prepare 5 at% Co doped anatase epitaxial TiO₂ thin films on LaAlO₃ and SrTiO₃ (001) substrates via pulsed laser deposition and study the structure and magnetic properties of the thin films.
- To further prepare Co doped TiO₂ thin films under various conditions and study how deposition parameters affect films' structure and magnetism.
- To examine the vacancy defects and ferromagnetic ordering in WS₂ powders treated via annealing.
- To study the doping impact of Nd on the magnetic properties of MoS₂ crystals by ion implantation.

1.3 Thesis outline

This thesis consists of seven chapters. An overview of previous studies about diluted magnetic semiconductors and the objectives of this thesis are given in chapter 1. Chapter 2 presents a literature review about current development of spintronics and three flagship examples of diluted magnetic semiconductors. Chapter 3 gives a brief introduction of the methods used to fabricate and characterize the samples. Chapter 4 presents a detailed investigation on the microstructure and magnetic origin of PLD grown TiO₂ thin film doped with cobalt. Chapter 5 further explores the growth mechanism of TiO₂ thin film and the relationship between film structure and magnetic properties. Chapter 6 depict the modulation of magnetic properties in WS₂ powders and MoS₂ single crystals via annealing and ion implantation, respectively. Finally, a summary of the results and findings can be found in chapter 7.

Chapter 2 Literature Review

2.1 Spintronics

Spintronics refers to an advanced electronic concept that mainly focuses on the intrinsic electron spin, magnetic moment and other associated physics phenomena. Spintronic devices utilize the unique spin degrees of freedom of electrons, rather than the charge degrees of freedom in traditional electronics. These devices possess potential advantages of higher integration density, higher processing speed, and lower power consumption [1]. In theory, ideal spintronic devices are operational even without applying an electric current, which unfortunately has not been accomplished yet.

The most successful industrial application of spintronics so far appears in the field of data storage, owing to the discovery of giant magnetoresistance (GMR) in multilayer thin films [2]. GMR is first unveiled in thin films made up of alternating metallic ferromagnetic and nonmagnetic layers. The electrical resistance of the thin films, dependent on the spin transport behaviors, is determined by the alignment of spins within the ferromagnetic layers. Later, a GMR-based device, spin valve, is demonstrated with a structure in which a nonmagnetic layer is sandwiched between two ferromagnetic layers [3]. Only one of the two ferromagnetic layers is spin free, while the other layer is pinned so its magnetization cannot be changed by an external magnetic field. The resistance of spin valve is controlled by the relative orientation of magnetizations in two ferromagnetic layers (parallel or antiparallel). The schematic diagram of a spin valve is shown in figure 2.1.

Read heads in hard disk drives were first benefited from GMR technology, enabling a significant improvement of the areal density. With the advantages of high sensitivity and large scale of integration, a wide range of applications have placed GMR devices as the

default magnetic sensors, such as integrated circuit (IC) current monitoring, automotive applications, non-destructive evaluation and biosensors for life science applications [4-7]. For instance, GMR effect can be used for biosensors by detecting the stray field of superparamagnetic nanoparticles attached to a target molecule hybridized to the specific proteins that are immobilized on the sensor surface [8].

Another major advanced application for GMR is magnetoresistive random-access memory (MRAM). For the past two decades, static RAM (SRAM), dynamic RAM (DRAM), and Flash have dominated the entire solid-state memory. While most SRAM applies six transistors in a flip-flop architecture, DRAM (volatile) and Flash (non-volatile) rely on stored electric charge. In contrast, MRAM utilizes the magnetic hysteresis associated with the ferromagnetic materials. The majority of current MRAM technologies are based on magnetic tunnel junction (MTJ) devices. Similar to spin valve, a typical MTJ also have two separated ferromagnetic layers, but the tunnel barrier between two layers is thin enough (< 2 nm) to let electrons pass through by quantum tunneling. The resistance of the device depends on the relative orientation of the magnetization in the free and fixed layers. For most materials, the resistance is lower when the two layers are aligned, similar to figure 2.1. To create an operational memory array, each memory cell consists of one MTJ which is grounded via a transistor. All current commercial MRAMs employ this architecture, known as 1-transistor-1-magnetic-tunnel-junction (1T-1MTJ) MRAM, as shown in figure 2.2 [9]. While the first generation of MRAM is field-switched, the second MRAM generation triggers magnetic switching by spin transfer torque (STT), enabling smaller MTJ devices for higher storage density. Recent STT-MRAM can achieve a high density of 4 Gb with compact cell structure (90 nm pitch) through optimizing parasitic resistance [10].

Low resistance state

High resistance state

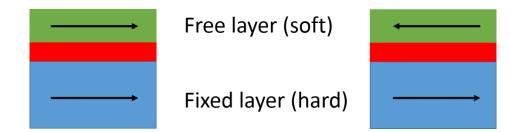


Figure 2.1 A schematic diagram of a pseudo spin valve. The fixed and free layer are magnetically hard and soft, respectively. When the magnetic layers are aligned (left), the electrical resistance is lower than when they are antiparallel (right).

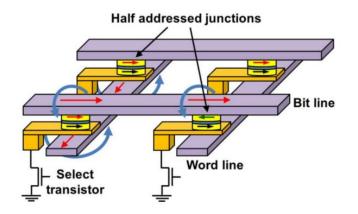


Figure 2.2 A schematic diagram of an array of MRAM cells [9].

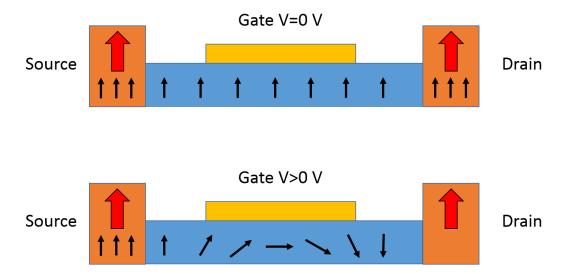


Figure 2.3 A schematic diagram of a spin-FET device. Spin can only transport through semiconductor channel with zero gate voltage (on state). Applying a gate voltage blocks the spin current and turns the device off.

The successful utilization of GMR has intrigued profound research interests in exploring

spintronics with more complex structures and functionalities, such as spin-field effect

transistor (spin-FET), spin-light emitting diode (spin-LED) [11-14]. However, unlike GMR-based devices which have been quickly commercialized after the discovery, the realization of these spintronics at room temperature has not yet to be accomplished due to many unsolved problems. Take spin-FET as an example, the operation of the device follows three steps: spin injection through the source contact, spin transportation via the semiconductor channel, and spin detection from the drain contact. None of above steps have been perfectly solved so far and researchers around the world are still working toward improvement [15]. Figure 2.3 presents a conceptual structure of a spin-FET [16], which consists of two ferromagnetic contacts connected by a semiconductor channel. Electrons are polarized by the source contact and then injected into the channel. During the propagation, the spins precess due to spin-orbit interaction. The final current depends on the relative angle between the spin direction at the end of the channel and the magnetization direction of the drain contact. Since the spin direction at the end of the channel relies on the strength of the spin-orbit interaction in the channel, which can be modulated by external electric field, the current of the spin-FET can be controlled by gating.

2.2 Diluted magnetic semiconductors

Diluted magnetic semiconductors (DMS) can be served as a key material in spintronic devices. DMSs consist of a semiconducting matrix doped with a small percentage of magnetic impurities. The doping elements include transition and rare-earth metals (Mn, Co, Fe, Gd and Nd), which have uncompensated electron spins in the partially occupied 3d or 4f orbitals in the ground state. These dopants may induce magnetism into the semiconductors, leading to potential integration of logic semiconductors and magnetic information-storage capability in a single electric component. As the trend of

miniaturization of electronic products continues, such multifunctional elements with combined semiconducting and magnetic properties are of great importance. Moreover, DMSs provide a platform to the observation and investigation of a series of fundamental physics phenomena, such as tunneling anisotropic magnetoresistance, spin torque and spin Seebeck effect [17]. Figure 2.4 shows a simplified model of a DMS.

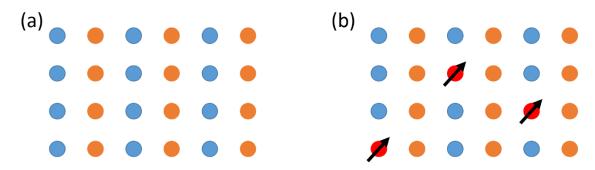


Figure 2.4 A schematic diagram of (a) a pure nonmagnetic semiconductor and (b) a diluted magnetic semiconductor.

According to the accumulated studies, most DMSs that support spontaneous polarization of spins can be divided into two categories. The first group is uniform DMSs, in which magnetically doped cations are randomly distributed within the semiconducting matrix. The most extensively investigated example in this group is Mn-doped III-V semiconductors. The spin-spin interactions in uniform DMSs are usually mediated by high concentration of carriers, either delocalized or weakly localized. The second class is heterogeneous DMSs, in which magnetic impurities are nonrandomly distributed. Magnetic dopants in these DMSs tend to aggregate in some particular nanoregions, such as interface or grain boundaries. To this class, it also exists a number of materials, in which magnetic properties are more related to various defects, rather than the doping regions. In both cases, the Curie temperature of heterogeneous DMSs appears obviously higher than uniform DMSs.

2.2.1 (Ga,Mn)As

Over the course of past decades, Mn-containing DMSs have been extensively investigated and related studies are the most mature. The flagship example is (Ga,Mn)As, which is widely considered as the prototype DMS, providing textbook magnetic and thermodynamic properties. GaAs is a mainstream semiconductor utilized in a variety of devices, including lasers and high-mobility transistors. As a consequence, (Ga,Mn)As holds great appeal if one can induce ferromagnetism by simply doping Mn into GaAs.

Generally, due to the low solubility of the magnetic dopants, dopants tend to aggregate and form dopant-rich nano compounds embedded in semiconducting matrix during the growth of DMSs at high temperatures. For instance, it was shown that at substrate temperature of 300-350 °C, MnAs nanoclusters with size in the range of 30-50 nm and height around 2-5 nm can be observed on the surface of pre-deposited GaAs buffer layer [18]. For (Ga,Mn)As with high Mn concentration (5 at%), the formation of MnAs nanocrystals can be characterized by an sharp transition from 2D to 3D features in reflection high-energy electron diffraction (RHEED), whereas the transition is rather gradual for low Mn doped GaAs (1 at%). Consequently, the growth of (Ga,Mn)As is more preferred under lower substrate temperature in the range of 200-300 °C and the approaches are commonly in the conditions far beyond thermal equilibrium, including pulsed-laser deposition (PLD), molecular beam epitaxy (MBE), and sputtering [19, 20].

More importantly, as the solubility of TM dopants in semiconductors is extremely low under thermal equilibrium, reducing the substrate temperature during epitaxy makes it possible to achieve ultra-high Mn concentrations (~10 at%). Another method to introduce a considerable density of doping elements is through ion implantation. In such cases, whether the high density of Mn ions can keep uniformly distributed in GaAs becomes a question. According to previous ab initio calculations [21, 22], a strong, short-range attractive interaction is expected between the nearest Mn neighbors. This confirms the tendency for Mn dopants to aggregate into small nanoclusters and the requirement of sophisticated low-temperature non-equilibrium fabrication methods to limit or totally eliminate phase separation. A number of advanced characterization methods indicate that the formation of secendary phases can be indeed suppressed by low-temperature epitaxy process. According to a state-of-the-art laser-assisted wide angle 3D atom probe technique, Mn dopants are uniformly distributed in $Ga_{0.963}Mn_{0.037}As$ along both the inplane and out-of-plane directions with 1 nm resolution [23]. This rules out the existence of large Mn aggregation inside GaAs or at the interface. However, the method cannot exclude the possibility of the formation of Mn dimmers or trimmers which are within the 1 nm resolution. In another study, high-resolution scanning tunneling microscopy (STM) was carried out to visualize electronic states of Mn dopants [24]. The STM images illustrate the wave functions of individual Mn dopant. Due to the unique topographic signatures, the figures not only show Mn dopants on the surface, but also those from the second and third subsurface layer. The result is in line with the fact that Mn ions can be uniformly distributed in (Ga,Mn)As without Mn aggregation.

In (Ga,Mn)As, Mn dopants substituted at Ga atom sites, serving both as acceptors and localized spins, the latter align with each other to give rise to long-range magnetic orderings. The valence state of Mn dopant in GaAs is Mn^{2+} and the configuration of its outermost d-shell electrons is half-filled d^5 [25, 26]. The half-filled valence d shell attributes to a variety of magnetic behaviors, depending on the way of interactions between neighbouring Mn ions. Direct exchange occurs when electron wave-functions of neighbouring Mn ions directly overlap, whereas indirect exchange refers to interactions mediated by carriers in GaAs.

To date, a number of models have been proposed to solve the mystery of ferromagnetism in DMSs. The mean-field Zener model is the most widely accepted model for the ferromagnetism in (Ga,Mn)As. Zener proposed the original model in the 1950s to address the ferromagnetism in transition metals [27]. According to Zener, three types of spin coupling, the direct exchange between half-filled d shells of neighbouring atoms, the exchange between inner d electrons and conduction electrons, and the Fermi kinetic energy of the conduction electrons are responsible for the magnetic properties of transition metals. The relative magnitudes of these types of spin coupling decide the final magnetic behavior in a specific metal. Nearly 50 years later, Dietl *et al.* discovered that the model can be extend to DMSs with mean-field approximation [28, 29]. In the new model, the free energy density can be divided into two parts, one related to the mere Mn spins without any carriers and another describes the free energy density of the carriers surrounding the Mn spins. Equilibrium magnetization is solved by minimizing the function of the free energy density with respect to the magnetization at given temperature and magnetic field.

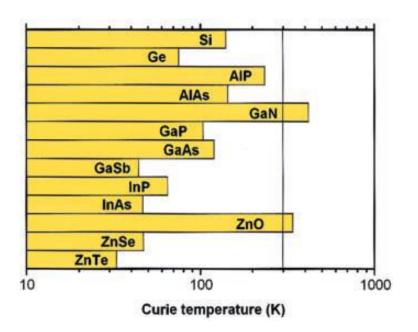


Figure 2.5 Theoretical Tc values for a variety of 5 at% Mn doping p-type semiconductors with hole density of 3.5×10^{20} /cm³ [30].

The limitation of the mean-field Zener model mainly originates from thermodynamic magnetization fluctuations, antiferromagnetic interactions, Kondo effect, and effects of disorder and localization [28] [31-34]. These limitations can eventually lead to an accumulated error of at most 25 at% in theoretical Tc. The Curie temperatures of various DMSs evaluated by the mean-field Zener model are illustrated in figure 2.5 [30]. The simulated Curie temperature of 5 at% Mn doped GaAs with hole density of 3.5×10^{20} /cm³ is ~120 K. Compared with experimental results, the calculation underestimates and overestimates the Tc value in low and high Ms range. It is also suggested by Zener modal that Curie temperature above room temperature can be achieved in (Ga,Mn)As containing 10 at% Mn in the absence of self-compensation. However, up to now the highest reported values, enhanced by post-annealing and nanostructure engineering, can only reach slightly above 200 K [35-38].

A key step towards the successful realization of spintronic devices is the manipulation of ferromagnetism in DMSs. As the hole density in (Ga,Mn)As can be tuned via extrinsic means, such as electric field and light, this provides us possible methods to manipulate the magnetic properties, including magnetization, Curie temperature and magnetic anisotropy, without changing temperature.

Gating, by applying an electric field to the semiconductor layer, is a highly mature technique in the semiconductor industry nowadays. For example, gating allows field-effect transistor (FET) to electrically switch between on and off states. Therefore, the possibility to turn on and off magnetic states through gating holds significant research interests since it is compatible with the existing commercial techniques. On early stage, a heterostructure consist of Mn doped GaAs and AlGaAs sandwiched by two GaAs layers was designed [39]. The heterostructure exhibits paramagnetic ground state due to low density of holes. By applying a negative gate bias, holes are electrostatically induced to

(Ga,Mn)As layer, resulting in a phase transition from paramagnetism to ferromagnetism. The operation temperature lays in the range of 100-117 K.

One big obstacle to modulate the magnetism in (Ga,Mn)As is the small modulation ratio. (Ga,Mn)As thin film has thickness dependent conductivity which transforms to an insulator with thickness thinner than 5 nm. However, employing thicker channel layer in order to lower conductivity is detrimental to the modulation ratio. One solution is adding a Al₂O₃ layer with high dielectric constant between (Ga,Mn)As and Au gate electrode. The modulation of magnetic anisotropy is evidenced by the observation of clear hysteresis loops at zero and negative electric field, while positive E leads to $\mu_0H_c=0$ [40]. Similar control of magnetic properties can be accomplished by metal-insulator-semiconductor (MIS) structures with a conducting ferromagnetic (Ga,Mn)As channel [33, 41]. However, the change of Tc is limited around 20 K as the Fermi level is frequently pinned by gap surface states, even Tc itself can vary from 45 to 165 K over a wide range of (Ga,Mn)As thicknesses and Mn concentrations [42]. A electric double layer transistor structure, with a polymer electrolyte dropping on top of (Ga,Mn)As channel, was proposed to increase the range of Tc modulation, but the improvement is insignificant [42].

Polarized illumination provides another effective method to tune magnetic properties of (Ga,Mn)As. Circularly polarized light generates spin-polarized holes that change the orientation of Mn spins [43]. This light ellipticity dependent phenomenon creates a large magnetization variation up to 15% of Ms. A memorization effect is demonstrated by the remanent Hall resistance after illumination. In addition, the number of polarized holes required for magnetization switching by illumination is much less that via gating. Further study demonstrates a 90 degrees switching of magnetization can be realized in (Ga,Mn)As microbar though a scanning laser magneto-optical microscope [44].

Magnetic anisotropy is extremely sensitive to strain, which is often unavoidable in epitaxial thin film due to lattice mismatch between film and substrate. Therefore, the magnetic properties of (Ga,Mn)As can also be tuned via strain engineering. In core-shell nanowires, in which GaAs core is wrapped by MBE grown (Ga,Mn)As shell, easy axis is oriented along the nanowire axis [45]. The observed magnetic anisotropy is strain induced, similar to the lithographically patterned nanobars. In the case of nanobars, the easy axis can be rotated by 90 degrees via nanopatterning [46, 47].

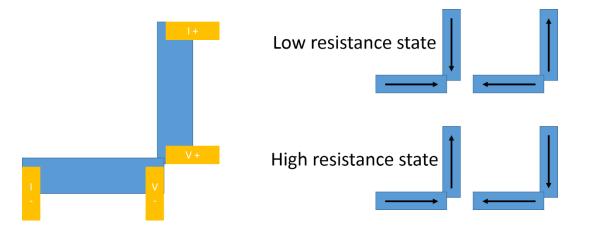


Figure 2.6 A schematic diagram of (Ga,Mn)As nanobar based memory cell. The resistance of the memory cell is controlled by the relative orientations of two perpendicular (Ga,Mn)As nanobars.

Upon the successful modulation of magnetic properties, spin injection, transportation and detection have been further investigated in (Ga,Mn)As based semiconductors [48-51]. By employing Esaki diodes with p⁺-(Ga,Mn)As/n⁺-GaAs structure as contacts, spin injection and detection can be accomplished in a single lateral semiconductor device [52]. Spin injection efficiency reaches ~50% under low bias currents, but higher bias can dramatically reduce the efficiency. The spin relaxation time of the device is ~4.3 ns. Based on the lithographically patterned nanobars discussed above, it is possible to design a prototype of (Ga,Mn)As based memory cell that operates in non-volatile regime [52-54]. The memory cell consists of two (Ga,Mn)As nanobars, electrically connected and perpendicular to each other. As the easy axis of nanobars is oriented along the long axis,

such structure can create different relative orientations of magnetization which can be set by writing process, as presented in figure 2.6. The reading is done through measuring the resistance depending on the relative orientations. Although the on/off ratio in the report is only 280%, much improvement is expected in future research.

2.2.2 (Ti,Co)O₂

Although Mn doped III-V DMSs provide perfect demonstrations for modulation of the magnetic properties in semiconductors, the Curie temperatures are generally below room temperature and the operation temperatures of correspondent devices drop even lower (< 50 K). Ferromagnetic ordering that can persist up to above room temperature is highly desirable for a vast number of applications. In the same paper in which Dietl calculated the Tc of 5 at% Mn doped GaAs, he also pointed out that some nitrides or oxides, such as GaN and ZnO, are possible to obtain a Tc above 300 K, as shown in figure 2.5. This leads to massive research exploration in the field of diluted magnetic semiconducting oxides (DMSO) and most research has been carried out focusing on two specific metal oxides, TiO₂ and ZnO [55].

Among several polymorphs of TiO₂, three exist in nature, namely rutile, anatase and brookite. Figure 2.7 shows the unit cells of rutile and anatase. Both rutile and anatase have a tetragonal crystal structure with wide indirect bandgaps of \sim 3 eV and rutile is thought to be more thermodynamically stable at high temperature. TiO₂ is definitely the most intensively investigated DMSO, due to a variety of well acknowledged interest properties or simply because it is the first DMSO reported to show ferromagnetism at room temperature.

