

# Role of surfaces and dopants in quantum devices and nanowire transistors

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## Role of surfaces and dopants in quantum devices and nanowire transistors

Author: Damon CARRAD Supervisor: A/Prof Adam MICOLICH

A thesis submitted in fulfilment of the requirements for the degree of Doctor of Philosophy



School of Physics Faculty of Science

December 2015

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Miniaturisation of electronic devices has driven development of high speed, high density processors and memory						
elements. This process has required extensive ontimisation of semiconductor materials and interfaces as the random						

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This thesis explores the role of p-type AlGaAs/GaAs heterostructures, self-assembled semiconductor nanowires and organic polymer electrolytes in this broad research programme. My research investigated the impact of background potentials generated by doping and surface states for quantum devices. I developed new wrap-gating techniques for InAs semiconductor nanowires towards economically viable arrays of III-V transistors on silicon substrates. This involved both conventional metal/oxide wrap-gates as well as nanoscale patterning of polymer electrolyte films to both improve the compatibility of organics with nanostructures and seek new functionalities for nanowire transistors. I then used polymer electrolytes to both act as an external dopant, and set the background potential for nanowire thermoelectrics. I also developed proof-of-principle complementary n- and p-type proton-to-electron transducers.

Throughout, I highlight the importance of dopants and surfaces. I show how these non-ideal aspects of semiconductor devices affect performance and attempt to find solutions where possible by, e.g., using sulfur-based surface passivation solutions or polymer electrolytes as an external dopant. Using these examples I illustrate that the drive to develop new technologies leads to new physics on both ends. Imperfections in research devices lead to new understanding of material physics, and once these are overcome, the new functionalities embodied by the devices can be used to study new aspects of nature.

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### Abstract

Miniaturisation of electronic devices has driven development of high speed, high density processors and memory elements. This process has required extensive optimisation of semiconductor materials and interfaces as the random nature of doping increasingly affects device performance and the influence of non-ideal surfaces and interfaces need to be counteracted. As Moore's law for silicon may soon reach its limit, there is a desire to harness electrically efficient III-V semiconductor materials in an economically viable way. There is also a desire to utilise new functionalities brought by quantum mechanics, thermoelectrics and organic materials.

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### Publications arising from this work

#### **Refereed Journal Publications**

D. J. Carrad, A. M. Burke, P. J. Reece, R. W. Lyttleton, D. E. J. Waddington, A. Rai, D. Reuter, A. D.Wieck & A. P. Micolich. 'The effect of (NH<sub>4</sub>)<sub>2</sub>S<sub>x</sub> passivation on the (311)A GaAs surface and its use in AlGaAs/GaAs heterostructure devices.' *Journal of Physics: Condensed Matter* 25, p. 325304 (2013)

My contribution: I worked closely with Dr Adam Burke on the fabrication and measurement of the electrical devices. I prepared samples for the photoluminescence and X-ray photoelectron spectroscopy and performed some of the photoluminescence measurements. I analysed the data and wrote the paper with A/Prof Adam Micolich. This paper is discussed in Chapter 3.

 D. J. Carrad, A. M. Burke, O. Klochan, A. M. See, A. R. Hamilton, A. Rai, D. Reuter, A. D. Wieck & A. P. Micolich. 'Determining the stability and activation energy of Si acceptors in AlGaAs using quantum interference in an open hole quantum dot.' *Physical Review B* 89, p. 155313 (2014)

My contribution: I fabricated and measured the electrical properties of the etched open hole quantum dot with Dr Adam Burke. I analysed the data and wrote the paper with A/Prof Adam Micolich. This paper is discussed in Chapter 3.

 D. J. Carrad, A. M. Burke, R. W. Lyttleton, H. J. Joyce, H. H. Tan, C. Jagadish, K. Storm, H. Linke, L. Samuelson & A. P. Micolich. 'Electron-Beam Patterning of Polymer Electrolyte Films To Make Multiple Nanoscale Gates for Nanowire Transistors.' *Nano Letters* 14, pp. 94–100 (2014)

My contribution: I developed the electron-beam patterning process, fabricated devices and conducted electrical measurements with the help of Dr Adam Burke and Roman Lyttleton. I analysed the data and wrote the paper with A/Prof Micolich. This paper is discussed in Chapter 4.

A. M. Burke, D. J. Carrad, J. G. Gluschke, K. Storm, S. Fahlvik Svensson, H. Linke, L. Samuelson & A. P. Micolich. 'InAs Nanowire Transistors with Multiple, Independent Wrap-Gate Segments.' *Nano Letters* 15, pp. 2836–2843 (2015)

My contribution: I assisted Dr Adam Burke and A/Prof Adam Micolich in developing the process for fabricating multiple-wrap gated devices. I also assisted with electrical measurements and development of the logic programs used to evaluate wrap-gate electrical balance. This paper is discussed in Chapter 4.

 S. Fahlvik Svensson, A. M. Burke, D. J. Carrad, M. Leijnse, H. Linke & A. P. Micolich. 'Using Polymer Electrolyte Gates to set-and-freeze Threshold Voltage and Local Potential in Nanowire-based Devices and Thermoelectrics.' Advanced Functional Materials 25, pp. 255–262 (2015)

My contribution: This work was highly collaborative. I worked with Sofia Fahlvik Svensson to fabricate the device. Sofia, Dr Adam Burke and myself worked together to obtain the electrical measurements. All authors contributed to the interpretation of the data. Sofia, A/Prof Adam Micolich and myself wrote the paper. This paper is discussed in Chapter 5.

D. J. Carrad, A. B. Mostert, A. R. Ullah, A. M. Burke, H. J. Joyce, H. H. Tan, C. Jagadish, P. Meredith & A. P. Micolich. 'n- and p-type proton-to-electron transducers based on III-V nanowires' *Manuscript in preparation*

My contribution: I fabricated devices with help from Rifat Ullah and Dr Adam Burke. I worked closely with Dr Bernard Mostert to obtain the electrical measurements. I am in the process of writing this paper with Bernard and A/Prof Adam Micolich. This paper is discussed in Chapter 6.

#### **Refereed Conference Proceedings**

 D. Carrad, A. Burke, S. Fahlvik Svensson, R. Lyttleton, H. Joyce, H. H. Tan, C. Jagadish, K. Storm, L. Samuelson, H. Linke & A. Micolich. 'Nanoscale polymer electrolytes: Fabrication and applications using nanowire transistors.' In: 2014 Conference on Optoelectronic and Microelectronic Materials Devices (COMMAD), pp. 286–289 (2014)

This presentation was based on the results collected for publications 3 and 6 above.

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# Abbreviations

2DEG	2-dimensional electron gas	
2DHG	2-dimensional hole gas	
AFM	Atomic force microscope/microscopy	
ALD	Atomic layer deposition	
CB	Conduction band	
CMOS	$\mathbf{C} \mathbf{o} \mathbf{m} \mathbf{e} \mathbf{t} \mathbf{a} \mathbf{l} \mathbf{-} \mathbf{o} \mathbf{x} \mathbf{i} \mathbf{d} \mathbf{e} \mathbf{-} \mathbf{s} \mathbf{e} \mathbf{m} \mathbf{i} \mathbf{c} \mathbf{o} \mathbf{d} \mathbf{u} \mathbf{c} \mathbf{t} \mathbf{o} \mathbf{r}$	
DSO	$\mathbf{D}$ igital <b>s</b> ignal <b>o</b> scilloscope	
EBL	Electron beam lithography	
EDL	Electric double layer	
FET	Field-effect transistor	
HEMT	$\mathbf{H}$ igh electron mobility transistor	
HHMT	$\mathbf{H}$ igh $\mathbf{h}$ ole $\mathbf{m}$ obility $\mathbf{t}$ ransistor	
LIA	Lock-in amplifier	
MCF	$\mathbf{M} agneto \mathbf{c} onductance \ \mathbf{f} luctuations$	
MOSFET	$\mathbf{M} etal\textbf{-o} xide\textbf{-s} emiconductor \ \mathbf{f} ield\textbf{-e} ffect \ \mathbf{t} ransistor$	
NMP	N-methyl-2- $p$ yrrolidone	
$\mathbf{NW}$	Nanowire	
NWFET	Nanowire field-effect transistor	
$\mathbf{PE}$	Polymer electrolyte	
PEO	$\mathbf{P}$ oly( <b>e</b> thylene <b>o</b> xide)	
PL	Photoluminescence	
PMMA	$\mathbf{P}$ oly(methyl methacrylate)	
QD	Quantum $dot$	

QPC	$\mathbf{Q} \mathrm{uantum} \ \mathbf{p} \mathrm{oint} \ \mathbf{c} \mathrm{ontact}$
SDBS	Source-drain bias spectroscopy
SEM	$\mathbf{S} \text{canning } \mathbf{e} \text{lectron } \mathbf{m} \text{icroscope}$
$\mathbf{SMU}$	Source measure unit
VB	Valence Band
XPS	${\bf X}\mbox{-}{\rm ray}\ {\bf p}\mbox{hotoelectron}\ {\bf s}\mbox{pectroscopy}$

# **Physical Constants**

Boltzmann constant	$k_B$	=	$1.381 \times 10^{-23} \text{ J/K},$
Donzinann constant			$8.617\times 10^{-5}~{\rm eV/K}$
Conductance quantum	$G_Q$	=	$\frac{2e^2}{h}$
Electron charge	e	=	$1.602 \times 10^{-19} \text{ C}$
Permittivity of free space	$\epsilon_0$	=	$8.854 \times 10^{-12} \frac{s^4 A^2}{m^3 kg}$
Planck constant	h	=	$6.626 \times 10^{-34} \text{ m}^2 \text{kg/s}$
Speed of Light	c	=	2.997 924 58 $\times 10^8 \ {\rm ms}^{-{\rm S}}$ (exact)

# Symbols

В	Magnetic field	tesla [T]
C	Capacitance	farads [F]
$C_{dot}$	Quantum dot capacitance	farads [F]
$C_G$	Gate capacitance	farads [F]
$D_S(E)$	Density of states at the surface	number of states per square centimetre per eV $[cm^{-2} eV^{-1}]$
$\Delta E_a$	Quantum dot addition energy	electron volts [eV]
$E_A, E_D, E_T$	Acceptor/donor/trap depth	electron volts [eV]
$E_C, E_V$	Conduction, Valence band energy	electron volts [eV]
$E_F$	Fermi energy	electron volts [eV]
$E_g = E_C - E_V$	Band-gap energy	electron volts [eV]
f	electrical frequency	hertz [Hz]
$I_d$	drain current	ampere [A]
$I_G$	gate leakage current	ampere [A]
$I_{on}, I_{off}$	transistor on/off current	ampere [A]
G	conductance	siemens [S]
l	mean free path	metres [m]
		number per cubic centimetre
n,p	electron/hole concentration	(3D) or square centimetre $(2D)$
		$[\rm cm^{-3}]$ or $[\rm cm^{-2}]$
Р	pressure	mbar
q	charge	coulombs [C]
R	resistance	ohms $[\Omega]$

	S	Seebeck coefficient	volts per kelvin $[V/K]$
	SS	subthreshold swing	millivolts per decade $[\rm mV/dec]$
	T	temperature	kelvin [K]
	$V_G$ or $V_g$	gate voltage	volts [V]
	$V_{in}, V_{out}, V_{supply}$	circuit input/output/supply voltage	volts [V]
	$V_{sd}$	source-drain voltage	volts [V]
	$V_{th}$	threshold voltage	volts [V]
	$V_T$	thermovoltage	volts [V]
	$V_{PE}$	voltage at the PE/NW interface	volts [V]
	$V_{po}$	pinch-off voltage	volts [V]
$\epsilon_r$		dielectric constant/	unitless
	$c_r$	relative permittivity	
$\kappa$		thermal conductivity	watts per metre kelvin
	10	onermar conductivity	$[W/(m{\cdot}K)]$
	$\lambda_F$	Fermi wavelength	nanometre [nm]
		charge carrier mobility	centimetre squared per
	$\mu$	charge carrier mobility	volt second $[\rm cm^2/Vs]$
	ν	optical frequency	hertz [Hz]
	Q	nanowire radius	nanometre [nm]
	-	aandustivity	siemens per centimetre
	σ	conductivity	[S/cm]
	τ	time constant	Seconds [s]

## Chapter 1

### Introduction and background

### **1.1** Introduction and chapter outline

The simple concept of the field-effect transistor has had a profound impact on scientific and industrial progress over the last half century through the proliferation of novel devices based on it.<sup>1–9</sup> The underlying field-effect principle is that the flow of electrons through a semiconductor is controlled electrostatically by a gate electrode. From the initial deployment of transistors in amplifiers and then on into switching in computers, the idea of electric fields controlling charge carrier density and tailoring the underlying band-structure has led to the engineering and control of quantum systems,<sup>5–7</sup> and the interfacing of manufactured electronics with biological systems.<sup>10,11</sup> In the meantime, the drive for more efficient technology has driven the size of devices to decrease well beyond many predicted physical limits, with Moore's law driving the exponential decrease in transistor size for five decades. Despite this, there is growing concern that the 7 nm long transistors in development at Intel are finally exposing the limits of silicon as the basis for transistors in integrated circuits. To increase energy efficiency further we need to seek out materials with better performance than silicon, or devices with increased functionality.<sup>4,12</sup>

At the level of fundamental research, substituting new materials and expanding functionality is seldom trivial. The discovery of non-ideal materials properties raises scientific questions that must be solved to fully exploit the new potential advantages. In particular, two chief problems arising as a result of reduction in device size involve dopant impurities and the proximity of the surface to the channel. Size reduction entails a decreasing number of dopant impurities within devices, meaning that non-homogeneous distributions and/or fluctuations in occupation increasingly affect the dynamics of the channel.<sup>13–15</sup> Countering this has involved efforts towards understanding dopant properties and developing strategies to altogether remove dopant impurities.<sup>16–20</sup> Secondly, the increase in surface/volume ratio with size reduction means surfaces increasingly affect device performance.<sup>4,21</sup> The disruption of the ordered crystal structure at surfaces and material interfaces makes them notoriously problematic, with roughness and long-lived charge traps causing mobility reduction and device instability,<sup>4,22–26</sup> along with issues obtaining ideal electrical behaviour of semiconductor/metal interfaces.<sup>27</sup> Once the problems are solved and new functionalities are realised, these devices can be turned back around to study nature, either by embodying a unique system or providing a different way of probing external systems.

My thesis covers this broad spectrum. At one end, we investigated the use of III-V semiconductor materials as the basis for novel transistor devices, including semiconductor nanowires,<sup>28–31</sup> and high-mobility planar heterostructure devices.<sup>32,33</sup> We studied non-ideal behaviours in these systems, including surface/interface charge traps and doping impurities and attempted to develop solutions for them. We also employed new techniques to observe and study these effects. In terms of adding functionality to devices, we explored the manufacture of hybrid organic/inorganic transistors, featuring semiconductor nanowire channels with polymer electrolyte gate dielectrics. This involved characterising the performance advantages of using polymer electrolyte dielectrics, and used them as the basis of an 'external doping' technique to provide a base-line charge carrier density without the need for dopant impurities. The dielectrics could generate quantum systems and also enhance their study at low temperatures in nanowire devices. In addition, we evaluated the thermoelectric properties of these devices, where the ultimate functional goal is conversion of waste heat into electrical energy, or efficiently cooling devices to cryogenic temperatures. Finally, the organic gate allowed us to develop complementary n- and p-type proton-to-electron transducers. These could be used to convert the proton signals typically seen in biological systems into electronic signals that can be read out by conventional electronics.

Chapter 1 gives the background behind my results. I start with the principles of operation of the field-effect transistor (FET), which forms the basis for almost all of the devices studied in this thesis. I then discuss the quantum physics of low-dimensional electron systems and how they can be addressed and controlled through field-effect methods. I discuss novel alternative methods for fabricating transistor elements including self-assembled semiconductor nanowires and polymer electrolyte gates. I also outline the non-ideal behaviours in research level devices that we seek to optimise, and explore the concept of biological interfacing.

Chapter 2 outlines the experimental procedures used in this thesis. This includes device fabrication and electrical measurement techniques at room temperature and low temperature,  $T \leq 4.2$  K.

Chapter 3 presents results on the cause of instability in *p*-type GaAs/AlGaAs heterostructures. High electron mobility *n*-type GaAs/AlGaAs heterostructures have formed the basis for a wealth of low dimensional quantum systems, but making the same kind of systems with holes in *p*-type heterostructures has been inhibited by gate instability. We explored the effect of surface states and dopant impurity charge migration on the operation of devices made from these structures. We attempted passivation of the (311)-oriented surface used for *p*-type heterostructures, and found why typical treatments were not as effective as for the more common (100)-oriented surface. We also fabricated a hole quantum dot in this material and used the unique signature of conductance fluctuations in response to an external magnetic field to capture a 'fingerprint' of changes in dopant impurity distributions. Chapter 4 focusses on novel techniques for gating self-assembled semiconductor nanowires. The results centre on gates that wrap around the nanowire to give the best control. We fabricated devices with multiple gates, using both traditional metal/oxide structures as well as polymer electrolyte gate dielectrics. In the latter case we developed a method to nanoscale pattern poly(ethylene oxide)-based polymer electrolytes and showed this organic/inorganic combination enhanced the functionality of both materials systems.

Chapter 5 explores the use of polymer electrolyte gates to enhance quantum systems in nanowires. The polymer electrolyte gates were used to 'set and freeze' the base-line charge density and the background potential for electrons traversing the nanowire at  $T \leq 4.2$  K. To help probe this system we used thermoelectric measurements, which can give additional information to complement traditional conductance measurements. In doing so we also showed that polymer electrolytes could play a vital role in thermoelectric cooling applications that require a specific charge density be set.

Chapter 6 explores the use nanoscale poly(ethylene oxide) dielectrics to facilitate proof-of-principle measurements for integrating nanowires with nano-biology. We discovered that these dielectrics supported proton transport and characterised the proton dynamics as a function of hydration level using nanowire transistors. This revealed a relatively high conductivity for protons in poly(ethylene oxide). The importance is that signalling in many biological systems occurs *via* proton transport rather than electron transport. As a result, interfacing manufactured electronics with the biological world requires transduction of these proton signals into electrical signals. Using *n*-InAs and *p*-GaAs nanowire channels, we fabricated complementary nanowire transistors gated by the field-effect arising from local proton concentration. The result was that a change in current caused by proton concentration fluctuations could be read out electronically. Pairing *n*-InAs and *p*-GaAs nanowires devices in functional circuits may open the possibility for size appropriate amplifiers of biological signals on chips compatible with Si-based microelectronics. Chapter 7 rounds out my thesis by summarising the key results and suggesting future directions. Using the examples from this thesis I show that the scientific study of device physics sits neatly between the motivations of understanding constituent material properties and the desire to generate novel devices that facilitate new understandings of physical systems from quantum to biological.

### **1.2** The field-effect transistor

The field-effect transistor (FET) consists of a semiconductor channel, usually doped *n*- or *p*-type, and three metal electrodes (see Fig. 1.1(a)). Device behaviour is determined by the different effect that voltages on these electrodes have on the bands and Fermi level within the semiconductor. The source and drain are electrically connected to the channel. Ideally the contacts are 'ohmic' – i.e. obey Ohm's law V = RI – with a low resistance such that the source/drain voltage  $V_{sd}$  drops mostly across the semiconductor channel (Fig. 1.1(b)). The contact between metals and semiconductors typically results in a potential barrier known as a Schottky barrier. This can be overcome by including a dopant in the contact metal and diffusing it into the semiconductor to form a thin, highly doped layer below the contact. The gate electrode is electrically insulated from the semiconductor; when an oxide layer is used, these devices are known as metal-oxide-semiconductor FETs (MOSFETs). Applying a voltage  $V_G$  to the gate capacitively depletes or induces charge carriers in the conduction channel. In energetic terms,  $V_G$  and the associated electric field alter the energy of the conduction and valence bands with respect to the Fermi level  $E_F$ , illustrated in Fig. 1.1(c). To pass a current  $I_d$ through the channel,  $V_{sd}$  is applied between source and drain to drive the device out of equilibrium, as in Fig. 1.1(d). In this way, the combination of  $V_{sd}$  and  $V_G$ determines the potential landscape for electrons/holes traversing the semiconductor channel, e.g. Fig. 1.1(e) illustrates how the gate is used to generate a potential barrier that stops the flow of current.

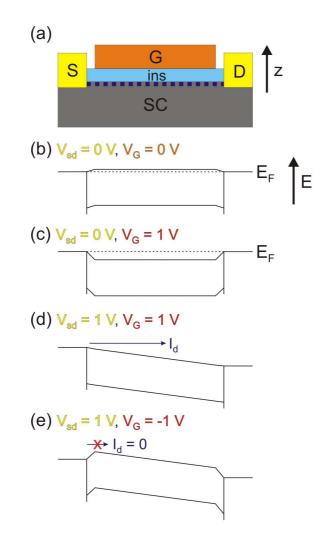


FIGURE 1.1: (a) Schematic of a field-effect transistor featuring metallic source (S), drain (D) and gate (G) electrodes along with the semiconductor channel (SC) and gate insulator (ins). The conduction channel is illustrated with the purple dotted line. (b) Band diagram of the device in (a) for  $V_{sd} = V_G = 0$  V. The source and drain have formed ohmic contacts with the semiconductor as the conduction band is aligned with the Fermi level  $E_F$  at S and D. (c) Positive applied  $V_G$  moves the position of the bands relative to  $E_F$ , increasing electron density. (d) An applied  $V_{sd}$  allows current  $I_d$  to flow through the device. (e)  $I_d$  is turned off by a negative  $V_G$ , as it causes a potential barrier for electrons.

Figure 1.2 shows the idealised  $I_d$  response to  $V_G$  applied to *n*- and *p*-type transistors at different  $V_{sd}$ . These plots are known as 'transfer characteristics'. More positive/negative  $V_G$  increases/depletes the electron/hole density, increasing/decreasing channel conduction; the device is said to be 'on'. Supplying a larger  $V_{sd}$  results in increased  $I_d$  for each  $V_G$ . Reversing the  $V_G$  direction depletes carriers.

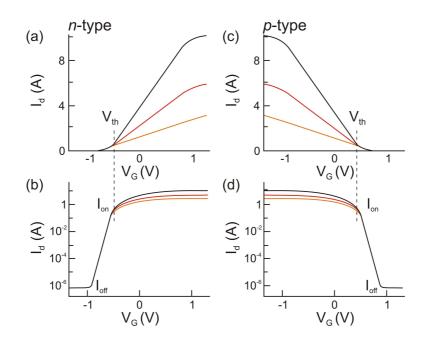


FIGURE 1.2: Transfer characteristics for n- and p-type transistors (a/b and c/d respectively) with linear and log  $I_d$  axes (top and bottom). The black, red and orange traces represent decreasing, constant  $V_{sd}$ . Below  $V_{th}$ ,  $I_d$  drops exponentially from  $I_{on}$  to  $I_{off}$ .

At the threshold voltage  $V_{th}$  the band diagram looks like Fig. 1.1(e), where a potential barrier blocks transport. Here,  $V_{sd}$  has no effect on  $I_d$  as the barrier height depends only on  $V_G$ . The current through the device goes to zero exponentially with  $V_G$ , turning off the device. The exponential turn-off in the subthreshold region can be seen more easily on a log-linear plot (Figs 1.2(b/d)), where  $\log(I_d)$  vs.  $V_G$  has a linear slope. The inverse subthreshold slope is called the subthreshold swing  $SS \equiv \frac{\partial V_G}{\partial \log(I_d)}$ , given in mV of  $V_G$  per decade change of  $I_d$ , i.e. mV/dec. It is an important metric as it quantifies the  $V_G$  swing required to turn the device off; a low SS is vital for low power applications. The value is determined by material parameters and energy considerations, and is thermodynamically limited to around 60 mV/dec at room temperature.<sup>1</sup> A good transistor combines a low SSwith a high on/off ratio,  $\frac{I_{onf}}{I_{off}}$ . This results in orders of magnitude changes in device resistance within a 1 V range of  $V_G$ . The next section discusses the importance of these features for switching and amplification.

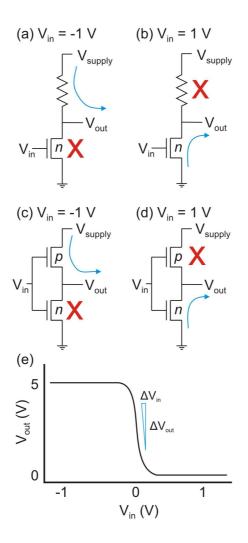


FIGURE 1.3: (a/b) An *n*MOS inverter circuit with *n*-type transistor and series resistor. (a) For  $V_{in} = -1$  V, the transistor is off, and  $V_{out} = V_{supply}$ . (b) For  $V_{in} = 1$  V, the transistor is on, making  $V_{out} = 0$  V. (c/d) a CMOS inverter with *p*- and *n*-type transistors. The  $V_{in}$  swing between -1 and 1 V takes the *p*-type/*n*-type devices from off/on to on/off, respectively. (e)  $V_{out}$  vs.  $V_{in}$  for a well matched CMOS inverter. The steep slope around  $V_{in} = 0$  V facilitates small signal amplification with high gain  $\Delta V_{out}/\Delta V_{in}$ .

#### **1.2.1** Basic applications: switching and amplification

Two examples of an inverter are shown in Fig. 1.3. This is the most basic circuit for transistor switching and amplification. In the first example (Figs 1.3(a/b)), a resistor is placed in series with the *n*-transistor channel, and  $V_{supply} = 5$  V is applied across the resistor and channel. A signal  $V_{in}$  is applied to the gate, and the response  $V_{out}$  is read out between the resistor and transistor source. Typical transistor ON and OFF resistances  $R_T$  are in the k $\Omega$  and G $\Omega$  ranges, respectively. The resistor value  $R_R$  is chosen to be between the ON and OFF resistances, e.g.,  $R_R = 1 \text{ M}\Omega$ . For  $V_{in} = -1 \text{ V}$ , the transistor is off, and since  $R_R \ll R_T$  by orders of magnitude, the  $V_{out}$  is linked to the  $V_{supply} = 5 \text{ V}$ . When the input is reversed to  $V_{in} = 1 \text{ V}$ ,  $R_R \gg R_T$ ; this essentially grounds  $V_{out} = 0 \text{ V}$  through the open transistor channel. In this way, switching  $V_{in}$  between two voltages switches  $V_{out}$ also. In addition, the 2 V signal (from -1 to +1 V) at  $V_{in}$  is amplified to a 5 V signal at  $V_{out}$ . The fact that a high  $V_{in}$  gave a low  $V_{out}$  and vice versa is the origin of the term 'inverter'.

Inverter performance is improved by using complementary *n*-type and *p*-type MOSFETs in the configuration of Figs 1.3(c/d). This is the cornerstone of the complementary MOS (CMOS) platform that is the basis for computer processor technology. A CMOS inverter simply replaces the resistor from the circuit in Figs 1.3(a/b) with a *p*-type transistor, and  $V_{in}$  is applied to both gates simultaneously. Here,  $V_{in} = -1$  V turns the *p*-transistor on and the *n*-transistor off, giving  $V_{out} = 5$  V. Similarly,  $V_{in} = 1$  V turns the *p*-transistor off and the *n*-transistor on, giving  $V_{out} = 0$  V. High amplification gain is achieved by applying small signals around the point where  $V_{in}$  swings  $V_{out}$  between  $V_{supply}$  and ground. Figure 1.3(e) shows that at this point very small changes in  $V_{in}$  result in large changes in  $V_{out}$ ; the origin of the voltage gain.

The switching action of inverters also gives a practical realisation of the NOT logic operation, since high voltages (logic state 1) on the input map to low voltages (logic state 0) on the output and vice versa. CMOS is used in computer processor technology due to its low power consumption: when  $V_{in} = \pm 1$  V, one or the other of the transistor channels is closed and no current flows through the circuit. This means power is only consumed very briefly during switching events.<sup>2</sup> More complex CMOS configurations yield practical realisations of logic operations such as AND and OR. However, demonstration of an inverter is typically at the edge of fundamental research as it serves as a proof-of-principle for a new potential CMOS platform. Scaling up the platform and optimising integrated circuits is mostly an engineering problem from that point onwards.

