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Improved irreversibility behavior and critical current density in MgB₂-diamond nanocomposites

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MgB₂-diamond nanocomposite superconductors have been synthesized by addition of nanodiamond powder. Microstructural analysis shows that the nanocomposite superconductor consists of tightly packed MgB₂ nanograins (~50-100 nm) with highly dispersed and uniformly distributed diamond nanoparticles ($\sim 10-20$ nm) inside the grains. The J_c-H and $H_{iir}-T$ characteristics have been significantly improved in this MgB2-diamond nanocomposite, compared to MgB2 bulk materials prepared by other techniques. Also, the J_c value of 1×10^4 A/cm² at 20 K and 4 T and the H_{irr} value of 6.4 T at 20 K have been achieved. © 2003 American Institute of Physics. [DOI: 10.1063/1.1606884]

Since the discovery of superconductivity at 39 K in MgB₂, significant progress has been made in improving the performance of MgB₂ materials.²⁻⁴ MgB₂ offers the possibility of wide engineering applications in the temperature range 20-30 K, where conventional superconductors, such as Nb₃Sn and Nb-Ti alloy, cannot play any roles due to their low T_c . However, the realization of large-scale applications for MgB2-based superconductivity technology essentially relies on the improvement of the pinning behavior of MgB₂ in high fields. As it has poor grain connection and a lack of pinning centers, MgB2 often exhibits a rapid decrease in critical current density, J_c , in high magnetic fields. Fortunately, through the formation of nanoparticle structures in bulk MgB₂²⁻⁴ and thin films,⁵ the problem of the poor grain connection can be solved, and the flux pinning force can also be significantly enhanced due to an increase of pinning centers served by grain boundaries. In order to improve further the performance of MgB₂, it is necessary to introduce more pinning centers, especially those consisting of nanosized second-phase inclusions, which often provide strong pinning

Nanodiamond, prepared by the detonation technique, has been widely used as an additive to improve the performance of various materials. 6 Yet, nanodiamond has never been used to increase the flux pinning force in MgB2 superconductors until the present study. The high dispersibility of the nanodiamond powder makes it possible to form a high density of nanoinclusions in MgB₂ matrix. In this letter, we have prepared the MgB2-diamond nanocomposite, which consists of

tightly packed MgB₂ nanograins (~50-100 nm) with diamond nanoparticles ($\sim 10-20$ nm) wrapped within the grains. This unique microstructure provides the composite with a good grain connection for the MgB2 phase and a high density of flux-pinning centers served by the diamond nanoparticles. Compared to the MgB2 bulk materials prepared with other techniques, the irreversibility line has been significantly improved and the J_c in high magnetic fields has been largely increased in the MgB2-diamond nanocomposite.

The MgB₂-diamond nanocomposites with compositions of $MgB_{2-x}C_x$ (x=0%, 5%, 8%, and 10%) were prepared by solid-state reaction at ambient pressure. Mg powder (99% purity, 325 meshes), amorphous B powder (99% purity, submicron-size), and nanodiamond powder (10–20 nm) were mixed and ground in air for 1 h. An extra 2% of Mg powder was added in the starting materials to compensate the loss of Mg caused by high temperature evaporation. The mixed powders were pressed into pellets with dimensions of 20 ×10×3 mm³ under a pressure of 800 kg/cm², sandwiched into two MgO plates, sintered in flowing Ar at 800 °C for 2 h, and then quenched to room temperature in air. In order to compare the substitution effect of carbon in boron in MgB₂ with the additional effect of the nanodiamond in MgB₂, a sample with an added 1.5 wt% of nanodiamond in MgB₂ was prepared. The sintering temperature and the sintering time for this sample were reduced respectively to 730 °C and 30 min in order to reduce the chemical reaction between the MgB₂ and the diamond. This sample has been referred to as 1.5 wt % C.

The crystal structure was investigated by powder x-ray diffraction (XRD) using an X'pert MRD diffractometer with

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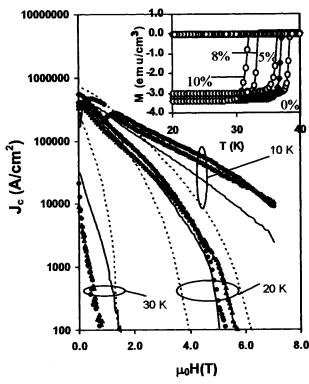


FIG. 1. Magnetic field dependence of J_c at 10, 20, and 30 K for ${\rm MgB_{2-x}C_x}$ with $x{=}0\%$ (dashed lines), 5% (solid lines), 8% (solid circles), and 10% (opened triangles). Inset: superconducting transition curves for the diamond-doped samples. The closed circles represent the results for the sample 1.5 wt % C.-