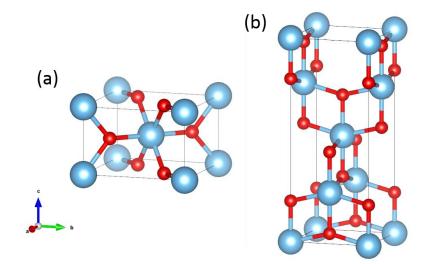


Figure 2.7 Structure of TiO_2 polymorphs: (a) rutile and (b) anatase. Ti and O atoms are represented by blue and red balls.

In the pioneering work published in journal Science in 2001 [56], Matsumoto *et al.* fabricated Co doped TiO₂ thin films, with cobalt concentrations between 0 and 8 at%, on LaAlO₃ (LAO) and SrTiO₃ (STO) substrates by the combinatorial laser MBE under oxygen pressure of 10^{-5} - 10^{-6} Torr and temperatures of 950-1000 K. XRD result shows only anatase structure and no trace of segregation of impurities can be found in TEM images throughout the whole range of Co concentrations. Thus, the authors drew the conclusion that the distribution of Co in anatase was homogeneous. Long range ferromagnetic ordering was demonstrated by the magnetic domain structure of ~20 µm in diameter observed in magnetic force microscopy (MFM). Magnetic moment deduced from the Ms of Ti_{1.93}Co_{0.07}O₂ is 0.32 µ_B per Co atom, which is in agreement with the low spin state of Co ions. The Curie temperature is estimated to be at least higher than 400 K.

Following the work of Matsumoto's group, Co doped TiO₂, especially thin films, has been fabricated using a wide variety of techniques. Similar to the growth of (Ga,Mn)As, the majority of the fabrication methods are in conditions far beyond thermal equilibrium to achieve DMSOs with high dopant concentrations that exceed solubility limits, such as laser MBE [56, 57], oxygen plasma assisted MBE [58, 59], reactive co-sputtering [60], and magnetron sputtering [61, 62]. However, these tremendous studies have failed to establish certain relationships between the growth conditions and the Co distribution in the thin films. In the epitaxial thin films, segregation of Co, in the form of metallic clusters, are observed as often as homogeneously distributed Co atoms.

The rather sensitive dependence of material structure on growth methods and conditions has raised fierce debate about the origin of observed ferromagnetism in Co-TiO₂. Is it intrinsic or extrinsic? The term diluted in DMS means dopant concentration should be diluted enough so that the distance between neighbouring dopants is mandatorily larger than the requirement for any conventional direct ferromagnetic exchange can take place. Thus, the intrinsic ferromagnetism in DMS can only be mediated through carrier. This is not a big issue in (Ga,Mn)As as Mn dopants are rather unlikely to form any secondary phases. However, in the case of Co-TiO₂, magnetic properties often originate from extrinsic ferromagnetism, methods are carried out to measure unique phenomena caused by the interaction between dopant and carrier spins, such as optical magnetic circular dichroism (MCD), anomalous Hall effect (AHE), and etc. These characteristics can be regarded as the signature of intrinsic ferromagnetism, but real issues are often complex beyond expectations, which will be discussed later.

Toyosaki *et al.* reported anomalous Hall effect in laser MBE grown rutile $Ti_{1-x}Co_xO_{2-\delta}$ on r-sapphire substrates [57]. The oxygen vacancy density δ can be controlled by adjusting the partial oxygen pressure during deposition. The carrier density *n* determined from ordinary Hall effect varies from 10^{18} to 10^{22} cm⁻³ with a rather stable mobility of ~ 10^{-1} cm²V⁻¹s⁻¹. The results show that the AHE resistance is dominant for lower magnetic field and develops with increasing carrier or oxygen deficiency concentrations. The authors

claimed the scaling behaviour of the $Ti_{1-x}Co_xO_{2-\delta}$ thin films is caused by a semiclassical nature of the charge carrier dynamics, which is not strongly affected by scattering from magnetic impurity, phonon and other defects. Later the temperature dependence of AHE in anatase $Ti_{1-x}Co_xO_{2-\delta}$ was studied over the range from 10 to 300 K [63]. While the anomalous Hall resistance and magnetization share the same magnetic field dependence, the former decreases at low temperature compared to nearly constant magnetization unaffected by temperature. Charge scattering may play an important role in AHE as its conductivity is proportional to the square of the mobility. Moreover, despite different conductivity σ_{xx} follows the same rule of the rutile $Ti_{1-x}Co_xO_{2-\delta}$ discussed previously [57, 64]. In a subsequent study, the group expanded measuring temperature to 300-600 K [65]. The AHE resistivity shows clear ferromagnetic hysteresis loop up to 600 K with negligible change in the magnitude, suggesting a Curie temperature higher than 600 K.

In addition to anomalous Hall effect, magneto-optic properties were examined using PLD grown anatase $Ti_{1-x}Co_xO_2$ ($0 \le x \le 0.1$) on LaSrAlO₄ (001) substrates [66]. By controlling the partial oxygen pressure during deposition, the transport behaviour of the thin films varies from metallic, through semiconducting to insulating, suggesting successful modulation of carrier density *n*. Meanwhile, deposition at lower temperature leads to higher film conductivity. Magneto-optical circular dichroism (MCD) spectra show ferromagnetic response with a peak at the band edge. The MCD intensity is enhanced with the increase of either carrier concentration *n* or Co concentration x, indicating the essential role of the interaction between Co and carriers in the observed ferromagnetism. Subsequently, Toyosaki *et al.* also carried out MCD experiments on their laser MBE grown rutile $Ti_{1-x}Co_xO_{2-\delta}$ and made an effort to establish a direct link between MCD and AHE [67]. The MCD signal is trivial at low doping concentration $x \le 0.1$, but for

 $0.03 \le x \le 0.1$, intensity increases sharply in low magnetic field and quickly saturates in high magnetic field, showing obvious ferromagnetic behavior. The magnetic field dependency of the MCD and AHE resistivity for Ti_{0.9}Co_{0.1}O_{2- δ} shows unmistakable coincidence. The Hall resistivity and MCD represent the ferromagnetic response of the itinerant charge carriers and photoexcited carriers, respectively. Thus, a similarity of AHE and MCD results indicates the existence of a single ferromagnetic source, most probably the charge carriers induce the ferromagnetism.

In a more systematic study, massive data was summarized by Fukumura *et al.* to clarify the role of charge carriers for the room-temperature ferromagnetism in Co doped rutile Ti_{1-x}Co_xO₂₋₆ [68]. The results show clear dependence of Ms on both Co concentration and carrier density *n*. For instance, when Co concentration x=0.03, the magnetization is negligible at low carrier density $n=7\times10^{18}$ cm⁻³, but gradually increases with higher *n*. Clear ferromagnetic hysteresis loops can be observed for $n\geq2\times10^{20}$ cm⁻³, while further increase in carrier density no longer improves Ms that is stabilized at ~1 µ_B/Co. For a constant carrier density of 2×10^{20} cm⁻³, higher Co density leads to higher Ms. The Ms of Ti_{1.95}Co_{0.05}O_{2-δ} and Ti_{0.9}Co_{0.1}O_{2-δ} are approximately 1.1 and 1.5 µ_B/Co. AHE resistivity and MCD of the Co doped rutile with different x and *n* show the same magnetic field dependence with Ms, indicating that the ferromagnetism is induced by charge carriers. The magnetic phase diagrams drew from the study suggest ferromagnetic ordering in Co doped TiO₂ is favored under higher Co concentration and carrier density, as shown in figure 2.8.

In addition to the above magnetic measurements, Fukumura *et al.* also evaluated electronic states of Co by XPS and XMCD. It is noted that Co ions are Co^{2+} with high spin configuration, which is in agreement with several previous reports [69, 70]. The multiplet characteristics appeared in XMCD measurements at the Co L_{2,3} edges on Co

doped rutile TiO₂ is claimed to originate from ferromagnetic Co²⁺ ions coordinated by O^{2-} ions [69]. In contrast, the XMCD of metallic Co is featureless. The multiplet feature matches the full atomic-multiplet calculation result of high-spin Co²⁺ state in a crystal field with D_{2h} symmetry, which is the local symmetry around the Ti site in rutile. This indicates Co²⁺ ions substitute on Ti sites. However, the orbital and spin moments estimated from XMCD are only 0.013 and 0.12 μ_B , respectively. The total magnetic moment (~0.13 μ_B) is much smaller than that evaluated by magnetization measurement (~1 μ_B) and the authors failed to explain the reason. The crystallographic position of Co in both rutile and anatase TiO₂ were further investigated by XRD utilizing anomalous scattering from K edge of Co [71]. Co should have the identical periodicity as the Ti ions in the lattice and generate a anomalous Bragg peak, if Co dopants are indeed substituting Ti site. However, unlike the results observed in Co-doped ZnO that support Co substitution, none of any anomaly was detected in Co doped rutile and anatase TiO₂, casting doubts on the exact whereabouts of Co in TiO₂.

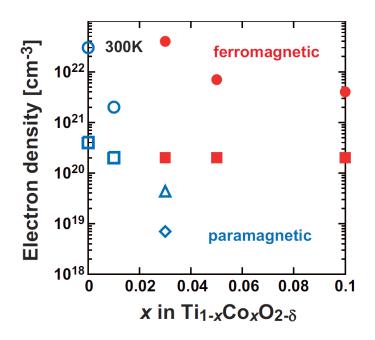


Figure 2.8 Magnetic phase diagram for $Ti_{1-x}Co_xO_{2-\delta}$ as functions of x and *n* (measured at 300 K). Circle, square, triangle, and diamond symbols represent sample grown under partial oxygen pressure $P_{02}=10^{-7}$, 10^{-6} , 10^{-5} , and 10^{-4} Torr, respectively [68].

Are all the evidences mentioned above sufficient enough to prove the intrinsic nature of the ferromagnetism observed in Co doped TiO₂ systems? Unfortunately, the results failed to cease the considerable debate regarding the origin of ferromagnetism. The presence of nanosized cobalt clusters has been frequently observed by using similar or exactly same growth conditions [72, 73], making previous conclusions about intrinsic ferromagnetism suspicious. AFM, SEM and TEM are commonly used to identify the existence of Co nanoclusters, but their accuracy remains questionable. Under particular growth conditions, segregation of Co may only take place in particular regions, such as interface, surface and grain boundaries that avoid certain characterizations. In some extreme cases, conclusions are drawn too casually that only surface-sensitive probes like XPS were used, but Co may have different valence states in deeper regions compared to the surface. More strangely, post-annealing seems to be able to lead to opposite redistribution of Co dopants. While some believe annealing can dissolve Co nanoclusters and create Co^{2+} ions [74, 75], others observed increased size of Co clusters, or precipitation of Co clusters on surface [72, 76]. The contradiction confirms the unreliability of judging Co states through mere STEM or XPS. Nevertheless, ferromagnetic orderings can be dramatically changed via annealing. For example, the Ms of 2 at% Co doped anatase TiO₂ increases from 0.3 to 1.1 μ_B/Co after 450 °C annealing under ultra-high vacuum ($< 10^{-8}$ Torr) [77].

Unfortunately, cobalt segregation is not the only possible source of extrinsic ferromagnetism. A number of magnetic contamination and instrumental artifacts can also generate ferromagnetic-like behaviours [78]. For example, any sample handling steps such as mounting and cleaning can accidentally contaminate the samples by forming magnetic compounds. Experiments must be treated carefully to minimize the contaminations.

To make things even worse, AHE was observed in rutile TiO_2 thin films with clear presence of 8-10 nm nanosized Co clusters at the interface [79]. The co-occurrence of AHE and superparamagnetism questions the employment of AHE as a signature of the intrinsic nature of the DMS. Although mere observation of AHE is not sufficient, some argue that the additional analysis of the magnitude of AHE and the comparison with theory may still be a reliable means.

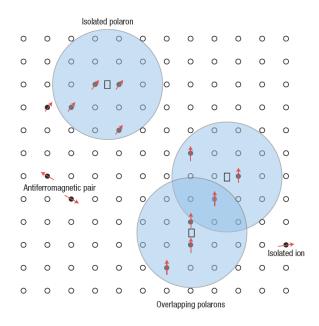


Figure 2.9 A schematic diagram of magnetic polarons. Cation and unnoccupied oxygen sites are represented by small circles and squares, respectively [80].

Due to the uncertainty of the dopant distribution, simulation of the Co-TiO₂ system using quantitative methods like the mean-field Zener model can be extremely inaccurate. Therefore, the system is often interpreted qualitatively and one of the most accepted models is bound magnetic polarons (BMPs) [80], as illustrated in figure 2.9. The concept was first proposed to explain the metal-insulator transition in EuO at low temperature [81]. In the BMPs model, defects like oxygen vacancies play the an important role, which not only donate electrons but also trap and bind the electrons to maintain insulating properties. The trapped electrons can couple and align the magnetic moments of the matrix, forming a bound polaron with a notable magnetic moment. The final magnetic

behaviour is determined by the interaction strength of neighbouring polarons. If the strength is strong, the material becomes ferromagnetic, while weak interaction can only leads to paramagnetic ordering [82, 83].

The ability to manipulate magnetism through gating is also considered as the most straightforward method to test the intrinsic ferromagnetism in DMSs. The first successful demonstration of modulation of ferromagnetism in Co doped anatase TiO_2 was done by Zhao's group [84]. A heterostructure consisting a bottom electrode SrRuO₃ layer (SRO), 7 at% Co doped TiO₂ layer, and a ferroelectric PbZr_{0.2}Ti_{0.8}O₃ (PZT) layer were deposited on LaAlO₃ substrate by PLD. The modulation effect both on Ms (15%) and coercivity (40%) were observed at room temperature. However, the mechanism explained by authors in terms of BMP remains obscure. The uniform distribution of Co dopants is suspicious as both Co concentration (7 at%) and growth temperature were very high (875 °C) in the study, which frequently leads to Co segregation. The authors did not examine the issue carefully enough in the paper, but rather believed the result from the studies about post-annealing of Co-TiO₂ which have been proved doubtful. More recently, a electric double layer transistor (EDLT) was developed in pursuit of an extremely high electric field effect [85]. In contrast to conventional transistors, EDLTs use a liquid electrolyte instead of solid one. Under an electric field, ions within liquid electrolyte assemble to form a nanometer-thick electric double layer. This alters the transistor from a low-carrier paramagnetic state to a high-carrier ferromagnetic state. Figure 2.10 presents the concept of the EDLT.

To date, only a limited number of work about Co-TiO₂ based spintronic devices have been reported. Toyosaki *et al.* reported magnetic tunnel junctions composed of Fe_{0.1}Co_{0.9} electrodes, AlO_x tunnel barrier and ferromagnetic semiconducting Ti_{1-x}Co_xO_{2- δ} layer [86]. The resistance of the magnetic tunnel junctions exponentially increases with the barrier thickness which can be controlled by stencil mask during the fabrication. The tunneling magnetoresistance and differential conductance in terms of bias voltage are extremely asymmetric at low temperature, which is probably caused by either degraded interface or asymmetric device structure.

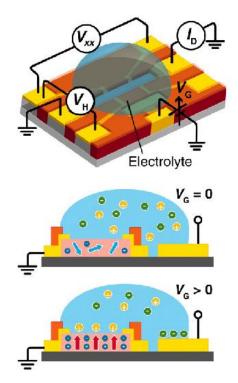


Figure 2.10 A schematic diagram of EDLT. When gate voltage is zero, the transistor is in paramagnetic state as localized spins are randomly oriented. With a positive voltage applied, ferromagnetism is induced due to the accumulated eletron carriers on the channel surface [85].

2.2.3 Transition metal dichalcogenides

While (Ga,Mn)As and (Ti,Co)O₂ based DMSs are under extensively exploration, scholars are still unceasingly searching for new kinds of semiconductors that are promising for spintronic applications. In 2004, graphene, a monolayer counterpart of graphite, was discovered and immediately attracted considerable research interest. Although at the moment graphene is still the most notable two-dimensional material, its zero bandgap restricts the potential applications in logical devices. Inspired by graphene, there has been renewed interest in graphene-like semiconductors, which have layered structure but with atomic thickness, such as boron nitride, transition metal dichalcogenides (TMDCs), and transition metal oxides (TMOs).

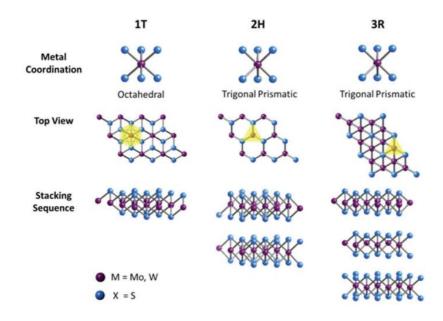


Figure 2.10 Structure of 1T, 2H and 3R MoS₂ and WS₂ unit cells [87].

TMDCs are a group of chemicals sharing a formula MX₂, in which M and X represent a transition metal and a chalcogen, respectively. Depending on the combination of different M and X, the TMDC family has more than 40 distinct members. MoS₂ and WS₂ are two most thoroughly studied TMDCs, especially for MoS₂, which is the only TMDC naturally existing in massive quantities. Both MoS₂ and WS₂ have sandwich-like layered structure. In each MoS₂ layer, a hexagonal plane of Mo atoms is sandwiched by two planes of S atoms. Like graphene, two neighboring MoS₂ layers are only connected by a weak van der Waals force, making it possible to obtain monolayer materials through exfoliation. Based on the type of Mo coordinations and stacking order of MoS₂ layers, three polymorphs of are MoS₂ commonly observed: 1T, 2H and 3R. The numbers refer to the number of MoS₂ layers in a primitive cell and the letters are abbreviations for the lattice symmetry (tetragonal, hexagonal and rhombohedral). Figure 2.10 shows the unit cell structures of three polymorphs.

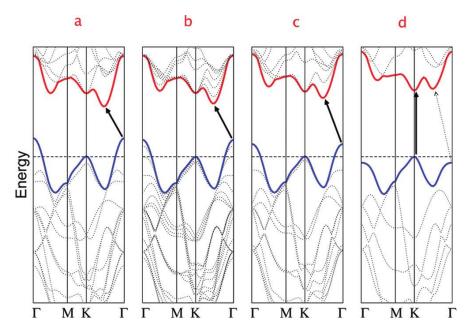


Figure 2.12 The evolution of the band structure of MoS₂: (a) bulk, (b) four-layer, (c) bilayer, and (d) monolayer MoS₂ [88].

One of the most intriguing features of 2D MoS₂ and WS₂ is the thickness-dependent band structure. For example, bulk MoS₂ has an initial indirect bandgap of 1.2 eV, which gradually increases with decreased layer number. However, an abrupt transition to direct bandgap occurs for monolayer MoS₂, with a much larger bandgap (~1.9 eV) located in the visible frequency region, as shown in figure 2.12. In addition to the direct bandgap, another significant property of monolayer MoS₂ and WS₂ is spin-orbit splitting. Due to the broken inversion symmetry, the band structure of monolayer MoS₂ at the top of valence band (TVB) splits into spin-up and spin-down bands with energy difference of 140-150 meV. The value is slightly bigger for WS₂ as W is heavier than Mo. More importantly, as a result of time-reversal symmetry, the splitting in opposite valleys must have opposite directions, indicating that the spin and valley carriers are inherently coupled. The modulation of the coupled spin and valley carries is the foundation of TMDC based spintronics and valleytronics.

Most methods used to fabricate graphene can be employed to 2D TMDCs with parallel effectiveness. The methods can be divided into two categories, top-down methods like

mechanical or chemical exfoliation [89-91], and bottom-up methods like chemical solution deposition or chemical vapor deposition (CVD) [92-97]. The simplest approach to obtain 2D TMDCs is through repeated cleavage, from initial bulk crystal to final fewlayer or even monolayer TMDCs, by using scotch tapes. In general, mechanical exfoliation produce best single crystalline nanosheets with high purity and surface cleanliness, which are favorable for the evaluation of fundamental properties. For example, mobility is often used to evaluate the quality of monolayer MoS₂ as various defects can considerably reduce the value. Based on a large number of experimental work, mechanical exfoliated MoS_2 exhibit the most superior mobility of several tens of cm^2/Vs [98-101]. Meanwhile, liquid-phase exfoliation can yield large quantity of TMD nanosheets. Using proper solvent such as N-methyl-pyrrolidinone (NMP), exfoliation of bulk MoS₂ to few-layer or monolayer nanosheets can be realized by one-step sonication [89]. Although the quality is usually low compared to mechanical exfoliation, it is extremely suitable for the applications that high crystallinity is not necessarily required and shape of the product has little impact on the performance, such as hydrogen evolution reaction, lithium ion batteries and supercapacitors [102, 103]. Li ion intercalation can dramatically increase the effectiveness of liquid-phase exfoliation process because sonication produces large microsheets as well as nanosheets and the yield of few-layer TMDCs is low [104]. More intriguing, Li ion intercalation causes a phase transition of MoS₂ from 2H phase to 1T phase. The two phases can coexist to form monolayer MoS₂ heterostructures [105]. Hydrothermal method, another approach often used for massive production, can also prompt the phase transition. With 1T phase occupies 25%, the density of electron carriers can increase by an order, leading to an intrinsic ferromagnetic response of 0.25 μ_B per Mo atom [106].