### **1.3** Non-ideal properties of real devices

The band structure of solid state materials arises from the regularity of atomic placement in a crystal lattice: the overlapping atomic orbitals hybridise to form energy bands that span the entire crystal.<sup>34</sup> While modern growth techniques such as molecular beam epitaxy have allowed manufacture of near perfect crystals with atomic precision,<sup>35</sup> the majority of materials contain imperfections and impurities. Indeed, the doping of semiconductors entails intentionally introducing impurities. Further, the crystal structure necessarily breaks down dramatically at the surface, or at interfaces with other materials.<sup>34</sup> Below I will briefly discuss how these defects can be understood as localised perturbations against the backdrop of the non-localised bands, and how the down-sizing of devices places increased importance on the distribution of dopant atoms and the effects of surfaces.

#### 1.3.1 Doping and bulk impurities

The most familiar microscopic modification to a band structure arises from doping impurities. A comparatively low concentration of atoms with one more or less proton than that of the host crystal is introduced.<sup>34</sup> Figure 1.4 illustrates how the atomic orbitals of the dopant atoms do not overlap like those of the host crystal. Hence they do not form a continuous band; rather their energy levels remain localised at the physical location of each dopant. For higher proton number Z, the attraction to the nucleus is stronger, giving more tightly bound atomic energy levels. This means that for, e.g., P in Si (Z = 29 and 28 respectively), isolated energy levels containing an electron sit just below the conduction band. If the difference in energy between the dopant level and conduction band is small compared to the thermal energy  $k_BT$ , the electrons will populate the conduction band and leave behind ionised dopant impurities. Similarly, the energy levels of Al are shifted slightly higher than Si, providing a localised empty energy state above the valence band. This traps electrons and leaves a non-localised hole in the valence band.

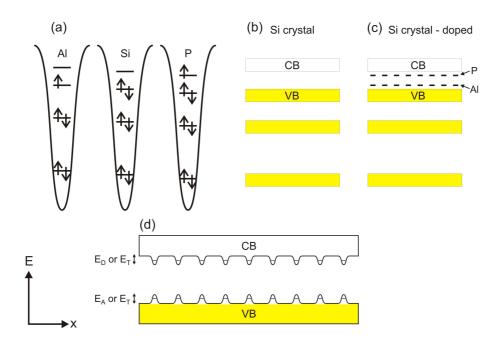


FIGURE 1.4: (a) Energy spectrum for isolated aluminium, silicon and phosphorous atoms. Increasing Z results in an additional electron (represented by spin arrows), and also generates a stronger attractive force, pulling energy levels down. (b) Hybridisation of discrete atomic orbitals leads to the formation of bands. In semiconductors, the bands are occupied until the top of the valence band (VB), leaving and empty conduction band (CB). A small density of Al or P impurities in the Si crystal act as if they are isolated atoms with discrete energy levels. (d) Impurity incorporation modifies the bands to add potential wells for electrons in the CB and holes in the VB. The wells can act as charge traps if the donor/acceptor/trap depth exceeds thermal energy,  $E_D, E_A, E_T > k_BT$ .

The incorporation of unintentional impurities is conceptually no different; the result is localised discrete energy levels dispersed throughout the crystal. Both unintentional and intentional impurities can possess undesirable features that contribute negatively to device performance. Since they provide potential wells just below/above the conduction/valence bands (Fig. 1.4(d)), impurities can act as electron/hole traps if the well depth  $E_T$  is large compared to  $k_BT$ . Here, a free electron trapped in the deep well is unable to escape due to insufficient thermal energy. If the trap depth  $E_T \sim k_BT$ , the charge state of the impurity may fluctuate between positive (unoccupied) and neutral (occupied) as electrons enter and leave the trap spontaneously. In larger devices the effect of many impurities rapidly switching charge state may be averaged out and not affect operation. By contrast, nm-scale devices often exist on the scale of the potential generated by the

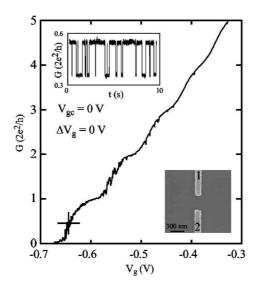


FIGURE 1.5: Data from an experiment by Pioro-Ladrière *et al.*<sup>15</sup> The main figure shows conductance  $G = \frac{I_d}{V_{sd}}$  vs. gate voltage  $V_g$  for the device shown in the lower inset. The device consists of split surface gates that generate a 1D quantum system in an underlying AlGaAs/GaAs heterostructure as described in Section 1.4.2. The data is noisy due to a two-level switching behaviour arising as a result of changing occupation of nearby Si dopant impurities. Upper inset: The two-level switching is clearly seen if the device is held at constant  $V_g \sim -0.64$  V, marked with a cross in the main figure.

impurity.<sup>13,14,36</sup> Such devices can be greatly affected if this potential switches between charge states. For instance, the current through a sensitive quantum device in Fig. 1.5 switches between two states as a nearby single impurity charges and discharges.<sup>15,37</sup> The importance of impurity configuration in setting a background potential for quantum devices is only now being fully appreciated; I will discuss this in Section 1.4.4 and present results from a device where background impurities dominate device behaviour in Chapter 3.

### **1.3.2** Surface states and interface states

The most dramatic rupture of the crystal structure comes at the surface. The ordered lattice structure is abruptly terminated, leaving surface atoms with unsatisfied dangling bonds, as in Fig. 1.6(a).<sup>2,21,34,39</sup> The dangling bonds seek to form surface bonds with each other as in Fig. 1.6(b), or capture oxygen atoms from the atmosphere and form a thin, amorphous oxide layer. Each of these surface bonds

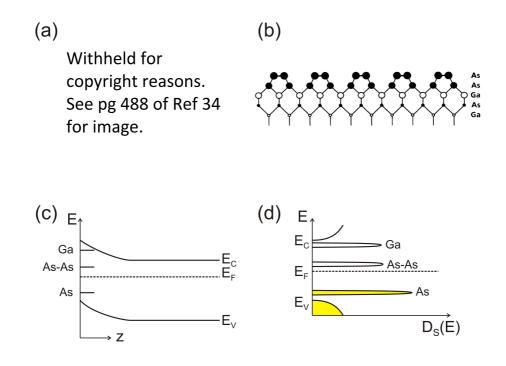


FIGURE 1.6: (a) The underlying order of a crystal is disrupted at the surface, leaving surface atoms with unsatisfied dangling bonds. (b) One possible reconstruction the (100) GaAs surface may take in order to satisfy dangling bonds, according to Ref. 38. The surface of semiconductors exposed to air consists of an amorphous layer of oxides. (c) Band diagram near the (100) surface of *n*-type GaAs. The Ga-dangling, As-dangling and As-As dimer bonds generate surface states that pin  $E_F$  mid-gap, but the doping level requires  $E_F$  to sit closer to  $E_C$  in the bulk. This generates a surface barrier that makes it difficult to inject carriers. (d) Density of states at the surface  $D_{SS}(E)$  illustrating the source of  $E_F$  pinning. The high density of surface states compared to conduction and valence band states means surface state occupation/density determines the Fermi level position. Note this diagram only applies at the semiconductor surface due to the localised nature of the surface states. Images in (a) and (b) from Refs 34 and 38.

are different to the bulk lattice bonds, and as such their associated electron energy states do not necessarily correspond to energies within the bulk bands. If the energies exist in the bulk band gap, these so called 'surface states' act as deep trapping sites localised at the surface.<sup>21,39</sup> Taking GaAs as an example, Fig. 1.6(c) shows that a clean (100) surface in vacuum consists of: Ga-dangling bonds associated with empty surface states; filled As- dangling bonds associated with filled states; and in some cases, As-As dimers associated with empty mid-gap surface states.<sup>21,40–43</sup> The surface states with energies in the band gap act as deep charge traps. By comparison, InAs has a smaller band-gap (0.345 eV vs. 1.42 eV for GaAs) and so the highest density of surface states is in the conduction band.<sup>40</sup> In reality this surface state spectrum is modified following exposure to air due to the growth of an amorphous native oxide layer.<sup>44,45</sup>

A high density of surface states presents numerous problems when depositing metal onto a semiconductor to act as source/drain contacts or gates. The problems arise from Fermi level position being determined by the occupation of these surface states, rather than the occupation in the bulk energy bands (see Fig. 1.6(d)). This pins the Fermi level of any metal deposited on the surface with limited dependence on metal composition and semiconductor doping concentration.<sup>27,46,47</sup> For GaAs, the Fermi level is pinned mid-gap, making it difficult to inject carriers into the conduction or valence band. For InAs, the Fermi level pinning in the conduction band.<sup>48</sup> Both situations are far from the ideal whereby the interface Schottky barrier can be tuned by metal composition and semiconductor doping.<sup>2</sup>

If the surface states are deep, long-lived traps (i.e.  $E_T > k_B T$ ) with a high density, they can cause instability and hysteresis when gating transistors. This is outlined in Fig. 1.7 for an *n*-type channel. A negative voltage on the gate may cause positive charge to be trapped in the surface states at the expense of depleting the channel underneath. This reduces the ability of the gate to deplete the channel, and in extreme cases may result in regions where  $I_d$  is not affected by  $V_G$  at all, as all of the energy goes into populating surface states. If a sweep to depletion is stopped,  $I_d$  will rise over time as the charge traps continue to be filled with positive charge that compensates the gate voltage. Once all the charge traps are filled, the Fermi level can be controlled again and depletion can continue. As the gate is swept towards zero, the charge traps will depopulate, but depopulate slowly. This means that initially the gate has good control over the channel, and  $I_d$  rises quickly, following a higher path than it did on the sweep to depletion. This produces the clockwise hysteresis loop for electron devices in Fig. 1.7. The characteristics for hole devices is the mirror image, since depletion occurs for positive voltages. This produces an anti-clockwise hysteresis loop that will be important in discussion of instability in *p*-type quantum devices in Sec. 1.4.5 and Chapter 3.

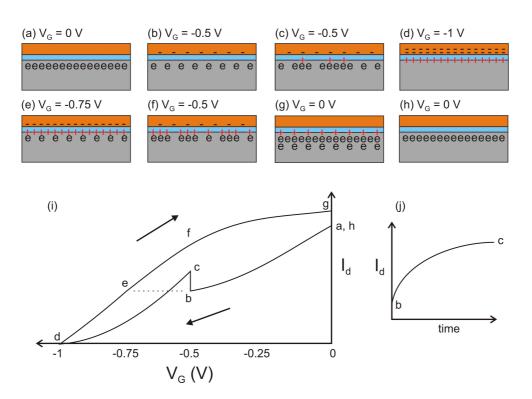


FIGURE 1.7: (a-h) Illustration of surface state charge trapping in a MOS-FET with metal gate (orange), insulator (blue) and semiconductor channel (grey).(i/j) Electrical characteristics corresponding to how each situation in (ah) generates hysteresis and instability in device characteristics. (a) Electrons, e, in the semiconductor channel carry current  $I_d$ . (b) Increasing  $V_G$  applies negative charge to the gate (-), depleting the channel of electrons. (c) Holding  $V_G = -0.5$  V causes trapping of positive charges at the semiconductor/oxide interface (+), leading to the re-accumulation of electrons. (d) The channel is fully depleted at  $V_G = -1$  V, but depletion may have required filling of all surface states. (e) As  $V_G \rightarrow 0$  V, the surface states de-trap slowly. As such, the electron density for  $V_G = -0.75$  V on this sweep is the same as it was for  $V_G = -0.5$  on the sweep towards  $V_G = 1$  V. (f) Similarly, the electron density for  $V_G = -0.5$  V on this sweep is higher than it was for  $V_G = -0.5$  on the sweep towards  $V_G = 1$  V. (g) Not all surface states have de-trapped at  $V_G = 0$  V, giving a higher current than in (a). (h) With time, surface states may de-trap, returning the device to the state it was at (a). (i) Hysteretic  $I_d$  vs.  $V_G$  traces for the sweeps to and from  $V_G = -1$  V. Points corresponding to figures (a-h) are indicated. The dashed line shows that  $I_d$  is the same at points b and e. (j)  $I_d$ increases over time from points b to c as surface traps fill slowly, compensating the gate voltage at constant  $V_G = -0.5$  V.

#### **1.3.2.1** Surface state passivation

The success of Si-based transistors has much to do with the fact that the bonds between Si and its native oxide  $SiO_2$  result in a low density of band-gap surface

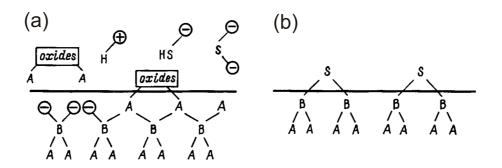


FIGURE 1.8: Chemical passivation for III-V semiconductors. A and B represent group-III and group-V elements, respectively. (a) The alkali solution etches the surface oxide layer. (b) Sulfur atoms form surface bonds, ideally with energy states outside the band gap. Adapted from Ref. 44.

states.<sup>4,39,40,49</sup> Unfortunately no such oxide is known for III-V's semiconductors. As a result, III-V surfaces and interfaces with gate oxides typically possess a high density of electrically active surface states in the band gap, which impede electrical and optical device performance.<sup>4,40,44,49</sup> Removing this impediment involves changing the bonding state of the surface/interface atoms with the aim of moving the surface states out of the band gap and into the conduction band, where they do not act as charge traps.<sup>21</sup> A favoured technique is the application of chemical passivation treatments. Passivation treatments act in two steps; first, they strip the natural oxide and secondly, they introduce an adsorbate atom which bonds in such a way that the associated surface states are non-band gap states (see Fig. 1.8).<sup>21,44,50</sup> A variety of treatments exist from both liquid and gas phases, with their suitability for different semiconductor surfaces dependent on adsorbate species as well as the chemical environment of the solution and semiconductor surface.<sup>21,51,52</sup>

One of the first passivation treatments for the (100) GaAs surface used spunon Na<sub>2</sub>S·9H<sub>2</sub>O.<sup>53</sup> This opened up a wealth of sulfur-based passivation solutions for this surface.<sup>21</sup> Aqueous and solvent-based ammonium sulfide solutions with or without extra sulfur loading – i.e.  $(NH_4)_2S$  or  $(NH_4)_2S_x$  – quickly emerged as the preferred solution due to the more effective native oxide removal.<sup>44,50–52,54</sup> These solutions were found to reduce the density of band gap states<sup>41,43</sup> and the associated surface Fermi level pinning<sup>55</sup> through the formation of Ga-S and As-S bonds at the exclusion of Ga-O/As-O bonds.<sup>56–59</sup> Annealing the sample to  $T = 300 - 360^{\circ}$ C after treatment was found to improve the passivation in some cases by transferring S atoms from As-S to Ga-S bonds.<sup>41,60,61</sup> References 44 and 21 provide excellent overviews of the passivation process, including the etching of oxides by the alkali solution and the surface adsorption of sulfur atoms. I will discuss the importance of Ga-S and As-S bonding for passivation in more detail in light of my results in Chapter 3.

One limitation of sulfur-based passivation is the low stability of Ga-S and As-S bonds when exposed to air; over a period as short as 30 mins, the sulfur desorbs, allowing a re-adsorption of O which returns the surface to its original state.<sup>57</sup> However, the passivated surface remains intact for many metal- and oxide-deposition techniques, and once covered the surface is protected from the environment and will not re-oxidise.<sup>40,41,62</sup> This ensures  $(NH_4)_2S_x$  is viable where the passivated surface can be quickly covered, such as in this thesis (Chapter 2 gives complete details).

# 1.4 Quantum physics of low-dimensional systems

The past few decades have seen the solid state realisation of many of the idealised quantum systems studied in undergraduate courses.<sup>5,35,63</sup> A combination of materials choice and selectively patterned surface gates can be used to modify the band structure. Quantum systems arise when the bands form potential wells that confine electrons/holes to regions on the order of the de Broglie wavelength at the Fermi energy (simply called the Fermi wavelength  $\lambda_F$ ). Within these wells, electrons/holes occupy discrete energy levels known as sub-bands. The robustness of these systems depends on two conditions: <sup>5,35,63</sup> first, the electrons/holes should not be scattered by, e.g., ionised impurities or phonons. This requires a high carrier mobility to ensure a long mean free path  $\ell$  between scattering events. Secondly, the sub-band spacing must be larger than the thermal energy  $k_BT$  to prevent

thermal smearing from making the sub-bands indistinct and returning the system to classical behaviour. For this reason, most devices require low temperatures, typically  $T \lesssim 4$  K. At these temperatures phonons are strongly suppressed and impurity scattering dominates mobility. By harnessing high mobility materials and cryogenic measurement techniques, 2D, 1D and 0D systems of various forms have been realised in the solid state using a combination of band engineering and selective field-effect depletion. The breadth of available systems combined with the precision of control and read-out of particle dynamics *via* the field-effect and associated electrical measurement techniques makes nanoscale electronic devices a powerful tool for testing the theoretical predictions of quantum mechanics.<sup>5,35,64,65</sup> There is also a strong drive to harness their power for future devices, e.g., in quantum computing applications.<sup>66,67</sup>

### 1.4.1 2D systems; two dimensional electron/hole gas

A two dimensional electron/hole system (also known as 2D electron/hole gas, 2DEG/2DHG) can be generated at the junction between two materials with different band-gaps.<sup>5,35</sup> The junction can be a semiconductor heterostructure or a semiconductor/insulator interface (see Fig. 1.9). Application of an electric field from, e.g., a gate on the opposite side of the higher band-gap material induces electrons or holes in the lower band-gap material. The charge carriers are unable to overcome the potential barrier at the interface, which confines them to a twodimensional sheet in the z-direction that extends in the x- and y-directions. To maximise the charge carrier mobility, the best choice is a lattice matched semiconductor heterostructure.<sup>5</sup> Electron mobility is much reduced for 2DEGs induced at semiconductor/insulator interfaces due to the interface/surface impurities outlined in Section 1.3.2.<sup>35</sup> Heterostructures can provide significantly less interface defects, but only if the two materials have matched inter-atomic spacing. Attempting to grow a heterostructure with different lattice constants results in significant strain, leading to defects at the interface, and even macro-scale fractures.  $^{1,35,69}$ Thankfully, alloying of III-V compounds, e.g.  $Al_xGa_{1-x}As$ ,  $Ga_xIn_{1-x}As$ , can be

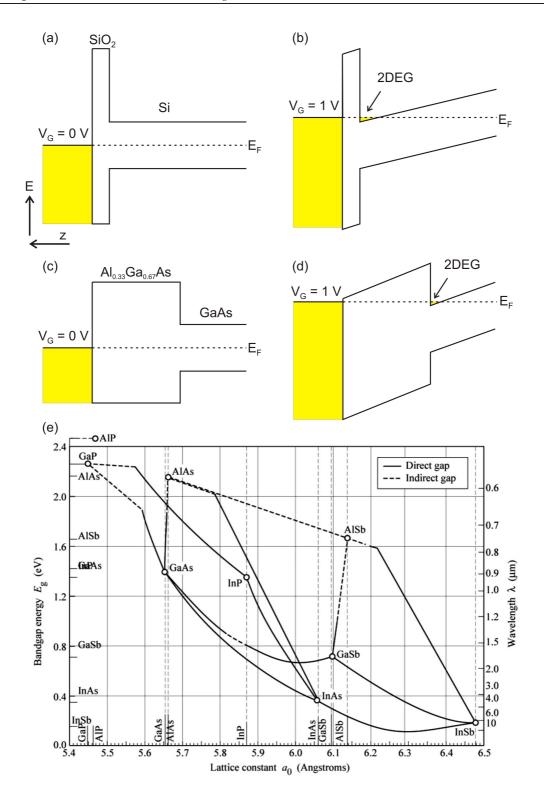


FIGURE 1.9: (a) SiO<sub>2</sub>/Si and (c) Al<sub>0.33</sub>Ga<sub>0.67</sub>As interfaces are used to induce 2D electron gases (2DEGs) via application of positive  $V_G$  (b, d). The 2DEG occupies a triangular well on a scale comparable to  $\lambda_F$ , confining electrons in the z-direction. (e) Band gap vs. lattice constant for III-V materials. Lines between pure compounds indicate the continuous variation of both parameters with alloying. Al<sub>x</sub>Ga<sub>1-x</sub>As is a popular choice due to the lattice matching across all alloys. Fig. (e) from Ref. 68.

used to vary both lattice constant and band gap. As Fig. 1.9(e) shows, there are a few materials combinations with the same lattice constant but different band gaps. The most well developed platform for quantum devices is GaAs with an  $Al_xGa_{1-x}As$  alloy barrier. Drawing lines of constant lattice spacing in Fig. 1.9(e) reveal other possible combinations of fixed alloys, e.g., InP/Ga<sub>0.47</sub>In<sub>0.53</sub>As is used for high speed electronic devices.<sup>35</sup> However, the GaAs/Al<sub>x</sub>Ga<sub>1-x</sub>As system provides much greater tunability as varying the alloy fraction x enables a continuous increase in band gap energy from 1.42 eV (GaAs) to 2.17 eV (AlAs) with a lattice mismatch of less than 0.15%.<sup>35,70</sup> Alloy fraction x = 0.33 is widely chosen as a trade-off between obtaining a sufficiently high barrier while minimising lattice strain. The devices in this thesis utilise a  $Al_{0.33}Ga_{0.67}As$  layer, which I will refer to as AlGaAs for brevity.

### 1.4.1.1 High electron/hole mobility transistors

Figure 1.9(d) illustrates a problematic aspect of electrostatically inducing the 2DEG or 2DHG; it requires a large gate voltage to turn the device on, and risks current leaking over the AlGaAs barrier into the device. This is overcome with a band engineering technique called 'modulation doping', where the AlGaAs layer is doped.<sup>71</sup> Rather than occupying the AlGaAs layer, the electrons or holes diffuse into the GaAs layer where the interface potential prevents them from returning to the AlGaAs. The AlGaAs is left with positively/negatively charged donor/acceptor atoms, which attract the electrons/holes back towards the interface. The combination of the attractive dopant potential and the interface barrier potential generates confinement in the z-direction at  $V_G = 0$  V (see Fig. 1.10).

The power of modulation doping lies in the fact that the dopant impurities and charge carriers are spatially separated. Ensuring the ionised dopant atoms are not directly in the path of electrons/holes maintains a high mobility<sup>71</sup> and leads to mean free paths of many microns at low  $T \sim 4$  K.<sup>5</sup> The high mean free path

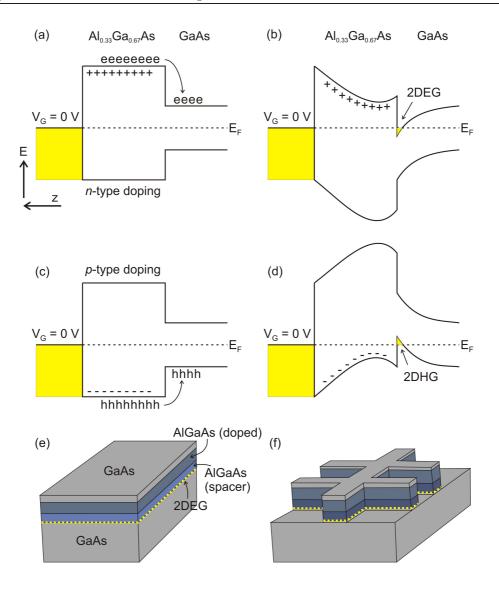


FIGURE 1.10: Modulation doping for (a/b) electrons and (c/d) holes. Electrons/holes diffuse from the doped AlGaAs layer into the GaAs layer. The electrons/holes leave positive/negatively charged dopant impurities in the Al-GaAs layer. The combination of the attractive dopant potential and AlGaAs potential barrier traps electrons/holes in a triangular well at the interface. (e) An undoped spacer layer increases mobility and GaAs cap prevents AlGaAs oxidisation. (f) This wafer is selectively etched into a Hall bar mesa, with contact arms that are ultimately connected to voltage sources and probes (full details in Chapter 2).

and existence of the 2DEG without applied bias has seen modulation doped Al-GaAs/GaAs heterostructures find extensive use as the basis for a wealth of lowdimensional systems discussed below. Mobility is enhanced further by the addition of an undoped AlGaAs spacer layer between the GaAs and doped AlGaAs layers.<sup>72</sup> Choosing the spacer depth involves a trade-off between mobility and carrier density; the latter is reduced as the dopants are further away from the 2DEG/2DHG. A thin GaAs cap is added at the surface as AlGaAs oxidises strongly in air. The complete wafer is shown in Fig. 1.10(e).

Constructing devices from this wafer entails the addition of low resistance ohmic contacts to the 2DEG/2DHG and surface gate(s) to modulate carrier density. First, the 2DEG/2DHG is restricted to a controlled region in the x- and ydirections by etching a 'mesa' structure like that in Fig. 1.10(f); confinement in the x- and y-directions is much larger than  $\lambda_F$  so the system is still two dimensional. The mesa typically features multiple arms, each with an ohmic contact that connects the circuitry to the 2DEG/2DHG with a fixed resistance. The multiple arms are used to perform 4-point and Hall voltage measurements, hence the structure is known as a 'Hall bar'. The gates can be deposited directly onto the GaAs since the combination of the surface barrier and AlGaAs layer is sufficient to insulate the 2DEG/2DHG from the gate for low voltage applications.<sup>5,73</sup> A gate that uniformly covers a section of the channel can be used to modulate the carrier density as in a transistor. The transistor-like behaviour of this high mobility system leads to the names high electron mobility transistor (HEMT) and high hole mobility transistor (HHMT).<sup>72</sup> The excellent characteristics of HEMTs encouraged development towards high-speed microwave applications.<sup>72,73</sup> In fundamental research, the interest lies in selectively patterning nanoscale gates on the surface and using the field-effect to provide additional confinement in the x- and/or y-directions. Doing so enables the engineering and manipulation of one- and zero-dimensional quantum systems.

## 1.4.2 1D systems; quantum wires and quantum point contacts

Generating one dimensional systems from two-dimensional systems such as 2DEGs entails generating additional confinement perpendicular to the direction of transport. This can be accomplished *via*, e.g., split surface gates that selectively deplete

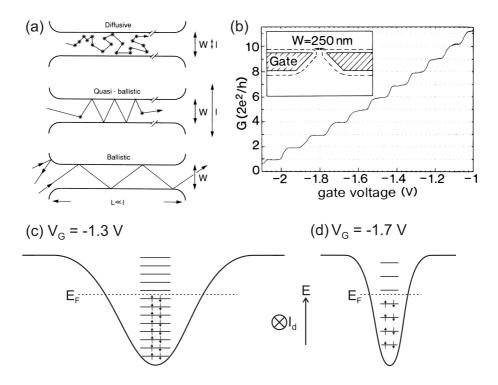


FIGURE 1.11: (a) 1D constriction with diffusive, quasi-ballistic and ballistic transport. Diffusive transport involves many scattering events and occurs when the length L and width W of the constriction are larger than the mean free path  $\ell$ . Ballistic transport occurs when  $\ell > W, L$  and carriers pass through the constriction without scattering. Electrons undergoing quasi-ballistic transport  $(\ell > W, l < L)$  will scatter only a few times, which can be viewed as a reduced transmission probability  $T_i$  (see text). (b) Split surface gates apply a 1D confining potential (dashed lines) to an underlying 2DEG. Ballistic transport through this potential results in conductance G quantised in units of  $2e^2/h$ . (c) Conduction band potential in the direction of current flow for  $V_G = -1.3$  V applied to the side gates in (b). Each electron carries  $I_d = (e^2/h)V_{sd}$ . (d) More negative  $V_G$  squeezes the confining potential, increasing the sub-band spacing. the sub-bands depopulate as they pass through  $E_F$ . (a/b) from Ref. 5

an underlying 2DEG. The conductance  $G \equiv I_d/V_{sd}$  and behaviour of electrons in 1D systems is dependent on the underlying transport regime; the three basic regimes are illustrated in Fig. 1.11(a). Diffusive transport occurs when device dimensions exceed the electron mean free path. Here, scattering from, e.g., dopant impurities dominates G according to classical electron transport theory. If no scattering occurs as electrons pass through this constriction they are said to undertake ballistic transport. In the ballistic regime, the conductance of the device is determined by the confining potential.<sup>74,75</sup> Operation of this device, known as a quantum point contact (QPC), is illustrated in Figs 1.11(b-d). The electric field from the two gates establishes a confining potential that leads to electrons traversing the constriction through 1D sub-bands, analogous to the modes of a wave-guide.<sup>5</sup> Increasing the voltage on the two electrodes increases the extent of the electric field, further increasing confinement. As the constriction narrows, the 1D sub-band spacing increases, which depletes the sub-bands of electrons as they are pushed up through the Fermi level. Depopulation of the 1D sub-bands occurs in steps, resulting in a quantised conductance  $G = i\frac{2e^2}{h}$ , where *i* denotes the *i*<sup>th</sup> sub-band. This expression arises as each sub-band contributes a current of  $I_d = \frac{2e^2}{h}V_{sd}$  to conduction. More generally,  $G = \frac{2e^2}{h}\sum T_i$  where  $T_i$  is the transmission probability, between 0 and 1, for the *i*<sup>th</sup> sub-band through the QPC. This accounts for quasi-ballistic transport through the constriction, which consists of ballistic motion between very few scattering centres.<sup>5</sup> Systems with a 1D confining potential are known more generally as quantum wires, irrespective of the transport regime. Quasi-ballistic and diffusive 1D transport are of relevance for self-assembled semiconductor nanowires discussed in Sec. 1.5.4.