Cu $K\alpha$ radiation. The microstructure was analyzed with a Philips CM200 field emission gun transmission electron microscope (FEGTEM). Dc magnetization measurements were performed in a superconducting quantum interference device (Quantum Design MPMS-7). J_c values were deduced from hysteresis loops using the Bean model. The sample's dimensions with typical values of $0.7\times2.1\times2.7$ mm³ are used in the calculation of J_c . The values of the irreversibility field, $H_{\rm irr}$, were determined from the closure of hysteresis loops with a criterion of 10^2 A/cm².

XRD results (not shown here) reveal a decrease of the lattice parameter along the a-axis due to a partial substitution of boron by carbon in MgB $_2$.^{7,8} The substitution effect can also be reflected by the gradual decrease of T_c with increasing carbon content (see the inset of Fig. 1). The values of onset T_c for these carbon-substituted MgB $_2$ samples are 38.6 K for x=0%, 36.1 K for x=5%, 33.0 K for x=8%, and 31.3 K for x=10%. The T_c for the sample 1.5 wt % C is 36.9 K, which is higher than that for the sample of x=5% (T_c =36.1 K), despite the former having a higher equivalent atomic percentage of carbon (x=5.4%).

Figure 1 shows the magnetic field dependence of J_c at 10, 20, and 30 K for the carbon-substituted MgB₂ samples. At 30 K, the undoped MgB₂ exhibits the highest J_c and the slowest decrease of J_c with H; whereas the sample of x=10% shows the lowest J_c and the quickest drop of J_c with H. It is evident that the J_c -H behavior at 30 K for these samples is positively correlated to their T_c values. However, when the temperature decreases to the values far below T_c , a totally different situation appears. For example, at 10 and 20 K, the diamond-doped samples show a much better J_c -H behavior. The J_c drops much more slowly in diamond-doped

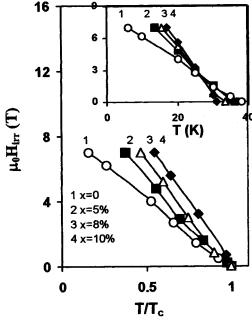


FIG. 2. Variation of $H_{\rm irr}$ with reduced temperature T/T_c for MgB_{2-x}C_x with x=0%, 5%, 8%, and 10%. Inset: $H_{\rm irr}$ -T plot for the same data shown in the main figure.

samples than in pure MgB₂. The best J_c at 20 K is found in the sample of x=10%, reaching a value of 6×10^3 A/cm² in a 4 T field, indicating that a strong flux pinning force exists in these diamond doped samples.

The $H_{\rm irr}-T$ relations for the diamond-substituted MgB₂ are shown in the inset of Fig. 2. The $H_{\rm irr}(T)$ curves get steeper with increasing doping level. The best value of $H_{\rm irr}$ reaches 5.7 T at 20 K for the sample of x=10%. As the T_c values vary with the diamond-doping level, only the $H_{\rm irr}-T$ relation cannot directly reflect the intrinsic irreversibility behavior for the samples of different doping levels. In the main panel of Fig. 2, the temperature dependence of $H_{\rm irr}$ is replotted using a reduced temperature, T/T_c . It is evident that the irreversibility field shifts towards higher temperatures with the increase of the diamond-doping level. The result clearly shows that the diamond doping does enhance the flux pinning in MgB₂ significantly.

However, the effect of diamond doping on the enhancement of flux pinning in MgB₂ may be counterbalanced by its suppression on superconductivity, as clearly shown in the situation of T=30 K (see Fig. 1). This counterbalancing effect may also exist at other temperatures, even when the effect of the J_c enhancement is dominant. The further increase of J_c depends critically on reducing the T_c -suppression effect in the MgB₂-diamond composite. This idea is confirmed by the results obtained in the diamondadded sample, 1.5 wt % C, which has a higher T_c than other diamond-doped samples (see inset of Fig. 1) and contains more nanodiamond inclusions as suggested by the XRD analysis and confirmed by our transmission electron microscopy analysis shown later. As shown in Fig. 4, the diamondadded sample shows a much better J_c -H behavior than the carbon-substituted sample. Its J_c reaches 1×10^4 A/cm² at 20 K and 4 T, and its $H_{\rm irr}$ reach 6.4 T at 20 K. In fact, at all temperatures below 35 K, the J_c –H behavior (results at 20 K are shown here only) and the H_{irr} -T relation (see the inset of Fig. 3) of the diamond-added sample are much better than