Defects can significantly modify the electronic and magnetic properties of single and fewlayer MoS₂. All kinds of defects have been studied by first-principles calculations, including vacancies, interstitials, antisites and adatoms [107-111]. It is noted that S vacancies are the most easily formed and abundant defects in MoS₂. Although all theoretical studies predicted ideal MoS₂ crystal has a nonmagnetic ground state, MoS₂ nanosheets prepared through a variety of methods frequently display clear ferromagnetic behaviours at room temperature, in contrast to bulk MoS₂ which only shows diamagnetism as simulations predict. Gao *et al.* prepared MoS₂ nanosheets via liquidphase exfoliation by sonicating MoS₂ powders in N,N-dimethylformamide (DMF) [112]. The size of the nanosheets can be tuned by the ultrasonic time. Higher Ms is obtained by smaller nanosheets and 10 h sonicated sample has a Ms of 1.1×10^{-3} emu/g at room temperature. Larger Ms is shown by CVD prepared MoS₂ and WS₂ nanosheets [113]. MoO₃ and WCl₆ were used as precursors as Mo and W source. The Ms of MoS₂ decreases from 1.08 emu/g at 10 K to 0.55 emu/g at 623 K, while the Ms of WS₂ is a bit lower. The ferromagnetic ordering persists to 865 and 820 K for MoS₂ and WS₂, respectively.

In spite of the fact that room-temperature ferromagnetism of TMDs has been observed in a large number of experimental studies, the understanding of its origin remains puzzled for researchers. Since bulk MoS₂ is pure diamagnetic, the ferromagnetism observed in MoS₂ nanosheets is absolutely attributed to defects. To date, the ferromagnetism is generally interpreted by two types of defects, vacancies and zigzag edges [112, 114, 115]. However, most vacancies, including V_S (monosulfur vacancy), are non-magnetic. According to theoretical studies, only a few vacancy complexes, such as V_{MoS2} (MoS₂ triple vacancy), can contribute to ferromagnetic orderings. On the other hand, simulation of MoS₂ zigzag nanoribbons indicates the magnetic moment decreases with enlarged width. When the ribbon is 6.43 nm wide, the value is only 0.037 µ_B per MoS₂ [116], which can be totally ignored since the size of nanosheets are much larger than several nanometres. Compared to zigzag edges, grain boundaries, another type of 1D defects, are more ubiquitous in TMDs. Recent first-principles studies suggest that dislocations or formation of antisite defects on MoS_2 grain boundaries can create a magnetic moment around 1 μ_B [117, 118].

Although vacancies and grain boundaries can theoretically introduce ferromagnetism into TMDCs, engineering of intrinsic defects has a number of drawbacks. First and foremost, it is next to impossible to experimentally evaluate the existence of the vacancies or grain boundaries in the material. Based on the first-principles calculations, ferromagnetism can only be caused by a few types of intrinsic defects, but the generation of these intrinsic defects is always accompanied by a bunch of other defects. For example, at least four types of vacancies and two types of antisites were observed in CVD grown monolayer MoS₂, but none of them can contribute to ferromagnetic orderings [119]. In contrast, magnetic doping already have mature techniques such as MBE and ion implantation that can achieve desired ionic substitutions with precise concentrations. Secondly, intrinsic defects usually have low magnetic moments less than 1 μ_B /defect, while the magnetic moments of transition or rare earth metals can easily exceed 3 µ_B/ion. Thus, high magnetization can be expected from magnetic doping. In addition, magnetic doping in 2D TMDCs has several advantages compared to DMS based on II-VI or III-V semiconductors. Generally, the single crystal quality is very high for 2D materials, so the defects in 2D materials can be more easily controlled by laser treatment, ion implantation and fabrication methods. Clustering or secondary phases are frequently observed and usually located in the interface between the thin film and substrate, destroying the intrinsic ferromagnetism in DMSs. For the magnetic dopants in TMDCs, the 2D nature of the thin film can avoid the formation of clusters or secondary phases.

Substitution of Mo atoms with transition and rare-earth metals has been investigated by a great number of research groups using first-principles calculations [120, 121]. Cheng et al. carried out a systematic simulation of 27 different transition metals and found that Fe, Co, Mn, Zn, Cd, and Hg doped monolayer MoS_2 are ferromagnetic [122]. Although some transition metals may not have direct contribution to magnetic moments, the result changes if they combine surrounding S vacancies to form a complex defect. For example, in V doped monolayer MoS₂, a V atom coupled a nearby S vacancy has a formation energy of 5.89 eV and can create a magnetic moment of ~ 1 μ_B [123]. Experimentally, either anions or cations in TMDCs can be easily substituted by elements from the same group. The alloys, such as $Mo_{1-x}W_xS_2$ and $MoS_{2x}Se_{2(1-x)}$, can be prepared by CVD with simply mixing two precursors together [124, 125]. However, substitutional doping of other transition metals has been proved to be far more difficult [102, 126]. In bilayer Codoped MoS₂ nanosheets grown by CVD, Co is only observed at the edge of the nanosheets [127]. Further increase deposition temperature from 680 to 750 °C leads to growth of CoS_2 layer on top of the MoS_2 nanosheets. In another study, it took more than 500 hours to prepared Nb-doped MoS₂ nanosheets by a chemical vapour transport method, but the concentration of Nb dopants could only achieve 0.5 at% [128].

2D TMDCs have great potential in applications of spintronics and valleytronics. Wang *et al.* designed first MoS₂ based spin valve by using monolayer MoS₂ and Permalloy (Py) as the nonmagnetic spacer layer and ferromagnetic layers [129]. However, due to strong hybridization with Ni and Fe atoms, the MoS₂ layer behaves more like conducting film rather than tunnel barrier. The magnetoresistance is very low, only 0.73% at 10 K. Further first-principles calculation indicates the value can reach 9% for a perfect Py/MoS₂/Py junction, suggesting much space for improvement.

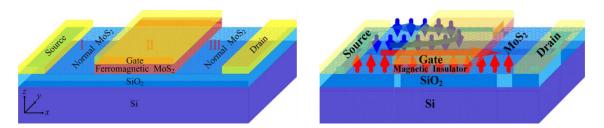


Figure 2.13 The conceptual structure of an all MoS₂ junction (left) and its operation principle (right) [130].

Most current research concerning 2D TMDCs related spintronics are based on nonmagnetic channel material, since the magnetic modulation of 2D TMDCs is still in its infancy. Li *et al.* theoretically proposed a junction structure that consists of a ferromagnetic MoS_2 layer sandwiched between two normal non-magnetic monolayer MoS_2 [130], as shown in figure 2.13. By applying a gate voltage, not only can the charge transport be manipulated, but also fully valley- and spin-polarized conductance can be simultaneously obtained with tunable amplitude and direction. A two-dimensional spin field-effect switch was experimentally demonstrated based on a $MoS_2/graphene$ heterostructure [131]. The spin current in the graphene channel can be electrically switched by adjusting the spin absorption into the MoS_2 layer. The device can be operative at temperatures up to 200 K.

Chapter 3 Experimental Methods

3.1 Material fabrication

3.1.1 Pulsed laser deposition

Pulsed laser deposition (PLD) is a type of physical vapor deposition (PVD) techniques. The concept of PLD is schematically illustrated in figure 3.1. A pulsed high-energy laser beam is focused through a set of optical apparatuses onto the surface of a target which is usually placed at an angle of 45° to the incident beam. After absorbing the laser energy, the target ejects a tiny amount of material in the form of a plume due to superheating. The plasma then accumulates on a substrate facing the target. During deposition, the chamber can maintain at ultra high vacuum below 10^{-8} torr or fill with certain background gas. In our case, oxygen is used to deposit TiO₂ thin films.

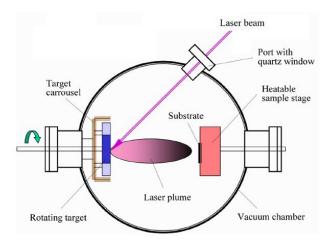


Figure 3.1 A schematic diagram of a PLD system.

While the basic concept is relatively simple, the physical mechanisms behind laser-target interaction and film growth are extremely complicated. The PLD process is generally modelled by Monte Carlo simulation [132], which consists of three crucial steps, including laser absorption and ablation of the target material, creation and dynamic of the plasma, and nucleation and film growth on the substrate.

The typical energy density and duration of a single laser pulse in PLD is approximately 1 J/cm^2 and 10 ns, leading to an instantaneous power density above 10^8 W/cm². The laser energy is then converted into heat and conducted into the target. During the thermal diffusion, the temperature in the target surface can rise at a rate as high as 10^{11} K/s that generates a gas pressure of 10-500 atm, leading to the vaporization of the target material in an extremely non-equilibrium state. The characteristic ablation plasma, or plume, is caused by multiphoton ionization of the material gas.

In the second step, the plume expands in the direction perpendicular to the target surface towards the substrate, as illustrated in figure 3.1. The composition of the plume is considerably complex, containing both stable and metastable species such as electrons, atoms, ions, molecules, clusters, particulates, and molten globules, with high kinetic energies in the range of 1-500 eV [133]. The spatial distribution of the plume is determined by the chamber pressure. For example, narrow shaped plume with no scattering with background gas only occurs at high vacuum. In general, the distribution of the plume is beamlike, leading to a restricted deposition area of only a few cm². The difficulty to accomplish large area deposition is commonly claimed as a drawback of PLD. However, if the substrate is not fixed but repeatedly rastered across the path of the plume, the problem can be solved and even turn to advantage [134].

When the plume reaches the substrate, some target atoms are sputtered out from the surface and further collide with the plume to form a condensed collision region, where the gas is supersaturated [135]. Afterwards, nucleation and coalescence of nuclei give rise to different thin film growth modes [136], as illustrated in figure 3.2. When the nuclei are strongly bound to the substrate, the deposition materials first grow into a single-layer film before the nucleation and growth of the second layer. This is called layer-by-layer growth mode. On the other hand, island growth mode occurs when the interatomic bonds within

the deposition material dominate. In such case, the nuclei continue to grow in three dimensions to form large islands. The last layer plus island growth mode starts with layer by layer deposition and transits to island mode during deposition. A number of parameters have critical influence over the nucleation process and growth kinetics, including laser fluence, pulse interval, substrate surface and temperature, background gas and pressure, and distance between target and substrate.

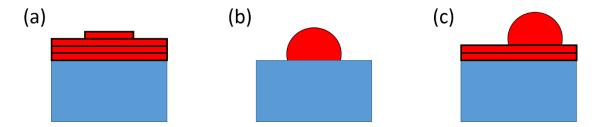


Figure 3.2 Schematic diagrams of three modes of thin film growth: (a) layer-by-layer mode, (b) island mode and (c) layer plus island mode.

In summary, PLD is a powerful technique and exhibits many practical advantages. Firstly, unlike other PVD methods such as thermal and e-beam evaporation, the energy source in PLD is located outside the vacuum chamber. This prevents contamination from the source and provides much more flexibility in equipment design. Secondly, owing to the ultrahigh temperature in the laser focus area, almost any condensed matter material can be melt and ablated. Moreover, the stoichiometry of the thin film can be controlled with extreme accuracy. Finally, it is convenient with the ability to control film growth rate and thickness by merely changing the pulse repetition rate while keeping the energy input. However, PLD technique has a major drawback. Particulates with size as large as a few micrometers, or labelled as laser droplets, are inevitably generated during the laser irradiation, which can seriously affect the growth and final quality of the thin film. New techniques aiming to effectively block the laser droplets are still under development. In this work, the PLD was performed in Australian National Fabrication Facility (ANFF) at UNSW.

3.1.2 Ion implantation

Ion implantation is a low-temperature technique to dope an element into a target material. The dopant atoms are first generated by heating an ion source, then electrostatically accelerated and directed at a target chamber after separating by the mass-to-charge ratios. The total amount of dopant implanted in the target, which is called the dose, can be calculated by integral of the ion current over time. The ion energies applied in implantation typically range from 10 to 500 keV. If shallow implantation of only a few nanometers are required, energies in the range of 1 to 10 keV can be used. Energies lower than 1 keV only lead to ion beam deposition rather than implantation. On the other hand, energies of several million electron volts can also be used but may cause severe damage to the lattice structure of the target.

After entering the target lattice, the ions gradually lose energy accompanied by a series of collisions with the target atoms and finally stop at some depth within the target. This process is called ion stopping, influenced by both nuclei and electrons. While nuclear stopping can be described by classical kinematics, the modelling of electronic stopping is quite complex. The energy loss by the the interaction between the implanted ions and the electrons is believed to dissipate into thermal vibrations of host atoms through the electron cloud. Although each implanted ion travels though the target by a random path, the ion trajectories can be solved by a statistical method. The distribution is often simplified as a Gaussian. In addition, annealing is usually required to repair the structure damage and move dopants from interstitial to substitutional sites, where dopants can be electrically activated.

In this work, the ion implantation was carried out using the low-energy ion implanter at GNS Science, New Zealand.

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3.2 Material characterisation

3.2.1 X-ray diffraction

X-ray diffraction (XRD) is one of the most significant techniques in material analysis because it can determine the atomic and molecular structure of a crystal [137]. X-rays are a form of electromagnetic radiation and the diffraction is based on Thomson scattering. When a beam of X-rays hits a material, its propagation can be modified by the electrons in the material while the kinetic energy, or wavelength, is conserved. If the material is a crystal, the incident X-rays is diffracted into certain directions based on the crystalline structure. The density of electrons within the three-dimensional crystal can be calculated from the angles and intensities of the diffracted beams. Further analysis can obtain more crystal information, such as the positions of the atoms, grain size, strain, and etc.

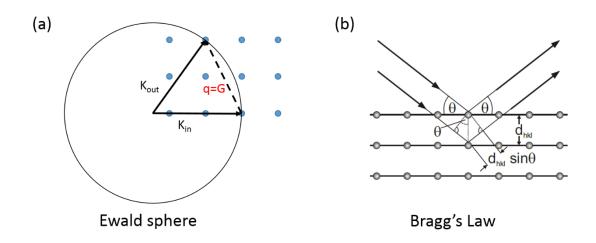


Figure 3.3 (a) The geometric construction of Ewald sphere. (b) Interpretation of the Bragg equation.

The diffraction of X-rays follows the rule $q=K_{out}-K_{in}=G$, in which K_{out} and K_{in} are wave vectors of diffracted and incident X-rays, and G is a reciprocal lattice vector. This condition is illustrated in figure 3.3(a), in which all the possible G are located with a sphere. The so called Ewald sphere has a diameter of $2|K|=4\pi/\lambda$. Assuming $G=nG_{hkl}$, one can eventually obtain the famous Bragg's Law $n\lambda=2d_{hkl}\sin\theta$, where the angle between K_{in}

and K_{out} is 20. The Bragg equation is usually interpreted by a simple way, as shown in figure 3.3(b). The incident X-rays are reflected by lattice planes (hkl). The diffraction only occurs when the path difference of waves reflected by two neighbouring planes is equal to an integer multiple (n) of the wavelength.

A basic X-ray diffractometer consists of three elements: an X-ray tube, a sample holder, and an X-ray detector. In the X-ray tube, X-rays are generated by heating a filament to produce electrons and then bombarding a target (Cu, Fe, Mo, Cr) with electrons. Cu is the most common target material for X-ray diffraction, with $\lambda(K_{\alpha})=1.5418$ Å. The X-ray generator and detector are located on the circumference of a graduated circle centred on the sample, as shown in figure 3.4(a). In some diffractometers, the sample holder can be rotated around three axes respectively (figure 3.4(b)), enabling a wide range of XRD scan modes. In the most commonly used $\theta/2\theta$ scan, the geometry of the diffractometer, also known as Bragg-Brentanno geometry, is that the rotation of the sample (ω/θ) equals to a half of the rotation of the detector (2 θ). Another example is Ω scan, also called rocking curve, in which sample is slightly rotated around the ω -axis while both generator and detector are fixed.

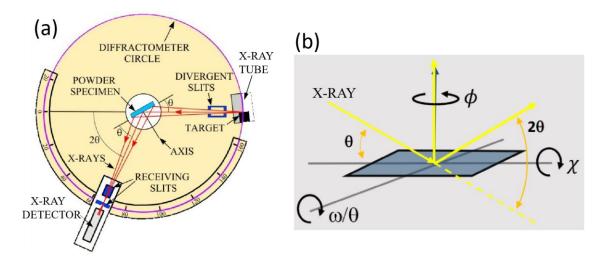


Figure 3.4 (a) A schematic diagram of an X-ray diffractometer. (b) Rotation axes of the sample holder.

The XRD data can be analyzed quantitatively. Rietveld refinement is a computational technique which compares the experimental data to a simulated diffraction profile. The Rietveld method employs a least squares based minimization algorithm. A variety of structural parameters of crystalline materials can be refined through the approach, including unit cell dimensions, preferred orientation, scale factors, etc.

In this work, Empyrean (PANalytical) Thin-Film X-ray Diffraction System and PANalytical Xpert Multipurpose X-ray Diffraction System was used for the phase identification of thin film and powder samples. X-rays are generated by the collision of high velocity electrons on a Cu target and then filtered to a monochromatic Cu K_{α} beam. The photon energy and wavelength of Cu K α line are 8.05 keV and 0.154 nm, respectively. The minimum step size is 0.013°. Data analysis was accomplished by X'pert HighScore Plus software with a built-in Rietveld refinement package.

3.2.2 Scanning electron microscopy

A scanning electron microscope (SEM) is a type of electron microscope, which produces various signals containing information about the topography and composition of the sample surface [138]. When a focused electron beam, with a tunable acceleration voltage of 5-30 keV, stikes on a sample, the energy exchange between the electron beam and the sample results in a variety of electron scattering and radiation, such auger, secondary and backscattered electrons.

In most cases, we are interested in collecting and capturing images of secondary electrons that are emitted from very top of the sample surface. Secondary electrons are extremely surface sensitive and can produce high-resolution images of the sample surface. In addition to secondary electrons, the energy of characteristic X-rays are often measured by Energy-dispersive X-ray spectroscopy (EDX) to identify elements in the sample and map the distribution. In this work, FEI Nova NanoSEM 450 was used to analyze the surface morphology of samples. The voltage of electron beam was set to 5 keV for high-resolution imaging and 20 keV for element identification.

3.2.3 Transmission electron microscopy

When a sample is thin enough (less than 100 nm), the beam of electrons interacted with the sample can transmit through it to form an image [139]. In a transmission electron microscope (TEM), the image is then magnified and focused onto a screen, either a fluorescent or a liquid-crystal display one. The major advantage of SEM and TEM over light microscopes is the ability to view samples at a considerably high resolution. This is attributed to the much shorter de Broglie wavelength of electrons (2.5 pm at 200 keV), especially in TEM where electrons are accelerated under several hundreds of kV, achieving an velocity (> 2×10^8 m/s) that approaches the speed of light. Practically, the resolution in TEM is limited by the objective lens system to ~0.1 nm. An advanced TEM may include various aberration correctors to reduce the degree of distortion in the image. To date, the state-of-the-art transmission electron aberration-corrected microscope has a spatial resolution as low as 0.05 nm, which is roughly half the size of an hydrogen atom.

Selected area electron diffraction (SAED) is a crystallographic technique that can be employed inside a TEM. The electrons in TEM are more wave-like than particle-like and can be diffracted by atoms similar to XRD. SAED patterns consist of sharp diffraction spots. Each spot represents a reciprocal lattice. As the area selected by SAED can be as small as several hundred nanometers in size, it can examine samples which are too small for XRD.

The thin film and powder TEM samples were prepared differently. The thin film sample was shaped by focused ion beam incorporated in FEI xT Nova NanoLab 200, and then

lifted out by a glass needle and transferred onto a copper grid. The powder sample was first dispersed in ethanol by ultrasonication, and then dropped onto a copper grid. The copper grad containing the sample was analyzed by JEOL 2200FS and Phillips CM200 TEM system.

3.2.4 Atomic force microscopy

Atomic force microscopy (AFM) is a type of scanning probe microscopy (SPM) [140]. During the scanning, a probe fixed to the open end of a small spring-like cantilever is brought into contact with the sample surface, either in direct contact mode or tapping at a tiny distance above (tapping mode). The force that the sample imposes on the probe can be measured and used to create a three-dimensional topography of the sample surface. Although the resolution in horizontal x-y plane is moderate, the resolution at vertical z direction can achieve the order of an angstrom. In this work, Bruker Dimension ICON SPM is used to examine the surface roughness of the thin film sample.

3.2.5 Raman spectroscopy

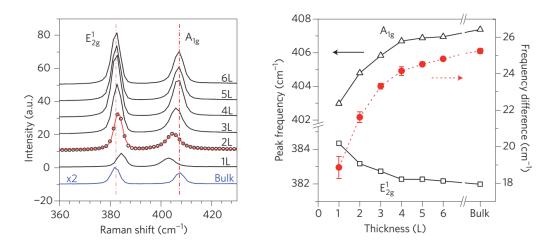


Figure 3.5 Raman spectra of few-layer and bulk MoS_2 (left), and peak position for the E_{2g}^1 and A_g^1 modes as a function of layer number (right) [141].

Raman spectroscopy is a technique used to detect vibrational, rotational, and other lowfrequency modes in a material. It is based on inelastic scattering of a laser covering near ultraviolet, visible, and near infrared range. The laser energy alters after interacting with molecular vibrations and rotations, offering a structural fingerprint specific to those modes of the system. Raman spectroscopy can be used to determine the thickness of fewlayer MoS₂. As the thickness of MoS₂ decreases, the E^{1}_{2g} vibration mode red-shifts while the A_{1g} vibration mode blue-shifts, as shown in figure 3.5 [141]. In this work, Raman spectra were collected in the Renishaw InVia Raman microscope equipped with 3000 I/mm grating and 532 nm line of Ar ion laser.