### 1.4.3 0D systems; quantum dots and billiards

Confinement in all three directions produces a quasi-zero dimensional system, usually referred to as a quantum dot. A common realisation of a quantum dot consists of an 'island' isolated from the source and drain leads by potential barriers. Electron transport through the system proceeds by tunnelling through the source barrier into discrete energy levels within the dot, and then subsequent tunnelling through the drain barrier. Two distinct effects contribute to the electrical behaviour of quantum dots. These are energy quantisation due to confinement and charge quantisation of the electron.<sup>63,77,78</sup> The former is analogous to the discrete energy levels that arise in square or parabolic potential wells and in atoms. Electrons within quantum dots behave somewhat like electrons in atoms in that they occupy degenerate 0D energy levels.<sup>64,76</sup> The second effect is known as Coulomb blockade, whereby the repulsive charge of electrons on the dot prevents subsequent electrons from tunnelling through the source barrier into the dot. Coulomb

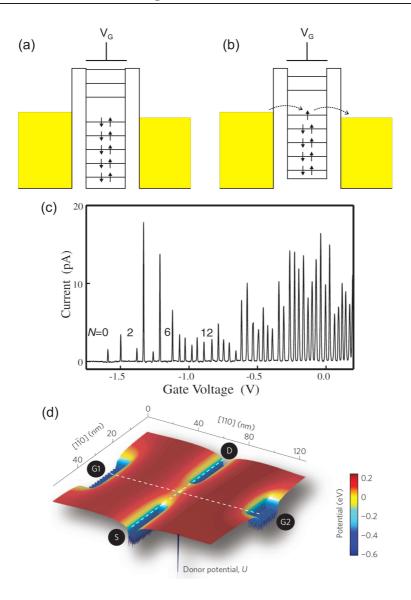


FIGURE 1.12: (a) Conduction band profile for a quantum dot with confinement in all three directions. (b) Transport takes place only when  $V_G$  is adjusted such that a discrete energy level is between the source and drain potentials. (c) A continuous sweep of  $V_G$  adds one electron at a time to the dot. The current resonances correspond to the situation in (b), and  $I_d = 0$  for the situation in (a). (d) Two dimensional potential plot for a device where a single phosphorous donor in Si acts as a quantum dot. Electrons tunnel through the barriers from source (S) into the potential well and out to the drain (D), while gates G1 and G2 are used to modify occupation. Data in (c) from Ref. 76, image in (d) from Ref. 6

blockade is a capacitive charging effect, and thus the energy required to overcome blockade is  $e^2/C_{dot}$  where  $C_{dot}$  is the capacitance of the dot. The dot geometry determines the relative energy scales for the two effects. For quantum dots defined in epitaxial GaAs heterostructures, the capacative charging energy associated with Coulomb blockade is often an order of magnitude *larger* than the separation of the energy levels associated with confinement.<sup>63,76-78</sup>

Figure 1.12(a) shows the spectrum of a quantum dot confined between two barriers that separate it from the source and drain on either side. The spectrum consists two contributions.<sup>63,76,77</sup> The first is the confinement in all three directions produces discrete, degenerate energy levels. The degeneracy arises from both spin, and symmetry effects that depend on the dot geometry. Secondly, the much larger energy separation near the source/drain potentials, above the highest occupied level, is due to Coulomb blockade. For the electron to have sufficient energy to tunnel into the dot, an empty dot energy state must be below the source potential. This can be accomplished by lowering the energy on the dot by varying  $V_G$  on a nearby gate electrode. Increasingly positive  $V_G$  leads to the situation in Fig. 1.12(b), where the lowest unoccupied level is brought between the source and drain potentials. Transport now occurs through the system via sequential tunnelling on to the dot from the source and off the dot into the drain, resulting in current through the device. Continuing the sweep to more positive  $V_G$  brings the energy level to below the drain potential where electrons now have insufficient energy to tunnel out. This traps a single additional electron within the dot and returns the current flow to zero. Continuing this process adds electrons to the dot one by one.<sup>79</sup> Figure 1.12(c) shows that this process can be sensed as a current at the drain, which exhibits sharp resonances when electrons pass through the QD (situation in b) and is zero otherwise (situation in a). $^{76}$ 

The electrical properties of quantum dots are almost entirely characterised by the energy required to add the next electron. If the next available energy state is in the same degenerate 0D energy level, the addition energy is simply the capacitive energy required to overcome Coulomb blockade,  $\Delta E_a = \frac{e^2}{C_{dot}}$ . If the 0D degenerate energy level is filled, the next electron must occupy a higher energy level. The addition energy is then  $\Delta E_a = \frac{e^2}{C_{dot}} + \Delta E_{0D}$ , where  $\Delta E_{0D}$  is the spacing between adjacent 0D levels. This explains the different spacing between the peaks for added electron number N = 2, 6, 12 in Fig. 1.12(c); a larger peak separation occurred when the added electron must start to fill the next 0D energy level.<sup>76</sup>

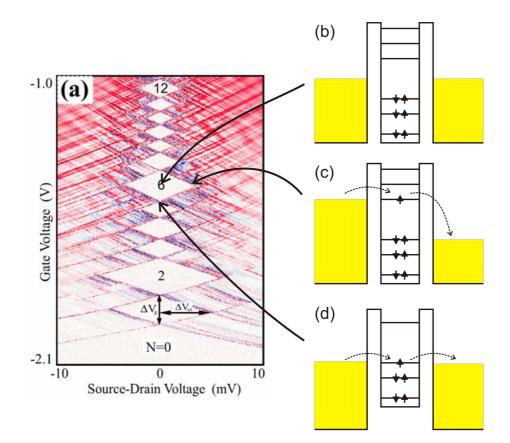


FIGURE 1.13: (a) Source-drain bias spectroscopy data. Differential conductance  $\frac{\partial I_d}{\partial V_{sd}}$  is shown in colour; white corresponds to  $\frac{\partial I_d}{\partial V_{sd}} = 0$ , red to positive  $\frac{\partial I_d}{\partial V_{sd}}$  and blue to negative  $\frac{\partial I_d}{\partial V_{sd}}$ . (b) In the central diamonds, no current passes through the dot and N electrons are on the dot. (c) At the left/right diamond vertices the source/drain potentials line up with two adjacent energy levels. Current flows for all higher  $V_{sd}$ . (d) At the top/bottom vertices of each diamond,  $V_{sd} \sim 0$  V, and is aligned with a dot energy level, allowing current to pass. A cut through the data at  $V_{sd} \sim 0$  V produces the data in Fig. 1.12(c). Data in (a) from Ref. 76

The degeneracy of the 0D levels depends on geometry. A large degeneracy can be observed above the third 0D energy level for this device. After N = 12, adding electrons consisted of simply overcoming Coulomb blockade.

Source-drain bias spectroscopy (SDBS) provides a powerful technique for mapping dot energetics, outlined in Fig. 1.13.<sup>80</sup> It consists of applying a variable DC bias  $V_{sd}^{dc}$  to split the source/drain potentials, while also varying  $V_G$ . Varying both  $V_{sd}^{dc}$ and  $V_G$  and constructing a 3-dimensional plot with  $I_d$  on the z-axis produces a series of diamonds, where transport is forbidden (permitted) inside (outside) the

diamonds.<sup>76</sup> The transport regimes arise in the following way. In the middle of the diamonds there is no energy level between the source and drain potentials and transport does not occur (Fig. 1.13(b)). Increasing source-drain bias splits the source/drain potentials. Transport occurs when this supplies sufficient energy for electrons to overcome the addition energy, i.e.,  $e\Delta V_{sd}^{dc} = \Delta E_a$ , where  $\Delta V_{sd}^{dc}$  is the  $V_{sd}^{dc}$  between the centre and corner of the diamond, indicated in Fig. 1.13(a). For higher  $V_{sd}^{dc}$ , transport is always possible as there is always at least one level between the source and drain potentials (colour regions in Fig. 1.13(a)). Again, modulating  $V_G$  increases/decreases the dot energy levels with respect to the source and drain potentials (Fig. 1.13(d)). Taking a cut of  $I_d$  vs  $V_G$  along  $V_{sd} \sim 0$  reproduces the Coulomb resonances in Fig. 1.12(c). The diamond pattern arises because larger  $V_{sd}^{dc}$ increases the  $V_G$  window where transport occurs. The coloured stripes outside the diamonds highlights a complex series of sharp transitions related to the excitation spectra of the dot. This is because the source/drain potential is greater than the Coulomb blockade energy and is thus resonant with the higher-level 0D states.<sup>76</sup> The power of SDBS is that it facilitates an almost complete characterisation of the ground state energetics of the dot. Charging energy  $\Delta E_a$ , dot capacitance  $C_{dot}$ and gate capacitance  $C_G$  are all obtained from the size of the diamonds, and from these the dot's physical size can also be calculated. This is done for a quantum dot embedded in a semiconductor nanowire in Chapter 5.

Quantum dots of this type are sometimes thought of as 'artificial atoms', due to their similarity to many atomistic features.<sup>64,76</sup> The ability to precisely address and control single electrons in an atom-like system gives researchers an opportunity to explore the spin properties of single electrons and the exotic behaviours resulting from interactions in a way that is not possible with isolated atoms.<sup>64,76</sup> Additionally, the spin of single electrons confined in a quantum dot is a prime candidate for a viable, scalable quantum bit (qubit).<sup>66,67</sup> Addressing single spins relies on first breaking the spin degeneracy in a magnetic field. This has driven strong interest in materials with a high g-factor, such that two-level systems can be generated at low fields.<sup>81</sup>

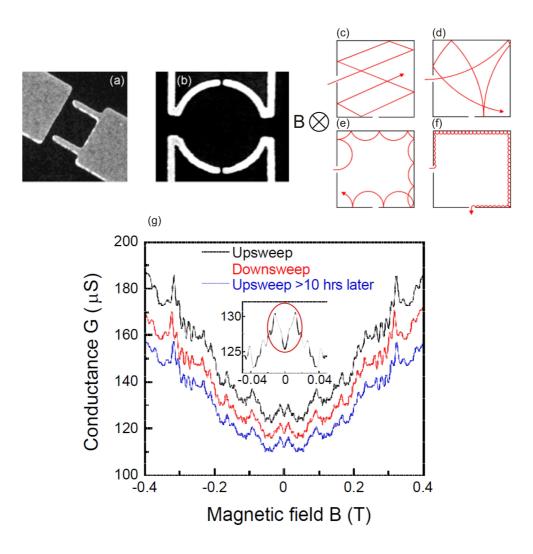


FIGURE 1.14: (a/b) Scanning electron micrographs of surface gates that define a  $1 \times 1 \mu m$  square billiard and  $1 \mu m$  diameter circular billiard on the underlying 2DEG, with entrance/exit QPCs. (c) Electron trajectories consist of ballistic paths between scattering events at the billiard walls. (d/e) The paths are curved under the influence of a magnetic field *B* perpendicular to the transport plane. (f) For high *B*, electrons undergo edge transport, consisting of skipping orbits along the billiard walls. (g) Magnetoconductance fluctuations (MCF) provide a unique fingerprint of the interference between possible electron paths through the dot. The fluctuations are reproducible for > 10 hours when the sample is kept at low temperatures (the vertical offset was introduced for clarity). Figures adapted from Ref. 82.

A second type of quasi-0D system is known as an open quantum dot, or billiard. Here, the dot is strongly coupled to the reservoir, usually by two QPCs at the entrance and exit that are set to  $G \ge 2e^2/h$  (Figs 1.14(a/b)).<sup>65,83</sup> Most billiards could also be considered a quasi-2D system, since x- and y-confinement typically exceeds  $\lambda_F$ , but is below the mean free path length. Additionally, since QPCs at these settings do not constitute tunnelling barriers, Coulomb blockade is not observed. Rather, the interest in billiards is on the trajectory of electrons in the billiard and quantum interference effects. The entrance QPC diffracts electrons, setting up multiple paths that interfere at the exit QPC.<sup>65,82,83</sup> In the ballistic case, the paths are determined by the billiard geometry as transport consists of electrons scattering off the billiard walls. The result of this interference determines the conductance G of the device. The paths can be altered by curving them with a magnetic field B which generates fluctuations in G(B) (Fig. 1.14). These magnetoconductance fluctuations (MCF) are clearly not random, or noise, since G(B) = G(-B) and they are reproducible for consecutive traces.<sup>65,82,83</sup> As such MCF provide a 'magneto-fingerprint' of the trajectories taken through the billiard.<sup>84</sup> Chaotic behaviour is observed with some geometries as the trajectories are very sensitive to the initial conditions.<sup>65,82</sup> Associated fractal behaviour has also been observed, where order on the large scale is repeated on smaller scales.<sup>82</sup>

### 1.4.4 Remote ionised impurities and quantum transport

The key assumption underlying ballistic transport studies in AlGaAs/GaAs heterostructures is that the remote ionised dopant impurities in the AlGaAs layer do not adversely impact electron trajectories. With dopants out of the way, electrons could be assumed to proceed in straight lines between confining potential walls, or the remaining large-angle scattering centres such as the inevitable unintentional impurity. This picture is only partially true. In reality, the 'tails' of the dopant potentials still cross the 2DEG, causing a random background potential on a scale smaller than the mean free path, as in Fig. 1.15.<sup>85</sup> The high mobility  $\sim 10^6 \text{ cm}^2/\text{Vs}$  typically achieved is due to the fact that the tails cause small-angle scattering, not large-angle scattering (c.f. Fig. 1.15(a)).<sup>86,87</sup> Large-angle scatterers make a much larger contribution to mobility degradation since by definition they knock electrons away from the drift direction. Nevertheless, small-angle scattering must take place between large-angle scattering events in 2DEGs, knocking electrons from their straight line trajectories (Figs 1.15(d/e)).

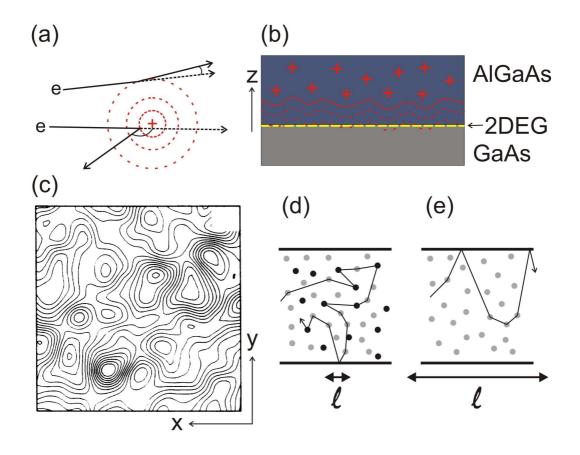


FIGURE 1.15: (a) Scattering from an ionised impurity potential. Red dashed lines represent potential contours, while solid black lines represent the electron trajectory and dashed black lines represent the path without scattering. Largeangle scattering from head on collisions impedes mobility more than smallangle scattering from glancing collisions. (b) The random placement of ionised impurities in the AlGaAs layer generates a random background potential which crosses the 2DEG. (c) Calculated non-uniform background impurity potential at the 2DEG for a 0.5  $\mu$ m<sup>2</sup> section (from Ref. 85). (d/e) Electron trajectory is altered by both large-angle and small-angle scattering sites (black/grey dots) for diffusive transport. (e) In the ballistic limit where device dimensions are less than the mean free path  $\ell$ , electrons still undergo small-angle scattering.

The question is, how does small-angle scattering affect the quantum transport studies mentioned above? Fortunately, the influence is likely to be negligible for many systems. In particular, transport through QPCs is not affected by *forward* scattering as this just causes the electrons to scatter between sub-bands; it doesn't change the number of occupied sub-bands.<sup>5</sup> However, probing the paths electrons take after leaving a QPC has shown that paths of electrons in a 2DEG *are* sensitive to small-angle scattering.<sup>88,89</sup> This is likely to strongly impact MCF in open quantum dots/billiards, as they are entirely dependent on electron trajectories.

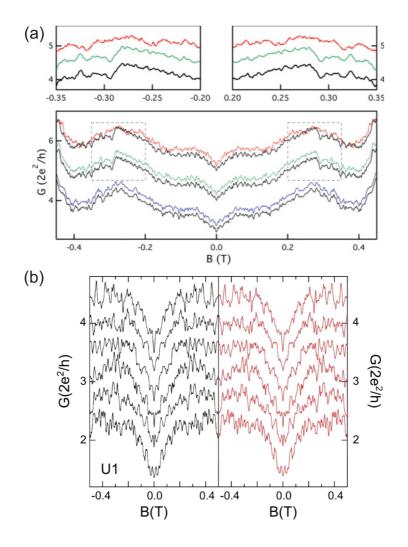


FIGURE 1.16: (a) MCF in a billiard populated by modulation doping. Traces taken before (black) and after cycling to  $T_i = 240$  mK (blue), 115 K (green), and 300 K (red). MCF are robust unless  $k_BT$  exceeds the donor trap depth. Traces offset vertically for clarity (from Ref. 91). (b) MCF for a billiard populated electrostatically, without modulation doping, before (left/black) and after (right/red) cycling the device to  $T_i = 300$  mK. Each trace is at a different topgate voltage  $V_G$ . This demonstrates that removing the dopants results in robust MCF at a range of electron densities. From Ref. 90.

The impact of remote ionised impurities was considered initially, but thought to not adversely impact the essential physics.<sup>65</sup> Since then, a compelling body of evidence has steadily built for the notion that impurity distribution *does* play a large part in determining MCF in billiards (for a review, see Ref. 84). Definitive confirmation of the importance of small-angle scattering in billiards has come recently in two related works, which I turn to now.<sup>90,91</sup>

It is a well known, but rarely acknowledged fact that while MCF traces are reproducible while the device is held at low T in the cryostat, the MCF are not reproducible upon thermally cycling the device to room temperature and then cooling back down.<sup>65</sup> This suggests that something changes in the device that causes changes in the scattering potential. To probe the source of this change, Scannell et al.<sup>91</sup> measured initial trace  $G_1(B)$  at 230 mK, held the device at intermediate temperature between  $T_i = 230$  mK and 300 K for 30 minutes before cooling to T = 230 mK and measuring final trace  $G_2(B)$ . They then examined the correlation between traces  $G_1(B)$  and  $G_2(B)$  using an autocorrelation function that measured the difference between traces  $G_1(B) - G_2(B)$  and normalised it to values between 0 (no correlation) and 1 (complete correlation).<sup>92</sup> The key results are shown in Fig. 1.16(a); for low  $T_i \leq 115$  K,  $G_1(B)$  and  $G_2(B)$  were found to be identical. Around T = 165 K, traces began to significantly decorrelate, and correlation was destroyed above T = 200 K. This is precisely the point where Si donors in the AlGaAs layer are known to de-trap.<sup>93,94</sup> The conclusion is that thermal cycling to  $T_i \ge 150$  K caused the charge configuration in the dopant layer to change, and the subsequent background potential to change. The related change in MCF gives strong evidence that small-angle scattering from remote impurities impacts the magnetofingerprint of the billiard. A follow-up work<sup>90</sup> performed the same experiment but for a device without modulation doping. The MCF in this device were robust to thermal cycling all the way up to room temperature (see Fig. 1.16(b)), further confirming dopants play a major role in determining electron paths through the dot. Turning this result around, the decorrelation of MCF in billiards has the potential to provide information on changes in remote ionised impurity configuration. In Chapter 3 I use the MCF of a hole billiard to probe dopant dynamics in p-type AlGaAs/GaAs heterostructures, in an effort to understand instability in these wafers.

## 1.4.5 Instability in *p*-type GaAs/AlGaAs heterostructures

Almost all of the studies mentioned above involved electron devices. This is not for a lack of interest in holes; on the contrary, holes in GaAs possess a number of interesting features. For example, their reduced spin-decoherence times compared to electrons makes them of great interest for quantum computation applications.<sup>95–98</sup> The presence of strong spin-orbit interactions also gives holes in GaAs a spin- $\frac{3}{2}$ nature which leads to geometrical anisotropies in spin-splitting,<sup>99,100</sup> and lends insight into other novel quantum phenomena.<sup>101–103</sup> Wafer growth is not the limiting factor either; although it was predicted to be non-trivial to grow p-type AlGaAs/-GaAs heterostructures,<sup>104</sup> it is now routinely done. In fact, the amphoteric nature of Si in AlGaAs facilitates the side-by-side growth of nominally identical p-type and n-type heterostructures. Si settles on Ga sites for (100)-oriented growth, and on As sites for (311)A-oriented growth, leaving the former *n*-doped and the latter p-doped.<sup>105</sup> This means that simply placing (100)-oriented and (311)A-oriented wafers side-by-side in the growth chamber gives matched n- and p-modulation doped structures.<sup>24,105</sup> C-doping has also been utilised as an alternative method for crystal growth on lower-index planes (e.g. (100) and (110)).<sup>106–108</sup>

The limiting factor for p-type heterostructures has been device instability under gating. In electron devices, using gates to confine electrons to quantum systems as discussed above is in general very robust. The devices operate without hysteresis, allowing very reliable measurements across long time periods. By contrast, gates on p-type heterostructures are plagued with instability and hysteresis for both Si- and C-doping. Some one- and zero-dimensional systems have been realised by using alternative etching or oxidation<sup>106,109</sup> techniques, or by working within the limits provided by gate instability.<sup>108,110</sup> However, data from these devices was generally more noisy and prone to drift than that from electron devices (see Fig. 1.17). That this is some of the best data from the literature highlights the significant problems faced in constructing hole quantum systems from modulation doped AlGaAs/GaAs heterostructures.

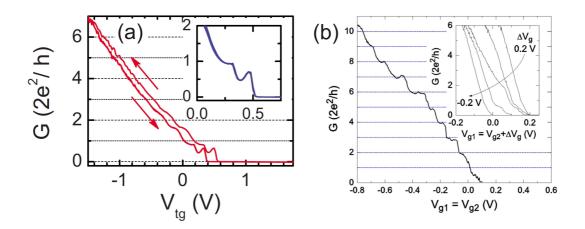


FIGURE 1.17: (a) Data from a hole QPC generated by two side gates at T = 100 mK (Ref. 108). The occupation is varied *via* a voltage applied to a separate top-gate,  $V_{tg}$ . Conductance steps are poorly defined and the device suffers hysteresis. Inset: Hystersis is reduced with a reduced operating range. (b) Data from a hole QPC defined by local anodic oxidation, with  $V_g$  applied simultaneously to two split gates,  $V_{g1} = V_{g2}$  at T = 50 mK. It is not clear if hysteresis was present, however there is more noise than would be expected from an electron device. From Ref. 109.

The data in Fig. 1.18(a) gives an indication of the magnitude of the problem. This device consisted of a Si-doped p-type AlGaAs/GaAs heterostructure with a top gate covering the entire Hall bar area to modulate the 2DHG density.<sup>24</sup> Sweeping the device to positive voltages saw the device go through a region where the gate was ineffective, before depletion could be completed. If the gate was stopped, the device would relax towards higher  $I_d$ . The sweep toward  $V_G = 0$  V saw  $I_d$  running at higher values. Similar results have been seen by other groups. The relaxation of  $I_d$  in particular destroys the hope for maintaining stable quantum systems, as the gates cannot provide a uniform, consistent, stable confining potential.

Prior work by our group suggested that both dopant impurities and surface states play a role in the instability of *p*-type heterostructures.<sup>24</sup> The work in Burke *et*  $al.^{24}$  was originally spurred by suggestions that charge carriers leaking into the heterostructure from the gates was the cause of the hysteresis.<sup>106,108</sup> The best check for this hypothesis is to fabricate devices with an insulating layer between the GaAs cap and the gate. Doing so did not fix the hysteresis; Fig. 1.18(b) shows that insulating the gates in fact worsened the hysteresis.<sup>24</sup> The fact that a surface modification produced such a clear change strongly suggests a contribution from

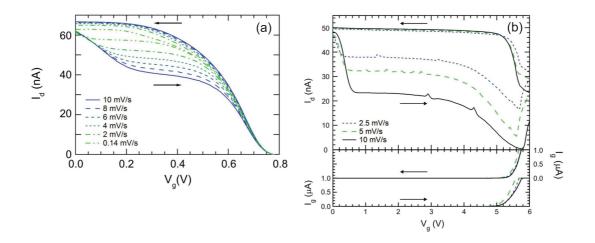


FIGURE 1.18:  $I_d$  vs  $V_g$  for HHMTs with (a) uninsulated and (b) insulated surface gates at T = 4.2 K. The hysteresis direction and sweep rate dependence is consistent with charge trapping/migration between the gate and 2DHG. The degraded performance in (b) is likely due to the high density of surface states at the GaAs/Al<sub>2</sub>O<sub>3</sub> interface. The device could not be completely depleted before current  $I_g$  leaked from the gate into the 2DHG. (c) Optical micrograph of a HHMT with uninsulated gates. Data in (a/b) from Ref. 24.

surface states. Long-lived charge traps between the 2DHG and the gate would explain the data in Fig. 1.18 (c.f. discussion in Sec. 1.3.2). Building on this, I applied surface passivation solutions underneath the (uninsulated) gates. The focus was on the widely used sulfur-based solutions that have proven effective for (100)-oriented GaAs surfaces. While this did cause nominal changes to device behaviour, there was only one device where hysteresis was appreciably reduced.<sup>24</sup> This called into question the efficacy of sulfur-based solutions for (311)A GaAs surfaces. In Chapter 3 I present a focussed study of the electrical behaviour of HHMTs treated with sulfur passivation solutions prior to gate deposition. I also used photoluminesence and XPS as independent characterisation tools to examine the chemistry and physics of the passivation solutions.

As part of the work in Burke *et al.*<sup>24</sup> we considered other potential explanations for the hysteresis. In particular, instability in the doping layer could lead to instability of the 2DHG if the distribution of ionised impurities is affected by the gate potential. The stability of *n*-type Si dopants in AlGaAs is in part due to incorporation of deep traps known as DX centres. These arise from lattice distortions caused when the Si dopant settles on the Ga site. The form of the potential of

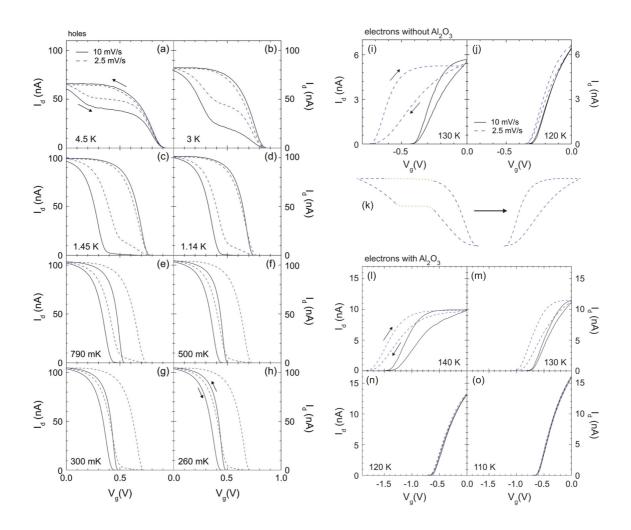


FIGURE 1.19: Temperature dependence for hole (left) and electron (right) modulation doped AlGaAs/GaAs heterostructure transistors. (a-h) Hysteresis is reduced for hole devices as temperature drops from T = 4.5 K to 260 mK. (i/j) Hysteresis was introduced by measuring electron devices at T > 130 K, where the gate causes charge migration between dopant sites in the AlGaAs layer. (k) The electron data looks similar to the hole data, except for the plateaus at intermediate  $V_g$ . These may be due to surface state differences between the (311)A and (100) surfaces, while the remaining hysteresis arises from charge migration in the dopant layer. The gates on uninsulated electron devices leaked above T = 130 K. (l-o) Electron devices with insulated gates allowed measurement up to T = 140 K to confirm the effect. From Ref. 24.