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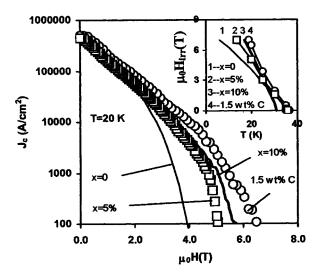


FIG. 3. Comparison of J_c-H relations at 20 K for diamond-added MgB₂ sample 1.5 wt % C with diamond-substituted MgB₂. The atomic percentages of carbon in the sample 1.5 wt % C and the sample of x=5% are almost the same. Inset: $H_{\rm irr}-T$ relations for the same samples shown in the main figure.

those of other samples in this study.

Figure 4 shows the typical results from microstructural analysis for the diamond-substituted MgB₂ and diamond-added MgB₂ samples. The diamond-substitutional sample mainly consists of relatively large MgB₂ grains (\sim 1 μ m or so in size) with a high density of dislocations. In some areas, discrete nanosized particles can be seen [Figs. 4(a) and 4(b)]. The diamond-added sample mainly consists of two kinds of nanoparticles: MgB₂ grains with a size of 50–100 nm and diamond particles with a size of 10–20 nm [see Fig. 4(c)]. In fact, this diamond-added MgB₂ forms a typical nanocomposite material. The nanodiamond particles are inserted into the MgB₂ grains. As the *ab* plane coherence length of MgB₂ is about 6–7 nm, ⁹ these 10–20 nm-sized diamond inclusions, with a high density, are ideal flux pinning centres and are responsible for the high performance in our samples.

The significant improvement of J_c and $H_{\rm irr}$ in the nanodiamond-added samples (1.5 wt%) can be attributed to their nanocomposite structure which consists of two kinds of nanoparticles: MgB $_2$ grains with a size of 50–100 nm and diamond particles with a size of 10–20 nm. The enhanced number of grain boundaries associated with the smaller grain size can enhance the flux pinning, as reported previously. $^{2-5}$

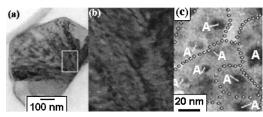


FIG. 4. FEGTEM micrographs for (a) a typical grain ($\sim 1~\mu m$) of diamond substituted MgB₂ with x=5% which shows the high density of dislocations (dark stripes) in the sample; (b) an enlarged view of the dislocations in Fig. 4(a); (c) diamond added MgB₂ with the carbon content of 1.5 wt %. The grain boundaries of MgB₂ are indicated by the guidelines. The diamond nanoparticles are marked by letter A beside it (for small ones) or on the particles (for big ones). The atomic percentages of carbon in these two samples are almost the same.

However, only this factor cannot fully explain the experimental results because the enhancement of flux pinning in the nanodiamond-added samples (1.5 wt%) is even much better than that in the Ti-doped MgB₂²⁻⁴ where the average grain size of MgB₂ reaches 8-10 nm. This indicates that there may be other mechanisms of flux pinning enhancement in the present system. One of the most likely candidates is the diamond nanoparticles which may play a similar role as Y₂O₃ nanoparticles did in Y₂O₃-doped MgB₂. ¹⁰ It is worth noting that, compared to the Y2O3 Y2O3-doped MgB2 the nanodiamond-added samples (1.5 wt %) has a higher J_c and $H_{\rm irr}$. This may be due to the advantage of the nanodiamond whose lattice contact (for the cubic diamond a = 0.356 nm) is very close to the c axis of MgB₂ (c = 0.352 nm). Therefore, these diamond nanoparticles may provide nucleation centres for MgB₂ and are tightly bound to them. It has been reported that some undoped MgB2 samples with a slight grain texture also show a high J_c and H_{irr} (see, for example, Narozhnyi et al.), 11 suggesting that achieving a textured microstructure is another effective way to improve J_c of MgB₂ because of a slight anisotropy existing in this system. Accordingly, it is expected that the performance of the MgB2-diamond nanocomposite may be further improved by optimizing the microstructure and the doping levels.

In summary, we have synthesized a MgB₂-diamond nanocomposite superconductor by adding nanodiamond powder into MgB₂. The nanocomposite consists of tightly packed MgB₂ nanograins (\sim 50–100 nm) with diamond nanoparticles (\sim 10–20 nm) inserted inside these grains. The J_c-H and $H_{\rm irr}-T$ characteristics have been significantly improved in this MgB₂-diamond nanocomposite, in comparison with MgB₂ bulk materials prepared with other techniques.

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