3.2.6 Secondary ion mass spectrometry

Secondary-ion mass spectrometry (SIMS) is a technique used to identify the elemental composition of a specimen by applying a focused ion beam on sample surface and detecting the mass/charge ratios of rejected secondary ions with a mass spectrometer. It is very effective to examine the isotopic composition of an element as isotopes like carbon-12 and carbon-14 have different mass. SIMS is even more surface sensitive than XPS, only signals from a depth of 1-2 nm below the surface are collected. However, since ion beam can also etch the surface at the same time, making it possible to analyze the indepth distribution of the elements. In this work, Cameca IMS 5FE7 system with oxygen ion source is used.

3.2.7 X-ray photoelectron spectroscopy

In X-ray photoelectron spectroscopy (XPS), core electrons are ejected from the material after absorbing photons with energy higher than the binding energy of the electron. The excess energy is converted to kinetic energy of ejected photoelectrons, which can be measured by an electron analyser. It is easy to calculate the binding energy if the energy of the incident photon and the work function of the spectrometer are known. XPS is very sensitive to the chemical state of an element. The shift of binding energy, also called chemical shift, as well as the shape of XPS peaks can be used to determine the local coordination in a material. In this work, Thermo ESCALAB250Xi X-ray photoelectron

spectrometer is used with a monochromatic Al Ka source (1486.68 eV). All binding energies were referenced to the C 1s peak at 284.85 eV.

3.2.8 X-ray absorption spectroscopy and X-ray magnetic circular dichroism

The X-ray absorption spectroscopy (XAS) tracks the absorption intensity as a function of the incident X-ray energy, which is tuned to a range of 0.1-100 keV that the core electrons of a certain element can be excited. Based on the principal quantum number n of the excited core electrons, the spectra are called K-, L-, M-edges for n=1, 2, 3, respectively. As the initial state of a core electron is well defined by quantum mechanics, the excitation must follow the selection rules. Generally, the most prominent features of the excitation are caused by electric-dipole transitions ($\Delta \ell = \pm 1$), for example, 1s to p-like final states for a K-edge XAS.

If the material is ferromagnetic, the X-ray absorption process becomes spin-dependent and can be detected by X-ray magnetic circular dichroism (XMCD). In XMCD, two XAS, one with left circularly polarized light and the other with right circularly polarized light, are measured in a magnetic field. Magnetic properties, such as the orbital and spin magnetic moments, can be obtained by analyzing the difference between two XAS spectra. In this work, XAS and XMCD were measured at Singapore synchrotron light source.

3.2.9 Superconducting quantum interference device

Superconducting quantum interference device (SQUID) is currently one of the most sensitive magnetometer employed to measure ultra-small magnetic fields with detection limit as low as 10^{-8} emu. The core of the SQUID is two parallel Josephson junctions. Each junction consists of two superconducting regions separated by a thin insulating barrier. Any change of magnetic flux through the SQUID can cause periodic voltage response. The periodicity of the voltage oscillation is equal to one flux quantum, Φ_0 , and is effective to evaluate the magnetic flux change. In this work, we use Quantum design-XL-7 to measure the magnetization of the samples as a function of external magnetic field at different temperatures. During a typical test, the external magnetic field starts from highest point (50 kOe) and gradually decreases to lowest value (-50 kOe), and then restores to 50 kOe. If the sample is ferromagnetic, the result is a hysteresis loop in which magnetization quickly saturates at Ms with a nonzero coercivity.

3.2.10 Polarized neutron reflectometry

Polarized neutron reflectometry (PNR) is a technique well suited for measuring the nanostructure of magnetic thin films. The principle is similar to the complementary technique, X-ray reflectivity (XRR). The basic idea behind the technique is quite simple, a neutron beam is incident on a thin film with flat surface at a grazing angle and reflected in the specular direction (reflected angle is equal to incident angle). The reflected intensity is then measured as a function of the scattering vector $q_z=4\pi \sin(\theta)/\lambda$ (λ is the neutron wavelength and θ is the angle of incidence). While XRR can only obtain the chemical compositional depth profile, PNR can provide additional magnetic information by polarization of incident neutrons.

Although reflected patterns contain structure and magnetic information of each layer above the substrate, such as thickness, density, interface roughness and magnetic moment, the exact mathematical relationship is very complicated, especially for thin films with multiple layers. Instead of direct calculation, the results must be extracted through mathematical analysis by manually fitting the parameters to get a simulated curve close enough to the measured data. In this work, PNR was measured in Australia Neutron Science Technology Organization (ANSTO) and Igor Pro 7 was used to fit the data.

3.2.11 Electron spin resonance

Electron spin resonance (ESR), also known as electron paramagnetic resonance (EPR), is the spectroscopic technique that unambiguously detects unpaired electrons in a material. Electron is a Dirac fermion with spin quantum number s=1/2. Under an external magnetic field with strength B₀, electrons align in the way with magnetic moments either parallel (m_s =-1/2) or antiparallel (m_s =1/2) to the field. Due to Zeeman effect, each electron has a specific energy proportional to m_s : E=m_sg_e μ_B B₀, where g_e is the g-factor for electrons. Thus, the splitting between the lower and higher energy state is Δ E=g_e μ_B B₀. Initially, more electrons stay in the lower energy level than in the upper level, but an unpaired electron can move from lower level to higher level by absorbing a photon with energy exact equals to Δ E (hv=g_e μ_B B₀). In ESR, the sample is exposed to microwave irradiation

with a fixed frequency while sweeping the external magnetic field. When the external magnetic field meets the resonance condition, there is a net absorption of energy which can be monitored by ESR. In the most cases, the first derivative of the absorption spectrum is used to analyze ESR result. In this work, Bruker EMX X-Band ESR spectrometer was used for the measurements.

Chapter 4 Microstructures and Magnetic Properties of Co Doped TiO₂ Thin Film Prepared by Pulsed Laser Deposition

4.1 Overview

Diluted magnetic semiconductors (DMSs) have drawn a great deal of research attention over the course of past few decades, in the hope of the realization of a technology that can integrate logic semiconductors with magnetic information-storage capability. Traditional II-VI and III-V DMSs have a typical Curie temperature (Tc) much lower than room temperature. Diluted magnetic nitrides and oxides (DMNs and DMOs) have become a brand new research highlight after Tc above room temperature was predicted in these materials by Diet1 [30].

TiO₂, with various crystal structures and properties, has great potential in applications including sensors, solar cells, photocatalyst, as well as spintronic devices [142-145]. Co-TiO₂ has demonstrated to be a remarkable promising DMS system with room-temperature ferromagnetism in both rutile and anatase forms in a variety of morphologies, such as thin films [146-148], nanoparticles [149], and nanorods [150]. However, even as researchers have struggled for years to clarify the origin of the ferromagnetism in the system, they failed to achieve a fully convincible explanation.

In previous studies of our group, 5 at% Co doped anatase TiO₂ epitaxial thin films are grown by PLD under different oxygen partial pressures (10^{-4} , 10^{-5} , and 10^{-6} torr) [151]. From the hysteresis loops, the magnetic moments from Co are 1.03 and 0.93 μ_B for samples grown under 10^{-6} and 10^{-5} torr, while the magnetization of the 10^{-4} torr sample is negligible. Further depth-resolved LE- μ SR measurement indicates the 10^{-6} torr sample is fully magnetic and controlled by oxygen vacancies.

The magnetic moment observed in Co-TiO₂ so far typically lays in the range of 0.01 and 1.7 μ_B /Co (value of bulk Co metal), while higher saturation magnetization surpasseing 2 μ_B /Co has only occasionally been reported. High magnetization materials are of importance for the applications of magnetic recording media, permanent magnet, transformer and bioapplications [152]. It is found that nanoclusters have shown enhanced magnetic moment compared with their bulk counterparts [153, 154].

However, due to the frustration of surface dangling spins, the Curie temperature of nanoclusters significantly decrease, which is unwelcome for practical applications [155, 156]. During the development of DMSs, giant magnetic moments have been occasionally witnessed in the systems with transition metal doping. For example, $7.5\pm0.5 \mu_B/Co$ was reported in Co doped SnO₂ grown by pulsed laser ablation [157], while a colossal magnetic moment of more than 4000 μ_B per Gd atom was achieved by Gd doped GaN grown directly on 6H-SiC (0001) substrates using reactive MBE with very low doping concentrations [158]. However, the mechanism for the high magnetic moments remains unsolved since in most studies, only magnetometer measurement is used and there is no direct evidence to link the observed high magnetic moment to the magnetic element.

In this work, we deposited 5 at% Co doped TiO₂ thin film on LaAlO₃ (001) (LAO) substrate by PLD equipped with a 248 nm KrF excimer laser. 5 at% Co doped TiO₂ target was prepared by first mixing Co₃O₄ and TiO₂ (99.99 % in purity) powders in a mortar. Afterward, the mixed powders were pressed into pellet and then sintered in a furnace at 1300 °C for 10 h in air. Prior to deposition, the LAO substrates were cleaned by alternative ultrasonication in acetone and ethanol for several times. The deposition was carried out at 650 °C and oxygen partial pressure of 10⁻⁵ torr. The laser fluence and deposition were

kept at ~2.5J/cm² and 5 Hz, and the final film thickness was controlled to be around 60 nm.

With the delicate control of deposition parameters, we achieve the microstructure of Co clusters embedded in TiO₂ matrix. XMCD measurement indicates that Co has a very high magnetic moment of $3.5 \mu_B$. In addition, O edge contributes significantly to the overall magnetization, whereas, Ti edge shows a much weaker XMCD signal. PNR measurement supports the high magnetization of the films and a saturation magnetization as high as 106 emu/cm³ has been achieved. We use first principles calculations to investigate metallic nanoclusters embedded in oxide semiconductor matrix by considering spin-orbit coupling. It is found that high magnetic moment (2.9 μ_B /Co) can be obtained with clusters containing 14 atoms, agreeing well with experimental results. We have demonstrated that magnetic clusters embedded in oxide semiconductor matrix are able to show high magnetic moment above room temperature.

4.2 Microstructure study of Co doped TiO₂ thin film

Based on the results of numerous studies, Co can achieve either uniform or non-uniform distribution in TiO₂. The three possible states of Co are illustrated in figure 4.1. Clustering of metallic Co or coexistence of clustering and uniform doping are the microstructure of TiO₂ thin films commonly observed due to the difficulty to obtain uniform doping. For our PLD grown 5 at% Co doped TiO₂ thin films, it is found that the microstructures are strongly associated with deposition parameters. In our previous work, magnetic uniformity with Ms between 20-40 emu/cm³ was achieved by the deposition with a laser fluence of 1.0-1.8 J/cm² under the oxygen partial pressure of 10⁻⁶ torr. Segregation of dopants starts to appear by increasing the oxygen partial pressure to 10⁻⁵ torr, leading to a decreased Ms around 20 emu/cm³. However, we further discover that although thin

films deposited at higher laser fluence under same pressure (10^{-5} torr) still show Co clusters, the Ms is enhanced more than 4 times. Therefore, in this work, we focus on Co-TiO₂ film deposited under this condition.

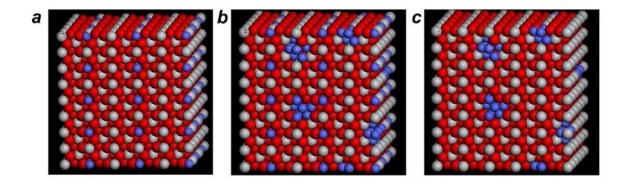


Figure 4.1 Three possible states of Co with TiO₂ matrix: (a) substitution, (b) coexistence of clustering and uniform doping and (c) metallic Co clusters.

Figure 4.2 shows the XRD pattern of 5 at% Co doped TiO₂ deposited under an oxygen partial pressure of 10⁻⁵ torr. Only two peaks can be observed in XRD, one is located around 37.8° and another tiny peak appears around 80.7°. This two peaks are identified as anatase (004) and anatase (008). Thus, the TiO₂ thin film shows pure anatase phase. For PLD with optimal parameters, the deposited materials tend to form a crystalline overlayer which has well-defined crystallographic orientations with respect to the crystal structure of the substrate. This process is called epitaxial growth. Because the substrate is served as a seed crystal, during initial stage of deposition, atoms of deposited film rearrange to match the lattice of substrate surface. LAO has a rhombohedral structure at room temperature (space group R-3c) with lattice *a* and *c* equal to 0.536 and 1.311 nm. However, it is more often represented as pseudocubic structure with a=0.3793 nm. Meanwhile, anatase is tetragonal with lattice parameters a=0.378 nm and c=0.951 nm. The lattice mismatch between anatase (001) and LAO (001) is only -0.26%. Thus, anatase is expected to grow on LAO with orientation relationships of (001)_{anatase}||(001)_{LAO} in out-

of-plane direction and $[100]_{anatase} \parallel [100]_{LAO}$ in in-plane direction, which is in agreement with the XRD result (out-of-plane).

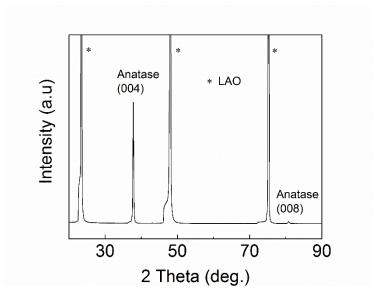
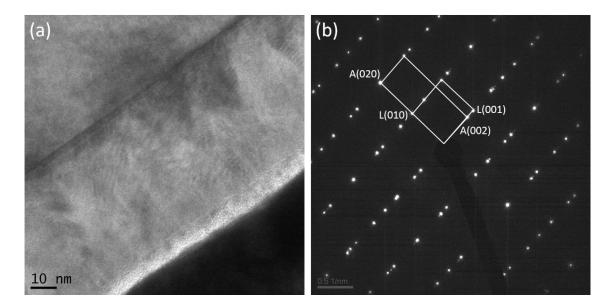


Figure 4.2 XRD pattern of 5 at% Co doped TiO₂ thin film.



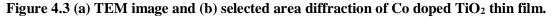


Figure 4.3(a) shows TEM micrograph of the thin film. It can be seen that the thickness of the film is around 60 nm. The SAED pattern shown in figure 4.3(b) reveal the perfect epitaxial structure of the thin film without any polycrystalline dots. Unlike XRD 2θ - ω scan which can only examine the epitaxial relationship in out-of-plane direction. SAED from TEM clearly shows that the in-plane orientation relationships between anatase TiO₂

thin film and LAO substrate is $[100]_{anatase} || [100]_{LAO}$ and $[010]_{anatase} || [010]_{LAO}$, agreeing well with the simulation result.

Nanoclusters with varied dimensions are observed near the interface by high-resolution TEM, as shown in figure 4.4(a). While large clusters with diameters close to 5 nm (marked by circles) are located at the interface, smaller nanocluster with size of just 1-2 nm can also be seen near the interface, as indicated by the arrows. EDS mapping was performed in order to evaluate the distribution of the nanoclusters in the entire film and the results are shown in figure 4.4(b) and (c). Ti and O are uniform in the entire film without precipitates. In contrast, Co is non-uniform in the whole area and most is segregated near the interface, which is consistent with the HRTEM result.

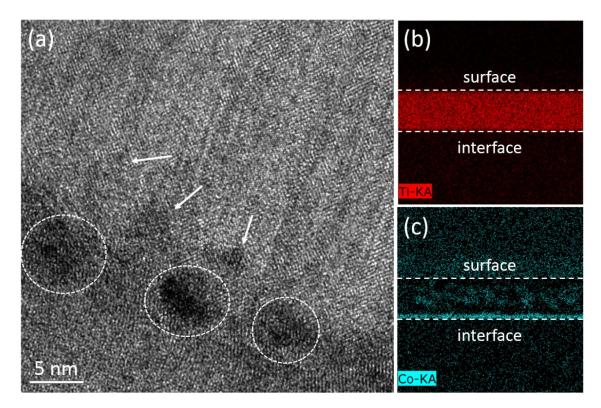


Figure 4.4 (a) High resolution TEM image. EDS mapping of (b) Ti and (c) Co element, respectively.

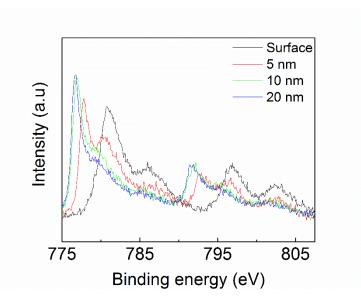


Figure 4.5 XPS spectrum of Co edge (2p core-level).

The presence of Co nanoclusters can be identified by XPS since Co valence state in the clusters should be Co⁰. The XPS spectra of Co 2p core level at different depths are shown in figure 4.5. Because XPS can only collect signals of electrons that escape from the top 5 nm of the thin film, an argon ion beam was employed to etch the thin film to reveal the layers beneath surface. The weak satellite characteristics ascribed to Co²⁺ are observed at the surface. However, the sharp Co⁰ peak with binding energy of ~778.2 eV and significantly spin-orbit splitting (Δ_{metal} =14.99eV) soon becomes dominant after etching. Only at shallow depth of 5 nm, Co shows a mixture of metallic and a tiny fraction of oxidized valence states. Below 10 nm from the surface, XPS only shows Co⁰ state, indicating that the majority of Co in TiO₂ is metallic. Thus, the ionized state of Co in the surface layer may be caused by surface oxidation of the thin film. In order to further identify the valence state of Co dopants in TiO₂ thin film, XANES was performed in Singapore Light Synchrotron [159]. As shown in figure 4.6(a), compared with CoO and Co reference samples, the spectrum of Co in TiO₂ is metallic state. Fourier transformation, illustrated in

figure 4.6(b), further confirms that Co dopants in TiO_2 is Co⁰ since it has the same Co-Co distance and coordination numbers as that of metallic Co.

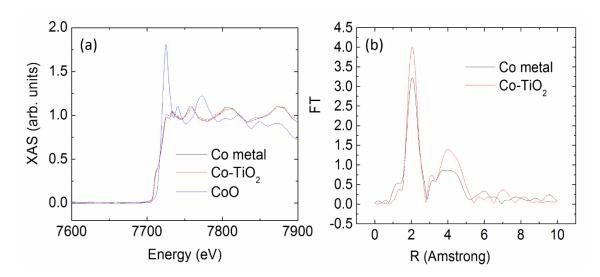


Figure 4.6 (a) NEXAS of Co K edge in Co doped TiO₂ and two reference samples, Co metal and CoO. (b) Fourier transformation of reference Co metal and Co in Co doped TiO₂.

4.3 Magnetic properties of Co doped TiO₂ thin film

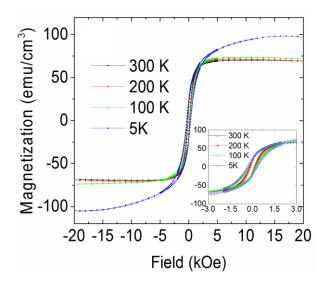


Figure 4.7 Hysteresis loops measured at different temperatures.

Figure 4.7 shows the M-H loops of Co doped TiO_2 thin film measured at different temperatures. The film shows a surprisingly high saturation magnetization of 70 emu/cm³ at room temperature, which is even higher at 5 K (98 emu/cm³). From the inset, the coercivity and remanence magnetization (~17 emu/cm³) of the hysteresis loops confirm

the ferromagnetic ordering at room temperature. The fast increase in magnetization at 5 K suggests there may be some paramagnetic phases at room temperature. Given that the magnetization is entirely contributed by the 5 at% Co dopants, each Co atom has a magnetic moment of 4.73 μ_B at room temperature. It is known that bulk metallic Co has a magnetic moment of 1.72 μ_B /Co, which is much lower than those observed in this work.

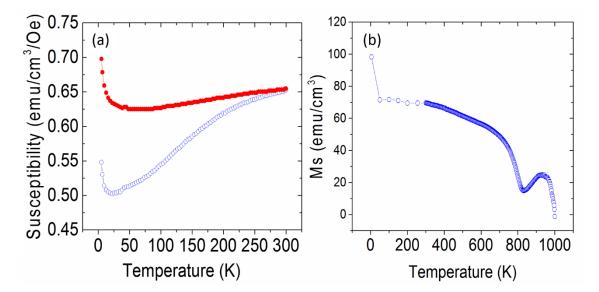


Figure 4.8 (a) ZFC/FC curves of Co doped TiO₂ thin film. The magnetic field applied is 500 Oe. (b) Temperature dependence of magnetization that shows the Curie temperature.

Figure 4.8(a) shows the ZFC and FC curves of the thin film. The relatively high magnetization at low temperature indicates the presence of paramagnetic orderings, consistent with hysteresis loop measurement. Superparamagnetism is frequently observed in systems with ultra-small magnetic nanoparticles (Fe, Co, Ni). For example, ion beam fabricated Co nanoparticles with size smaller than 3 nm lead to superparamagnetic behaviour in SiO₂ film while clusters above 7 nm only result in ferromagnetism [160]. Superparamagnetism can also be observed in Co ion-implanted TiO₂ thin film, which disappears after postannealing due to enlarged cluster size [161]. However, no superparamagnetic feature can be observed in the ZFC/FC curves, suggesting the estimation of nanocluster size from TEM may not accurate and the smaller nanoclusters are still above the critical diameter. In order to further understand the mechanism of the

ferromagnetism, Curie temperature was measured in a furnace attached to a SQUID system. Two critical temperatures, 800 and 1000 K, are observed in the curve (figure 4.8(b)). From analysis above, Co dopants in TiO₂ have complex compositions, including at least large and small nanoclusters in the interface and oxidation at shallow depth, which may have different critical temperatures. It should be noted that 1000 K is still much lower than the Curie temperature of bulk Co (1400 K), suggesting nanostructured Co can reduce the Curie temperature.