DX centres is unusual in that it features a barrier to electron trapping.<sup>93,111,112</sup> This barrier, combined with the depth  $E_T = 250$  meV makes it very unlikely that electrons will migrate between trapping sites at  $T \leq 4.2$  K. By contrast, little is known about Si-acceptors in AlGaAs; previous evidence suggests they form shallow potential wells without a trapping barrier.<sup>113</sup> In this case, the acceptors may

be active even at T = 4.2 K and trapped holes may be able to migrate by hopping between acceptor sites in response to an external potential. Since hopping is thermally activated, the hysteresis should have a temperature dependence. We performed temperature dependent studies of both p-type and n-type heterostructure devices to test whether dopant migration contributed to hysteresis, shown in Fig. 1.19. In the *n*-type devices, we saw hysteresis emerge for  $T \ge 130$  K, where DX centres are known to begin to de-trap charge,<sup>91,93,94,114</sup> similar to results from semiconductor billiards in Sec. 1.4.4. Note that the apparent activation temperature in this experiment was  $\sim 20 - 30$  K lower than for the billiards, because the gate potential constitutes a strong perturbation that causes charge migration between dopant sites. By contrast, the billiard experiments relied on temperature-induced spontaneous re-organisation alone. Returning to Fig. 1.19, the form and temperature dependence of the hysteresis in n-type devices above T = 130 K mirrored that of p-type devices below T = 1 K. This suggests strongly that the shallow acceptor traps are active at low temperatures and contribute to the hysteresis.<sup>24</sup> In Chapter 3 I present data related to MCF in a hole billiard fabricated from the p-AlGaAs/GaAs wafer. I follow the methodology of Scannell et al.<sup>91</sup> to estimate the potential well depth  $E_A$  for Si acceptors in Al<sub>0.33</sub>Ga<sub>0.67</sub>As, using a measured activation temperature  $\sim 2$  K. I also show that the gate altered the charge configuration in the dopant layer even at T = 40 mK, well below the activation temperature.

### 1.5 Self-assembled semiconductor nanowires

### 1.5.1 Motivation

Si-based microelectronics has been a hugely successful technology platform in the manufacture of integrated circuits, solar cells and LEDs. The wide use of Si comes down to two factors: Firstly, the natural oxide,  $SiO_2$ , produces a very clean interface with Si that features a low defect density and the lack of a comparable oxide for III-V semiconductors has hampered device development in this area (as

discussed in Sec. 1.3.2).<sup>4,39,40,49</sup> Secondly, Si is relatively low cost and plentiful in supply compared to III-Vs, making it more economically viable.<sup>49</sup> However, Si has properties that restrict the capacity for devices with increased energy efficiency: its low electron/hole mobility imposes restrictions on transistor switching speed and  $I_{on}$  and its indirect band gap limits efficient use in opto-electronic applications such as LEDs and solar cells. Instead there is a strong desire to use III-Vs, which generally feature higher electron/hole mobilities and band gap tunability through alloying.<sup>4,49</sup> Although many prototype devices have been made, there is an economic hurdle for commercial use of III-Vs. The fabrication of integrated circuits proceeds from selectively etching Si wafers to create islands that form the basis for individual transistor channels. The relatively thick underlying wafer acts as a physical support for the overlayed devices. It is simply not economically viable to grow mm-thick wafers of e.g. GaAs and then utilise only the top 10-100 nm of material for device fabrication.<sup>3</sup> A compromise would be to grow a mm-thick Si wafer and then thin layers of III-V's on top, but this can only be done directly if the III-V material has the same lattice constant as Si, limiting the choice of materials according to Fig. 1.9(e). Two solutions to this problem are the growth of buffer layers that make the transition between the Si and chosen III-V lattice constants either gradually to prevent dislocations<sup>12</sup> or via intentionally introduce dislocations,<sup>115</sup> and wafer-bonding techniques that use van der Waals forces to join separately grown wafers.<sup>116,117</sup> However, self-assembled nanowires facilitate crystalline III-V material growth directly on Si substrates. This has opened up other possibilities for device applications and quantum system exploration.

### 1.5.2 Growth principles and advantages

Self-assembled semiconductor nanowires were first realised using particle-assisted, vapour-liquid-solid techniques,<sup>119–123</sup> such as metal-organic chemical vapour deposition (MOCVD). To grow nanowires by MOCVD, Au nano-particles are spread across a substrate surface (Fig. 1.20(a)); they act as a catalyst for nanowire growth. The substrates are placed in a high temperature growth chamber ( $T \sim$ 

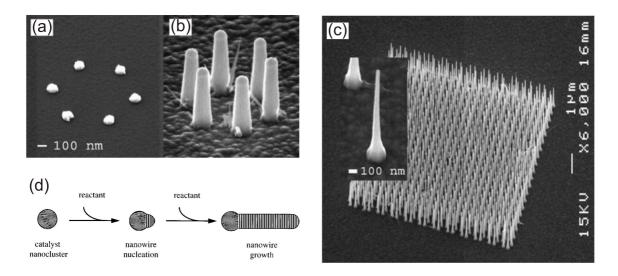


FIGURE 1.20: Scanning electron micrographs of (a) 100 nm diameter Au particles prior to nanowire growth, (b) as-grown nanowires from the particles in (a) and (c) a larger array of nanowires. From Ref. 118 (d) Nanowires grow preferentially underneath the catalyst, after an initial nucleation process. From Ref. 119

 $500 - 700^{\circ}$ C) and exposed to a flow of precursor gas. For InAs, the gases are trimethylindium and arsine, which facilitate atomic In and As incorporation into the liquid Au droplet. This essentially forms a tiny melt, where In and As crystallise into InAs beneath the droplet (Fig. 1.20(d)).<sup>124</sup> Progressively more InAs material grows under the Au particle, ultimately forming a nanowire where the size of the Au particle determines the nanowire diameter – typically 20 - 200 nm – and the growth time determines the length – typically 100 nm - 20  $\mu$ m. The Au particles can be patterned before nanowire growth to facilitate high density arrays (e.g. Fig. 1.20(b/c)).<sup>118</sup> An number of methods have since been developed to grow very high quality nanowire arrays using, e.g., molecular beam epitaxy without the need for catalyst particles.<sup>125,126</sup>

The biggest advantage of the self-assembly process is that the very high aspect ratio means the III-V material is not constrained to a substrate and can tolerate a high strain relaxation without defects.<sup>127,128</sup> As such, III-V crystals can be grown directly on Si substrates with minimal use of expensive materials.<sup>129–131</sup> The high strain relaxation also significantly reduces lattice matching constraints on the nanowire itself, which has facilitated growth of a large range of defectfree heterostructure combinations not possible for planar structures, e.g. Si/Ge, GaAs/InAs. This can be done in both the radial<sup>132</sup> and axial<sup>133</sup> directions, or even a combination of the two<sup>134</sup> to facilitate new device functionality. This has opened up a wealth of possibilities for band-engineering that has been exploited for quantum system generation, novel transistor devices and optoelectronic devices.<sup>124,135</sup> In particular, nanowire solar cells and LEDs are already being considered for commercially viable technologies.<sup>136–138</sup>

### 1.5.3 Nanowire FETs

The geometry of nanowires offers a particular advantage towards FET development; they can be used as the basis for a 'gate-all-around', or 'wrap-gated' transistor. A gate that wraps around the transistor channel can apply the electric field symmetrically from all sides, in contrast to uni-directional planar devices. This helps to maintain a low SS and  $I_{off}$  in short channel devices.<sup>4</sup> Wrap-gated nanowire FETs (NWFETs) were first realised with as-grown nanowires standing vertically on the growth substrate, as in Figs 1.21(a/b).<sup>23,139,140</sup> Optimisation of this structure for InAs nanowires has lead to switching operation at frequencies well into the GHz range.<sup>141,142</sup> While the vertical configuration is the choice for efforts towards industrial deployment, it is time consuming to contact and study individual nanowires in this way. For fundamental research purposes, it is actually more more convenient to break the nanowires off the growth substrate and deposit them horizontally on a secondary substrate.

One immediate advantage to using the horizontal/lateral orientation is that the substrate itself can be used as a global gate, without the need for a wrap-gate.<sup>144</sup> Typically the substrate is highly doped Si, insulated from the nanowire by a SiO<sub>2</sub> layer, and  $V_G$  is applied directly to the substrate (Fig. 1.21(c)). The simple fabrication makes substrate gating a popular choice for fundamental research, whether it be a simple materials characterisation or probing embedded quantum dots. In many cases a substrate gate can be sufficient, but in general it is not an ideal

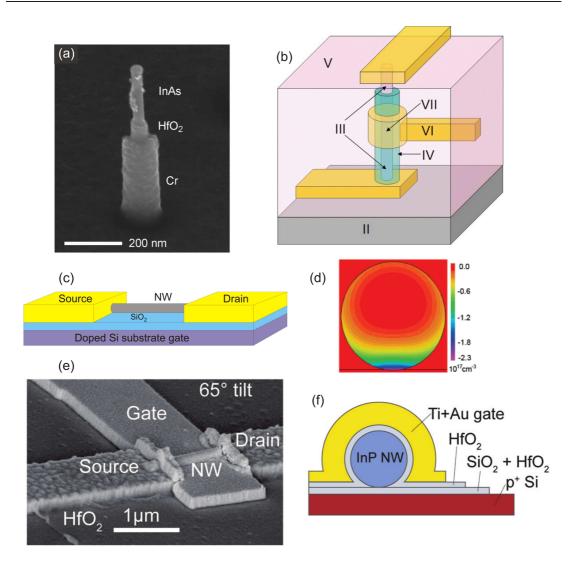


FIGURE 1.21: (a) Scanning electron micrograph of vertical InAs nanowire with Cr wrap-gate insulated by HfO<sub>2</sub>. (b) Schematic of completed vertical wrap-gate nanowire transistor with: II; Si substrate, III; ohmic contact to source/drain, IV; HfO<sub>2</sub> layer, V; polymer filling layer to facilitate deposition of top contact, and VI; Cr wrap gate. (c) Schematic of substrate gated lateral NWFET. (d) Cross-section of non-uniform electron density (colour scale) in a substrate gated NWFET. (d) Scanning electron micrograph and (e) schematic cross-section of an omega gated lateral NWFET. (a/b) from Ref. 23, (d) from Ref. 143, (e/f) from Ref. 19.

strategy. First, there is no local control over individual nanowires on the same substrate, let alone the possibility of multiple gates on the same nanowire. This is essential to increase transistor density and generate and manipulate quantum systems with high precision.<sup>135,145</sup> Secondly, the low dielectric constant of SiO<sub>2</sub>, combined with the fact that it is usually 100 nm thick to insure against break-down, means large voltages  $\sim 10$  V or more are required to fully deplete the

nanowire.<sup>22,30,145,146</sup> The accompanying large subthreshold swings of  $SS \sim 1V/dec$  are clearly not optimal for efficient, low power operation. Thirdly, gating a device with radial symmetry from only one side invariably introduces a non-symmetrical potential and therefore non-symmetrical depletion/enhancement of charge carriers.<sup>143</sup> Figure 1.21(d) shows that the charge carriers are pushed against the rough, unevenly charged nanowire surface on one side, reducing mobility and presenting a non-uniform system.

Local gating can be achieved by back gates patterned prior to nanowire deposition, or post-deposition patterned 'omega' gates on top of the nanowire (Figs 1.21(e/f)). Omega gates give much improved gating – in some cases with subthreshold swings approaching the thermal  $limit^{19,147}$  – and the local patternability of both backand omega-gates has facilitated quantum system engineering.<sup>148,149</sup> However, the fact that both approaches still yield a non-uniform radial potential highlights a desire for lateral wrap-gating. Despite this, realising lateral wrap-gated nanowires proved elusive as it was not clear how to easily fabricate a wrap-gate in this orientation.<sup>145</sup> I will discuss this in more depth in Chapter 2, but the eventual fabrication required an innovative use of chemical etchants in work by Kristian Storm and others from Lund University, Sweden collaborating with my supervisor Adam Micolich.<sup>145</sup> Stepping this up to multiple lateral wrap-gates entailed a nontrivial re-think of the entire fabrication process, which I worked on during my research for this thesis. In doing so, we demonstrated the first laterally-oriented nanowire FETs with multiple wrap-gates.<sup>29</sup> We also showed that laterally-oriented devices with multiple wrap-gates may have scalability advantages over vertical structures because once the initial step is made from one to two gates, adding more doesn't entail additional fabrication steps. This means it takes the same time to fabricate two, three or four wrap-gates. By contrast, each gate added to a vertical NWFET entails a repetition of fabrication steps, which adds a time and yield cost. This is discussed in detail in Chapter 4.

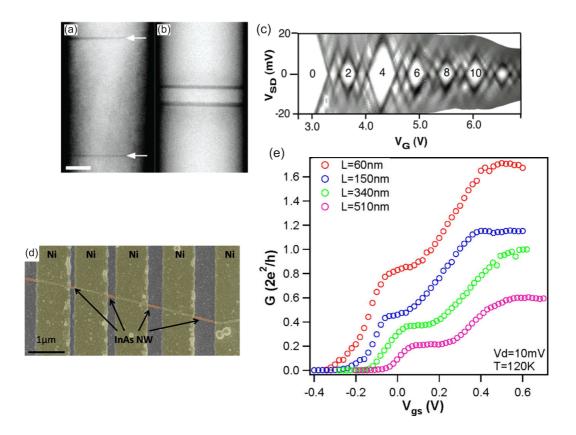


FIGURE 1.22: (a/b) Dark-field scanning transmission electron microscopy image along the axis of an InAs nanowire with ~ 3–4 nm thick InP heterostructure barriers (dark regions). The barriers define InAs quantum dots of 100 nm and 20 nm length. The scale bar depicts 20 nm. (c) Source-drain bias spectroscopy data for a 10 nm dot with  $V_G$  applied to the substrate gate. From Ref. 150. (d) InAs nanowire with multiple ohmic contacts defining different channel lengths. (e) Quasi-ballistic transport for different lengths defined in the device in (d). A small amount of scattering events disrupt ballistic transport, reducing the transmission co-efficient and leading to steps  $G < 2e^2/h$ . The step height increases as the channel length approaches the mean free path length. From Ref. 151.

### 1.5.4 Quantum systems in nanowires

The radial confinement of nanowires and the ability to grow heterostructure barriers resulted in the very swift realisation of embedded quantum dots based on, e.g., InP barriers in an InAs nanowire. This included single-electron transistors, few-electron quantum dots and double dots with signatures observable up to T = 60 K.<sup>150,152,153</sup> The ability to easily define dots in materials like InAs and InSb makes nanowires of great interest in addressing electron spins for quantum

computing applications due to the comparatively large effective g-factor  $g^*$  of electrons in these materials. The effective g-factor is the constant of proportionality for the magnitude of spin-splitting in a magnetic field of otherwise degenerate energy states.<sup>63</sup> Addressing individual spins with high accuracy at low B is aided considerably by materials by a large g-factor, e.g.,  $g^* = -14.7$  for bulk InAs. Additionally, the g-factor can be tuned with quantum dot size<sup>154</sup> and occupation level.<sup>81,155</sup> This opens up the possibility for multiple individually addressable dots on the same nanowire by selecting different g-factors. For quantum dots in InSb nanowires, g-factors of up to 70 have been observed.<sup>81</sup>

By contrast, ballistic 1D transport in nanowires has been elusive despite the natural radial confinement on the order of 20 - 100 nm. Initial indications suggested a Coulomb staircase superimposed on a disordered background,<sup>156</sup> with recent measurements showing this background could be partially overcome by suppressing backscattering via an applied magnetic field.<sup>157</sup> Similarly, there exists evidence for diffusive,<sup>158,159</sup> and quasi-ballistic 1D sub-bands<sup>151,160</sup> – i.e. carriers undergo scattering events – that with high quality growth, surface passivation and very short channel length  $\sim 60$  nm can approach the ballistic limit.<sup>151</sup> These results in Fig. 1.22 are also promising in that sub-band splitting can be observed at relatively high temperatures  $T \sim 120$  K. Nevertheless, manipulation of a truly robust ballistic 1D system has not been convincingly achieved in nanowires to date. The continued drive towards 1D ballistic transport in nanowires is due to some interesting potential applications in which nanowires are predicted to play a pivotal role. One example is the search for Majorana fermions, particles that are their own anti-particle. They are predicted to arise in a very specific solid-state system that consists of a superconducting contact to one half of a nanowire with underlying 1D transport.<sup>161,162</sup> A second example, there is a long-standing prediction that the figure-of-merit for thermoelectric power conversion should be significantly enhanced by 1D transport in nanowires;<sup>163,164</sup> this latter application is relevant for this thesis and I will discuss this further in the section below.

The problem for ballistic transport in nanowires is that the surface necessarily defines the radial confinement. Since the surface states provide a non-uniform background potential, the confinement is not uniform along the channel length. This non-uniform potential is superimposed on any additional gate potential, meaning the potential is nothing like the ideal smooth, uniform barrier that steadily confines charge carriers. In addition, the electron mobility in nanowires is typically less than  $10^4 \text{ cm}^2/\text{Vs}$ , and mean free paths are expected to be 200 nm or less.<sup>151,157,165</sup> Compare this situation to QPCs in GaAs/AlGaAs heterostructures, where the well defined surface gates can provide a very uniform confining potential to a system with a mobility of  $\sim 10^6 \text{ cm}^2/\text{Vs}$  and mean free paths over  $10 \ \mu m$ . Observing true 1D ballistic transport in single crystal nanowires is likely to require fabrication of gates with lengths  $\leq 200$  nm that are located in a region where the surface potential is uniform. A possible alternative is modulation doped nanowires, with a GaAs core and doped AlGaAs shell.<sup>147,166</sup> This would ideally provide the high electron mobility and background potential required for ballistic 1D transport.

### 1.5.5 Thermoelectrics and nanowires

The direct conversion of heat to electricity and vice versa *via* the Seebeck and Peltier effects has long held interest for applications such as generating electricity from waste heat or solid-state cooling.<sup>167–169</sup> The basic principle is that a thermal gradient  $\Delta T$  causes charge carrier diffusion from hot to cold, which establishes an open circuit voltage called the thermovoltage  $V_T$ .<sup>1</sup> The Seebeck coefficient  $S \equiv -V_T/\Delta T$ , also known as thermopower, links these two essential properties to quantify the magnitude of the thermoelectric effect. In metals S is only tens of millivolts per Kelvin as diffusion quickly cancels out the carrier imbalance that defines  $V_T$ .<sup>169</sup> However, insulators are not an effective solution either since extracting usable power from the device requires that a current flows. In fact, optimising thermoelectric efficiency is a balancing act governed by the thermoelectric figure

<sup>&</sup>lt;sup>1</sup>I have used  $V_T$  for thermovoltage rather than the more commonly used  $V_{th}$  to avoid confusion with threshold voltage.

of merit  $ZT = S^2 \sigma T/\kappa$ , where  $\sigma$  is the charge carrier conductivity, T the average temperature of the sample and  $\kappa$  the thermal conductivity.<sup>170</sup> The thermal conductivity of solids consists of phonon and charge carrier (electron/hole) contribution  $\kappa = \kappa_{ph} + \kappa_{eh}$ . The balancing act arises because  $S, \sigma$  and  $\kappa$  are almost always interlinked. S and  $\sigma$  both need to be high to extract power from the device, but as illustrated by the comparison between metals and insulators, S and  $\sigma$  are mostly inversely proportional to each other. Similarly,  $\kappa$  should be low such that the thermal gradient  $\Delta T$  is not equalised across the material. However, a high carrier conductivity  $\sigma$  tends to result in a high carrier thermal conductivity  $\kappa_{eh}$  by the Wiedemann–Franz law,  $\frac{\kappa}{\sigma} = LT$ , where L is the Lorenz number (a constant). The difficulty in maximising ZT in bulk materials generally results in ZT < 1at room temperature. Values of 3 of more are considered to be necessary for a thermoelectric material to provide efficient conversion in practical applications.<sup>169</sup>

At the fundamental research level, nano-structures, and in particular nanowires have been a strong focus in obtaining a useful ZT > 3. This is for two reasons: Firstly, heat flow is disrupted by the proximity of the rough surface, giving a much reduced  $\kappa$  that is largely independent of S and  $\sigma$ .<sup>172</sup> Secondly, 1D quantum confinement is predicted to result in increased  $S^2\sigma$  for nanowires based on tellurides and III-V semiconductors.<sup>163,164</sup> This relies on operating the device just below the lowest sub-band, where transport only takes place from the hot to cold reservoir.<sup>163,173</sup> Despite this promise and a handful of results indicating small enhancements,<sup>159,174</sup> significant ZT enhancement via 1D confinement in nanowires has been frustrated by surface disorder prohibiting clear observation of 1D ballistic transport. Additionally, the requirements of very thin nanowires with diameters < 10 nm<sup>163,164</sup> and careful Fermi level control<sup>163,171,173</sup> are rarely addressed properly.

However, recent work by Wu *et al.*<sup>171</sup> has highlighted that the precise restrictions of the 1D regime are not necessarily required to achieve significant  $S^2\sigma$  enhancement. In fact, this work uses the disordered nanowire surface as an advantage,<sup>171</sup> as the disorder potential essentially consists of a number of barriers along the nanowire channel.<sup>175</sup> This means that electron transport can be thought of as a mixture of propogating modes coupled to quantum dot-like states. This was shown recently

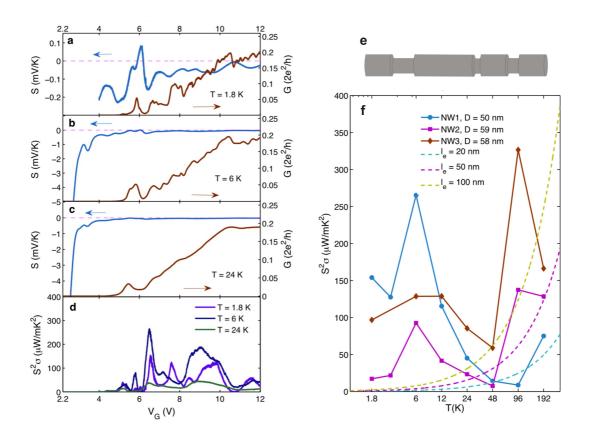


FIGURE 1.23: (a/b/c) Seebeck coefficient S and conductance G vs  $V_G$  for a substrate-gated InAs nanowire at T = 1.8, 6, 24 K. The fluctuations in G are quantum interference effects due to the coupling of disorder defined quantum dot states with propagating modes. The oscillations in S are also suggestive of quantum dot signatures, since they approximate  $\frac{\partial G}{\partial V_G}$  and undergo sign changes. (d) The oscillations in S and G produce peaks in  $S^2\sigma$  over a wide  $V_G$  range. (e) Representation of the radial confinement potential for nanowires at low temperature. (f) Peak value of  $S^2\sigma$  for three different nanowire devices at different T (dots) compared to  $S^2\sigma$  for bulk InAs (dashed lines).  $S^2\sigma$  is enhanced in nanowires for  $T \leq 24$  K. Adapted from Ref. 171.

both experimentally<sup>171</sup> and theoretically<sup>176–178</sup> to enhance  $S^2\sigma$  at T < 20 K. Figure 1.23 shows the advantage is that the enhancement occurs over a wider range of carrier concentration. This relaxes the strict requirements on Fermi level positioning demanded in the 1D case and broadens the scope for practical applications where careful positioning of the Fermi level in large device arrays is not typically feasible.<sup>171,173</sup> Although, some optimisation will be required to exploit this effect at higher temperatures.

Beyond seeking materials with high thermoelectric efficiency, thermovoltage measurements hold interest as an electrical characterisation tool. Since  $V_T$  is related

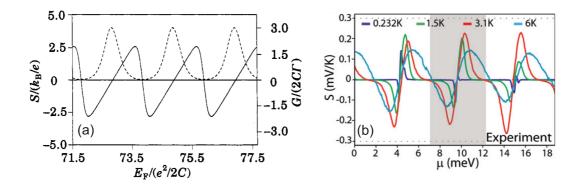


FIGURE 1.24: (a) Seebeck coefficient  $S = V_T/\Delta T$  in the sequential tunnelling regime (solid trace/left axis) and conductance G (dashed trace, right axis) as a gate voltage sweeps quantum dot states through the Fermi level. The Gresonances correspond to allowed transport through the discrete dot energy levels.  $V_T$  and S follow the energy of the closest dot level. From Ref. 179. (b) Experimental data of S vs gate potential  $\mu V_G$  at average temperature T = 0.232, 1.5, 3.1 and 6 K. Reducing temperature drives S (and therefore  $V_T$ ) towards a derivative-like lineshape. From Ref. 180.

to the average energy of carriers in a quantum system it can contains more information than conductance  $G^{179,181-184}$  For sequential tunnelling in quantum dots at  $V_{sd} \sim 0$ , the thermovoltage dependence on plunger gate voltage follows the rise of the closest dot level,  $V_T(V_G) \sim \mu_N - \mu_S$ . Here,  $\mu_S$  and  $\mu_N$  are the chemical potentials of the source contact and the dot level with energy closest to  $\mu_S$ , respectively. This produces a sawtooth function that crosses  $V_T = 0$  when the dot/contact levels align, and switches sharply back to negative  $V_T$  halfway between  $I_d$  resonances (see Fig. 1.24(a)).  $V_T$  also can contain fine structure, carrying information about the excitation spectrum of the dot that is absent from a simple G measurement.<sup>179,181,182</sup> When higher order tunnelling processes start to dominate transport,  $V_T$  makes a smooth transition towards a derivative-like lineshape  $V_T \sim \frac{1}{G} \frac{\partial G}{\partial V_G}$ , shown in Fig. 1.24(b).<sup>180,183,185</sup> Employing quantum dots or QPCs can also facilitate accurate measurement of the electron temperature, which is often distinct from the lattice temperature at low T.<sup>186–188</sup> In classical/diffusive systems,  $V_T$  can be used to distinguish between carrier types<sup>189</sup> and estimate carrier mobility and concentration.<sup>190,191</sup> This is particularly important for nanostructures where methods such as the Hall effect are problematic.<sup>192</sup>

In Chapter 5, I present results that use an organic polymer electrolyte gate dielectric to set base-line carrier density and modulate the disorder potential in an InAs nanowire for low T measurements. I show this can increase the influence of quantum dot-like states. This suggests this could be a viable method for both setting an optimal disorder potential and carrier concentration for thermoelectrics. I also use complementary G and  $V_T$  measurements to show that  $V_T$  carries extra information about multiple quantum dot systems.

## **1.6** Organic electronics and ionic transport

An exciting new direction in electronics research involves the use of organic materials. They offer a wealth of new functionalities not readily facilitated by inorganic materials such as Si and III-V semiconductors. In particular, organic devices can be deployed on flexible substrates for displays, lighting and plastic electronics.<sup>11</sup> Many of the materials feature novel, easily scalable fabrication methods such as inkjet printing.<sup>193</sup> In addition to electron/hole transport, many biological materials facilitate ionic transport. Ion transport has long been of interest for, e.g., Li-ion batteries, but is also relevant to biological applications.<sup>9-11,194</sup> Using ions in an electrolyte to facilitate gating of organic transistors and inorganic nanostructures is gaining attention where traditional metal/oxide methods are incompatible, or where the organic route provides additional functionalities.<sup>8</sup> Liquid electrolytes were first used at Bell labs to demonstrate Fermi level control in semiconductors<sup>195</sup> and continue to be employed in some specialised situations. However, polymer electrolytes in the solid or gel phase are most popular as they are more easily contained within device architecture.

#### **1.6.1** Polymer electrolytes

A polymer electrolyte (PE) consists of mobile ionic species – often derived from a salt such as  $LiClO_4$  – hosted in a polymer backbone, such as poly(ethylene oxide)

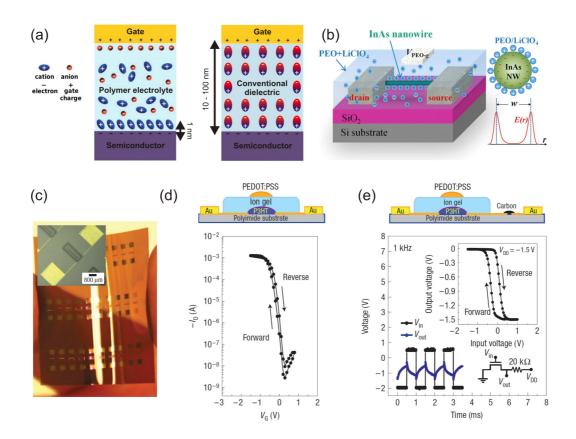


FIGURE 1.25: (a) Operation of a polymer electrolyte gate dielectric. Positive charge on the gate drives ionic motion to establish an electric double layer 1 nm away from the semiconductor channel. This contrasts with a conventional dielectric where the gate voltage is dropped over 10 - 100 nm by dipole alignment. Adapted from Ref. 8. (b) Ionic motion provides a radially symmetric potential to nanowires, facilitating wrap-gating. The PE layer covers the source/drain and entire substrate due to previous patterning resolution restrictions. From Ref. 196. (c) Flexible array of organic transistors. (d) The channel is an organic semiconductor P3HT, with an ion gel gate dielectric which facilitates a operating range  $\Delta V_G < 2$  V. (d) An inverter based on the transistor in (c) with a carbon resistor load. Switching occurs at 1 kHz, but fidelity is low. From Ref. 197.

(PEO).<sup>8,198</sup> Figure 1.25(a) shows how polymer electrolytes are operated as high capacitance gate dielectrics. Application of  $V_G$  to the gate electrode causes dissociated Li<sup>+</sup> and ClO<sub>4</sub><sup>-</sup> ions to move through the polymer *via* chain relaxation.<sup>199–202</sup> This results in electric double layers (EDL) at the channel/PE and PE/electrode interfaces. EDL formation ideally transfers the gate charge to approximately 1 nm away from the semiconductor channel, meaning  $V_G$  essentially drops across a 1 nm thick dielectric.<sup>203</sup> At this distance, the dielectric constant of an EDL becomes independent of the host material, and takes a value  $\epsilon_r \sim 10 - 20$ . The high  $\epsilon_r$ 

and low dielectric thickness produces a large capacitance and strong electric field which facilitates very effective gating.<sup>8</sup>

These features made polymer electrolytes favourable dielectrics for organic transistors. Organic surfaces typically have a high oxide content which produced nonideal surface/interface effects when in contact with oxide dielectrics, similar to the issues seen in III-V semiconductors (recall Sec. 1.3.2). The solution processability of polymer electrolytes also improved their compatibility with organic channels.<sup>8</sup> Using PEs made it possible to reduce the  $V_G$  operating range to 1 or 2 V, down from the tens of volts required using other architectures (see Figs 1.25(c/d)).<sup>197,204,205</sup> PEs were also used to induce very high carrier densities to dramatically improve device conductivity.<sup>203,206–208</sup> In some cases this was assisted by the anions passing into the organic transistor channel and doping it; an additional functionality not possible with inorganic materials. Incorporating the salt-based solid electrolytes in industrial devices was hampered by the slow switching speed.<sup>8</sup> This arises due to the fact that EDL formation takes a finite time, limited by the ionic conductivity. The relatively rigid chains in, e.g., PEO at room temperature results in a low ionic conductivity and slow switching speeds.<sup>201,202</sup> The desire to increase switching speed led to exploration of ion gel gate dielectrics, <sup>197,209–211</sup> which consist of ionic liquids such as 1-ethyl-3-methyl imidazolium bis(trifluoromethane sulfonyl)imide gelated by polymerisation of, e.g., polymethyl methacrylate (PMMA) or block co-polymers.<sup>212–214</sup> Ionic liquids on their own have a very high conductivity, and gelating them with < 10% polymer ensures orders of magnitude higher conductivity at room temperature compared to salt-doped PEO, facilitating faster switching speeds (Fig. 1.25(e)).<sup>8,215</sup> However, the low viscosity of ion gels means they are not suited to all applications.

The benefits brought by ionic transport have seen polymer electrolytes applied to nanostructures such as carbon nanotubes<sup>216–218</sup> and semiconductor nanowires.<sup>196,219</sup> Here, PEs also bring the advantage that ions will follow the surface profile of the channel. This provides a simple way to achieve a uniform potential in unusual geometries. In particular, they open a novel route for wrap-gating nanowires (see Fig. 1.25(b)).<sup>196,219</sup> In my research I furthered the use of polymer electrolytes with

nano-materials by demonstrating nanoscale patterning of a PE for the first time. Most previous devices using nanostructures have used the polymer electrolyte as a continuous thin film that covers the entire substrate, and therefore globally gates all devices on the chip. While inkjet printing,<sup>193,197</sup> photolithography<sup>211</sup> and injection into microfluidic channels<sup>217</sup> have been used to pattern PEs, all of these techniques had resolutions > 1  $\mu$ m. Broadening the scope for using PEs with nanomaterials requires nanoscale patterning to enable local control of individual devices and the possibility of multiple gates on the same device. Eliminating PE overlap of the source/drain contacts is also important to reduce parasitic behaviours.<sup>8</sup> In Chapter 4 I present a technique to pattern the LiClO<sub>4</sub>/PEO polymer electrolyte on the nanoscale using electron beam lithography. I used this to fabricate devices with multiple gates on the same nanowire. Fabrication of NWFETs with multiple polymer electrolyte gates required even less steps than the fabrication for metal/oxide wrap-gates. Additionally, the basic performance parameters, e.g., subthreshold slope were comparable. In Chapter 5 we used the PE gate as an external doping strategy for low temperature T < 4 K measurements of an InAs nanowire. In addition to setting the base-line carrier density we found that the PE gate could strongly alter the background potential and generate quantum-dot like systems.

## 1.6.2 Proton transport and biological applications

One area that organic electronic devices have found great application in is interfacing with biological systems. Biological signalling occurs *via* ions and protons, rather than the electronic signals of manufactured electronics. Interfacing with these systems requires transduction of ionic/protonic signals into electronic ones that can be read out *via* conventional electronics.<sup>9–11,194</sup> Proton signals in particular are central to a number of important biological processes.<sup>220</sup> In solid state organic materials with high water content, water molecules can form proton wires where H<sup>+</sup> hops between H<sub>2</sub>O sites in response to an external field, by the Grotthuss mechanism. In other materials they proceed by chain relaxation, as

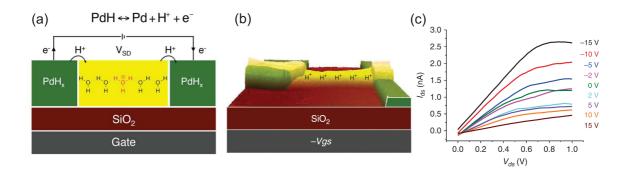


FIGURE 1.26: (a) Schematic of a protonic field-effect transistor.  $PdH_x$  contacts inject and collect  $H^+$  ions, which hop along hydrogen bonded water chains that act as proton wires. (b) Combined AFM/schematic of a proton transistor with maleic–chitosan proton channel. The backgate modulates the  $H^+$  concentration in the channel. (c) More negative  $V_G$  (colour) causes a higher proton current  $I_d$ . Adapted from Refs. 220 and 223.

for Li<sup>+</sup> in PEO.<sup>11</sup> In terms of sensing proton currents, direct proton-to-electron and electron-to-proton transduction can be facilitated by  $PdH_x$  in contact with the proton conductor.<sup>220–222</sup> Figure 1.26(a) shows that applying a bias to an external circuit causes a chemical reaction at the injection contact, such that  $PdH_x \rightarrow Pd + H^+ + e^-$ , with the proton injected into the channel and the electron entering the external circuit. The reverse occurs at the collecting contact, and a proton current is detected.

An interesting expansion of this idea is a proton-channel field-effect transistor (Figs 1.26(b/c)). It consists of a proton conducting maleic–chitosan channel between two PdH<sub>x</sub> contacts and a doped Si substrate gate that is used to electrostatically deplete or accumulate protons in the channel.<sup>220</sup> A complementary device has been constructed using OH<sup>-</sup> as 'proton holes', and the junction between the two materials behaves like a p-n junction.<sup>223</sup> In a similar vein, npn and pnp ionic bipolar junction transistors based on the organic semiconductor PEDOT:PSS and electrolyte interfaces have been fabricated.<sup>224,225</sup> These devices are aimed at selective drug delivery more than ion-to-electron transduction.

A more general method for ion/proton-to-electron transduction involves using FETs with electron/hole channels that are sensitive to changes in the ionic environment. This can occur in a couple of ways; the ions may cause a chemical

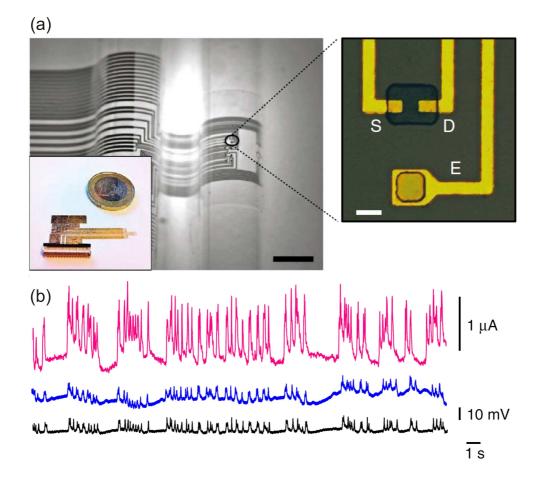


FIGURE 1.27: (a) Organic transistor channel (between source and drain S and D) and metal electrode G used to sense neuron signals. (b) Current through the channel (pink) and potential at two nearby electrodes (blue and black) responds to neuron signals in a rat brain. Signals have been normalised to the noise level, illustrating the better signal-to-noise ratio of the organic transistor. From Ref. 226.

interaction at the surface that alters the surface charge state, changing the current in the channel. These ion sensitive FETs (ISFETs) are used to detect pH changes in solution.<sup>227,228</sup> Sensing may also occur by simple field-effect, without a chemical reaction. These devices essentially operate like the polymer electrolyte gated transistors; a signal causes polarisation of ions within an electrolyte dielectric, gating the channel.<sup>194</sup> Alternatively, the biological sample may be placed directly on top of the channel. An exciting strand of this research involves engaging with neuron firings; Fig. 1.27 shows data from an organic transistor placed directly on a rat brain. The spikes correspond to neural signal events. Transistors

offer a much higher signal-to-noise ratio than more commonly used metal electrodes due to their inherent ability to amplify small signals.<sup>226</sup> In this particular device the ions permeate the semiconductor channel and dope it. It is also possible to turn this around and eject ions from the channel. These pass through a polymer electrolyte to trigger the neuron.<sup>229</sup> Such devices could form a platform for stimulating and detecting neural responses on chip.

A number of steps are needed to establish viable proton to electron transducer circuits. Amplification of the biological signals at the source is of great interest; this would have increased signal-to-noise compared with amplification using external circuitry. Complementary n- and p-type transducers coupled to the signal would provide the most effective amplification. *p*-type organic transducers with high-fidelity kHz operation have been made,<sup>230,231</sup> but manufacturing complementary n-type devices with the same performance is difficult. Additionally, switching speed may ultimately be limited by the low electron/hole mobility  $\mu_{eh} < 1 \text{ cm}^2/\text{Vs}$  of the organic transistor channels.<sup>11</sup> This has seen renewed interest in using nanowires: the large surface/volume ratio offers scope for enhanced signal output,<sup>194,228</sup> and their size is suited for the accurate detection of biological events.<sup>227,232</sup> In Chapter 6 I present results suggesting that nanoscale patterned PEO supports proton transport with an unusually high conductivity. I use devices featuring PEO dielectrics without salt doping to simultaneously explore the properties of proton conduction in PEO and use them as a proof-of-principle for III-V nanowires as effective ion/proton-to-electron transducers with both n-InAs and p-GaAs channels.

## 1.7 Summary

This chapter outlined the physics and previous research underpinning the results presented in this thesis. Included was a description of the field-effect transistor and how the field-effect in general is a powerful tool for generating and probing various systems, e.g., low-dimensional quantum systems and biological materials. Evaluating emergent technologies in a research usually requires overcoming materials complications. Surface states and dopants play a central role to the operation of both p-AlGaAs/GaAs heterostructure devices and nanowires. Both cause gate instability and hysteresis and can influence the properties of quantum systems. The goal is to mitigate their influence, work within the provided limits, or think about how they could be used to provide additional functionality. Examples of each are demonstrated throughout Chapters 3 - 6. In the next chapter I present the device fabrication methods and measurement techniques used during my research.

# Chapter 2

# Methods

# 2.1 Fabrication and micro-/nano-lithography background

Fabrication of the micro-/nanoscale devices in this thesis was enabled by lithography processes that use thin, patternable polymeric films known as 'resists'. The lithography proceeds by defining micro-/nanoscale windows in the resist layer to allow access to the substrate in only these defined regions. Subsequent deposition of thin layers of material (e.g. metals and oxides) or chemical etches are then only applied to the substrate in these windows. The resist are patterned by exposing selected areas of the resist to high energy photons (photolithography) or electrons (electron beam lithography), which causes either the scission or crosslinking polymer chains, depending on the resist properties. The exposed areas become either more or less soluble in particular 'developer' solutions. The behaviour of polymer chains under irradiation determines the tone of the resist:

• Positive-tone resist: polymer chain scission occurs under irradiation, increasing the solubility of the exposed regions.

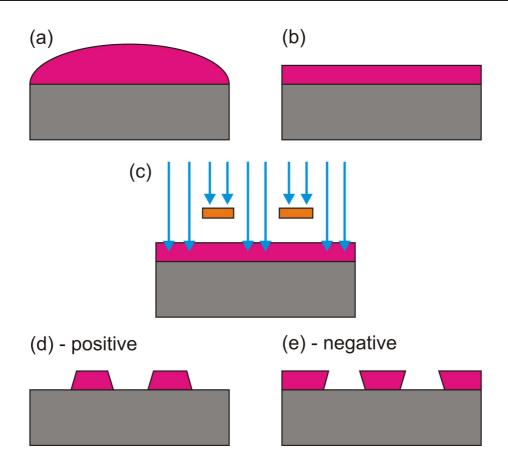


FIGURE 2.1: Photolithography process. (a) Polymeric resist (pink) is deposited onto substrate (grey). (b) Spinning and baking results in a thin film with thickness  $\sim 1 - 2 \mu m$ . (c) Sample is exposed to UV light through a chromeon-glass mask (orange). (d) UV irradiation breaks polymer chains in a positive resist. Exposed regions are removed in developer solution, transferring the mask image to the resist layer. (e) UV irradiation cross-links polymer chains in a negative resist. Exposed regions remain after development, transferring the negative image of the mask.

• Negative-tone resist: polymer chains are cross-linked under irradiation, strengthening these regions against dissolution.

The developer is chosen such that it dissolves the regions with the weakest chains; the exposed (unexposed) regions for positive (negative) resist. This creates the windows in the resist layer through which the selective lithography takes place *via* etching or materials deposition.

### 2.1.1 Photolithography

Photolithography utilises resists that are sensitive to UV radiation. Selectivity is usually provided by a pre-patterned chrome-on-glass mask, where the chrome sections block the underlying resist from UV exposure. The resists used in this thesis were MicroChem S1813 (positive tone)<sup>233</sup> and AZ nLOF2020 (negative tone),<sup>234</sup> which both consist of a polymer blend dissolved in a solvent to facilitate spincasting. Figure 2.1 illustrates the photolithography process, which followed the steps:

- A drop of the resist was spin-cast onto the sample substrate to form a thin film. S1813 was spun at 5000 rpm for 60 s, and AZ nLOF2020 was spun at 10000 rpm for 60 s.
- 2. The sample was baked on a hotplate at 110°C for 60 s to evaporate the solvent. The final resist thickness was  $\sim 1-2 \ \mu m.^{233,234}$
- 3. The sample was exposed to UV radiation with power density 10 mW/cm<sup>2</sup> through a chrome-on-glass mask. Mask alignment and UV exposure was performed using a Quintel Q4000 with exposure times of 6.5 s for S1813 and 4 s for nLOF2020.
- 4. A post-exposure hotplate bake at 110°C for 60 s was conducted for AZ nLOF2020 only. This strengthened the unexposed regions of this resist.
- 5. Development was performed in solutions of 2.38% tetramethylammonium hydroxide (TMAH) in H<sub>2</sub>O for 60 s. The process was terminated with a H<sub>2</sub>O rinse and the sample was dried with a flow of N<sub>2</sub> gas.

For the positive-tone S1813 resist, the remaining resist is in the pattern of the mask used (Fig. 2.1(d)). For the negative-tone nLOF2020 resist, the removed regions take the pattern of the mask (Fig. 2.1(e)). Note that for any given chrome-onglass mask, the negative of that mask can be also be created. This means that a nominally identical pattern can be obtained in both resist tones. That is, the

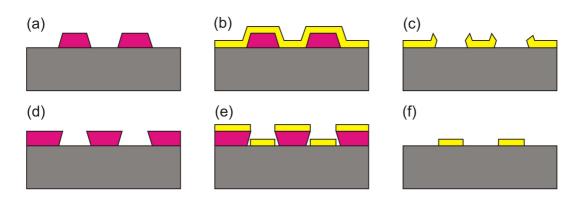


FIGURE 2.2: Metal deposition on a sample featuring (a-c) positive photo-resist with sloped side-walls and (d-f) negative photo-resist with undercut side-walls. (b) Metal deposition on positive resist results in a continuous film at the pattern edges. (c) Lift-off tears the metal, producing poorly defined edges. (e) The undercut profile of negative resist means that (e) the metal is discontinuous at the pattern edges. (f) This results in clean edges on a well defined metal pattern after lift-off.

same pattern is obtained in a negative resist as a positive resist if the negative mask pattern is also used.

The type of resist used depends on practical considerations, since the pattern itself does not determine the resist type. The most common consideration is the differing side-wall profile between the tone types (see Figs 2.1(d/e)). This arises because the UV radiation intensity is greater at the resist surface than at the resist/sample interface.<sup>235</sup> The profile is a particularly vital consideration in the deposition of metal films, pictured in Fig. 2.2. In this process, a 30-200 nm thick metal layer is deposited uniformly across the sample, adhering to the substrate surfaced only in the resist windows. Completing the metal pattern entails a step known as 'lift-off', where the resist is dissolved by immersion in a solvent such as acetone or N-methyl-2-pyrrolidone (NMP). This removes the resist layer and the metal regions not in contact with the substrate. For lift-off to be effective, the solvent must be able to access the resist layer underneath the metal through discontinuous sections of the metal layer. The undercut profile of the negative-tone resist results in discontinuities of the metal layer at the edges of the resist windows (Fig. 2.2(e)), and therefore facilitates effective lift-off (Fig. 2.2(f)). Conversely, the continuous metal layer over the positive resist prevents the solvent from accessing

the resist layer (Fig. 2.2(b)). This means that lift-off is either completely prevented, or metal tearing at the edges of the pattern occurs (Fig. 2.2(c)). As such, negative photo-resists are commonly used for metal deposition, although strategies do exist to make positive resists compatible with lift-off by, e.g., soaking the resist in chlorobenzene prior to development. The chlorobenzene hardens only the very top layer of the resist and thereby generates an undercut profile.

A second consideration is that the mask usually needs to be aligned to pre-existing features on the sample. This is more easily done if the chrome pattern covers a minimal area on the mask and therefore facilitates a less restricted view of the sample. In applications that have no preferred side-wall profile, e.g., chemical etching, the choice of mask pattern and resist tone is usually guided by the ease of alignment.

The minimum feature size for photolithographically defined patterns is typically  $\sim 1 \ \mu m$ , limited by the UV radiation wavelength, i.e., hundreds of nanometers, and properties of the resist. Electron beam lithography is used for features requiring a smaller pattern size.

### 2.1.2 Electron beam lithography

Electron beam lithography (EBL) uses the highly focussed electron beam of a scanning electron microscope (SEM) to selectively modify the crosslinking of chains in the resist at the nanoscale. With some optimisation, patterns smaller than 10 nm can be made in the resist.<sup>237</sup> EBL in this thesis was primarily done using the positive-tone resist poly(methyl methacrylate) (PMMA), with molecular weight MW = 950 k, dissolved in anisole. The anisole content is used to increase the viscosity and modify the film thickness vs. spin speed relationship. Thicknesses of 150 and 300 nm are expected for the MicroChem PMMA-A3 and PMMA-A5 films at the spin speeds used here.<sup>236</sup> PMMA-A3 was used generate the etch pattern for the hole billiard used in Chapter 3, and PMMA-A5 was used for the fabrication of nanowire FETs in Chapters 4, 5 and 6. The EBL process was:

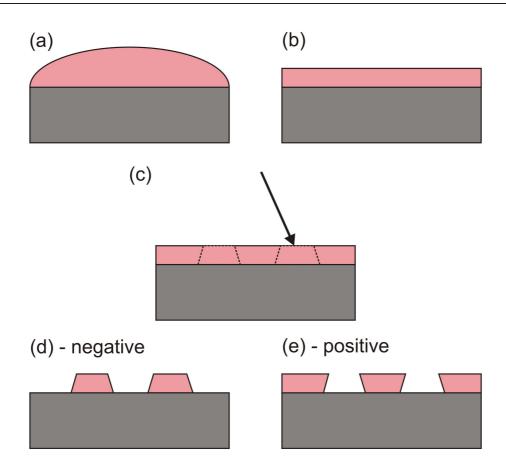


FIGURE 2.3: Electron beam lithography (EBL) process. Resist is (a) deposited and (b) spun onto the substrate to form a film  $\sim 50 - 300$  nm thick.<sup>236</sup> (c) The electron beam of an SEM traces out the desired exposure pattern. The proximity effect exposes the underside of the resist outside the region defined by the beam. This results in (d) sloping side-walls for negative-tone resists and (e) an undercut profile for positive-tone resists.

- 1. The PMMA resist was spun on at 5000 rpm for 60 s.
- A 3 min/5 min bake at 180°C was used to evaporate the anisole for PMMA-A3/A5, respectively.
- 3. The desired pattern was exposed using either an FEI Sirion (hole billiard) or Raith 150-Two (NWFETs). For PMMA-A3, an accelerating voltage of 30 kV and electron area dose of  $300 400 \ \mu\text{C/cm}^2$  was used depending on feature size. PMMA-A5 was exposed with a 20 kV accelerating voltage and dose of  $260 520 \ \mu\text{C/cm}^2$ .

4. The pattern was developed by immersion in a 1:3 mix of methyl-isobutylketone:isopropanol for 60 s. The sample was rinsed in isopropanol and dried under an N<sub>2</sub> flow.

There are some notable differences between EBL and photolithography. In EBL, a pattern generator deflects the electron beam directly to the regions of the resist to be exposed rather than using a pre-patterned mask as in photolithography. Typically the EBL exposure is done with the sample in a constant position relative to the SEM aperture, and the beam is deflected to trace out the specified pattern. This means that sections of the pattern are exposed consecutively, rather than concurrently. As a result, exposing larger pattern areas requires a longer time to complete. This contrasts with mask-based processes like photolithography, where the single exposure step takes the same time no matter the pattern size.

A second difference between EBL and photolithography is the resulting side-wall profiles. in EBL, the side-wall profiles arise from the proximity effect, which occurs as follows: primary electrons from the SEM beam penetrate well into the substrate, where backscattering and secondary electron generation causes electrons to be fired back into the underside of the resist.<sup>238</sup> This results in the resist close to the substrate being exposed outside of the areas scanned by the electron beam. Therefore, development produces sloped side-walls for negative-tone EBL resists, and an undercut profile for positive-tone resists like PMMA (Figs 2.3(d/e)). The proximity effect poses a difficulty in the optimisation of many negative-tone EBL resists as it can generate significant pattern broadening. An example of this is discussed in detail in Chapter 4 as part of the results on optimising the resolution of EBL patterned polymer electrolyte thin films.

## 2.2 *p*-AlGaAs/GaAs device fabrication

The p-AlGaAs/GaAs HHMTs in this thesis were fabricated on high quality custom wafers grown using molecular beam epitaxy by Ashish Rai and Dirk Reuter

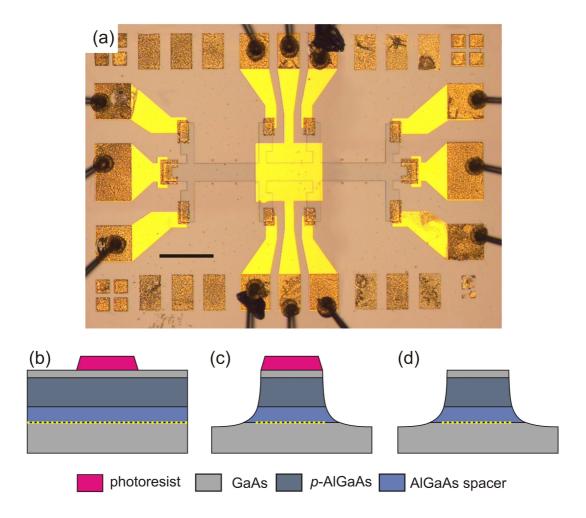


FIGURE 2.4: (a) Optical micrograph of a completed HHMT and (b-d) process used to define the the Hall bar mesa. (a) The HHMT features a central Hall bar mesa (grey outline), AuBe ohmic contacts (grainy gold) on each of the Hall bar arms, and Ti/Au gate (bright yellow square) in the middle of the Hall bar. The AuBe bond pads surrounding the device were defined and deposited as part of the ohmic contact pattern. The Au wires used to connect the device to the chip package can be seen around the outer edge of the image. The interconnects between bond pads and ohmic contacts were deposited with the Ti/Au gate. (b) Cross-section of the heterostructure, including the 2DHG (yellow dotted line) at the AlGaAs/GaAs interface. Positive-tone S1813 was patterned in the shape of the Hall bar on the heterostructure surface. (c) A HF-based solution was used to etch the unprotected heterostructure regions to a depth 10 nm below the 2DHG. (d) The resist was removed using acetone.

in Andreas Wieck's group at Ruhr-Universität Bochum, Germany. Both n-type and p-type wafers were grown simultaneously on (100)- and (311)A-oriented semiinsulating GaAs substrates supplied by AXT. Epitaxial growth began with 50 nm of undoped GaAs followed by a twenty-period superlattice, where each period consisted of 2 nm GaAs and 2 nm AlAs. The superlattice was included to trap impurities at these interfaces and thereby reduce the background impurity density close to the 2DEG/2DHG. The active region was then grown, which was a 650 nm undoped GaAs buffer layer, a 35 nm undoped AlGaAs spacer layer, a 80 nm Si-doped AlGaAs layer with Si density  $N_{Si} = 3.5 \times 10^{16}$  cm<sup>-3</sup> and a 5 nm undoped GaAs cap. The Si-dopants settle on Ga sites as a *n*-type dopant for (100)oriented growth and As sites as a *p*-type dopant for (311)A-oriented growth,<sup>105</sup> and provide the electrons/holes for the 2DEG/2DHG that resides at the interface between the AlGaAs spacer and GaAs buffer. All devices in this thesis were fabricated from the *p*-type (311)A-oriented wafer. The *n*-type wafer was used without additional device fabrication as part of the surface passivation characterisation in Chapter 3. The hole density and mobility in the 2DHG were measured to be  $p = 1.63 \times 10^{11}$  cm<sup>-2</sup> and  $\mu = 1.03 \times 10^6$  cm<sup>2</sup>/Vs at 4.2 K using the Hall effect.

A completed HHMT is shown in Fig. 2.4(a). The first step in device fabrication was to create a Hall bar mesa that defined the bounds of the 2DEG in the xand y-directions. This was accomplished using photolithography and wet etching as illustrated in Figs 2.4(b-d). First, positive-tone S1813 resist was patterned into the shape of a Hall bar at the heterostructure surface. The regions of the heterostructure not covered by resist were then etched away by immersion for 37 s in a 2:1:20 solution of buffered HF:H<sub>2</sub>O<sub>2</sub>:H<sub>2</sub>O, where the buffered HF was 7:1 NH<sub>4</sub>F:HF. This etch proceeds as the H<sub>2</sub>O<sub>2</sub> oxidises the AlGaAs or GaAs material and these oxides are etched away by the HF acid. The heterostructure was etched to a depth of 130 nm, which was 10 nm below the 2DHG. The resist was removed by soaking the device in acetone for 10 mins at room temperature. An isopropanol rinse was used to remove any residual acetone and the sample was dried with N<sub>2</sub> gas.