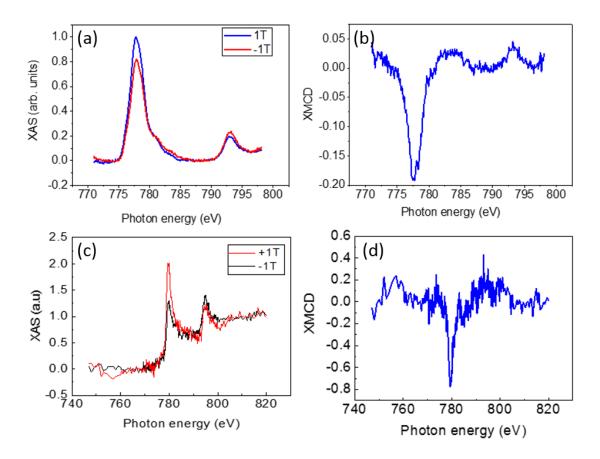


Figure 4.9 XAS and XMCD spectra of Co L edge. The signals are collected by (a, b) total yield mode and (c, d) Florence mode.

SQUID measurement can only provide the overall magnetization of the entire sample. To uncover if Co dopant alone contributes to the high magnetic moment, XMCD was employed to evaluate the individual moment contribution from each element in Co doped TiO₂ thin film. Figure 4.9(a) present the XAS spectra of Co L edge by applying a magnetic field of 1T and -1 T using the total yield mode and the correspond XMCD results of Co are shown in figure 4.9(b). From sum rules [162], in the transition of XAS

from 2p
$$\rightarrow$$
3d, the lz sum rule can be written as: $\langle l_z \rangle = -2 \frac{\int d\varepsilon (I_+ - I_-)}{\int d\varepsilon (I_+ + I_- + I_0)}$, where $\langle l_z \rangle$

is the average of the magnetic quantum number of the orbital angular momentum per 3d hole. I_0 is the XAS with the linearly polarized light whose polarization vector is parallel to the magnetization. The integration $\int_{j=3/2} (\int_{j=1/2})$ is taken over the whole $2p_{3/2}(2p_{1/2})$ absorption region. The S_z sum rule is written as: $\langle S_Z \rangle + \frac{7}{2} \langle t_Z \rangle = -\frac{3}{2} \frac{\int_{j=3/2} d\varepsilon (I_+ - I_-) - 2 \int_{j=1/2} d\varepsilon (I_+ - I_-)}{\int d\varepsilon (I_+ + I_- + I_0)}$, where t_Z is the z component

of the magnetic dipole operator $t = s - 3r(r \cdot s) / |r|^2$. The integration is taken only over the absorption region.

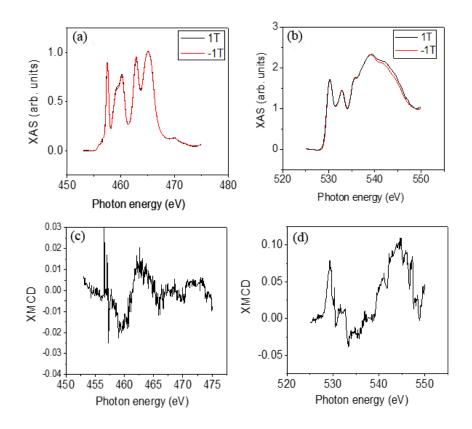


Figure 4.10 XAS spectra of (a) Ti L edge and (b) O K edge. XMCD spectra of (c) Ti L edge and (d) O K edge. The XAS is collected by total yield mode.

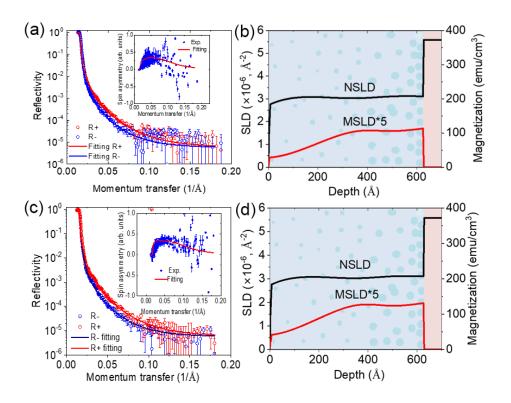


Figure 4.11 PNR spectra of Co doped TiO_2 film. (a) Reflectivity versus moment transfer and (b) SLD curves at 300 K; (c) Reflectivity versus moment transfer and (d) SLD curves at 5 K.

Considering the beam polarization is 0.9, the calculated orbital moment μ_L and spin moment μ_S are 0.62 and 0.96 μ_B , respectively. The total magnetic moment (1.58 μ_B) is much lower than that obtained from magnetic hysteresis loop measurement (4.73 μ_B at room temperature). The XAS signals in the total yield mode are mainly collected from surface and oxide based diluted magnetic semiconductors often have a dead layer on the surface, leading to an underestimated magnetic moment [163]. Hence, XMCD is further measured by the Florence mode, as shown in figure 4.9 (c). From the calculations based on sum rules, μ_S =3.06 μ_B and μ_L =0.43 μ_B (Using step function to remove the background). Hence, the total magnetic moment of Co is approximately 3.49 μ_B . However, it is still much lower than that obtained from hysteresis loop measurement by SQUID. XAS and XMCD of Ti and O element were further measured and the results are shown in figure 4.10. While XMCD signal from Ti L edge is insignificant, a large XMCD response can be obtained from O K edge. Due to the lack of hole information, we can no longer calculate the magnetic moments of Ti L and O K edge by using sum rules, but they may contribute to the high magnetization of the thin film in addition to Co nanoclusters.

As discussed above, the Co doped TiO_2 thin film is not uniform, in which the formation of precipitation tends to take place in the interface and a magnetic dead layer may form in the surface. The absence of spatial uniformity has been reported in numerous studies, but how the unevenly distribution of dopants influence the magnetic properties remains unclear [163-166]. To identify the magnetization in the interface and the dead layer is critically important for understanding the magnetism in the film. In this work, the depth profile of the magnetism is evaluated by PNR.

Although the thickness of 60 nm is estimated from TEM image, the reflectivity curves shown in figure 4.11 (a) and (c) do not display clear oscillations, indicating the surface is very rough. We fitted the curves considering the thickness measured by TEM, saturation magnetization measured by SQUID, and microstructure identified by TEM and XPS. Figure 4.11 (b) and (d) are nuclear scattering length density (NSLD) and magnetic scattering length density (MSLD) at 300 and 5 K, respectively. Slightly lower NSLD in the surface is expected due to surface roughness. Interestingly, a sunken area appears at round 38 nm in NSLD curve, which corresponds the peak of MSLD, the area with the highest magnetization. The sunken area in NSLD suggests relatively high concentration of Co clusters since Co $(2.265 \times 10^{-6}/Å^2)$ has a lower NSLD than that of TiO₂ $(2.684 \times 10^{-6}/Å^2)$. From TEM and EDS mapping, in addition to the relatively large Co clusters located near the interface, there are also very small Co clusters distributed inside the film, which may be attributed to the magnetization peak. On the other hand, the surface area of the thin film present minor Ms from Co²⁺, indicating that Co nanoclusters play a decisive role

in the high magnetization. The average Ms of the thin film obtained from MSLD is around 75 emu/cm^3 at 300 K and 99 emu/cm^3 at 5 K, well agreed with the SQUID results. Moreover, the highest Ms from the peak of MSLD reaches around 106 emu/cm^3 .

4.4 First-principles calculations

In order to further understand the mechanism behind the high magnetic moment, we asked collaborators (Dr. Xiangyuan Cui and Prof. Simon P. Ringer) performed first-principles spin-polarised calculations under density functional theory (DFT). Considering the challenging nature of magnetic moment involved, especially the orbital contribution [167, 168], we employed different schemes for the electron exchange and correlation functionals, namely generalized gradient approximation (GGA) [169], the mean field approximation GGA+U (with a large Hubbard U parameter, U=6 eV for both Co-3d and Ti-3d [168, 170]) and the screened hybrid functional of Heyd, Scuseria and Ernzerhof (HSE) [171] (the Hartree-Fock mixing parameter is set to 0.2), as implemented in the VASP code [172]. The spin-orbit coupling (SOC) was included self-consistently. The cutoff energy of plane-wave basis set and the energy convergence criterion between two electronic steps were 500 eV and 10⁻⁴ eV, respectively. The first Brillouin zone was sampled by a k-mesh of 2 \times 2 \times 1 for the 3 \times 3 \times 1 anatase TiO₂ supercell containing 108 atoms. In anatase TiO₂, each Ti atom is coordinated to six O atoms, and each O atom to three Ti atoms. To construct the metallic clusters in TiO₂ supercell, we first removed some central adjacent Ti and O atoms and then filled some Co atoms. For instance, we removed one Ti atom and its six bonding O atoms, and filled with 6 or 7 Co atoms. To accommodate larger Co clusters (11Co, 14 Co and 15 Co), we further removed three or four Ti atoms. This leads to strong local structural distortion as revealed by internal atomic relaxation. Subsequently, the volume relaxation by keeping c/a constant and

lattice angles as 90 ° was conducted. The specific numbers of removed and added atoms, and the relaxed lattice constants of the supercells are shown in Table 4.1. Atomic positions were optimized in all the calculations until the maximum force on each atom is less than 0.02 eV/Å.

Figure 4.12 illustrates the spin density of various Co nanoclusters along [100], [010] and [001] spin axis directions by HSE-SOC. It is to note that for 11Co, the HSE-SOC calculation did not converge. Calculations show that the easy spin axis for Co nanoclusters is along [001] direction with an isosurface value of 0.02 electron/Å³, while isosurface value is 0.0001 electron/Å³ for [100] and [010].

The calculated ground state spin and orbital contributions to the magnetic moments of metallic Co clusters in TiO₂ along the easy spin axis [001] direction by different methods are summarized in Table 4.2. It is evident that orbital polarization contributes sizably to the overall magnetic moment, consistent with experimental results. Moreover, it is seen that both spin moment and orbital moment calculated from GGA+U and HSE are substantially larger than those of from GGA. As expected, due to the constrained interatomic electron motion [173], the magnetic moments of the embedded metallic Co clusters are smaller than isolated Co clusters. However, as discussed afore, the isolated clusters have low Curie temperature and are unsuitable for practical applications [168]. For these three methods in our work considering spin-orbit coupling, both the spin and orbit moments give similar trends regarding the spin and orbital moments for the small clusters.

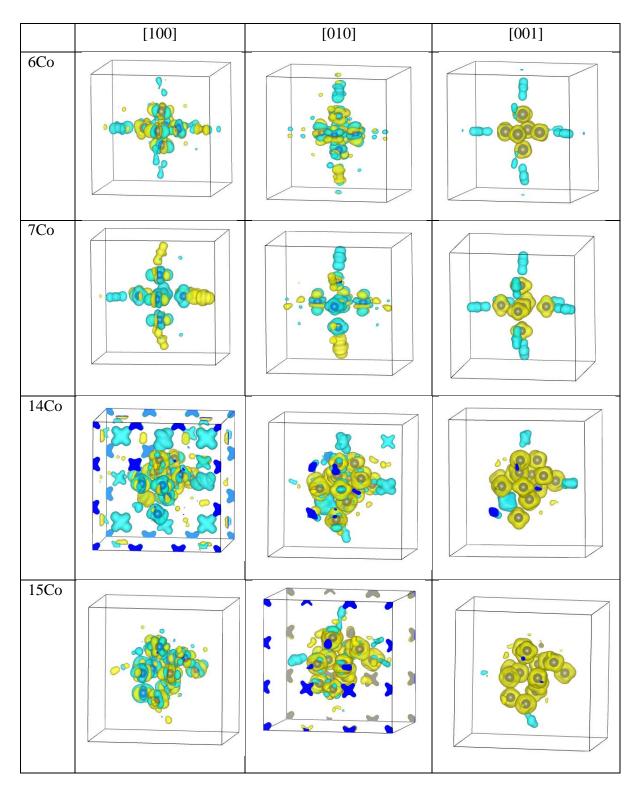


Figure 4.12 The calculated spin density plots along different spin axis directions for various metallic Co clusters embedded in TiO₂ matrix (HSE-SOC). Yellow (blue) isosurfaces denote positive (negative) spin polarization. For simplicity, the Ti and O atoms are omitted.

Based on the limited number of Co clusters considered, the magnetic moment seems to increase with the metallic Co cluster size for GGA calculation. However, for GGA+U+SOC calculation, the magnetic moment reaches maximum for 14 Co nanoclusters, which has the same trend as that calculated in Ref. [174]. For HSE-SOC method, the magnetic moment also reaches maximum for 14 Co atoms and 15 Co atoms. A general feature regarding the magnetic structure of these clusters is that while the magnetic interaction among Co atoms are always ferromagnetic, there are some antiferromagnetic ones between the Co clusters and their surrounding Ti and O atoms. Such a cancellation effect leads to the slightly decreased moments observed in the smaller clusters. However, from XMCD results, O edge shows relatively large moment, which may be from the oxygen atoms surrounding Ti vacancies.

Systems	Number of Ti	Number of O	a=b (Å)	c (Å)
	atoms	atoms		
	removed	removed		
0 Co	0	0	11.474	9.661
6 Co	1	6	11.212	9.441
7Co	1	6	11.578	9.749
11Co	4	6	11.678	9.833
14Co	5	6	11.822	9.954
15Co	5	6	12.102	10.190

Table 4.1 Composition and lattice constants of different metallic Co clusters embedded in TiO₂ matrix (GGA).

From the table, we can also see that the contribution of the orbital moment cannot be negligible. However, compared with XMCD measurement, the DFT calculated orbital moments are considerably smaller. Such an underestimation by DFT has been well documented in the literature [168, 175-177]. XMCD measurement shows that 5 at% Co

in TiO₂ has a magnetic moment as high as 3.5 μ_B /atom, which is close to that calculated from GGA+U-SOC and HSE-SOC, but much higher than GGA-SOC. It is known that for transition metal or metal oxide, GGA calculated result is always underestimated. Hence, the first-principles calculations based on GGA+U+SOC and HSE+SOC support the possible high magnetic moments achieved in Co doped TiO₂ thin film.

systems	GGA-SOC		GGA+U-SOC		HSE-SOC	
	Spin	Orbital	Spin	Orbital	Spin	Orbital
	moment	moment	moment	moment	moment	moment
	(µ _B)					
6Co	1.21	0.14	2.24	0.21	1.89	0.17
7Co	1.21	0.16	2.29	0.25	1.94	0.21
11Co	1.41	0.19	2.48	0.27	-	-
14Co	1.70	0.20	2.62	0.25	2.42	0.24
15Co	1.94	0.19	2.61	0.22	2.44	0.22

Table 4.2 Calculated average spin and orbital moments in μ_B /atom along the easy spin axis [001].

We also performed the calculation without SOC. The calculated magnetic moments are listed in table 4.3. As one can see, for most clusters (all except 15 Co), GGA predicts slightly larger spin moments than GGA-SOC. In contrast, including the strong correlation effect in the embedded systems, both GGA+U and HSE consistently predict that the calculated spin moments are much larger when the SOC is included compared to the case when those without SOC for all the embedded clusters are considered. Hence, SOC is one of the factors to affect the magnetic properties of clusters.

To understand the mechanism of high Curie temperature in Co doped TiO_2 , we performed further DFT calculations by comparing confined Co clusters embedded in TiO_2 matrix and free Co clusters. We studied free Co clusters under two states: One is to remove TiO_2 matrix and allow the Co clusters relax and the other is to freeze the atomic position of Co after removing TiO₂ matrix. By comparing the total energy of the different spin states by considering the diverse initial spin configurations, we observe that Co clusters embedded in TiO₂ matrix always leads to robust ferromagnetic ordering with high Curie temperature. In contrast to confined clusters, free Co clusters, either the relaxed and the fixed states, both show various energetically competitive local spin states besides ground states, suggesting that unstable states with low Curie temperatures. Therefore, it can be concluded that confinement effect in the Co clusters embedded in TiO₂ matrix is responsible for the high Curie temperature above room temperature. Compared to previous experimental results, Co clusters embedded in TiO₂ matrix actually have slightly lower magnetic moment compared to free Co clusters. Hence, the enhancement of magnetic moment comes from the nanostructure of Co itself. It is to note that Ti and O edge also contribute to the high magnetization. Nevertheless, the confinement effect induced high Curie temperature above room temperature may pave ways to achieve practical high magnetization materials if these nanoparticles are embedded in oxides. In addition, the embedded Co clusters are structurally, chemically and magnetically protected by the matrix, hence more stable magnetic properties, compared with free metallic clusters.

systems	$GGA(\mu_B)$	$GGA+U(\mu_B)$	HSE (μ_B)
6Co	1.25	2.01	1.78
7Co	1.27	2.21	1.85
11Co	1.45	2.42	2.01
14Co	1.79	2.51	2.22
15Co	1.89	2.47	2.27

Table 4.3 Calculated average spin moments in µ_B/atom without spin-orbit coupling (SOC).

4.5 Conclusions

5 at% Co doped TiO₂ thin film was deposited on LaAlO₃ (001) substrate using a pulsed laser deposition system. XRD result indicates the film is pure anatase phase. Further SAED analysis suggests the orientation relationships between the anatase thin film and LAO substrate are (001)_{anatase}||(001)_{LAO} in out-of-plane direction and [100]_{anatase}||[100]_{LAO} in in-plane direction. Segregation of Co with different sizes are observed near the interface. Larger clusters with diameters around 5 nm are located close the interface, while 1-2 nm small nanoclusters are embedded inside the film. Both XPS and XAS confirm most Co exists as metallic Co⁰ except a small amount of Co²⁺ in the surface layer.

Assuming that all the magnetization comes from Co, SQUID measurement gives an extremely high magnetic moment of 4.73 μ_B per Co atom at room temperature, while the result calculated by XMCD is ~3.5 μ_B /Co. Moreover, observed from XMCD, although Ti edge contributes little to the overall magnetization, the signal from O edge cannot be ignored. Further PNR measurement supports the high magnetization of the films and the highest saturation magnetization at the peak area is around 106 emu/cm³.

First principles calculations by GGA+U and HSE support the high magnetic moment of Co when Co clusters are embedded in TiO₂ matrix considering spin-orbit coupling, though GGA method underestimates the magnetic moment. A 15-atom Co cluster can obtain a high magnetic moment of 2.9 μ_B /Co, demonstrating that magnetic clusters embedded in oxide matrix are able to show high magnetic moment above room temperature.

Chapter 5 Cluster-dependent Magnetism in Co-doped TiO₂ Thin Film

5.1 Overview

 TiO_2 , with various crystal and band structures, has great potential to serve as host material in diluted magnetic semiconductors. Doped with cobalt, Co-TiO₂ has demonstrated to be a remarkable promising DMS system with room-temperature ferromagnetism observed in both rutile and anatase thin films. However, after two decades of extensive research, several issues concerning the microstructure and magnetic properties of Co doped TiO₂ thin films are still not thoroughly understood.

The first issue is the solubility limit of Co in TiO₂. In the initial work of Matsumoto's group, the uniform distribution of Co can be achieved with concentration as high as 8% [56]. Several later studies further push the solubility limit up to 12% [60, 178]. However, the solubility limit in other reports are much lower. For example, PLD grown Co doped TiO₂ thin film in Shinde's group can only tolerate the solubility of Co up to 2%, while higher dopant concentrations clear lead to formation of Co clusters [74]. Based on current collective data from numerous studies, uniformly distributed Co in TiO₂ can be accomplished at least at low concentrations of less than 2-3% under certain growth conditions.

The second issue is the lack of clear relationship between the growth methods and film microstructure. In an early work, Chambers *et al.* investigated oxygen-plasma-assisted MBE grown Co doped anatase TiO₂ thin film at different growth rates and temperatures varied between 300 and 750 °C [58]. The epitaxial thin films with highest crystalline quality and most evenly dopant distribution are deposited at temperatures in the range of

550-600 °C with a slow growth rate around 0.01 nm/s. In the case of higher deposition rate (~0.04 nm/s), rutile nanocrystals appear on the anatase thin film surface due to segregation of Co. Further alter growth conditions can cause all Co dopants segregate to nanoclusters [59].

The final problem is the uncertainty of the relationship of microstructure and magnetic properties, which often leads to contradictory results. For example, in ref.[59], the homogeneous Co doped anatase TiO₂ thin film shows no ferromagnetic properties, while tiny magnetic moment of ~0.1 μ_B /Co was shown in ref.[72]. In contrast, our previous work discovered the magnetic uniformity with Ms of ~1.03 μ_B /Co by deposition with a laser fluence of 1.0-1.8 J/cm² under the oxygen partial pressure of 10⁻⁶ torr [151]. Similarly, no consensus has been reached for Co doped rutile TiO₂, although many believe each Co atom possesses a magnetic moment of ~1 μ_B /Co by low spin configuration. The typical magnetic moments observed in Co-TiO₂ so far range widely from 0.01 μ_B /Co to 1.7 μ_B /Co, suggesting immature understanding and control of ferromagnetism in the system.