Ohmic contacts were then added to each arm of the Hall bar to facilitate electrical contact with the 2DHG according to the process shown in Figs 2.5(a-c). Windows at the end of each arm were patterned in negative-tone nLOF-2020 resist, which had the desired undercut side-wall profile. After development, the sample underwent an 'ashing' step, with the purpose of removing any residual resist in the defined windows. Depositing metal onto the surface when there is residual

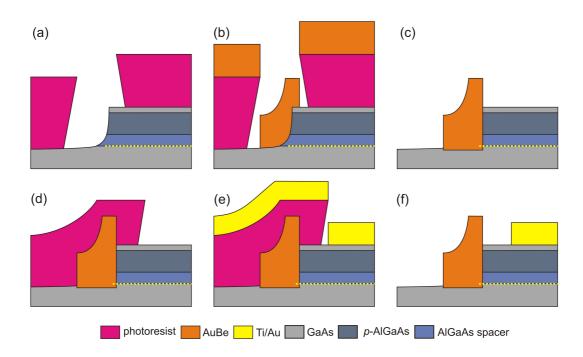


FIGURE 2.5: Cross-section of half of the Hall bar showing the contact metal deposition process. (a) Openings in negative-tone nLOF-2020 were made on each arm of the Hall bar prior to (b) AuBe deposition. (c) Lift-off removed unwanted metal, and annealing caused diffusion of the AuBe into the semiconductor and facilitated electrical contact to the 2DHG. (d) Negative-tone nLOF-2020 was again used to define the pattern for deposition of (d) Ti/Au gates. (e) The device was completed by lift-off in acetone.

resist can result in poor surface adhesion of the metal and high contact resistance. As such, the top few nanometers of resist layer were stripped by a short 30 s etch in an O<sub>2</sub> plasma. This was done using a Denton plasma asher with an O<sub>2</sub> pressure of 340 mTorr and incident power of 50 W. The 150 nm thick 99:1 AuBe ohmic contacts were then deposited by thermal evaporation under high vacuum  $\sim 10^{-6}$  mbar. The source material was placed in a tungsten or molybdenum 'boat' facing the sample. The boat was resistively heated by applying a current  $\sim 5-6$  A. The source metal first melts and then evaporates, which causes the metal atoms to travel ballistically through the vacuum and solidifying on the sample. Lift-off was performed by soaking the sample in acetone for 10 mins at room temperature before rinsing with isopropanol and drying with N<sub>2</sub>.

The as-deposited AuBe forms a Schottky barrier with the underlying GaAs, since the high density of surface states pin the metal Fermi level in the middle of the GaAs band gap (recall Fig. 1.6 in Sec. 1.3.2).<sup>27,46,47</sup> The most common way to

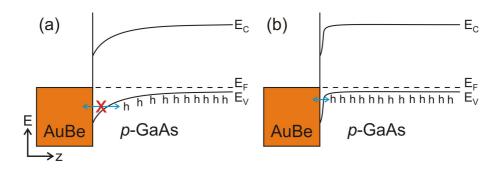


FIGURE 2.6: AuBe interface with uniformly doped *p*-GaAs (a) before and (b) after thermal annealing, illustrating the the general physics behind ohmic contact formation. (a) The as-deposited metal Fermi level is pinned mid-gap by GaAs surface states, which generates a Schottky barrier that suppresses charge transport across the interface. (b) Annealing causes Be diffusion into the GaAs sub-surface layer, highly doping this region and reducing the Schottky barrier width. The resultant tunnelling transport facilitates linear, low resistance electrical contact between the metal and semiconductor.

overcome the Schottky barrier and obtain linear I - V characteristics is to cause the contact metal and dopant to diffuse into the GaAs via rapid thermal annealing. (Fig. 2.5(c)) For these devices, the contacts were annealed at 490°C for 90 s in a 5% H<sub>2</sub>/N<sub>2</sub> atmosphere. The rapid diffusion of Be in GaAs<sup>239</sup> causes the Be to diffuse further than the Au and degenerately dope a thin layer of the GaAs in the vicinity of the metal contact. This does not affect the surface pinning; rather it causes the Schottky barrier to narrow as the high level of doping brings the valence band close to the Fermi level. This is shown in Fig. 2.6. The reduced barrier thickness enhances tunnelling transport through the barrier to the point where electrons/holes can pass between the metal and semiconductor with a low resistance. Ultimately this process results in linear, ohmic contacts.

The final fabrication step was the addition of a gate to modulate current flow in the 2DHG (Figs 2.5(d-f)). The photolithography process was similar to that used for the ohmic contacts. The pattern was defined in a nLOF-2020 resist, and plasma ashing post-development ensured that all resist was removed from the patterned windows. Thermal evaporation was again used to deposit a 15 nm Ti adhesion layer and a 85 nm Au layer. Lift-off was performed using a 10 min acetone soak, isopropanol rinse and N<sub>2</sub> dry. Finished devices were affixed to an LCC20 chip package using a small drop of PMMA or conductive silver paint (Fig. 2.7). The

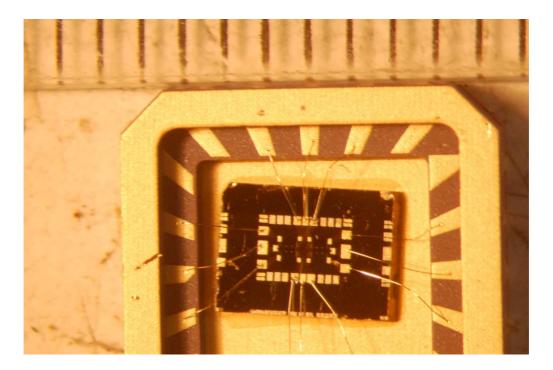


FIGURE 2.7: Finished HHMT bonded in an LCC20 chip package. The Au wire connects the bond pads on the device to the LCC20 pins. The ruler at the top of the image has 1 mm graduations.

LCC20 package features 20 pins that were connected to bond pads on the device by a thin Au wire using a Kulicke & Soffa Au ball bonder. The standardised LCC20 packages fit into the sockets on the measurement systems outlined below, and thereby facilitate interfacing the micro-scale device structures with electrical measurement equipment.

The above process was used to fabricate standard HHMTs. I undertook several variations on this throughout my research. Notably, I fabricated an etched hole billiard with an over-all top gate. Fabrication of this device involved EBL and wet etching to define the billiard trenches prior to deposition of a top-gate that was insulated from the device by a polyimide layer. The specifics of this fabrication will be discussed in Chapter 3. I also applied surface sulfur treatments prior to gate deposition on selected HHMTs; I discuss this now.

#### 2.2.1 Sulfur passivation

The success of  $(NH_4)_2S_x$  solutions in passivating low-index GaAs surfaces was outlined in Sec. 1.3.2.1. Effective passivation is achieved in two steps: first, the amorphous surface oxide is removed and secondly, adsorbed S atoms covalently satisfy all dangling bonds such that no electronic states are present in the surface band-gap.<sup>21,44</sup> We investigated applying these treatments to HHMTs in an attempt to mitigate the gate hysteresis and instability in these devices.<sup>24,32</sup> An important consideration in incorporating this treatment to the HHMT fabrication process is that the S-bonded GaAs surface has a low stability in air. Exposure to light and oxygen for approximately half an hour breaks the Ga-S and As-S bonds, and subsequent oxygen adsorption returns the surface to an oxide-covered, unpassivated state.<sup>44,57</sup> As such, we needed to apply the treatment immediately prior to gate deposition, after development and ashing of the gate photo-pattern. After gate deposition, the metal layer is expected to protect the surface from re-oxidisation.<sup>41,62</sup>

We initially considered both aqueous and alcoholic  $(NH_4)_2S_x$  solutions as potential passivation treatments.<sup>44</sup> However, I show in Chapter 3 that alcoholic solutions were incompatible with our processing since alcoholic solvents tend to dissolve photoresist. The remainder of the work focussed on aqueous treatments, which left the photoresist intact. All treatments were based on a stock solution, which was was prepared by dissolving 2.4 g of elemental S in 25 mL of 20%  $(NH_4)_2S$ in H<sub>2</sub>O using a magnetic stirrer at room temperature in the dark for at least 12 hours. Two treatments were applied:

- Strong treatment: 10 min immersion in  $\sim 3$  mL of the stock solution.
- Weak treatment: 2 min immersion in a 0.5% dilution of the stock solution in H<sub>2</sub>O.

Both treatments were performed under illumination with the solution at  $T = 40^{\circ}$ C.<sup>44,62</sup> These conditions enhance S-bonding and assist in the removal of excess

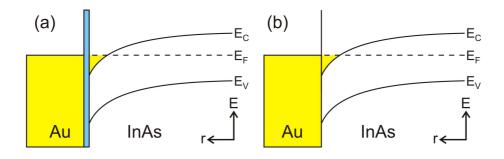


FIGURE 2.8: Radial band diagram for Au metal deposited on InAs nanowire surfaces (a) without sulfur passivation and (b) with sulfur passivation. (a) The native amorphous oxide (blue) increases the contact resistance as it provides an injection barrier. (b) Sulfur passivation removes this barrier and prevents reoxidisation prior to metal deposition, facilitating low resistance ohmic contact formation.

electrons from the surface.<sup>44</sup> Treated samples were rinsed with  $H_2O$  and transferred to the evaporator as quickly as practicable. The evaporator typically achieved a vacuum of < 1 mTorr in under 5 minutes. Gate deposition and device packaging then proceeded as described above.

The sulfur passivation treatment was also applied to InAs nanowires, but with a markedly different purpose. The InAs surface states pin the surface Fermi in the conduction band,<sup>48</sup> in strong contrast to the GaAs surface states that pin the Fermi level mid-gap. For InAs, this aligns the Fermi level of deposited metals directly with the conduction band and facilitates direct injection of electrons. However, the presence of a thin native oxide presents an injection barrier, which increases the contact resistance (Fig. 2.8(a)). Sulfur passivation of InAs nanowires serves to remove the native oxide and re-pin the Fermi level at a slightly different energy within the conduction band (Fig. 2.8(b)). The mono-layer of S atoms preserves this oxide-free surface for long enough to perform metal deposition, which enables the formation of low resistance ohmic contacts.<sup>240</sup>

In this section I describe the fabrication of substrate-gated and wrap-gated NWFETs. These processes form the basis of the advanced gating techniques developed as part of my research. The fabrication processes we developed for multiple wrapgated NWFETs and polymer electrolyte gated FETs are described in Chapter 4. Nanowire growth was done at Lund University, Sweden, the Australian National University (ANU), and the Niels Bohr Institute, University of Copenhagen, Denmark. The ANU nanowires were pure wurtzite InAs nanowires, 50 nm in diameter and 3 – 6  $\mu$ m in length, grown using metal-organic chemical vapour deposition. These were used for the polymer electrolyte gated NWFETs in Chapters 4 and 6. Lund University supplied two separate batches of InAs nanowires grown by chemical beam epitaxy. The first had diameter 90 nm and length 3  $\mu$ m. These were coated with 12.5 nm Al<sub>2</sub>O<sub>3</sub> using atomic layer deposition, 16.5 nm of W and 11.5 nm of Au deposited *via* dc sputtering, and a final 12.5 nm Al<sub>2</sub>O<sub>3</sub> layer deposited using atomic layer deposition. This batch was used for fabrication of multiple wrap-gated devices in Chapter 4. The second batch consisted of InAs nanowires which were 50 nm in diameter and 3  $\mu$ m in length. One of these nanowires was used for the device in Chapter 5. The University of Copenhagen supplied Be-doped *p*-GaAs nanowires grown using molecular beam epitaxy, with acceptor concentration  $N_A = 1 \times 10^{18}$  cm<sup>-3</sup>, diameter 100 nm and length 5–10  $\mu$ m. These were used in for complementary PE-gated NWFETs in Chapter 6.

## 2.3 NWFET fabrication

#### 2.3.1 Substrate-gated NWFETs

The first step in NWFET fabrication was transferring the as-grown nanowires from the growth substrate to the measurement substrate (Fig. 2.9(a)). This was accomplished using the corner of a clean-room tissue. Lightly tapping the tissue corner on the growth substrate collected the nanowires as they were broken from the substrate. Subsequently tapping the tissue corner on the measurement substrate randomly deposited the nanowires on this surface.

The measurement substrate was based on a degenerately doped  $n^+$ -Si wafer which functioned as a global back-gate. The  $n^+$ -Si was insulated from the nanowires by a 100 nm thermally grown SiO<sub>2</sub> layer and a 10 nm HfO<sub>2</sub> layer deposited by atomic layer deposition. The latter was included as the slow etch rate of HfO<sub>2</sub> protected

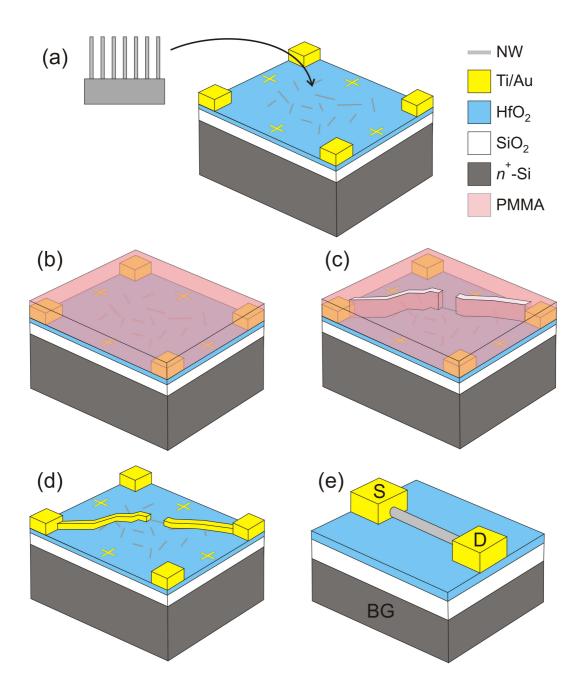


FIGURE 2.9: Fabrication process for substrate-gated nanowire FETs. (a) Asgrown nanowires are transferred from the growth substrate to the measurement substrate. The measurement substrate consisted of degenerately doped  $n^+$ -Si which was used as a global back-gate, insulated by 100 nm of SiO<sub>2</sub> and 10 nm of HfO<sub>2</sub>. Substrates were pre-patterned with a series of 100  $\mu$ m ×100  $\mu$ m writefields. The alignment crosses were at each side and the Ti/Au interconnects at each corner. (b) The sample was covered in PMMA-A5 for EBL processing. The beam was deflected according to the desired pattern within each write-field. (c) Development removed the exposed regions. (d) Metal deposition and lift-off finalised the process. (e) Close-up of the contacted NWFET featuring source (S), drain (D) and back-gate (BG).

the  $SiO_2$  layer from the acid etches used in wrap-gate fabrication (see next section, 2.3.2). The substrates were patterned with Ti/Au alignment crosses and interconnects prior to nanowire deposition in order to facilitate accurate pattern alignment and high through-put processing, shown in Fig. 2.9(a). The interconnects extend beyond the region shown in Fig. 2.9 and run out to bond-pads that ultimately enable electrical connection in a similar fashion to the HHMT devices. The alignment crosses were patterned to utilise the high precision alignment afforded by the Raith 150-Two EBL tool. These pre-patterned features divided the substrate into sets of 24 adjacent 100  $\mu m \times 100 \mu m$  'write-fields'. In EBL, the write-field size defines the maximal area over which the beam may be deflected to expose a pattern in the resist. Writing outside this area entails first moving the stage such that the SEM aperture is centred over a different section of the sample. Scanning the beam over the new area then constitutes writing in a separate write-field. Chips were typically processed with multiple sets of the 24 adjacent write-field patterns to enable concurrently applying an identical fabrication process to up to 72 NWFETs.

The EBL process for NWFETs was:

- After nanowire deposition, PMMA-A5 was spun onto the substrate according to the process in Sec. 2.1.2.
- An optical micrograph was taken of each write-field, and custom patterns were generated to link the source/drain of a single nanowire in each writefield to the Ti/Au interconnects.
- 3. The substrate was placed on a movable sample stage in the Raith 150-Two.
- 4. 'Global' sample alignment was performed, which consisted of recording the substrate position and orientation in the EBL software. Subsequent positioning was done relative to this global alignment step.
- 5. With the beam off, the stage was moved such that the SEM aperture was immediately above the centre of the first write-field.

- 6. The beam was turned on and scanned over each of the four alignment crosses. These images are displayed to the user, who places a cursor at the centre of each cross within the EBL software. This precisely specifies the position/alignment of the current write-field.
- 7. The beam was deflected within the 100  $\mu$ m<sup>2</sup> write-field to expose the contact pattern in the PMMA layer.
- 8. The beam was then turned off, and the stage was moved to the next write-field.
- 9. Steps 6 8 were repeated until all patterns had been exposed in each write-field.
- The sample was removed from the EBL tool and the resist was developed in 1:3 methyl isobutyl ketone:isopropanol for 60 s, rinsed with isopropanol and dried with N<sub>2</sub>.

After development, the sample was ashed in  $O_2$  plasma for 30 s to remove residual resist. The nanowire contacts then underwent the weak sulfur treatment outlined in the previous section immediately prior to contact deposition. The ohmic contacts were either Ti/Au or Ni/Au deposited by thermal evaporation; specific layer composition and thickness are given for each presented device. Lift-off was accomplished by soaking in NMP at 80°C for 6 to 18 hours. Samples were rinsed in acetone and isopropanol and dried with N<sub>2</sub> gas. The finished chips were divided into smaller sections using a diamond scriber. Each smaller piece was affixed to an LCC20 chip package using conductive silver paint, and selected devices were contacted using the Au ball bonder. One of the pins on the LCC20 was bonded directly to the bottom of the package. This facilitated electrical contact to the  $n^+$ -Si substrate via the conductive silver paint.

This process was used the basis for polymer electrolyte-gated NWFETs, with the full details given in Chapter 4. I also studied polymer electrolyte-gated NWFETs with p-GaAs channels. In these devices, at 200 nm thick AuBe layer was used for the ohmic contacts rather than Ni/Au. The nanowires underwent a 30 s etch

of 20% HCl in  $H_2O$  to remove the native oxide prior to metal deposition. After deposition and lift-off, the contacts were annealed at 300°C for 30 s to facilitate ohmic contact formation.

### 2.3.2 Wrap-gated NWFETs

The fabrication procedure for lateral wrap-gated NWFETs shown in Fig. 2.10 was developed by Storm *et al.*<sup>145</sup> In this method, the wrap-gate layers were coated onto the as-grown nanowires in the vertical orientation. The nanowires were then transferred to the measurement substrate and wrap-gate segments are selectively etched in this lateral orientation. The coatings consisted of a 12 nm insulating  $Al_2O_3$  layer deposited by atomic layer deposition (ALD), the metal wrap gate, which was 16.5 nm of W and 11.5 nm of Au deposited by dc sputtering, and an outer 12 nm  $Al_2O_3$  layer. The outer  $Al_2O_3$  layer was deposited solely to facilitate nanowire transfer; metal-coated nanowires do not adhere to the cleanroom tissue. Fabrication of wrap-gated NWFETs then proceeds as follows:

- 1. Nanowires were transferred to the measurement substrate, as for substrategated NWFETs (Fig. 2.10(c)).
- 2. EBL was used to open windows in the PMMA resist at both ends of the nanowire (Fig. 2.10(d)). This pattern ultimately defined the source/drain leads, and therefore the windows extend to the pre-patterned interconnects.
- 3. A series of etches were used to selectively remove the outer oxide and wrapgate coatings in the source/drain contact regions. An important aspect of this step was that the etches work along the nanowire length from the resist windows and remove portions of the coating layers that were covered by resist (Fig. 2.10(e)). This is central to the success of the fabrication process; if the etches only removed material in the resist windows, depositing the ohmic contacts into these windows would cause them to electrically short directly to the wrap-gate. An additional advantage of the etch working underneath the resist is that it allows the gate length to be set by the etch time. Longer

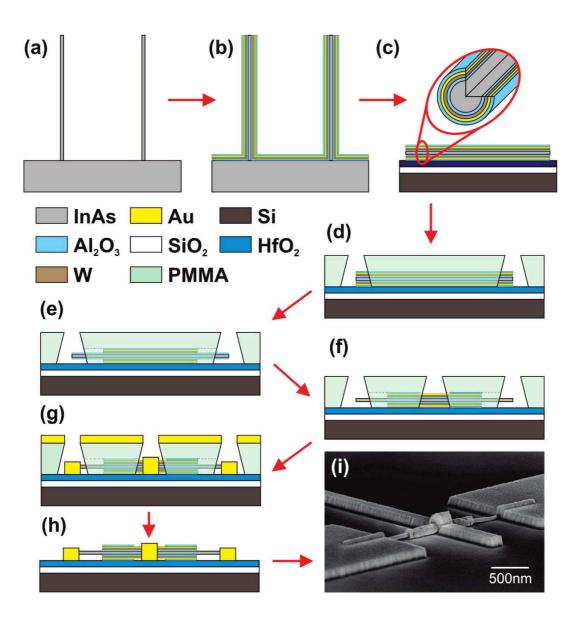


FIGURE 2.10: Fabrication for lateral wrap-gated NWFETs. (a/b/c) As-grown nanowires are coated with the wrap-gate material in the vertical orientation before being transferred to the measurement substrate. The layers are shown in the inset to (c), and are an inner Al<sub>2</sub>O<sub>3</sub> insulator, the W and Au wrap-gate, and an outer Al<sub>2</sub>O<sub>3</sub> layer to enable nanowire transfer in (c). (d) EBL is used to define the source/drain contact regions. (e) A series of chemical etches remove the outer oxide and wrap-gate metal in the contact regions. The etches undercut the resist, with the wrap-gate length determined by the dilution and etch time of the Au etch. (f) A window is opened in the resist for the gate contact. A second HF etch removes the outer oxide at the gate contact and the inner oxide at the source/drain contacts. (g) The Ni/Au contacts are deposited, and (h) lift-off completes the device. (i) Scanning electron micrograph of completed device. Figure from Ref. 145.

etch times remove more gate material, and result in shorter gate lengths. The etches were:

- (a) A 30 s etch in 1:7 buffered HF:H<sub>2</sub>O to remove the Al<sub>2</sub>O<sub>3</sub> layer. The need to remove an oxide with an acid etch explains the choice of oxide in the wrap-gate coating and on the measurement substrate. HfO<sub>2</sub> has a much slower etch rate in HF solutions than Al<sub>2</sub>O<sub>3</sub> and SiO<sub>2</sub>. Therefore HfO<sub>2</sub> was used as a protective layer for the measurement substrate, as only a minimal amount is removed during the 30 s required to completely remove the Al<sub>2</sub>O<sub>3</sub> wrap-gate insulator.
- (b) A KI/I<sub>2</sub> etch to remove the Au layer. The composition and time of this etch determines the gate length. All etch solutions were based on a stock solution of 4 g KI and 1 g I<sub>2</sub> dissolved in 80 mL of H<sub>2</sub>O. Storm *et al.*<sup>145</sup> used dilutions from 20% stock to 1% stock in H<sub>2</sub>O. The highly diluted etches provided the slowest etch rate and therefore the most accurate gate length control. Typical etch times were 60 s to 240 s.
- (c) A solution of 31%  $H_2O_2$  in  $H_2O$  at  $T = 40^{\circ}C$  was used to etch away the W layer.
- 4. A second EBL step on the same resist opened a window for deposition of the gate contact.
- 5. The outer oxide in this window and inner oxide at the source/drain were etched using the buffered HF solution (Fig. 2.10(f)).
- 6. The source/drain regions of the NW were passivated with the weak  $(NH_4)_2S_x$  treatment immediately before loading the device into the thermal evaporator.
- 7. 25 nm Ni and 75 nm Au are deposited to contact the device at the source, drain and gate (Fig. 2.10(g)). Liftoff in NMP completes the device, shown in Fig. 2.10(h/i).

A particular advantage of this fabrication procedure is that it uses only one resist layer. This ensured that the nanowire remained in the same position throughout processing, which is vital to accurately complete high-precision nanofabrication. The concern was that using two separate resists for gate definition and contact deposition would cause the nanowire to shift position during lift-off or subsequent processing.

A major disadvantage of the single-resist approach is that it cannot be used to fabricate devices with multiple wrap-gates. This is because the final step involves depositing metal wherever a region of the resist has been opened. For example, a natural place to start with a single-resist approach would be to open the resist in the middle of the nanowire at the same time as the source/drain contact etch regions are defined at Step 2. The middle of the wrap-gate would then be etched away along with the wrap-gate portions at the source and drain in Step 3, splitting the gate into two segments. However, metal would ultimately be deposited into the resist opening at Step 7, directly onto the nanowire. This is incompatible with a number of key potential applications where a small spacing between adjacent wrap-gates would be desirable. An unnecessary ohmic contact between the gates severely limits the spacing between segments and radically increases the chance of the ohmic contacts shorting directly to the gates. A small wrap-gate separation is highly desirable for down-scaling applications. It would also be beneficial for, e.g., a quantum dot defined using three wrap-gate segments. In the latter, the potential on the two outer wrap-gates would define the tunnelling barriers and the central gate could be used as a plunger gate to probe the dot energetics. Here, having ohmic contacts within the quantum dot is highly undesirable, especially for InAs nanowire devices where the invasive ohmic contacts themselves can constitute a significant potential barrier.<sup>30,241</sup> Developing a general method for the fabircation of multiple wrap-gated NWFETs with only a source and drain at either end of the wire entailed using a two-resist process. I present the process we developed in Chapter 4, which was used to realise devices with up to four independent wrap-gate segments with separations of down to 200 nm.

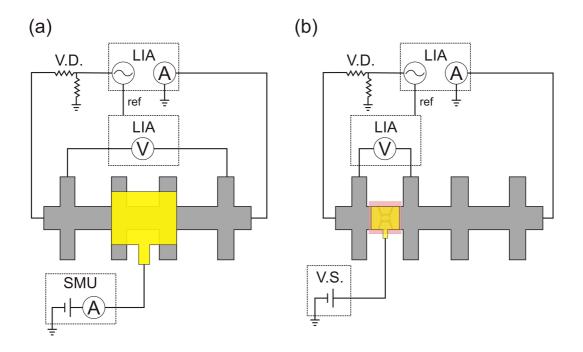


FIGURE 2.11: Electrical circuits used for (a) HHMTs and (b) the etched hole billiard in Chapter 3. The Hall bar is shown in grey, the Ti/Au gates in gold. The polyimide layer in (b) is shown in pink, with the etched billiard depicted underneath the gate and polyimide. Two lock-in amplifiers (LIA) were used to facilitate phase-sensitive detection in a four-terminal configuration. The upper LIA sourced ac excitation  $V_{sd}^{ac}$  and measured  $I_d$  while the lower was used to measure four-terminal voltage  $V_{4T}$ . A reference signal (ref) was provided for the second LIA to phase-lock to the  $V_{sd}^{ac}$  signal. A passive voltage divider (V.D.) was used to reduce  $V_{sd}^{ac}$  to 100  $\mu$ V. For HHMTs, a Keithley K2400 source-measure unit (SMU) was used to supply  $V_G$  and simultaneously measure gate leakage currents  $I_G$ . A Yokogawa 7651 voltage supply (V.G.) was used to source  $V_G$  to the gate of the hole billiard due to the superior signal-to-noise ratio.

# 2.4 Experimental methods and measurement systems

### 2.4.1 Electrical characterisation

The purpose of electrical measurements throughout this thesis was to examine the influence of the gate potential on conductance through a semiconductor channel of some form. All electrical set-ups and experiments therefore proceeded along the same basic lines. A constant voltage  $V_{sd}$  was applied to the source contact, and the gate voltage  $V_G$  was used to modulate the current  $I_d$  through the channel, as

measured at the drain. In the interests of maximising the signal-to-noise ratio, a lock-in amplifier was used to source  $V_{sd}$  and sense  $I_d$  where possible, enabling phase sensitive detection. A lock-in amplifier sources a sinusoidal ac excitation  $V_{sd}^{ac}$  at known frequency  $\omega_0$  and measures the root-mean-square value of a separate input signal  $V_{in}$ . In general the input signal is also a voltage, but most lock-in amplifiers contain a current pre-amplifying stage to convert a current signal to a voltage signal if this is desired. The phase sensitive detection occurs when the input ac signal is multiplied by the output excitation, and this signal is sent through a low-pass filter. This process returns a dc value  $\frac{1}{2}V_{sd}^{rms}V_{in}^{rms}$  when the input signal frequency is exactly the output frequency  $\omega_0$ ; otherwise it returns the dc value of zero.<sup>242</sup> In general, the input signal will consist of the desired  $I_d$  signal generated by  $V_{sd}^{ac}$ , plus noise in the form of sinusoidal components at other frequencies. The phase sensitive detector in a lock-in amplifier 'selects' only the component at the correct frequency, and rejects any components arising from noise. This enables a very high signal-to-noise ratio. In addition, the fact that the lock-in amplifier returns the root-mean-square  $I_d^{rms}$  means the measurement is directly comparable to a dc measurement.<sup>242</sup>

The circuit diagrams used for measurements on *p*-AlGaAs/GaAs heterostructure devices are shown in Fig. 2.11. An SR830 lock-in amplifier (LIA) was used to source  $V_{sd}^{ac}$  and measure  $I_d$  at either end of the Hall bar. A passive voltage divider (V.D.) was used to reduce the 1 V ac output of the LIA to 100  $\mu$ V at the device source contact. The LIA served as the virtual ground for the device at the drain. Note that this means the 2DHG was grounded, with the ac excitation  $V_{sd}^{ac}$  oscillating about zero. A four-terminal configuration was used to accurately measure the channel resistance/conductance. This was accomplished using a second LIA (EG&G 5210) that measured the voltage drop  $V_{4T}$  between the two outer arms of the Hall bar. A reference signal at the excitation frequency was sent to the second lock-in to facilitate phase sensitive detection of  $V_{4T}$ . Measuring the four-terminal voltage drop  $V_{4T}$  and resistance  $R = V_{4T}/I_d$  excludes the non-zero resistances of the ohmic contacts and the instrument cables, and gives an accurate measurement of the resistance of the channel itself.<sup>243</sup> This was particularly vital in the case of the hole billiard (Fig. 2.11(b)), which required accurate measurement of conductance G. Finally, the gate voltage  $V_G$  was modulated by either a Keithley K2400 sourcemeasure unit (SMU) or Yokogawa 7651 voltage supply (V.S.). The SMU sources a voltage  $V_G$  and measures the current to ground  $I_G$ . Ideally  $I_G = 0$  as the gate should be floating, however a high applied  $V_G$  can cause charge to 'leak' over the Schottky barrier at the surface, and through the heterostructure into the grounded 2DHG. Gate leakage currents also arise for devices with an insulating layer if  $V_G$ exceeds the breakdown voltage of the insulator (see Fig. 1.18 in Sec. 1.4.5). The K2400 features a built-in current limiting feature to prevent further increments in  $V_G$  if a user defined  $I_G$  is reached, typically 1-10 nA. This prevents potential damage to the device caused by large leakage currents. The K2400 was used for all measurements of HHMT devices in Chapter 3. The Yokogawa voltage source was used in measurements of the hole billiard due to the superior signal-to-noise ratio compared to the K2400 SMU. However, the SMU was used prior to collecting the measurements given in this thesis to check that  $I_G < 0.5$  nA throughout the  $V_G$ measurement range.

The electrical circuits used to measure NWFETs in Chapters 4 and 6 are presented in Fig. 2.12. For the multiple wrap-gated NWFETs and polymer electrolyte-gated NWFETs presented in Chapter 4, an SR830 LIA was again used to source  $V_{sd}^{ac}$ and measure  $I_d$ . The SR830 also houses four voltage sources which were used to supply  $V_G$  to each of the gates on the multiple wrap-gate devices (Fig. 2.12(a)). This was done so that nominally identical voltage sources were used for each gate. For polymer electrolyte-gated NWFETs, a low-noise Yokogawa GS210 voltage source was used to supply  $V_G$  (Fig. 2.12(b)). Measurements in Chapter 6 were conducted at the University of Queensland. For these experiments, a Keithley K2450 was used to supply a  $V_{sd}^{dc}$ . This SMU also houses a high accuracy current module that was used to measure  $I_d$  at the drain. As mentioned above, the rms values obtained from LIA measurements are directly comparable to those obtained from dc measurements. A Keithley K2400 was used to supply  $V_G$  and measure  $I_d < 0.1$  nA (Fig. 2.12(c)). We also explored the transient response of the PE-gated NWFET to square pulse train applied to  $V_G$  using the set-up shown in Fig. 2.12(d).

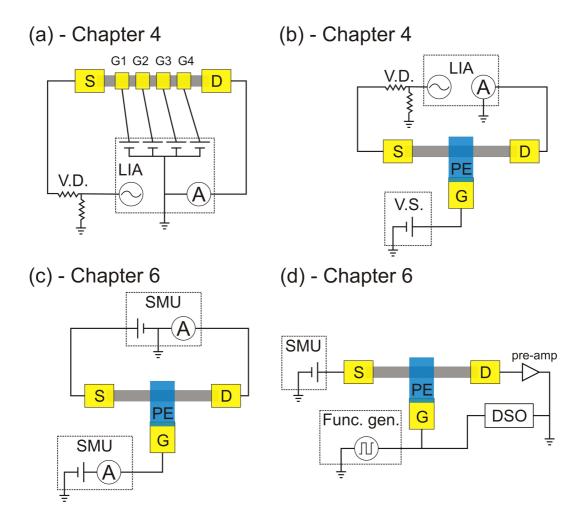


FIGURE 2.12: Electrical circuits used to measure NWFETs in (a/b) Chapter 4 and (c/d) Chapter 6. (a) For NWFETs with multiple metal/oxide wrapgates, an SR830 provided  $V_{sd}^{ac}$  and measured  $I_d$ . The SR830 contains four voltage sources, which were used to supply  $V_G$  to each of the wrap-gates. (b) For NWFETs with patterned polymer electrolyte gate dielectrics, a Yokogawa GS210 voltage source (V.S.) was used to supply  $V_G$ . An SR830 was used to provide  $V_{sd}^{ac}$  and measure  $I_d$ . (c) For humidity dependent measurements at the University of Queensland, two separate modules within a Keithley K2450 SMU were used to source  $V_{sd}^{dc}$  and measure  $I_d$ . A K2400 SMU sourced  $V_G$ . (d) In a separate configuration, an HP33120A function generator (func. gen.) supplied a series of voltage pulses to  $V_G$ . The resulting  $I_d$  was converted to a voltage using a Femto DLPCA200 current pre-amplifier (pre-amp) and measuring using an Agilent DSO-X 3024A digital signal oscilloscope (DSO). The K2450 was used to supply  $V_{sd}^{dc}$ .

The pulse train was supplied by a H.P. 33120A function generator. A low-noise Femto DLPCA200 current pre-amplifier converted the  $I_d$  signal to a voltage signal with a gain of  $10^5 - 10^7$  V/A, and this was measured by an Agilent DSO-X 3024A digital signal oscilloscope (DSO). The DSO also directly monitored the  $V_G$  pulses. Four-terminal measurements were not done on NWFETs as the morphology of the nanowires and the invasive, low resistance contacts means the contacts are either difficult to fabricate<sup>192</sup> or prevent a true four-terminal measurement due to challenges in correctly measuring the voltage of one dimensional modes.<sup>156,244</sup> In any case, the low contact resistance means that the 2-terminal and 4-terminal resistances of InAs nanowires differ by only ~ 5%.<sup>135,245</sup> The circuits used to measure  $I_d$  and thermovoltage due to quantum dot-like states in InAs nanowires built on the set-ups used in Chapter 4. They are presented in the relevant section of Chapter 5.

### 2.4.2 Cryogen dewar and dipstick

Many of the devices in this thesis were studied at low temperature to observe quantum effects, and/or to reduce noise and gate instability. A very simple and effective way to cool a sample is to immerse it in a cryogenic liquid. The most commonly used cryogens are liquid N<sub>2</sub> (boiling point T = 77 K) and liquid <sup>4</sup>He (boiling point T = 4.2 K). The cryogens were kept in double walled, vacuum shielded dewars to minimise thermal contact with the surrounding lab. The packaged samples were loaded into a socket on the end of the 'dipstick' pictured in Fig. 2.13. The pins on the socket connected to the pins on the LCC20 package. Electrical wires from each pin ran up through the dipstick to a break-out box with BNC connections. BNC cables were then connected between the break-out box and instruments.

The dipstick/dewar set-up could be used to measure devices in a continuous temperature range between the cryogen temperature and room temperature, inclusively. For measurement at T = 77 K or 4.2 K, the dipstick was lowered into

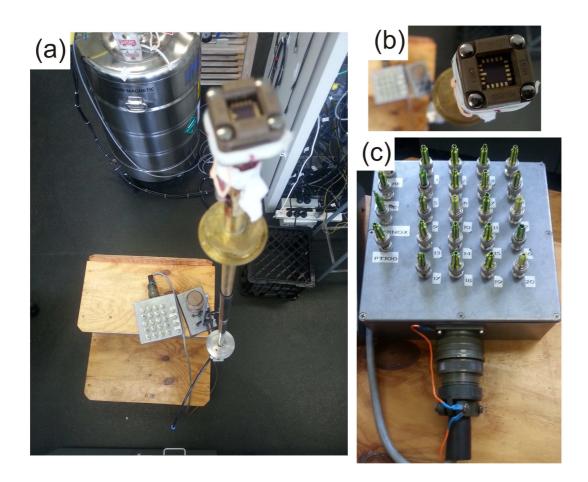


FIGURE 2.13: Photographs of the cryogen dewar and dipstick. (a) The dipstick features a socket to hold the LCC20 chip packages, which is electrically connected to the break-out box. The dipstick was inserted into the silver-coloured dewar in the background to obtain measurement temperatures down to 4.2 K. (b) Close-up of the underside of an LCC20 chip package contained within the dipstick socket. (c) Close-up of the break-out box, with 20 connectors corresponding to the 20 pins on the LCC20 packages.

the dewar until the sample was immersed in the cryogen. Stratification of the atmosphere above the liquid produces a vertical temperature gradient between the liquid and the top of the dewar. Therefore, fixing the sample position above the liquid facilitated measurement at temperatures 4.2 K < T < 300 K. The dipstick was also used outside the dewar in ambient to conduct the room temperature measurements of polymer electrolyte-gated NWFETs presented in Chapter 4.

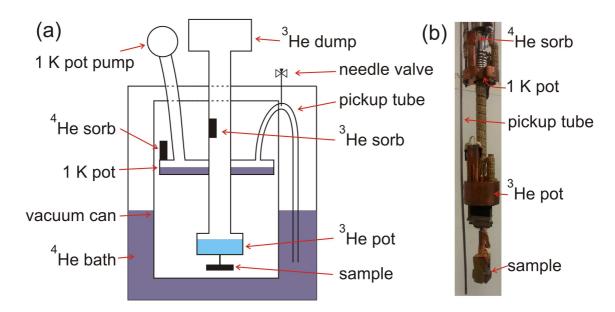


FIGURE 2.14: (a) Schematic of the Heliox <sup>3</sup>He cryostat. The system is contained within a vacuum can with a small amount of <sup>4</sup>He exchange gas. The cryostat is lowered into a <sup>4</sup>He bath, which brings the system to 4.2 K and causes the exchange gas to be absorbed by the <sup>4</sup>He sorb. Liquid <sup>4</sup>He is admitted to the 1 K pot by controlling the needle valve. Pumping on the evaporated liquid cools the 1 K pot to  $T \sim 1.5$  K. The 1 K pot is in thermal contact with the closed <sup>3</sup>He system, which condenses <sup>3</sup>He into the <sup>3</sup>He pot. Absorption of evaporated gas by the <sup>3</sup>He sorb effectively pumps on the <sup>3</sup>He and cools the pot to base temperature  $T_B = 275$  mK. The sample is kept in thermal contact with the <sup>3</sup>He pot. (b) Image of the lower section of the Heliox, showing the <sup>4</sup>He sorb wrapped in copper mesh, 1 K pot, pick-up tube, <sup>3</sup>He pot and sample socket.

### 2.4.3 Heliox <sup>3</sup>He cryostat

Temperatures  $T < 4.2 \ K$  can be obtained *via* evaporative cooling techniques. The principle is that the latent heat involved in evaporation of a liquid necessarily cools the remaining liquid. The cooling power is determined by the evaporation rate, which can be enhanced by pumping out the evaporated gas. By pumping on <sup>4</sup>He it is possible to cool the liquid to  $T = 1.2 - 1.5 \ K$ . Lower temperatures  $T \sim 230 \ m$ K can be obtained by using <sup>3</sup>He, which has a faster evaporation rate than <sup>4</sup>He, and thereby increased cooling power.<sup>246</sup>

The Heliox cryostat utilises three stages of immersion in a liquid <sup>4</sup>He bath and both <sup>4</sup>He and <sup>3</sup>He evaporative cooling to obtain  $T \sim 275$  mK.<sup>246,247</sup> A schematic of the equipment is given in Fig. 2.14. The closed <sup>3</sup>He section consists of a pot where the liquid will eventually settle and a dump where the gaseous <sup>3</sup>He is contained at room temperature. Having this section of the system tightly closed off facilitates simple operation and entails low risk of leaking expensive <sup>3</sup>He. It does mean, however, that pumping is not done using a typical mechanical pump. Rather, the behaviour of the <sup>3</sup>He is mediated by a charcoal 'sorb' mounted between the <sup>3</sup>He pot and dump, which absorbs <sup>3</sup>He gas for T < 10 K. In this way the sorb effectively acts as a miniature, automatic pump. The pumped <sup>4</sup>He aspect of the Heliox surrounds the '1 K pot'. A thin vacuum shielded pick-up tube runs from the 1 K pot into the <sup>4</sup>He bath to draw liquid <sup>4</sup>He into the 1 K pot. The flow is regulated by a needle valve. A pumping line runs out of the cryostat and is connected to a rotary pump, which is used to pump the 1 K pot to vacuum < 1 mbar before loading the cryostat. The entire cryostat is contained within a vacuum sealed can. A small amount, ~ 1 cm<sup>3</sup>, of <sup>4</sup>He exchange gas is admitted into the vacuum can to facilitate the initial cooling to 4.2 K. The sample is mounted in thermal contact with the <sup>3</sup>He pot.<sup>246,247</sup>

Cooling the Heliox to base temperature  $T_B \sim 275$  mK follows these four steps:<sup>247</sup>

- 1. Cooling to T = 4.2 K: The cryostat is loaded into the liquid <sup>4</sup>He dewar and immersed in the bath. The exchange gas in the vacuum can brings the cryostat into thermal equilibrium with the bath, at T = 4.2 K. A second charcoal sorb was included within the vacuum can – but outside the other cryostat components – to absorb the exchange gas as the cryostat cools. This restores vacuum between the cryostat and <sup>4</sup>He, which is desirable as the following steps cool the cryostat below 4.2 K. Thermal contact with the liquid <sup>4</sup>He bath from this point on would constitute an unwanted heat source.
- 2. Cooling the 1 K pot: A small amount of liquid <sup>4</sup>He is drawn into the 1 K pot *via* the pick-up tube. Pumping on the 1 K pot while drawing in liquid <sup>4</sup>He cools the 1 K pot to  $T \sim 1.5$  K.
- 3. Condensing <sup>3</sup>He: The 1 K pot is in thermal contact with the central section of the closed <sup>3</sup>He portion of the cryostat. This causes condensation of <sup>3</sup>He

into the pot, since 1.5 K is less than the <sup>3</sup>He boiling point of 3.2 K. The <sup>3</sup>He sorb is heated to T = 30 K to ensure that all <sup>3</sup>He is released, and ultimately condenses into the pot.

4. Cooling to base temperature  $T_B \sim 275$  mK: To cool the <sup>3</sup>He from 1.5 K, the sorb heater is turned off. The sorb slowly cools to  $T \sim 2$  K and begins to absorb <sup>3</sup>He. This essentially pumps on the <sup>3</sup>He and cools the pot to base temperature  $T_B \sim 275$  mK.

The Heliox holds base temperature for approximately 48 hours, limited by the finite supply of <sup>3</sup>He. As the <sup>3</sup>He evaporates and is absorbed, the thermal mass of the remaining liquid is slowly reduced and the <sup>3</sup>He pot temperature rises. Cooling the <sup>3</sup>He pot back to  $T_B$  requires repeating Steps 4 and 5 to expel all <sup>3</sup>He from the sorb, condense the gas into liquid and then cool the system to base temperature.

It is possible to obtain stable intermediate temperatures between 275 mK and 4.2 K by working in two distinct regimes, depending on whether the desired temperature is higher or lower than the temperature of the 1 K pot. Temperatures up to 1.5 K can be achieved by heating the charcoal <sup>3</sup>He sorb to moderate temperatures < 10 K. Heating reduces the pumping efficiency of the sorb, and therefore the cooling power of the evaporation process. The advantage of this method over directly heating the <sup>3</sup>He pot is that the latter accelerates evaporation and significantly reduces the available measurement time at  $T_B$ .<sup>246,247</sup> To obtain T > 1.5 K, the charcoal sorb is first heated to 20 K to release some of the <sup>3</sup>He gas and bring the <sup>3</sup>He pot into thermal equilibrium with the 1 K pot at T = 1.5 K. Heat is then applied directly to the <sup>3</sup>He pot until the required temperature is reached.<sup>246,247 3</sup>He evaporation in this regime is less of a concern as heating the sorb releases excess <sup>3</sup>He gas which is condensed when the sorb heater is turned off after measuring.

The Heliox cryostat was used to collect the results on quantum dot-system in InAs nanowires presented in Chapter 5.

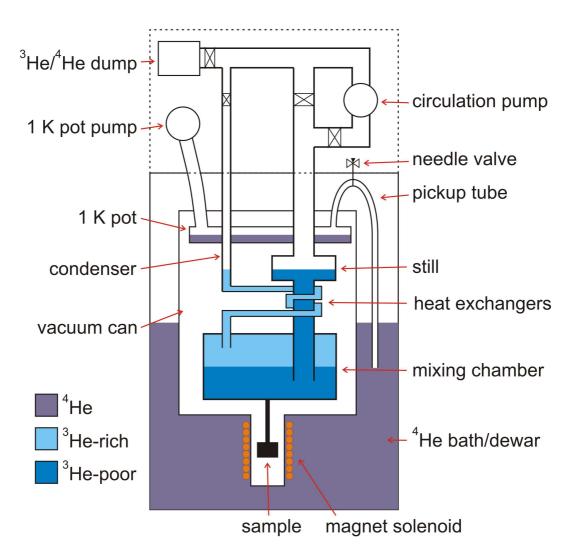


FIGURE 2.15: Diagram of Kelvinox K100 dilution refrigerator at base temperature,  $T_B = 40$  mK. The dilution unit is contained within a vacuum can secured in a dewar with <sup>4</sup>He bath. The bath initially cools the unit to 4.2 K and supplies <sup>4</sup>He to the 1 K pot via the pick-up tube. Pumping the 1 K pot cools it to  $T \sim 1.2$  K. The boundary of the phase separated <sup>3</sup>He/<sup>4</sup>He mixture is in the middle of the mixing chamber. The still draws only on the <sup>3</sup>He-poor phase. The circulation pump extracts evaporated <sup>3</sup>He and feeds it back to the condenser, where the gas is re-condensed by the 1 K pot (see Fig. 2.16 for more detail). The condenser tubes are wrapped around the pipes below the still to act as a heat exchanger; this cools the re-condensed gas to the temperature of the mixing chamber. The sample is secured to the mixing chamber on a post that fits within the bore of a superconducting solenoid magnet. The magnet is held within the <sup>4</sup>He bath. The gaseous <sup>3</sup>He/<sup>4</sup>He mixture is stored in the dump when not in use. The dump is part of a room temperature gas handling system, contained within the dashed lines.

### 2.4.4 Kelvinox K100 dilution refrigerator

Obtaining even lower measurement temperatures requires more complex cooling methods. One such method relies on drawing <sup>3</sup>He across a phase boundary between two different dilutions of a <sup>3</sup>He/<sup>4</sup>He mixture. The operating principle is as follows. Below a critical temperature, liquid <sup>3</sup>He/<sup>4</sup>He mixtures separate into two phases with different dilutions; a <sup>3</sup>He-rich and <sup>3</sup>He-poor phase. The comparatively light <sup>3</sup>He-rich phase floats on the heavier <sup>3</sup>He-poor phase. <sup>3</sup>He atoms can be induced to cross the phase boundary in a process called mixing. Transferring <sup>3</sup>He across the phase boundary requires energy, in the form of latent heat. This can be understood by analogy to evaporative cooling. The transfer of <sup>3</sup>He from the <sup>3</sup>He-rich phase to the <sup>3</sup>He-poor phase is similar to the phase transition as atoms evaporate from a liquid; the latent heat required to facilitate the phase transition necessarily cools the <sup>3</sup>He/<sup>4</sup>He mixture.

Dilution refrigerators enable the realisation of this powerful cooling process.<sup>82,248</sup> A diagram of the K100 used in this thesis is shown in Fig. 2.15. The dilution unit contains the mixing chamber, which is connected on one side by condenser lines, and on the other side by a chamber called the still. The sample is mounted in thermal contact with mixing chamber, and is positioned inside the bore of a superconducting solenoid, capable of delivering magnetic fields up to  $B = \pm 10$  T. The solenoid is secured in the surrounding <sup>4</sup>He bath. Initial cooling occurs in the same manner as the Heliox. The unit is contained within a vacuum can that features a small amount of exchange gas and a charcoal sorb. The exchange gas cools the dilution unit to 4.2 K when brought into thermal contact with the <sup>4</sup>He bath. The sorb automatically adsorbs the exchange gas while this occurs, thermally isolating the dilution unit for T < 4.2 K. The <sup>3</sup>He/<sup>4</sup>He mixture is condensed via thermal contact with a 1 K pot. A pick-up tube feeds <sup>4</sup>He into the 1 K pot and pumping on the <sup>4</sup>He in the pot cools it to  $T \sim 1.2$  K. The room temperature gas-handling system directs the gaseous  ${}^{3}\text{He}/{}^{4}\text{He}$  mixture flow through the system. Briefly, the values are initially set to direct the  ${}^{3}\text{He}/{}^{4}\text{He}$ from the dump into the dilution unit for condensing. The configuration is then

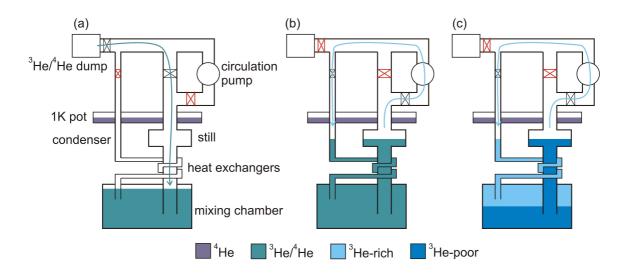


FIGURE 2.16: Cooling process for the K100 dilution refrigerator. Open (closed) valves are represented in black (red) and the arrows indicate the direction of gas flow. (a) The 1 K pot temperature is set to ~ 1.2 K and <sup>3</sup>He/<sup>4</sup>He gas flow is directed into the mixing chamber on the still side. Thermal contact with the 1 K pot condenses the mixture. (b) After condensation, the mixture  $T \sim 1.2$  K. The dump is sealed, and gas is directed from the still to the condenser through the circulation pump. Pumping on the still reduces the mixture temperature through evaporative cooling. <sup>3</sup>He is preferentially evaporated due to the superfluid state of the heavier <sup>4</sup>He atoms. (c) Below the critical temperature  $T_C = 870$  mK, phase separation occurs. The circulation pump draws <sup>3</sup>He from the still, which causes <sup>3</sup>He in the mixing chamber to cross the phase boundary, cooling the mixing chamber to base temperature  $T_B = 40$  mK.

switched to circulate the mixture and cool the system to base temperature. In detail, reaching base temperature in the dilution refrigerator is achieved with the following steps, following Fig. 2.16:

1. Condensing the mixture: The 1 K pot temperature is set to T = 1.2 K by adjusting the needle valve and pumping on the <sup>4</sup>He. The initial condensing process is actually done *via* the still. The larger pipes on this side make the process faster and ensure that any contaminants, e.g., oxygen and nitrogen, do not solidify in the thin condenser tubes and cause a blockage. Figure 2.16(a) shows the valves are set to flow <sup>3</sup>He/<sup>4</sup>He gas through the still. Thermal contact with the 1 K pot condenses the mixture into the mixing chamber.

- 2. Circulation: Once all the mixture has been condensed into the mixing chamber, the  ${}^{3}\text{He}/{}^{4}\text{He}$  dump value is closed, and the gas system set up to circulate the mixture (Fig. 2.16(b)). A rotary pump is used to extract evaporated He from the still, reducing the temperature of the mixture by evaporative cooling. As the temperature is reduced, the faster evaporation rate of  ${}^{3}\text{He}$ compared with <sup>4</sup>He means that gas flow consists almost entirely of <sup>3</sup>He. This occurs because the heavier, bosonic <sup>4</sup>He atoms condense into a strongly interacting superfluid that resists evaporation. Meanwhile, the weaker interactions between the lighter <sup>3</sup>He fermions means they continue to undergo evaporation. The <sup>4</sup>He component can essentially be considered a static and inert background for the movement of <sup>3</sup>He.<sup>82,248</sup> The evaporated <sup>3</sup>He is fed through liquid  $N_2$  and liquid <sup>4</sup>He traps to solidify and remove contaminants before the gas is returned to the condenser pipes. Once in the condenser lines, thermal contact with the 1 K pot condenses the <sup>3</sup>He into the mixing chamber. The condenser tubes are wrapped around the pipe below the still. This acts as a heat exchanger to ensure that the temperature of the re-condensed <sup>3</sup>He quickly comes to equilibrium with the liquid already in the mixing chamber.
- 3. Phase separation: The process of evaporation and circulation cools the mixture to below the critical temperature for phase separation,  $T_C = 870$  mK. The component size and mixture volume are matched to meet three conditions: First, the phase boundary occurs in the mixing chamber. Secondly, the still draws on only the <sup>3</sup>He-poor phase. Thirdly, the condenser returns liquid directly into the <sup>3</sup>He-rich phase (Fig. 2.16(c)). The continuing circulation process evaporates <sup>3</sup>He at the still and consequently draws <sup>3</sup>He across the phase boundary in the mixing chamber. This occurs because the <sup>3</sup>Hepoor phase has a minimum <sup>3</sup>He concentration, even at 0 K.<sup>82,248</sup> As a result, any <sup>3</sup>He evaporated from the <sup>3</sup>He-poor phase at the still must be replaced from the <sup>3</sup>He-rich phase at the mixing chamber. The subsequent <sup>3</sup>He transfer across the phase boundary cools the mixing chamber to  $T_B = 40$  mK.

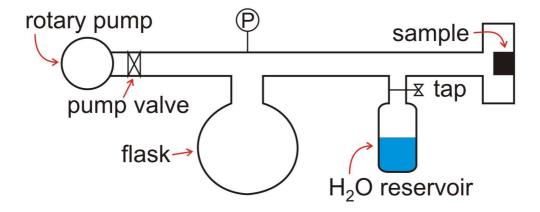


FIGURE 2.17: Hydration dependent measurement chamber schematic. The rotary pump evacuated the chamber and was then isolated by the pump valve.  $H_2O$  vapour was admitted to the chamber from the reservoir *via* the bleed valve, which hydrated the sample. The flask increased the chamber volume to ensure that the pressure P read on the gauge matched that at the sample location.

The circulation and cooling process continues indefinitely once the mixing chamber reaches base temperature, assuming there are no unintentional heat loads on the mixing chamber or 1 K pot. This is a highly favourable aspect compared to, e.g., the Heliox, as it facilitates continuous measurement at base temperature for as long as the system is stable. The indefinite circulation process also means that higher temperatures can be obtained by directly heating the mixture chamber. However, this is limited to  $T \sim 900$  mK due to the critical temperature of the phase separation. Above this temperature, the system becomes very unstable, meaning the next viable measurement temperature is at T = 4.2 K, in thermal equilibrium with the <sup>4</sup>He bath.

The Kelvinox K100 was used to collect data from the hole billiard, presented in Chapter 3.

### 2.4.5 Hydration dependent measurement chamber

Chapter 6 presents results on hydration-dependent transport properties of polymer electrolyte-gated NWFETs. These measurements were obtained in a hydration controlled measurement chamber housed at the University of Queensland, Brisbane. A schematic of the equipment is shown in Fig. 2.17. The sample was mounted inside a vacuum chamber connected to a  $H_2O$  reservoir by a bleed valve. The deionised Millipore  $H_2O$  was degassed prior to measurements by three freeze-pump-thaw cycles.<sup>249–251</sup> This procedure operates as follows: pumping on the reservoir while the  $H_2O$  is frozen removes gas in the reservoir. Thawing the  $H_2O$  releases any gas, e.g.,  $N_2$ ,  $O_2$ , previously contained in the liquid. The  $H_2O$ is then re-frozen to facilitate pumping out this gas. The procedure was repeated until the gas was removed from the  $H_2O$ , and the reservoir was left at vacuum.