In the last chapter, we achieve an ultrahigh magnetic moment of 4.73 μ_B /Co at room temperature from metallic Co nanoclusters embedded in TiO₂ matrix. In order to further clarify how the film microstructure influences magnetic properties, we fabricated Co clusters doped TiO₂ thin film with different cluster sizes and distributions by delicate control of the growth parameters during PLD fabrication. We used the same 5 at% Co doped TiO₂ target but adopted SrTiO₃ (001) substrate this time. The lattice mismatch between anatase (001) and STO (001) is -3.1%, much larger than using LAO (001) (-0.26%), so the tendency of Co segregation is higher. The laser fluence and oxygen partial pressure were kept constant at ~3 J/cm² and 10⁻⁵ Torr. Two deposition rate and a wide range of substrate temperatures were selected for the growth of four films, as shown in

table 5.1. Results show that the magnetic moment of Co can exceed 2 μ_B even with cluster size larger than 20 nm. This work provides a new strategy to design high magnetization for heterogeneous DMSs with a nonrandom distribution of dopants.

Film number	Substrate temperature (°C)	Deposition rate (nm/s)
1	500	0.1
2	500	0.01
3	650	0.01
4	800	0.01

Table 5.1 Growth parameters of Co doped TiO₂ thin films.

5.2 Results and discussion

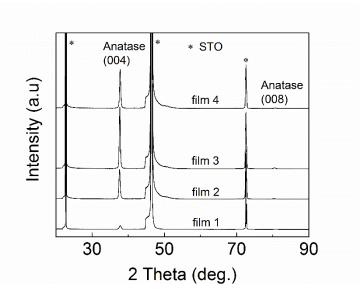


Figure 5.1 XRD patterns of 5 at% Co doped TiO₂ films with different growth parameters. Figure 5.1 displays the XRD patterns of 5 at% Co doped TiO thin films deposited under various temperatures and speeds. Regardless of the difference in deposition temperature and rate, only diffraction peaks of anatase phase are observed in all the films. However, the intensity of (004) anatase peak of film 1 is almost 10 times lower than other three films, indicating high deposition rate can reduce the crystalline quality of TiO₂ films. The

peak becomes both higher and shaper with reduced growth rate and higher temperature, suggesting better degree of crystallinity under those conditions.

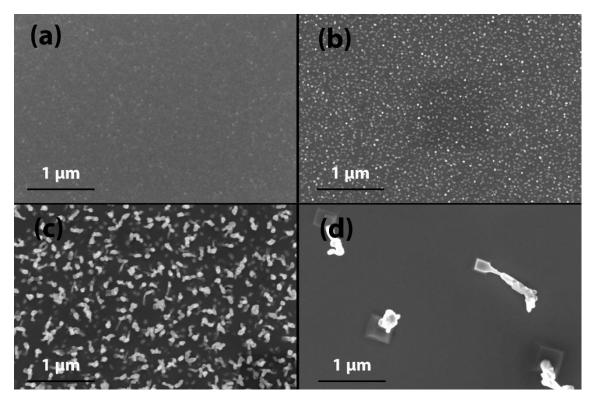


Figure 5.2 SEM images of Co doped TiO₂ thin films, (a) film 1, (b) film 2, (c) film 3, and (d) film 4.

Although all four films share the same crystalline structure, there are huge discrepancies among their surface morphologies. Only film 1 has a rather smooth surface, as shown in figure 5.2. With lower deposition rate, small nanoparticles with typical size of 30~40 nm begin to appear in film 2. As growth temperature rises, not only does the particle size remarkably enlarges, but the distance between neighbour particles becomes notably larger. The densely packed nanostructure vanishes when deposition temperature reaches 800 °C as particles are separated by more than 1 micrometre and each particle is attached to a rectangular-shaped pyramid which should be the anatase grain. This sudden change of shape, from rectangular to irregular, is likely resulting from the change of surface energy due to Co aggregation at grain boundaries. The surface morphology of the films is further demonstrated by AFM, as shown in figure 5.3. The root mean square (RMS) roughness of film 1 is smaller than 1 nm. The height of nanoparticles on film 2 and 3 lays in the range of 20-50 nm and 100-250 nm, respectively, while the particles on film 4 can even reach larger than 800 nm. It should be noted that this result is in direct contradiction to the conclusion in ref.[58], in which the epitaxial thin films with highest crystalline quality and most evenly Co distribution are deposited with a slow growth rate around 0.01 nm/s using MBE method. This total opposite tendency suggests the mechanism of film growth is very complex and may not have certain rules to guide the deposition.

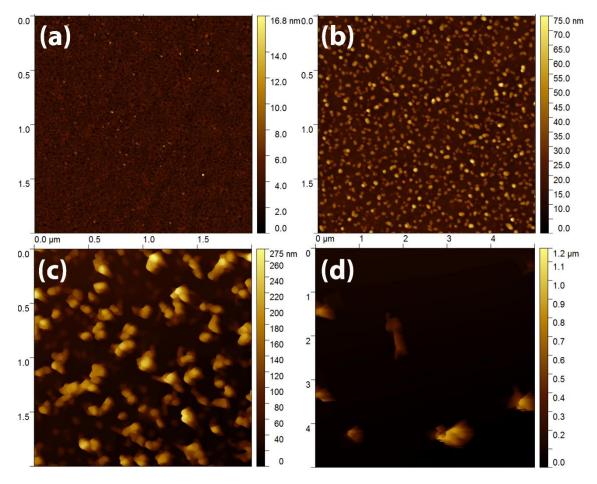


Figure 5.3 AFM images of Co doped TiO₂ thin films, (a) film 1, (b) film 2, (c) film 3, and (d) film 4.

Cross-section bright-field TEM images and SAED patterns of four thin films are presented in figure 5.4. In agreement with SEM and AFM, only film 1 has a rather smooth

surface. Film 2, 3 and 4 are mainly composed with a continuous anatase film plus various particles on top. SAED pattern of film 1 shows a polycrystalline feature, while the epitaxy gets better crystallinity when higher temperature and slower growing rate are applied.

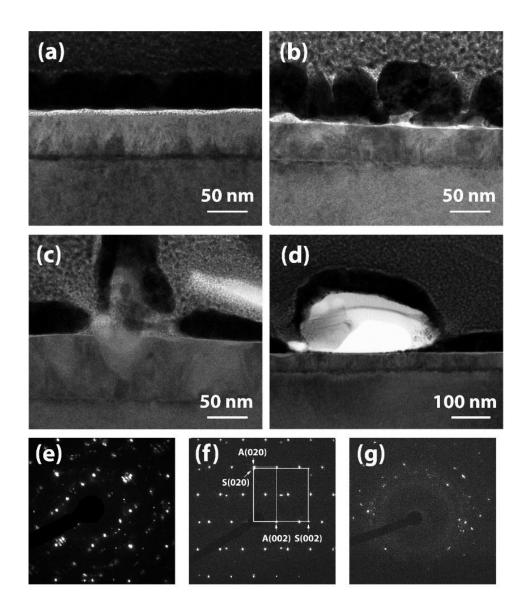


Figure 5.4 TEM images of (a) film 1, (b) film 2, (c) film 3, and (d) film 4. SAED of (e) film 1, (f) the continuous film part of film 2, 3, and 4, and (g) the particles on the surface of film 3 and 4.

From figure 5.4(f), the epitaxial relationship between film 2, 3, 4 and STO substrate are $[100]_{Anatase} \parallel [100]_{STO}$, $[010]_{Anatase} \parallel [010]_{STO}$ (in-plain), and $(001)_{Anatase} \parallel (001)_{STO}$ (out-of-plain). We also observe that [020] Anatase and [020] STO do not overlap in SAED, which differs from the film grown on LAO substrate, as shown in figure 4.3. In epitaxial growth,

if the film is mismatched to the substrate, then the film is driven to match the substrate lattice in the lateral direction. Based on the degree of distortion, the film can be fully strained if the lateral film lattice is equal to the substrate, fully relaxed if the lattice parameters are identical to the bulk values, or in a partially strained intermediate state. Figure 5.5 displays the schematic diagram of strained and relaxed film. The lattice mismatch between anatase (001) and STO (001) is -3.1%, which is equal to the difference of [020] Anatase and [020] STO in figure 5.4(f), indicating the epitaxy is fully relaxed. SAED obtained from mere particles in film 3 and 4, as shown in figure 5.4(g), suggests the particles are polycrystals.

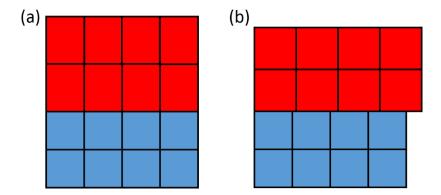


Figure 5.5 A shematic diagram of (a) strained film and (b) relaxed film.

The crystalline nature of films can be also obtained from the fast Fourier transform (FFT) of high-resolution TEM images, as shown in figure 5.6. While film 1 is polycrystalline, some regions (figure 5.6(b)) are better crystallized than others (figure 5.6(a)). Figure 5.6 (c) and (d) are places in film 2 near and away from the particle on the surface. It is observed both locations are single crystalline, but the points in the FFT of former area are not as sharp as those in the latter, indicating film structure near the particles suffers slight distortion. Both film 3 and film 4 are perfect single crystalline, as shown in figure 5.6 (e) and (f).

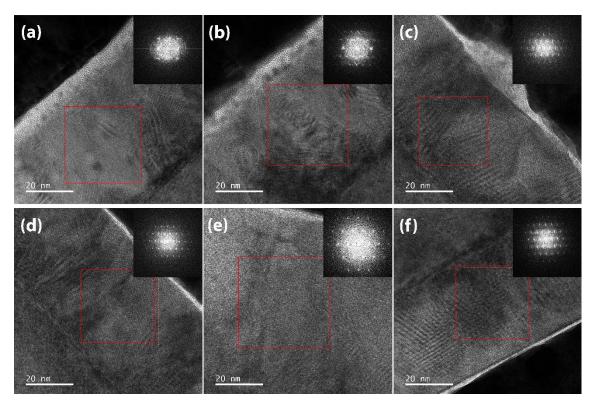


Figure 5.6 HRTEM of Co doped TiO₂ thin films, insets are FFT images of the areas marked by red rectangular. (a) and (b) film 1, (c) and (d) film 2, (e) film 3, (f) film 4.

EDS mapping and SIMS were carried out to examine the spatial distribution of the Co in the films, as shown in figure 5.7. Co tends to aggregate as nanoclusters near the interface in film 1, confirming by the intensity maximum near the interface in SIMS. In film 2, Co nanoclusters tend to grow bigger and migrate toward surface. Large clusters can be observed in the vicinity of the particles on film surface, while some nanoclusters still remain near the interface, as two peaks can be seen in the SIMS of film 2. Nearly all Co elements are blended into the surface particles in film 3, except for a minority of smaller clusters within the film near the particles. Therefore, the SIMS intensity in film 3 quickly reaches the maximum near the surface. Co can be hardly detected through EDS in film 4 (The red dots in figure 5.7(d) are from noises in gold and other elements). From crosssection view of TEM and SEM images, there is a hole inside nearly all the particles in film 4. This might be caused by evaporation of Co due to long-time exposure on the surface at high temperature, which explains the absence of Co dopants in EDS. In addition, the extremely low SIMS intensity of film 4 also suggests Co may have been evaporated during deposition, at least to some extent. The schematic of film growth is illustrated in figure 5.8. In an early work, Chambers *et al.* investigated oxygen-plasma-assisted MBE grown Co doped anatase TiO₂ thin film at different growth rates and temperatures [58]. The epitaxial thin films with highest crystalline quality and most evenly dopant distribution were deposited at temperatures between 550-600 °C with a slow growth rate around 0.01 nm/s. However, similar parameters lead to Co segregation in our work, suggesting film growth is extremely sensitive to the equipment.

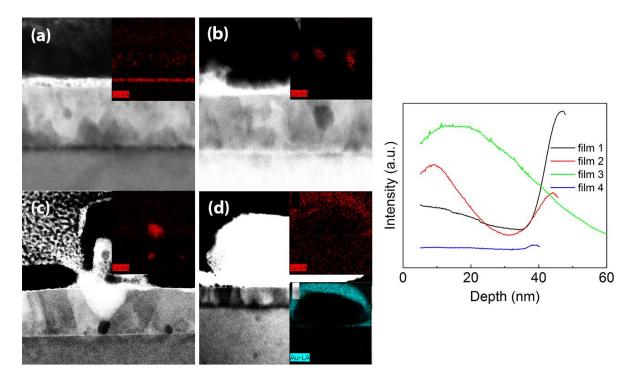


Figure 5.7 ESD mapping of Co in the thin films (left). (a) film 1, (b) film 2, (c) film 3, (d) film 4. SIMS depth profiles for Co (right).

The valence state of Co is of vital importance to the film's magnetic performance. Co 2p core-level XPS collected at the surface of film 2-4 show the characteristic peaks which can be ascribed to the 2+ valence state (Figure 5.9(a)), with strong satellite peaks and a spin-orbit splitting of ~16.0 eV. The Co 2p peaks of film 1 shifts to lower binding energy. In addition, the spectrum has a smaller spin-orbit splitting (~15.3 eV) and decreased satellite to main peak intensity ratio. These features indicate the appearance of Co³⁺ ions

in film 1. As the cobalt precursor in the target is Co_3O_4 , this suggests the growth condition of film 1 can prevent the reduction of Co_3O_4 at the surface. The peak intensity of film 4 is considerably lower than the other three, which is consistent with partial evaporation assumption. However, XPS can only collect signals of electrons that escape from the top layer of the films, resulting in a quite shallow characterization depth. To unveil the valence state of Co dopants in deeper layers, an argon ion beam was used to etch the films for 90 s to reach ~20 nm below original surface. The XPS spectra, as presented in figure 5.9(b), show totally different features compared to those taken at the surface. The sharp Co^0 peak is highly consistent with Co metal nanoclusters in TiO₂ grown on LAO investigated in last chapter. Therefore, Co^{2+} can only be observed at the very surface, likely due to surface oxidation. Moreover, the clear Co^0 signal from film 4 suggests Co dopants cannot be totally evaporated.

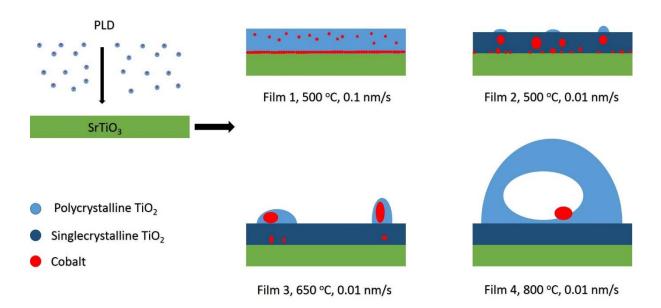


Figure 5.8 Schematic of film growth during PLD.

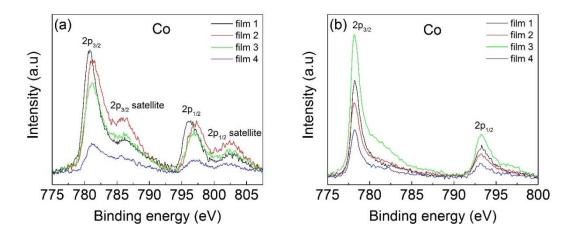


Figure 5.9 XPS spectra of Co doped TiO₂ thin films: (a) original surface, (b) after 90s ionbeam etching.

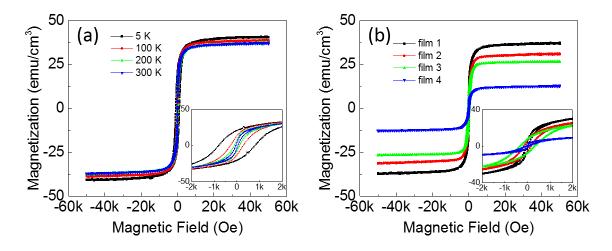


Figure 5.10 (a) M-H loops of film 1 measured at different temperatures. (b) M-H loops of four thin films measured at room temperature.

The hysteresis loops of film 1 measured at various temperatures are shown in figure 5.10(a). The film shows obvious ferromagnetic features with a Ms of 40.5 emu/cm³ at 5 K. The ferromagnetic ordering is further confirmed by the coercivity of the M-H loops, as shown in the inset. Increasing the temperature does not lead to a dramatic drop in Ms, which remains mostly at the same level of 36.9 emu/cm³ above 200 K. The persisted Ms to 300 K suggests the Tc of the film is above room temperature. Considering the entire magnetic moment is contributed by Co, the dopant has a magnetic moment of 2.49 μ_B /Co at room temperature. Figure 5.10(b) shows the hysteresis loops of different films at room

temperature, with the Ms constantly decreases from film 1 to 4. Despite the reduced Ms, film 2 and 3 have significantly enlarged coercivity compared to film 1, as shown in the inset of figure 5.10(b). The coercivity of film 3 measured at low temperature is even larger, reaching ~1k Oe.

It is well acknowledged that the presence of Co clusters gives rise to ferromagnetism persisted to above room temperature in inhomogeneous films. It is further confirmed with recent polarized neutrons reflectometry results that magnetization merely originates from Co-rich regions and shares similar in-depth pattern to Co concentration [179]. However, magnetic moments induced by Co nanoclusters can hardly reach above 2.5 μ_B/Co regardless of the states of the clusters, including Co clusters embedded in Cu and SiO₂ matrixes, and free-standing clusters measured via molecular beam deflection [153, 180]. With very small sizes (fewer than 30 atoms), Co shows atom-like magnetic moments (~2.4 μ B/Co), which decrease toward the bulk limit with increased cluster size. According to first principles calculation, magnetic moments of free Co clusters can vary based on different geometries, but all below 2.6 $\mu_{\rm B}$ /Co [181, 182]. Other Co nanostructures, such as nanowire arrays, generate similar magnetic moments [183]. Magnetic moments between 3-4 μ_B /Co have only been reported in ultra-small free Co nanoclusters with less than 20 atoms at extremely low temperature, contributed from extra unquenched orbital moments [184-186]. However, such moments are significantly reduced with increased cluster size and temperature. Furthermore, whether the enhanced moments can persist after dispersing in a matrix remains doubtful as interactions between ultra-small magnetic particles frequently lead to superparamagnetic or spin-glass behaviours [160, 187]. The magnetic moments observed in our Co-doped TiO₂ (2.49 μ_B /Co in film 1) is consistent with previous studies. Furthermore, although most Co clusters observed in film 2 have diameters in the range of 10-20 nm, the film still shows a Ms higher than 2 μ B/Co,

indicating the Ms may be more sensitive to the distribution of Co clusters than the mere size. The thin films can only maintain high magnetic moments when Co clusters are distributed inside the films, demonstrated by the sharp drop of Ms in film 4. Moreover, it is believed a variety of defects, such as Ti³⁺ ions and oxygen vacancies, can also induce magnetic moments [149, 188, 189]. Thus, the high magnetization is likely attributed to the combination of Co nanoclusters and various defects.

5.3 Conclusions

Controllable formation of Co nanoclusters embedded in TiO₂ are achievable through pulsed laser deposition on SrTiO₃ (001) substrate. The film is polycrystalline with high deposition rate at 500 °C, in which Co segregates into small clusters at the interface. When grown at slow rate and high temperatures, fully relaxed epitaxial films are obtained. Meanwhile, the Co nanoclusters increase in size with increasing temperature and move toward surface, causing the formation of nano- and even micro- particles on the surface. XPS result suggests that the Co dopants mainly exist as Co⁰, which is consistent with the film grown on LAO. SQUID measurement shows a highest magnetic moment of 2.49 $\mu_{\rm B}$ /Co at room temperature. Although the magnetic properties are sensitive to the size and distribution of Co nanoclusters, large magnetization can still be achieved by Co nanoclusters with size bigger than 20 nm. A part of the high magnetic moments may originate from Ti and O defects.

Chapter 6 Defect Engineering and Magnetic Properties of TMDCs

6.1 Overview

2D materials with atomic thickness have attracted considerable attention since the discovery of graphene, the monolayer counterpart of graphite. Currently, graphene is still the most eminent 2D material due to its extraordinary electrical mobility at room temperature (larger than 1×10^5 cm²V⁻¹s⁻¹), Young's modulus of 1 TPa and intrinsic strength of 130 GPa [190, 191]. However, the lack of nonzero bandgap seriously limits its application in logical circuits where a high on/off current ratio is required [192]. Although several techniques, such as chemical doping or strain modulated bilayer graphene, have been proposed to manually introduce a bandgap into graphene, these modifications inevitably yield either low electrical conductivity or complex design [193, 194]. Inspired by graphene, there has been renewed interest in materials with layered structure, including transition metal dichalcogenides (TMDs), transition metal oxides (TMOs), boron nitride and silicene, *etc* [195-198].

Using top-down or bottom-up approaches to fabricate 2D TMDs similar to those of graphene, 2D TMDs can be synthesized via a wide variety of methods, including mechanical and chemical exfoliation, chemical solution deposition, chemical and physical vapor deposition. TMD materials have demonstrated outstanding performances in the field of hydrogen evolution reaction, energy storage, diagnostic imaging, electronic and optoelectronic devices, leading to extensive studies involving both theoretical and experimental research [199-202]. However, the magnetic properties of layered TMDs remain less explored compared to the electronic and optical properties. Theoretical

studies based on first-principles calculations reveal a nonmagnetic ground state of perfect crystalline MoS₂ [120, 203]. Ferromagnetism can be introduced into MoS₂ by strain, the formation of vacancies, zigzag edges in nanoribbons, doping with transition metals, and edge passivation [204-207]. While some simulations are straightforward, others are very complex. For example, the magnetic moment of zigzag MoS₂ nanoribbons decreases when only one edge is passivated with C, but increases with passivation of both edges [208].

The vast majority of studies related to TMDs so far focus on MoS₂. Compared to MoS₂, WS₂ has advantages over thermal stability and electrical properties. Like MoS₂, WS₂ nanosheets also display room-temperature ferromagnetism [209]. First-principles simulations indicate the magnetic state of WS₂ can be modulated by defects and transition metal doping [210-213]. Recently, robust valley polarization and valley polarization coherence, due to giant spin-orbit coupling and spin-valley coupling, have been observed in bilayer WS₂, making it a promising candidate for spintronics and valleytronics devices [214].

Rare earth element has played important roles in the properties of magnetic materials for its strong spin-orbit coupling [215, 216]. Nd is a well-known rare earth element that can effectively tailor the electrical and magnetic properties of semiconductors like ZnO and TiO₂ [217-219]. Evidence indicates that Nd-doped ZnO nanoparticles show enhanced photoluminescence and ferromagnetic properties compared to bare ZnO [220]. Moreover, high coercivity and magnetization have been observed in Nd doped ZnO nanowires [221]. So far, however, there is no theoretical or experimental work of Nd doped TMDC materials.

In this work, demonstrations were made to manipulate the magnetic properties of TMDCs

by defect engineering. Firstly, vacancies were introduced into WS₂ powders by heat treatment. WS₂ powders (99.9%) were purchased from Zaibang Chemical Industry CO., LTD, China. Annealing was carried out in a tube furnace and alumina crucible was used to transfer WS_2 powders into the center of the furnace. The furnace was purged by forming gas (5% Hydrogen, 95% Argon) for half an hour prior to annealing, then it was heated up to annealing temperature (400 °C, 700 °C and 1000 °C) by a rate of 5 °C/min with continuous gas flow of 50 sccm. The annealing time was 1 h and the annealed samples were labeled as 400-WS₂, 700-WS₂, and 1000-WS₂. In the second part, Nd with different concentrations was doped into MoS₂ single crystals. The ion implantation took place at GNS Science, New Zealand. A low-energy ion implanter was used with a voltage of 30 keV at a fluence of 5×10^{14} , 1×10^{15} , 4.8×10^{15} Nd-ions/cm². The projected range was set to ~12.6 nm and the average concentrations of Nd over 50 nm are 0.5, 1, 5 at%, respectively. It was found that annealing and ion implantation both lead to enhanced ferromagnetism by the formation of defects. Defect density, which is strongly dependent on annealing temperature and Nd dose, plays a crucial role in the magnetic behaviors of WS₂ and MoS₂.

6.2 Magnetic properties of annealed WS₂ powders

The XRD patterns of pristine and annealed WS₂ samples are presented in figure 6.1. The shape of the patterns indicates a mixture of hexagonal 2H phase (P6₃/mmc) and the rhombohedral 3R phase (R3mH). Analysis of phase content was carried out via Rietveld refinement approach built in HighScore Plus software. The proportions of 2H phase in four WS₂ samples vary from 45.8% to 47.2%, almost identical. This suggests annealing has little impact on the composition of WS₂ powder. The SEM images of the pristine and annealed WS₂ powders are shown in figure 6.2. It reveals that annealing process has little

impact on the powder morphology. All samples consist of micro-sized particles accumulated by various forms of nanostructured WS_2 . Thin WS_2 nanosheets and irregular nanoparticles can be observed in the samples, as well as nanorods with a width of around 500 nm and length of several micrometers.

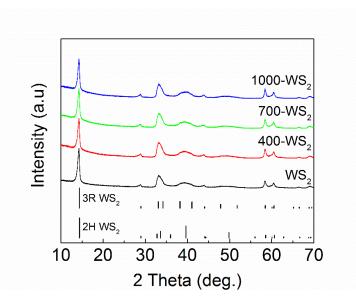


Figure 6.1 XRD spectra WS₂ powders annealed at different temperatures.

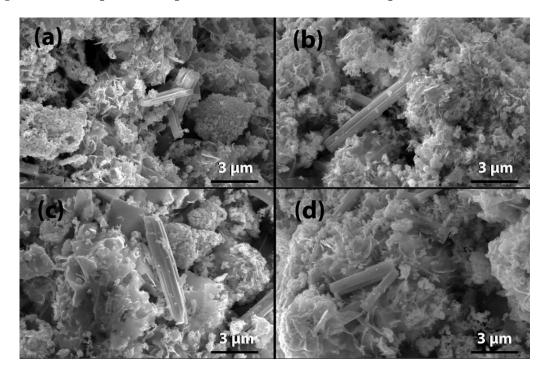


Figure 6.2 SEM images of WS_2 powders, (a) pristine WS_2 , (b) 400- WS_2 , (c) 700- WS_2 , (d) 1000- WS_2 .

Raman spectroscopy has emerged as a powerful non-destructive method to explore various properties of 2D TMDs such as MoS₂ and WS₂. For example, the layer numbers, as well as doping conditions of 2D flakes, can be derived from peak shifts of the firstorder E_{2g}^{1} and A_{1g} optical modes [222]. As presented in figure 6.3, WS₂ powders have two dominated Raman peaks. The peak at ~420 cm⁻¹ corresponds to out-of-plane vibrational A_{1g} mode, whereas the peak at ~350 cm⁻¹ is the result of the overlapping of in-plane vibrational E¹_{2g} mode and longitudinal acoustic 2LA mode. The thickness WS₂ can be identified by analysing the frequency differences between the $A_{1g} \mbox{ and } E^1{}_{2g} \mbox{ modes}.$ In our case, the frequency difference of 70 cm⁻¹ matches the value of bulk WS_2 [223]. Previous studies also suggest that A1g mode of MoS2 shows blue/red shift by p/n-type doping while the E¹_{2g} mode remains at the same position [224, 225]. Annealing can induce sulfur vacancies due to sulfur decomposition at high temperatures, which introduce unpaired electrons and n-type doping [226]. Tiny red shifts can be seen from A_{1g} modes in 400-WS₂ (419.98 cm⁻¹) and 700-WS₂ (419.91 cm⁻¹) compared to pristine WS_2 (420.23 cm⁻¹) and 1000- WS_2 (420.33 cm⁻¹) as shown in figure 6.3(b), but the shifts are too small to confirm the n-doping effect.

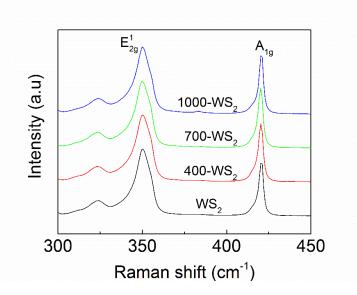


Figure 6.3 Raman spectra of WS₂ powders.

XPS is sensitive to chemical composition and stoichiometry of WS₂ powders. The binding energy profiles for W 4f are illustrated in figure 6.4(a). Three peaks observed at 33.6, 35.8, and 39.4 eV of untreated WS₂ powders are attributed to the W $4f_{7/2}$, W $4f_{5/2}$, and W 5p_{3/2}, respectively. The positions of these peaks demonstrate a 4+ valence state of W from pure WS_2 phase, which is consistent with other reports for WS_2 crystals [227]. Compared to pristine WS_2 powder, all annealed powders exhibit certain peak shift. 400- WS_2 and 700- WS_2 display a shift to the lower binding energy side. It is an evidence of the redistribution of localized charges due to the formation of sulfur vacancies, which agrees with studies of sulfur vacancies in MoS_2 [228]. Recent studies have shown that hydrogen annealing below certain temperature can lead to both micrometer and nanometer-scale holes, while large cracks and even sample decomposition are expected at high temperature [229]. These may explain the back-shift of W4f peaks of 1000-WS₂. From figure 6.4(b), a similar trend can be observed for S $2p_{1/2}$ and S $2p_{3/2}$ doublet with binding energy of ~163.3 and ~164.4 eV. Furthermore, an increase of vacancy density in annealing samples is also demonstrated by ESR spectra. As shown in figure 6.5, the paramagnetic signals around g≈1.977 in ESR first increase and reach the maximum by

700-WS₂, then drop to bottom. The ESR result provides further proof of the existence of sulfur vacancies as anion vacancies can provide unpaired electrons [230]. The low ESR intensity of 1000-WS₂ indicates annealing at 1000 °C has negative effects on sulfur vacancy generation, possibly caused by structure disruption, which agrees with XPS results.

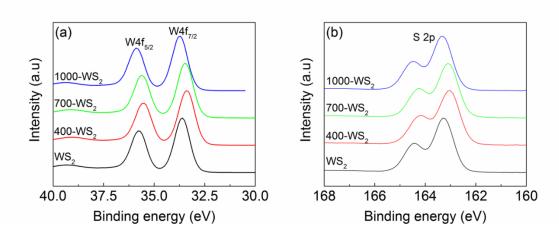


Figure 6.4 XPS core-level spectra of (a) W4f and (b) S2p.

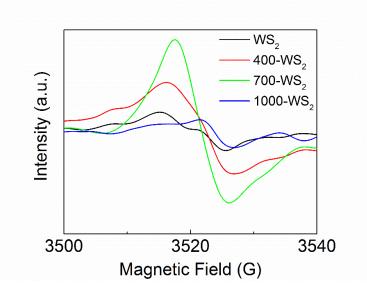


Figure 6.5 Room-temperature EPR spectra of WS₂ samples.

The magnetization versus applied field (M-H) loops of different WS₂ powders are shown in figure 6.6. Unlike bulk WS₂ crystals with merely diamagnetic nature, all WS₂ powders present clear additional ferromagnetic features at 5 K. The ferromagnetism can be attributed to various dangling bonds or vacancies. Further first-principles calculations indicate that the ferromagnetic ordering can be realized in the 2D network of dangling bonds, and collapses if the dangling bonds are saturated by H ions [231]. After annealing, the magnetization of WS₂ powders significantly increases, with the highest magnetization achieved by 700-WS₂ taken at 5 K as shown in figure 6.6(e). Although diamagnetic signals dominate the magnetic response when the temperature reaches above 50 K, ferromagnetic hysteresis loop can still be observed above the diamagnetic background up to room temperature. After deducing the diamagnetic part, the ferromagnetic loops of four powders measured at room temperature are illustrated in figure 6.6(f). 700-WS₂ still shows the highest saturation magnetization (Ms) with ~ 0.002 emu/g. 1000-WS₂ seems to possess the second-best room-temperature Ms, but its zero coercivity indicates non-ferromagnetic phase.

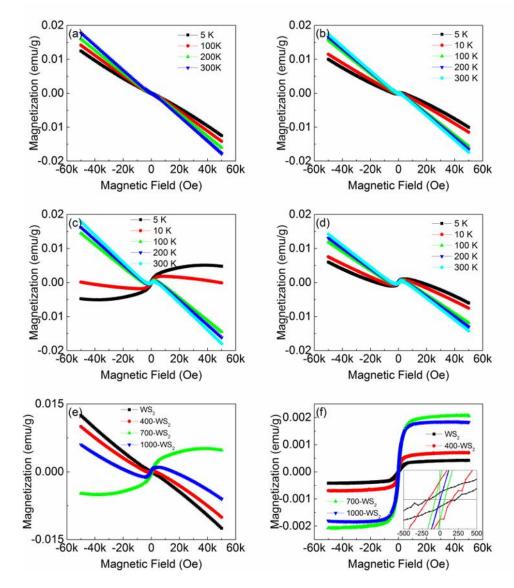


Figure 6.6 M-H measurement of (a) original WS₂, (b) 400-WS₂, (c) 700-WS₂ and (d) 1000-WS₂ measured at different temperatures. (e) M-H loops of different WS₂ powders measured at 5 K. (f) Hysteresis loops obtained by deducting the diamagnetic signal, measured at 300 K. The inset is the loops in a small field range.

According to first-principles calculations, the ferromagnetism arising from nanostructured TMDs is mainly contributed by defects, either point defects like vacancies or one-dimensional defects such as edges and grain boundaries. In several studies of defect-induced ferromagnetism in MoS_2 [106, 232], authors claimed that the induced

ferromagnetic ordering is due to the exchange interactions between sulfur vacancy and the Mo⁴⁺ 4d bandgap state. For WS₂, while simple vacancies like single tungsten and sulfur vacancies (V_w and V_s) are non-magnetic, a number of complex vacancies, such as three-tungsten vacancy (V_{3w}), six-sulfur vacancy (V_{6S}) and 1W+2S vacancy (V_{1W+2S}), show ferromagnetic behaviours caused by the S p states and W d states near vacancy defects [233].

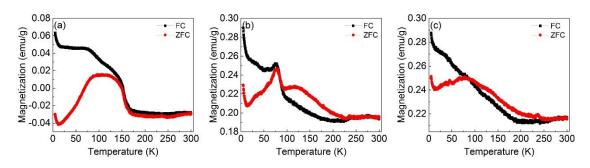


Figure 6.7 ZFC and FC curves of (a) 400-WS₂, (b) 700-WS₂ and (c) 1000-WS₂. The magnetic field applied is 500 Oe.

From XPS and ESR, annealing below 1000 °C can increase the density of vacancies, leading to higher magnetization. Further increase in annealing temperature can disrupt WS₂ structure and destroy ferromagnetic ordering. Figure 6.7 displays zero field cooling (ZFC) and field cooling (FC) curves of annealed WS₂ powders. The relatively high magnetization below 15 K suggests the presence of paramagnetic phase. It is obvious there is a phase transition around 75 K for all samples, although it is unreported in previous studies. WS₂ annealed at various temperatures show totally different curve features near the transition point. ZFC/FC curves of 400-WS₂ show a broad transition range between 75 and 150 K. ZFC/FC curves of 700-WS₂ and 1000-WS₂ intercross near the critical temperature and FC curve continues to decrease below ZFC curve. This negative irreversible magnetization is often observed in random-field systems [234]. The defects and impurities are the sources of random field. In the model, even weak random fields can destroy long-range order. For WS₂ annealed at high temperatures, WS₂ tends

to decompose into small nanosheets and create massive defects like edges and boundaries, which has drastic effects on phase transitions. The random field generated from the defects compete with the long range ferromagnetic order. Imry and Ma first suggested the possibility that the ordered state could be broken up into certain sized domains [235]. The absence of long-range ferromagnetic ordering has later been explained by domain wall pinning in metastable states [236]. This might be the cause for the disappearance of ferromagnetism in 1000-WS₂, but the exact random field induced disordering mechanism in a disordered ferromagnet is still under debate. Moreover, hydrogen passivation may have some impacts on the magnetization as hydrogen adsorption can be observed on MoS₂ and WS₂ [237, 238]. However, theoretical studies have predicted extremely complex magnetic properties of WS₂ nanoribbons with edge hydrogenation [239]. Therefore, we have not taken hydrogen passivation into consideration.

6.3 Magnetic properties of Nd-doped MoS₂ single crystals

Figure 6.8(a) displays the XRD patterns of pure and Nd-doped MoS₂ with various Nd concentrations. All diffraction peaks of the pure MoS₂ crystal have indexes parallel to the [001] orientation of hexagonal MoS₂ (P63/mmc), showing a perfect single crystal structure. No peaks of any other impurities are detected for 0.5 at% and 1 at% Nd-doped MoS₂, indicating that no secondary Nd phases are generated during ion implantation. An extremely weak peak emerges at 22.16° for 5 at% Nd-doped MoS₂ with intensity of 32 compared to 1.56×10^6 of the main peak, which may belong to the (002) diffraction peak of NdS₂.

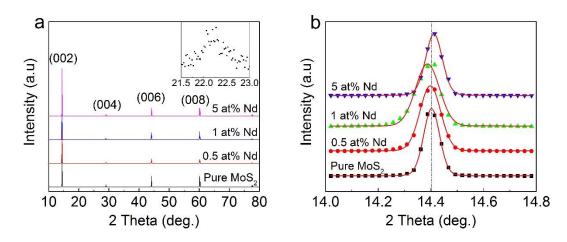


Figure 6.8 (a) XRD patterns of undoped and Nd-doped MoS₂ crystals. The inset shows NdS₂ peak in 5 at% Nd-doped MoS₂. (b) Enlarged (002) diffraction peaks.

(002) diffraction peaks of 0.5 at% and 1 at% Nd-doped MoS₂ shift towards lower diffraction angle as compared to undoped MoS₂ crystal (As shown in figure 6.8(b). All peaks are fitted using a pseudo-Voigt function), demonstrating a lattice constant expansion. The enlargement of the lattice parameters is caused by the ionic radius difference between dopant and matrix atoms. Because Nd³⁺ (0.98 Å) has a larger radius than Mo⁴⁺ (0.65 Å), the expansion suggests that Nd³⁺ ions replace the Mo⁴⁺ lattice or stay in interstitial sites. Although the (002) diffraction peak of 5 at% Nd-doped MoS₂ has the highest diffraction angle, its lattice parameter *a* (3.172 Å) obtained through refinement is still slightly larger than 0.5 at% Nd-doped MoS₂ (3.186 Å). The abnormal lattice parameters of 5 at% Nd-doped MoS₂ in contrast to its highest dopant concentration can be explained by the secondary Nd phase and abundant surface vacancies which may distort unit cells. Another intriguing finding is that the expansion of unit cell is anisotropic. The relative variation along c-axis is less than one tenth of that along a-axis, suggesting the cell is mainly stretched in the preferred in-plane direction during the doping.

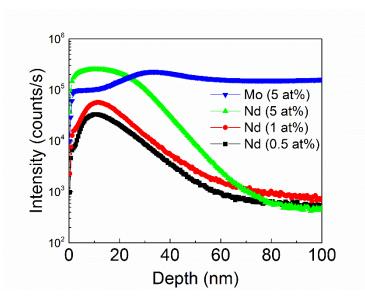


Figure 6.9 SIMS depth profiles for Nd element in three doped crystals and Mo in 5 at% Nd-doped MoS₂.

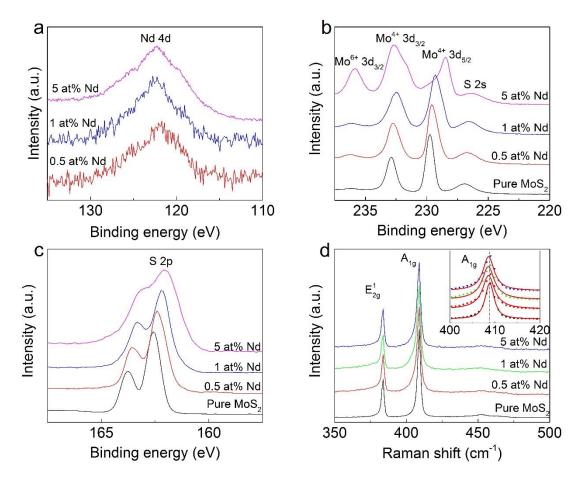


Figure 6.10 XPS core-level spectra of (a) Nd 4d, (b) Mo 3d and S 2s and (c) S 2p for pure and Nd-doped MoS_2 crystals. (d) Raman spectra of MoS_2 . The inset is enlarged A_{1g} modes with peak fittings.

The distribution of Nd dopants in MoS_2 is examined by SIMS, as presented in figure 6.9.

Rather than uniformly distributed within the top 50 nm, the depth profiles can be approximated as Gaussian function with the centre located around 12.6 nm, which is well agreed with the projected range. However, this also means most Nd dopants only accumulate in a restrict space with near double concentration than the original design. Take 5 at% Nd-doped MoS₂ for example, the concentration can reach 10 at% within the top 20 nm. Such high concentration may be the cause for the formation of secondary phase.

During the ion implantation, some defects are inevitably introduced into MoS_2 crystals because the energy of Nd implantation (30 keV) is far above the formation energies of various kinds of defects, especially the sulfur vacancy with a formation energy as low as ~6.0 eV [122]. Therefore, we investigated the MoS_2 crystals before and after ion implantation via XPS to identify possible defects. The first concern is to determine the valence of Nd element. However, high-resolution scan finds no trace of Nd on the surface of 0.5 at% Nd-doped MoS₂. Distinct Nd 4d core-level signal with concentration of 1.1 at% can only be observed after 30s ion-beam etching to reach depth below 10 nm, confirming the Gaussian distribution of Nd from SIMS (figure 6.10(a)). From XPS, 1 at% Nd-doped MoS_2 also displays higher Nd concentration (1.4 at%) after etching than the original surface (0.64 at%). 5 at% Nd-doped MoS_2 is an exception. Its original surface has an unexpectedly higher dopant concentration (19.1 at%) than post-etching (18.3 at%) and both values considerably exceed the estimation from SIMS. This suggests doping with heavy dose can even cause Mo ions to escape from surface, supported by the SIMS depth profile of Mo in 5 at% Nd-doped MoS₂, in which the intensity of top 10 nm is only ~60% compared with that of 100 nm. The peak position of Nd around 122.1 eV in 0.5 at% Nddoped MoS₂ indicates Nd mainly exists as Nd³⁺ ions [240, 241]. However, binding energy decreases with increased doping concentration, to ~122.9 eV for 5 at% Nd-doped MoS₂,

which suggests the possible formation of Nd^{4+} ions and agrees with the appearance of NdS_2 in XRD pattern.

Figure 6.10(b) and (c) show the high-resolution Mo 3d and S 2p core-level XPS spectra. The characteristic peaks of $Mo^{4+} 3d_{3/2}$ and $Mo^{4+} 3d_{5/2}$ doublet of pure MoS_2 locate at 232.9 eV and 229.7 eV, which shift to the lower binding energy side after Nd doping. The peak shifts of $Mo^{4+} 3d_{3/2}$ and $Mo^{4+} 3d_{5/2}$ orbitals for 0.5 at% Nd-doped MoS₂ are around 0.2 eV and 0.3 eV, respectively, and further shifts can be observed for MoS₂ crystals with higher dopant concentration. In previous studies, it is proposed that the $Mo^{4+} 3d_{5/2}$ and $Mo^{4+} 3d_{3/2}$ peaks can be split into two parts: one part of intrinsic MoS₂ and another part of MoS₂ containing sulfur vacancies [242]. The spectra of vacancy-containing MoS₂ appear at lower binding energies compared to the intrinsic MoS₂ part due to the redistribution of charges localized on the more electronegative sulfur atoms to the first neighbouring atoms (Coulomb screening effect) [228]. Therefore, the peak shifts to the lower binding energy side can be considered as a sign of sulfur vacancies. The XPS spectra of S²⁻ $2p_{3/2}$, $2p_{1/2}$, and 2s have similar feature to Mo 3d, with the peaks to the lower binding energy region as the dopant concentration increases. The sulfur deficiency is further confirmed by the calculation of atomic ratio of S/Mo obtained from XPS spectra. The S/Mo ratio decreases from 2.24 in pure MoS₂ single crystals to 1.82 and 1.80 in 0.5 and 1 at% Nd-doped MoS₂, which further drops to mere 1.09 in 5 at% Nd-doped MoS₂.

In addition to peak shift, a prominent peak at ~235.7 eV shown up in figure 6.10(b) is attributed to the Mo^{6+} state. Although Mo^{6+} also contains double peaks like Mo^{4+} , the $3d_{5/2}$ peak at lower binding energy is entirely covered by the $Mo^{4+} 3d_{3/2}$ peak when Mo^{6+} ions only occupy a small proportion, like in 0.5 and 1 at% Nd-doped MoS_2 . This Mo^{6+} peak has been observed in previous reports, which is explained by absorption of oxygen [242, 243]. The loss of Mo atoms on the surface may be another cause for the strong Mo^{6+}

peak observed in 5 at% Nd-doped MoS₂.

Raman spectroscopy has emerged as a powerful non-destructive means to examine the structure of MoS₂. As shown in figure 6.10(d), the peak at ~383.47 cm⁻¹ corresponds to E^{1}_{2g} mode due to the displacement of Mo and sulfur atoms within the basal plane, whereas the A_{1g} mode at ~408.82 cm⁻¹ originates from the vibration perpendicular to the basal plane. Several studies suggest that A_{1g} mode of MoS₂ is considerably sensitive to doping while the E^{1}_{2g} mode is almost inert. A p/n-type doping can cause A_{1g} peak to blue/red shift because of strong electron-phonon coupling [224, 225]. From Raman spectra, a small red shift of ~0.3 cm⁻¹ can be observed for A_{1g} mode of Nd-doped MoS₂ crystals, indicating n-type doping. Theoretically, Nd³⁺ substitutional doping should be p-type while sulfur vacancies are n-type. Therefore, sulfur vacancies have a greater impact on the electronic structures of MoS₂ than Nd³⁺ ions. Moreover, full width at half maximum (FWHM) of A_{1g} peak increases from 2.4 for pure MoS₂ crystal to 3.0-3.3 for Nd-doped MoS₂, suggesting decreasing crystallinity caused by surface defects.

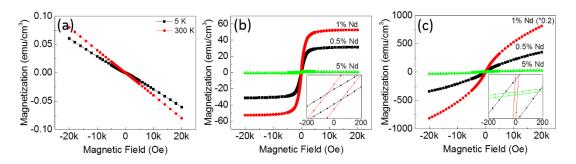


Figure 6.11 (a) M-H loops of pure MoS₂ measured at 5 K and 300 K. Hysteresis loops of Nd-doped MoS₂ after deducting the diamagnetic signals, measured at (b) 300K and (c) 5 K. The inset is the loop in a small field range.

To investigate the magnetic properties of the Nd-doped MoS_2 crystals, SQUID measurements were performed at various temperatures. Figure 6.11 shows the magnetization versus magnetic field (*M*–*H*) curves of Nd-doped MoS_2 crystals at 300 and 5 K. Pure MoS_2 crystal exhibits sole diamagnetic feature at both temperatures. For the Nd

doped MoS₂, the magnetization should only come from doping area since pure MoS₂ crystal does not make any contribution to the ferromagnetism. After deducting the diamagnetic signal from pure MoS₂, the M-H loops of Nd-doped MoS₂ samples are shown in figure 6.11(b). Room-temperature ferromagnetism in Nd-doped MoS₂ except 5 at% is confirmed by evident coercivities shown in the inset. The Ms of 0.5 and 1 at% Nd-doped MoS₂ at room temperature are ~33.6 and 28.3 μ_B /Nd, given that all the magnetic moments are contributed from Nd dopants. At 5 K, both 0.5 and 1 at% Nd-doped MoS₂ show strong ferromagnetic plus paramagnetic signals, supported by the non-saturated M-H loop at relatively high magnetic field in figure 6.11(c). For 5 at% Nd-doped MoS₂, a small hysteresis loop can be observed at 5 K, suggesting that a tiny amount of ferromagnetic ordering remains.

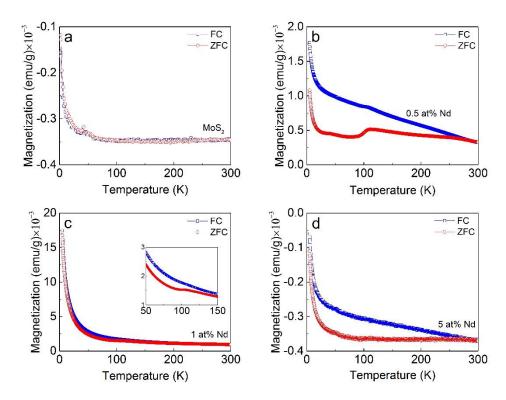


Figure 6.12 ZFC and FC curves of MoS₂ with different dopant concentrations. (a) pure; (b) 0.5 at%; (c) 1 at%; and (d) 5 at% Nd-doped MoS₂. The inset in (c) presents the same curves in (c) in the range of 50 K to 150 K. The applied field for ZFC and FC measurement is 500 Oe.

Although neither theoretical nor experimental studies about Nd-doped TMDs systems

have been reported so far, several works on Nd-doped transition metal oxides has been published [218, 220, 244]. Strong ferromagnetic ordering is reported in the above literatures, ascribed to the strong spin-orbit coupling of rare earth elements. Figure 6.12 illustrates the zero field cooling (ZFC) and field cooling (FC) curves of all samples, which further confirms pure MoS₂ crystals do not make any contribution to the ferromagnetism. As shown in figure 6.12(a), there is no divergence between ZFC and FC curves from 5 to 300 K. It is to note the magnetization of the ZFC and FC curves is negative, due to the diamagnetic nature of the sample. In addition, the hump at around 50 K is associated with the signal from oxygen. For 0.5 at% Nd-doped MoS₂, there is an obvious separation between ZFC and FC curves (figure 6.12(b)) and the divergence reaches temperature close to 300 K, suggesting that weak ferromagnetic ordering with relatively high Curie temperature exists in the sample, which is consistent with M-H loop measurement. A transition temperature at around 110 K can be clearly seen in both ZFC and FC curves, which is associated with defective MoS₂ having a phase transformation at this temperature, as previously reported [207]. For 1 at% Nd-doped MoS₂, ZFC and FC appears to overlap due to the high magnetization at low temperature caused by paramagnetism. However, the enlarged part shown in the inset clearly shows similar feature to 0.5 at% Nd doped sample. For 5 at% Nd-doped MoS₂, only weak ferromagnetism is observed and the strong diamagnetic signal from undoped MoS₂ part dominates.

The observed ferromagnetic behaviour of Nd-doped MoS_2 crystals may be originated from the exchange coupling of sulfur vacancies with Nd^{3+} ions, which is often explained by the bound magnetic polarons (BMP) [80]. It is suggested the sulfur vacancies in Nddoped MoS_2 can create localized electrons because of the redistribution of the localized charges. The localized electrons can polarize surrounding magnetic Nd^{3+} ions, leading to

the formation of magnetic polarons [244]. When the density of magnetic polarons increases, the areas of magnetic polarons eventually overlap and induce the spin-spin interaction between Nd³⁺ magnetic ions, giving rise to the long-range ferromagnetism in Nd-doped MoS_2 crystals, as illustrated in figure 6.13. Moreover, Nd^{3+} ions may not be the only contributor to the colossal magnetic moment of $\sim 30 \mu_B/Nd$. Sulfur vacancies and other defects within the magnetic polarons may also contribute to the high magnetic moment. In this case, 5% Nd doping should exhibit the highest saturation magnetization since it possesses the highest concentration of defects due to ion implantation. On the contrary, it shows the weakest ferromagnetism among the three samples. Hence, the high magnetic moment is not originated from sulfur or Mo vacancies but from more complex Nd^{3+} related defects. Nd^{3+} should play the dominant roles for the giant magnetic moment. The maxima of ZFC/FC curves around ~110 K is probably also caused by the magnetic polarons since the phase transformation may affect the distribution of dopants inside MoS₂. Some isolated magnetic polarons in particular regions may only be able to remain ferromagnetic ordering at the temperature below 110 K. A similar curve feature can be observed in graphene, explained by the coexistence of ferromagnetic and antiferromagnetic states [245].

When the concentration of Nd dopants exceeds certain threshold, antiferromagnetic ordering replaces ferromagnetic ordering as neighbour magnetic Nd³⁺ ions are close enough to create antiferromagnetic pairs for reducing system energy. In the samples of 0.5 and 1 at% Nd-doped MoS₂, ferromagnetic ordering dominates. However, when the doping concentration reaches 5 at%, Nd dopants segregate and form NdS₂ secondary phase due to the non-uniform distribution. The BMP model is no longer valid at this situation. Hence, diamagnetic signals dominate. However, unlike the overlapped ZFC/FC curves in pure MoS₂, the divergence in 5 at% Nd-doped MoS₂ indicates weak

ferromagnetism remains in the sample. The source of the weak ferromagnetism may be from the region with relatively lower doping concentration.

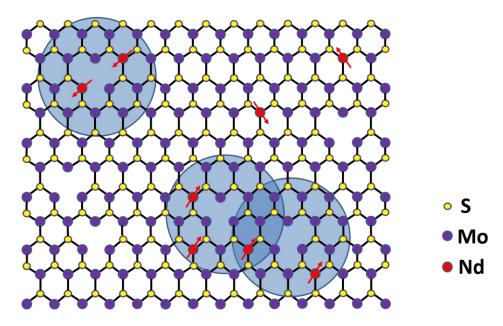


Figure 6.13 A schematic diagram of bound magnetic polarons model for Nd-doped MoS₂ system. The radius represents a hydrogenic orbital confined by an electron associated with sulfur vacancies.

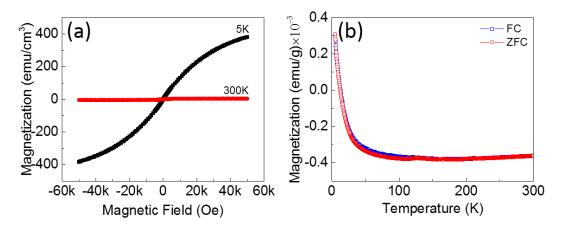


Figure 6.14 (a) M-H loops (after deducting the diamagnetic signals) and (b) ZFC and FC curves of post annealed 0.5 at% Nd-doped MoS₂. The magnetic field applied is 500 Oe.

A post annealing procedure frequently, if not always, follows ion implantation which aims to remove lattice stress and repair defects such as interstitials. To see whether it has any effect on the magnetic properties of Nd-doped MoS₂, we performed a quick annealing (10 min) for 0.5 at% Nd-doped MoS₂ at 400 °C under forming gas (95% Ar,5% H₂). The M-H and ZFC/FC curves of post-annealed sample are presented in figure 6.14, indicating that the room-temperature ferromagnetism disappears after annealing. Due to the high energy of ion implantation, vacancy generation is thought to be non-equilibrium, random and isolated. However, thermal treatments like annealing can prompt the formation of vacancy clusters with different configurations, such as vacancy lines, according to both theoretical and experimental studies [246, 247]. The diffusion of Nd dopants and sulfur vacancies destroys the formation of magnetic polarons, leading to the vanishing of ferromagnetism.

6.4 Conclusions

In summary, annealing of WS₂ powders was performed to explore how defects affect the magnetic properties. It is found that annealing can effectively change the density of sulfur vacancies, which increases with increased annealing temperature but diminishes at 1000 °C. The annealed powders show ferromagnetic ordering with Tc above room temperature, in addition to the existence of paramagnetic phases at low temperature. The highest saturation magnetization of ~0.002 emu/g is achieved by the WS₂ powder annealed at 700 °C with the highest volume of vacancies, suggesting vacancies play a significant role in enhanced magnetization. The magnetism of 2D TMDs can be modulated via defects engineering, which offers valuable guides for the magnetic performance in potential spintronic devices.

Doping of Nd into MoS_2 crystals was successfully carried out by ion implantation. The microstructure and magnetic properties of Nd doped MoS_2 crystals were studied systematically. Results reveal that Nd dopants are at a trivalent chemical state as Nd³⁺ ions without the formation of secondary phase at low doping concentrations. Meanwhile, sulfur vacancies are also introduced into the samples by Nd doping. Giant magnetic moment has been observed in the samples with 0.5 at% and 1 at% Nd. The

ferromagnetism in Nd doped MoS_2 crystals can be explained by the BMP model, in which magnetic polarons are created by the exchange coupling of sulfur vacancies and magnetic Nd^{3+} ions.

Chapter 7 Conclusions and Future Work

In this dissertation, the magnetic properties of a variety of diluted magnetic semiconductors, including TiO₂ based, MoS₂ based and WS₂ based materials, were studied. Firstly, we prepared 5 at% Co doped TiO₂ thin film on LaAlO₃ (001) substrate by pulsed laser deposition. XRD pattern shows a pure anatase phase with (001) plane parallel to the substrate surface. SAED analysis confirms the orientation relationships between the anatase thin film and LAO substrate is $(001)_{anatase}||(001)_{LAO}$ in out-of-plane direction and further indicates in-plane orientation relationships are $[100]_{anatase}||[100]_{LAO}$ and $[010]_{anatase}||[010]_{LAO}$. Co nanoclusters with different sizes are observed near the interface from high-resolution TEM image. Larger clusters with diameters around 5 nm are located close the interface, while 1-2 nm small nanoclusters are embedded inside the film. Except a small amount of oxidized Co in the surface layer, most Co dopants exist as metallic Co⁰.

An ultra-high magnetic moment of $3.5 \ \mu_B$ /Co at room temperature was obtained from Co doped anatase TiO₂ thin film through XMCD measurements, given that all the magnetization come only from Co. In addition, XMCD measurement suggests the contribution from oxygen cannot be ignored. PNR measurement further indicates that the nanoreigns near the interface with higher Co concentration generate higher magnetization and the highest saturation magnetization at the peak area is around 106 emu/cm³. To clarify the mechanism of observed magnetic moment, first-principles calculations were performed considering spin-orbit coupling. Results show that that the easy spin axis for Co nanoclusters lays along [001] direction. A Co cluster containing 15 atoms can obtain a high magnetic moment of 2.9 μ_B per Co atom, demonstrating that magnetic Co nanoclusters embedded in TiO₂ matrix are able to show high magnetic moment above room temperature. To further investigate the relationship between microstructure and magnetic properties of Co doped TiO₂ system, a series of thin films were prepared on SrTiO₃ (001) substrate under different temperatures and growth rates. It was found deposition parameters have great impact on both size and distribution of Co nanoclusters, as well as the crystallinity of the anatase thin film. Only polycrystalline film can be obtained at 500 °C with high deposition rate, in which Co segregates into small clusters at the interface. Higher temperatures and slow growth rate lead to fully relaxed epitaxial thin films. In the meantime, Co clusters become larger in size and move to the surface. SQUID measurements show a highest magnetic moment of 2.5 μ_B /Co at room temperature. Moreover, large magnetization exceeding 2 μ_B /Co can be achieved by Co nanoclusters with size bigger than 20 nm.

The second half of this dissertation was to study the magnetic properties of two transition metal dichalcogenides, WS₂ and MoS₂. Firstly, WS₂ powders were annealed to increase the concentration of defects. XPS and ESR results indicate that density of sulfur vacancies significantly increases with increased annealing temperature below 1000 °C. All annealed powders show room-temperature ferromagnetism. The highest saturation magnetization of ~0.002 emu/g is achieved by the WS₂ powder annealed at 700 °C with the highest volume of vacancies, suggesting vacancies play a significant role in enhanced magnetization. In the following work, Nd was doped into MoS₂ crystals by ion implantation. Results reveal that Nd dopants are existed as Nd³⁺ ions without the formation of secondary phase at low doping concentrations. Meanwhile, Nd doping can introduce sulfur vacancies or even Mo vacancies at the surface. Giant magnetic moment around 30 $\mu_{\rm B}$ /Nd has been observed in the samples with 0.5 at% and 1 at% Nd doped samples. The ferromagnetism in Nd doped MoS₂ crystals can be explained by the BMP model, in which magnetic polarons are created by the exchange coupling of sulphur

vacancies and magnetic Nd³⁺ ions.

This dissertation has demonstrated the great potential of TMDs in advanced magnetic semiconductors with the accomplishment of simultaneous high Curie temperature and high magnetic moment. Moreover, the dopants in TiO₂ and MoS₂ are structurally, chemically and magnetically protected by the matrix, and hence offer more stable magnetic properties. However, within the framework of this thesis, it is impossible to cover all the magnetic properties of TiO₂-based and TMDCs based DMSs. In order to fully clarify the exact mechanism of ferromagnetism in the DMSs, further research work needs to be done in the future. In Co-TiO₂ grown on LAO, a large XMCD response has been obtained from O K edge in addition to Co L edge, suggesting O element is another source for the high magnetization. However, how O contributes to the magnetic moment remains unclear in this work. A number of previous reports suggest O vacancy, one of the most crucial and supposed to be the prevalent defect in many metal oxides, is able to induce ferromagnetism in TiO₂ even without extra magnetic dopants. The existence of O vacancy in the Co-doped TiO₂ thin film and the relation of concentration with magnetic properties need to be further investigated. Since O vacancy alters the electronic structure of TiO₂, it can be detected by methods like electron spin resonance and UV/visible light absorption. In addition to high magnetization, magnetic properties that can be tuned through extra fields are of great interest. Based on our previous study, relatively high deposition rate and low partial oxygen pressure lead to a more spatially uniform distribution of Co dopant. This work has identified that temperature is also a critical factor for the distribution of magnetic dopants. A magnetic phase diagram for Co-doped TiO_2 or other oxide based diluted magnetic semiconductors as functions of partial oxygen pressure and temperature can be drawn in the future by the variation of experiment parameters. Additionally, the choice of matrix material to embed Co should not be limited

to only TiO₂. Whether Co can maintain the high magnetization when switching TiO₂ to other semiconducting oxides is worth studying. Furthermore, dopants are not limited to Co either. Fe, Ni and other transition metals may be employed as the dopants. For Co-TiO₂ grown on STO, further complementary techniques in addition to STEM needs to be carried out to evaluate the exact size of Co nanoclusters. The impact of growth parameters on the concentration and magnetization of defects like O vacancies may also be an interesting topic. It is shown that higher temperature and slower growth rate lead to larger cluster size. The establishment of an exact mathematical relation between cluster size or saturation magnetization and growth parameters can be further investigated. The TiO₂ epitaxial thin film obtained on STO in this work is fully relaxed with no strain. As magnetic properties have been manipulated through strain engineering in numerous reports, preparation of fully or partially strained TiO_2 thin films and the variation of the magnetic properties can be the next step of the research. For WS₂ powders, annealing is not the only method to induce vacancies. The magnetic properties can be discussed with the samples treated by other approaches, such as plasma treatment. In addition, more magnetic properties like the Curie temperature need to be further investigated. Heat treatment can also be applied to other TMD materials such as MoSe₂ to study the impact of anion on magnetic properties. For Nd-doped MoS₂, the exact mechanism behind the astonishingly huge magnetic moment of Nd needs further investigation. First-principles spin-polarised calculations under density functional theory with different schemes for the electron exchange and correlation functional can be employed for the discussion. For the minimization of semiconductor devices, few-layer or monolayer MoS₂ will serve as matrix material in the future work. Chemical vapor deposition is currently one of the most promising methods to realize the growth of 2D TMDCs with large-scale, uniform thickness and usually regular shapes. 2D MoS₂ can be synthesized using Mo and sulfur

powders as precursors and then doped with a variety of transition and rare earth metals. The fabrication of electronic devices should then be followed through photolithography or e-beam lithography and several semiconducting properties such as the carrier type and density need further examination. Moreover, the possible magnetic modulation of the device through external fields needs to be tested, as well as a number of fundamental magnetic properties like the Curie temperature. The materials can also seek applications in certain fields such as nanomotor and sensors.

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