To take measurements, the chamber was evacuated overnight using a rotary pump. Once the chamber was at vacuum, the pump valve was closed. Water vapour was then admitted from the H<sub>2</sub>O reservoir by opening the bleed valve. The water vapour pressure was monitored on the pressure gauge, and the bleed valve closed when the desired pressure was reached. Electrical measurements were then performed at constant water vapour pressure. In this way the device hydration was increased in discrete steps until the saturation vapour pressure was reached,  $P_{sat} = 24$  mbar at  $T = 20^{\circ}$ C. The large round bottom flask increased the chamber volume to ensure a uniform atmosphere. A small chamber volume raises the probability of non-uniformities in atmospheric pressure. Increasing the chamber volume ensured the pressure at the gauge was equal to the pressure at the sample.<sup>249</sup>

### 2.4.6 Photoluminescence and X-ray photoelectron spectroscopy

In Chapter 3 I explore the efficacy of  $(NH_4)_2S_x$  treatments in passivating (311)A GaAs surfaces. In addition to measuring the electrical properties of treated HHMTs, I characterised the treated surfaces using photoluminescence (PL) and X-ray photoelectron spectroscopy (XPS). PL and XPS are two of the most commonly used methods to evaluate the efficacy of passivation treatments<sup>44,53,252</sup> and examine the surface chemistry.<sup>56–58,253</sup>

XPS uses the photoelectric effect to determine the binding energy  $E_B$  of inner shell electrons.<sup>254</sup> The underlying physics is depicted in Fig. 2.18. X-rays incident on

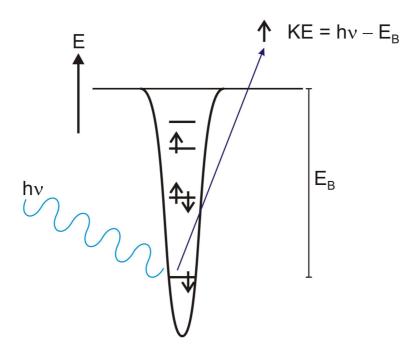


FIGURE 2.18: X-ray photoelectron spectroscopy is enabled by the photoelectric effect. Incident X-rays with known energy  $h\nu$  eject inner shell electrons. Measurement of the ejected electrons' kinetic energy  $KE = h\nu - E_B$  allows the binding energy  $E_B$  to be determined

the sample eject inner shell electrons by absorbing the photon. The electron first expends energy to overcome the binding energy of the atomic potential, and any excess energy manifests as kinetic energy. The binding energy  $E_B$  can be inferred by measuring the kinetic energy KE of emitted electrons, using:

$$E_B = h\nu - KE \tag{2.1}$$

where the incident X-ray energy  $h\nu$  is accurately known.

The use of XPS as a chemical analysis technique centres on the fact that different chemical bonds shift the electron binding energy relative to the binding energy in the isolated atom.<sup>254</sup> In the context of sulfur passivation of the GaAs surface, it is possible to distinguish between the binding energies of electrons emanating from, e.g., As-O bonded species and As-S bonded species. I used XPS in Chapter 3 to ensure that  $(NH_4)_2S_x$  treatment removed the surface oxides and generated a sulfur bonded monolayer. This process has been well established for (100) GaAs.<sup>56–58,253</sup>

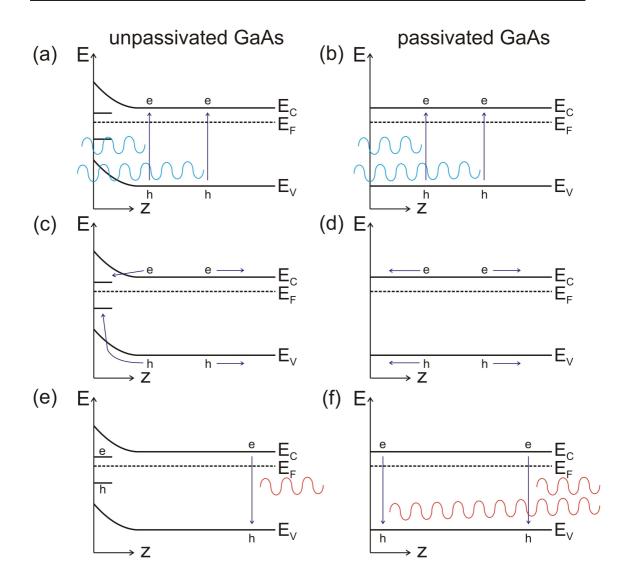


FIGURE 2.19: Photoluminescence of (a/c/e) a bare GaAs surface with a high surface state density and (b/d/e) a passivated GaAs with low surface state density. (a/b) A laser is used to generate electron-hole pairs. (c/d) The electrons/holes diffuse randomly in the sample and may become trapped in surface states. (e/f) Electron/hole recombination emits photons at the band gap energy  $h\nu = E_C - E_V$ . (e) The recombination life-time of charges trapped in surface states is much longer than that of electron/hole pairs in the continuous bands. (f) This leads to a higher PL intensity for surfaces free of band-gap states.

XPS was performed by Dr Bill Bin Gong at the Mark Wainwright Analytical Center at UNSW. Measurements were done using a ThermoScientific ESCALAB250Xi system with a monochromated Al K $\alpha$  X-ray source ( $h\nu = 1486.68$  eV at 164 W power). The penetration depth of XPS is such that only the top 1 – 10 nm of the material is probed. Binding energies were measured in reference to the C(1s) binding energy at 285.0 eV, and the expected uncertainty in all measurements was  $\pm 0.1$  eV. The measured binding energy peaks were fitted using the Advantage software package, based on the expected binding energy of electrons arising from each particular chemical species.

Photoluminescence (PL) was used to compare the relative surface state density of treated and untreated samples. The process is illustrated in Fig. 2.19. A laser is used to excite electron/hole pairs across the GaAs band gap. After diffusing for a short time, the pairs may recombine, or become trapped in surface states. Pair recombination emits luminescence consisting of photons with energy at the GaAs band-gap energy,  $E_g = 1.42$  eV. The luminescence intensity is directly proportional to the number of radiative recombination events at the band edges. Charges trapped in surface states do not contribute to this signal. As such, a sample with a low surface state density will have a high PL intensity compared to a sample with a high surface state density.<sup>44,53,252</sup>

It is worth noting that an increased PL intensity can also arise if the treatment modifies the surface state spectrum such that the surface band-bending increases. This presents as a surface barrier that inhibits tunnelling into the surface states. The consequence is that PL intensity may increase without a corresponding decrease in overall surface state density. As such it is important to conduct secondary experiments to corroborate PL findings. In this thesis, I show that results obtained from PL were consistent with the electrical performance of HHMTs. PL experiments were performed using the 488 nm line of an Ar ion (Coherent Innova 70). Luminescence was detected using a CCD camera coupled to a Spex 270M grating spectrometer with spectral resolution of 3 nm.

### 2.5 Summary

This Chapter outlined the fabrication and experimental techniques utilised in this thesis. This included photolithography and electron-beam lithography based strategies for generating micro- and nano-scale patterns *via*, e.g., metal deposition and selective etching. I then described how these techniques were applied to the specific GaAs/AlGaAs heterostructure devices and nanowire transistors investigated in this thesis. I also discussed the electrical measurement techniques employed to evaluate the field-effect related behaviour of these devices – including phase-sensitive detection of signals using lock-in amplifiers and four-terminal methods for accurately measuring sample resistance – and how these were applied to each device. Finally, I outlined the operation of each measurement system used, from room temperature down to T = 40 mK, and explained the principles behind photoluminescence and X-ray photoelectron spectroscopy, which were used to characterise sulfur-treated GaAs surfaces. The following chapters present the experimental results collected during the course of my research.

### Chapter 3

## Results: The impact of surfaces and dopants on instability in *p*-type AlGaAs/GaAs heterostructures

Section 1.4 outlined how the AlGaAs/GaAs materials system has facilitated the realisation and study of a wealth of quantum phenomena including one dimensional ballistic transport and artificial atoms. There is a great interest in studying the corresponding hole systems, but doing so has been severely hampered by the instability of hole devices manufactured from both Si-doped and C-doped *p*-AlGaAs/GaAs modulation doped wafers<sup>24,107–110,255</sup> as outlined in Sec. 1.4.5. During my honours year I contributed to a study which suggested that both charge trapping in (311)A surface states and charge migration between acceptor impurities in the AlGaAs layer contribute to gate hysteresis and instability in these wafers.<sup>24</sup> Our subsequent interest was in exploring the properties of both the (311)A GaAs surface and *p*-type Si dopants in AlGaAs. The aim was to more closely understand the underlying physics behind mechanisms causing gate instability and hysteresis, and potentially help to enable successful fabrication of stable devices based on *p*-type modulation doped AlGaAs/GaAs heterostructures. For the surface aspect, we explored the use of  $(NH_4)_2S_x$  treatments that have had extensive use as surface passivation for (100) GaAs.<sup>32</sup> As discussed in Sec. 1.3.2, sulfur passivation strips the native oxide and covalently satisfies both Ga- and As-related dangling bonds to result in a surface free of band-gap states.<sup>21,41,43,44,55</sup> To study dopant dynamics we fabricated an etched hole billiard in the hysteretic wafer, and studied changes in magnetoconductance fluctuations (MCF) induced in response to applied  $V_G$ and thermal cycling.<sup>33</sup> MCF in billiards is a phase-sensitive phenomenon arising from interference of electron/hole trajectories between entrance and exit.<sup>82,83</sup> The MCF are determined by both device geometry and background disorder potential; as such they can be used to provide a 'fingerprint' for the charge configuration of remote ionised dopants in the AlGaAs layer (see Secs 1.4.3 and 1.4.4).<sup>84,90,91</sup> I present results from both of these studies in this chapter.

### 3.1 Efficacy of $(NH_4)_2S_x$ solutions on the (311)A GaAs surface

### 3.1.1 Experiment and sample outline

The best case scenario towards improving (311)A AlGaAs/GaAs heterostructure performance would be gaining device stability through a simple surface treatment that provides effective surface state passivation. This was not necessarily going to be straightforward; despite significant work on lower index planes such as (100), (110) and (111) due to their importance for industrial device development, passivation of the (311)A GaAs surface has received little to no attention. Nevertheless, the success of  $(NH_4)_2S_x$  solutions for (100) GaAs made these a natural choice in our attempt to reduce the (311)A GaAs surface state density. Additionally, dipping the sample in a solution prior to gate deposition constitutes simpler fabrication compared to, e.g., gas deposition<sup>256</sup> or specialised oxide deposition.<sup>257</sup> Applying  $(NH_4)_2S_x$  to a new surface gave me the opportunity to explore the efficacy of the

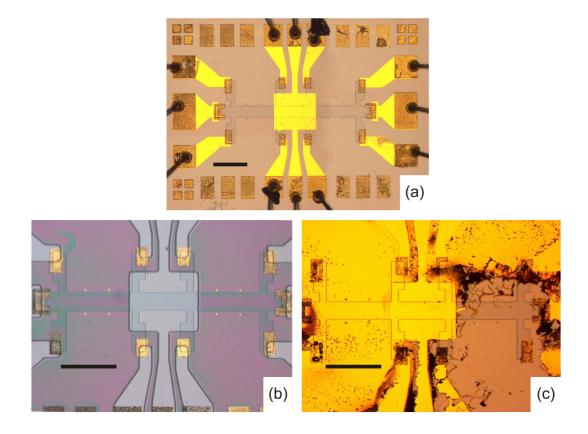


FIGURE 3.1: Optical micrographs of HHMTs at various stages. (a) Completed HHMT featuring Hall bar (light brown outline) contacted with AuBe Ohmic contacts (grainy gold) and a Ti/Au gate (central yellow square). (b) HHMT with gate pattern defined in photoresist layer (purple) prior to passivation and metallisation. (c) HHMT after treatment with alcoholic  $(NH_4)_2S_x$  solution and subsequent metallisation. The alcoholic solution dissolved much of the photoresist, leaving unwanted metal over large regions of the device. Scale bars represent 300  $\mu$ m.

treatment. In addition to measuring the electronic properties of treated AlGaAs/-GaAs devices, we characterised the effect of  $(NH_4)_2S_x$  treatments using X-ray photoelectron spectroscopy (XPS) and photoluminescence (PL).

The devices used for electrical characterisation were HHMTs fabricated from (311)A-oriented *p*-AlGaAs/GaAs heterostructures, with an overall top gate that modulated the 2DHG density (see Fig. 3.1(a)). Heterostructure layer depths, doping levels and fabrication methods were outlined in Sec. 2.2. We used XPS to confirm that the  $(NH_4)_2S_x$  treatment stripped the native oxide layer and produced Ga-S and As-S bonds at the (311)A and (100) surfaces. This was done using samples from the (311)A-oriented *p*-AlGaAs/GaAs and the matched (100)

*n*-AlGaAs/GaAs heterostructure grown alongside it. XPS probes only the top 5 nm of the sample, and thus interacted only with the GaAs capping layer. Finally, we utilised PL as a secondary tool to evaluate the passivation efficacy of  $(NH_4)_2S_x$ . An effective surface passivation reduces the number of non-radiative recombination pathways, increasing PL intensity.<sup>252</sup> The laser penetration depth of ~ 80 nm meant that the heterostructures could not be used here; I show in Sec. 3.1.4 below that band bending in the underlying structure caused recombination to take place well away from the surface. As a result electron/hole recombination and PL signal were not limited by surface effects for modulation doped heterostructures. Therefore we used bulk (100) and (311)A wafers with no underlying heterostructure. These wafers were similar to those used as the basis for the epitaxial heterostructure growth.

Two treatments were prepared as outlined in Sec. 2.2.1 based on the  $(NH_4)_2S_x$ stock solution. The 'weak' treatment consisted of a 2 min dip in a 0.5% dilution of the stock solution in  $H_2O$ . The 'strong' treatment involved immersion in the stock solution for 10 mins. The weak treatment was used on most devices due to concerns about the ability of  $(NH_4)_2S_x$  to etch GaAs. Etching the thin 5 nm GaAs cap on the heterostructure wafer would risk exposing the AlGaAs layer, which strongly oxidises if exposed to air. A 30 s dip in 31% HCl:H<sub>2</sub>O prior to sulfur treatment was used to ensure the surface was de-oxidised. We restricted our attention to aqueous  $(NH_4)_2S_x$  treatments due to compatibility issues with photoresist. To limit the exposure to light and oxygen, the gate pattern was defined in the photoresist layer prior to treatment (see Fig. 3.1(b)). This enabled the devices to be loaded into the metal evaporator and under vacuum of < 1 mTorr in less than 5 minutes. Alcoholic solutions based on, e.g., isopropanol and 2-methyl-2-propanol have been suggested as more effective in some circumstances.<sup>21,44</sup> We found that immersion in these solutions dissolved significant portions of the photoresist during the 2 minute weak treatment; often the resist was removed entirely. Figure 3.1(c) shows that attempting metal evaporation after treatment in a 2% mixture of the stock solution in isopropanol prior to gate deposition resulted in unwanted metal covering large regions of the device.

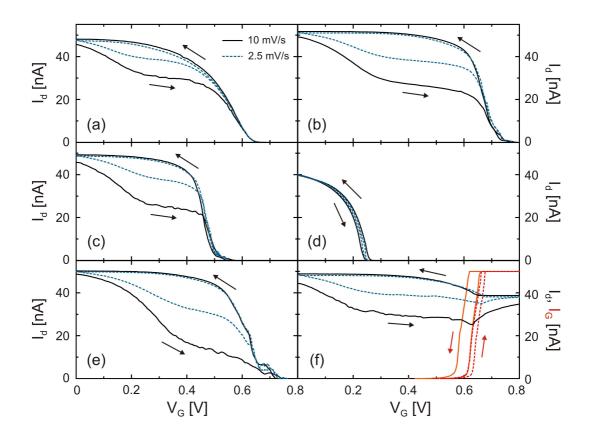


FIGURE 3.2:  $I_d$  vs  $V_G$  for devices with (a) no treatment, (b,c,d) weak treatment, (e) strong treatment and (f) strong treatment plus annealing at 360°C prior to gate deposition. The gate was incremented at 10 mV/s (black solid line) and 2.5 mV/s (blue dashed line). The device in (f) exhibited gate leakage with current  $I_G$  (red solid and dashed lines) above  $V_G = 0.61$  V.

## 3.1.2 Performance of $(NH_4)_2S_x$ treated heterostructure devices

The  $I_d$  vs  $V_G$  characteristic for an untreated *p*-AlGaAs/GaAs device at T = 4.2 K is shown in Fig. 3.2(a). An ac excitation  $V_{sd} = 100 \ \mu\text{V}$  at 73 Hz was applied to the source and  $I_d$  detected at the drain by a lock-in amplifier.  $V_G$  was supplied by a Keithley K2400 source measure unit, which simultaneously monitored gate leakage current  $I_G$ . The anti-clockwise, rate-dependent hysteresis loop is consistent with trapping of charge in long lived surface states and charge migration amongst dopants in the AlGaAs layer.<sup>24</sup> The data can be compared with Fig. 1.7, albeit with a reversed hysteresis direction due to the reversal of the sign of the charge carriers. A striking feature of the data in Fig. 3.2(a) is the  $I_d$  plateau at intermediate  $V_G$ 

on sweeps towards depletion. Comparison of these devices with *n*-AlGaAs/GaAs HEMTs at  $T \ge 120$  K in Ref. 24 led us to conclude that the  $I_d$  plateau was due to the gate induced filling of a high density of surface states.<sup>24</sup> This is supported by the results to follow. During this plateau, the  $I_d$  dependence on  $V_G$  is reduced significantly as the electric field causes charge to be trapped in surface states at the expense of causing depletion of holes in the 2DHG. Depletion resumed only once all the surface states were filled, allowing the device to reach pinch-off, i.e., where  $I_d = 0$ . The plateau length and pinch-off voltage  $V_{po}$  therefore provide a measure of relative surface state density between devices; a longer plateau and higher  $V_{po}$  implies a higher surface state density. One example of this was shown in Fig. 1.18 of Sec. 1.4.5, where adding an Al<sub>2</sub>O<sub>3</sub> insulating layer between the gate and semiconductor increased the plateau length to 3 - 4 V, consistent with increased Ga-O and As-O bonding and other interface defects.<sup>40,258</sup>

To investigate the effectiveness and reliability of  $(NH_4)_2S_x$  treatments on (311)AGaAs, the weak treatment was applied to five separate devices. Electrical characteristics from three of these devices are shown in Figs 3.2(b/c/d). There is significant variability in device characteristics, especially  $I_d$  plateau length and  $V_{po}$ . The latter varied between  $V_{po} = 0.25$  to 0.8 V across the five measured devices, despite each undergoing a nominally identical surface treatment prior to gate deposition. The  $I_d$  vs  $V_G$  characteristics of the two other devices we measured, but have not shown here, most resemble the device in Fig. 3.2(b). This same variability is not seen in devices with no sulfur treatment; the characteristics all strongly resemble those in Fig. 3.2(a). This suggests that the treatment altered (311)A GaAs surface states, and supports the designation of the  $I_d$  plateau as arising from surface states. However,  $(NH_4)_2S_x$  treatments clearly produced a highly variable effect on mid-gap surface state density. The longer  $I_d$  plateau and higher  $V_{po}$  for the device in Fig. 3.2(b) suggests an *increase* in surface state density. By contrast, the  $I_d$  plateau length – and therefore surface state density - was reduced for the device in Fig. 3.2(c) and eliminated entirely for the device in Fig. 3.2(d). The dramatic reduction in hysteresis for this latter device unfortunately was an isolated occurrence; although the data was repeatable for the same

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device on separate cool downs, the low hysteresis was not reproducible for any other subsequently fabricated device. Additionally, it is not clear that the data obtained from this device was a result of complete surface state elimination. It could be that the surface state spectrum was shifted away from the Fermi level at  $V_G = 0$  V. This could have facilitated pinch-off at lower  $V_G$ , before the Fermi level reached the highest surface state density. The two devices in Figs 3.2(b/c) also underwent increased depletion before the onset of the  $I_d$  plateau compared to the untreated device in Fig. 3.2(a). The device in Fig. 3.2(d) may simply represent an extreme variation of this behaviour.

A possible concern about the devices in Figs 3.2(b-d) is that the weak treatment was insufficient to provide complete passivation. To test this, the strong treatment was applied to the device in Fig. 3.2(e). This treatment modified the form of the hysteresis, facilitating increased depletion at low  $V_G$  and weakening the  $I_d$ plateau. However,  $V_{po}$  was not significantly altered. This is indicative of a change in the surface state distribution rather than an appreciable reduction in the overall density. A further issue is that instability increased at low  $I_d$  around pinch-off. The QPCs and quantum dots outlined in Sec. 1.4 all operate in this region, e.g., quantum dots operating down to the last electron typically have  $I_d$  in the nA to pA range. The introduction of noise/instability at low  $I_d$  would make any treatment incompatible with low dimensional electrostatic confinement.

One commonly used method to further alter the surface state distribution is postpassivation annealing.<sup>44</sup> The differing energetics and stability of Ga-related and As-related bonds means that annealing the treated (100) GaAs suface to over 350°C transfers sulfur atoms from As-S bonds to Ga-S bonds.<sup>59–61</sup> This can be beneficial or detrimental, depending on the relative importance of these bonds for a particular application. To ascertain the effect this has on the (311)A-oriented devices, we annealed one device to 360°C for 10 mins in Ar after treatment with the strong solution. Annealing at this temperature would not be possible with the photoresist already on the surface; the resist would either strongly adhere to the surface or undergo pyrolysis. As such we performed treatment and annealing prior to photoprocessing. Gate definition and metallisation was conducted as soon as possible afterwards, ensuring the device was kept in the dark as much as possible. I showed in Ref. 32 that the passivation effect can be preserved on (100) surfaces throughout the photoprocessing process by minimising exposure to illumination. Doing so is not ideal, hence our earlier desire to limit treatments to those compatible with photoprocessing. The data in Fig. 3.2(f) suggests that annealing was detrimental to device performance, especially in the re-emergence of a prominent  $I_d$  plateau compared to Fig. 3.2(e). Significant gate leakage occurred for  $V_G > 0.61$  V, possibly due to additional diffusion of Be from the ohmic contacts shorting to the gate. The fast diffusion of Be in GaAs is well established,<sup>239</sup> and although the temperature of this anneal was lower than that used to form ohmic contacts, the 10 minute annealing time was much longer. These results point to importance of As-S bonding in the partial passivation observed for the other (311)A devices.

The device characteristics in Fig. 3.2 show that the  $(NH_4)_2S_x$  treatment modifies the surface properties of the (311)A GaAs surface. However, it did not provide effective passivation and the resulting device behaviour was highly variable. We turned to XPS and PL on both (100) and (311)A GaAs surfaces to ensure that our treatment was correctly formulated and reproduced literature results. PL was also used to provide a secondary evaluation of the effectiveness of the solution in terms of reducing (311)A GaAs surface state density.

### 3.1.3 XPS of $(NH_4)_2 S_x$ treated surfaces

XPS was performed on four different samples: an untreated reference sample from each of the (100) and (311)A heterostructure wafers, and one sample each from the (100) and (311)A wafers that underwent the weak treatment. The samples were kept in the dark for the  $\sim 10$  mins prior to being loaded into the XPS chamber and placed under vacuum. XPS and peak fitting was conduced by Dr Bill Bin Gong at the Mark Wainwright Analytical Centre, UNSW. The As(2p<sub>3/2</sub>), Ga(2p<sub>3/2</sub>) and S(2p) core level spectra are most relevant for GaAs S-passivation, and data from each of the four devices are shown in Fig. 3.3.

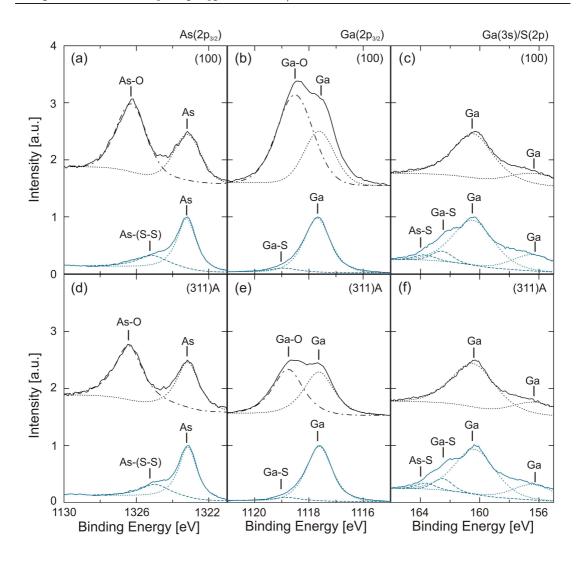


FIGURE 3.3: X-ray photoelectron spectra for (a/b/c) (100) oriented and (d/e/f) (311)A oriented GaAs surfaces without (black) and with (blue) weak  $(NH_4)_2S_x$  treatment. Solid lines represent the measured XPS signal. Dashed and dotted lines are fits to the marked peaks, assigned according to the expected bond binding energy. Each signal was normalised to the relevant Ga/As peak, and the black traces were offset vertically by 1.5 for clarity.  $(NH_4)_2S_x$  treatment stripped the surface oxide and resulted in Ga-S and As-S bonding for both (100) and (311)A surfaces.

Looking first at the As $(2p_{3/2})$  in Figs 3.3(a/d), both untreated samples show clear peaks related to GaAs and As<sub>2</sub>O<sub>3</sub> at 1323.2 eV and 1326.3 eV, respectively. The latter peak was eliminated in the treated samples and replaced by a peak at 1324.5 eV that is commonly associated with As-(S-S), i.e., the formation of disulfide bridges bonded to surface As.<sup>57</sup> The As-(S-S) peak intensity was the same for both (100) and (311)A GaAs. Similarly, the untreated surfaces in Fig. 3.3 show Ga(2p<sub>3/2</sub>) peaks related to GaAs and Ga<sub>2</sub>O<sub>3</sub> at 1117.4 eV and 1118.5 eV, respectively. The peak height for Ga<sub>2</sub>O<sub>3</sub> on the (311)A surface was twice as small as that for (100). This likely reflects the different valency of Ga surface atoms on the two surfaces;<sup>105,259,260</sup> (100) Ga surface atoms are divalent and hence bind twice as much oxygen as the monovalent Ga atoms at the (311)A surface. After treatment, the Ga-O peaks were removed. A small peak at higher binding energy, ~ 1119 eV emerged. Due to the low intensity it is not clear whether this can be conclusively assigned to Ga-S bonds.<sup>253</sup> It may have arisen from a residual Ga-O component also. Finally, the Ga(3s) spectra is not sensitive to oxygen bonding, and thus only shows two GaAs related bonds for untreated samples (Figs 3.3(c/f)). The interest in this binding energy range is that the S(2p) spectra overlap the Ga(3s) spectra. The spectra of treated samples includes Ga-S and As-S related bonds, with a similar peak intensity on both treated (100) and (311)A GaAs.<sup>253,261</sup> The As-S signal is possibly related to the disulfide bridges seen in the As(2p<sub>3/2</sub>) spectra.<sup>261</sup>

The XPS results show that the weak  $(NH_4)_2S_x$  treatment provided the expected result for (100) GaAs surface chemistry; the oxide layer was removed and replaced by Ga-S and As-S bonds. The similarity of results for (311)A strongly supports the notion that de-oxidisation and sulfidisation occurs equivalently for (311)A also. However, the ineffective passivation observed from device performance in Fig. 3.2 suggests that the treatment does not result in a low density of band-gap surface states for the (311)A surfaces. To provide independent evidence for this, I turn now to photoluminescence studies.

### 3.1.4 Photoluminescence of $(NH_4)_2S_x$ treated wafers

Figures 3.4(a/b) show PL spectra for (100) and (311)A GaAs wafers after receiving the weak  $(NH_4)_2S_x$  treatment. The signal arises from band-to-band recombination, with the peak intensity occurring at the room temperature GaAs band gap energy  $E_g = 1.42 \text{ eV} = \frac{hce}{873 \text{ nm}}$ . The peak intensity for each of the (100) and (311)A samples was normalised to the peak height obtained from an untreated reference sample from the same wafer. The raw, un-normalised data from these reference

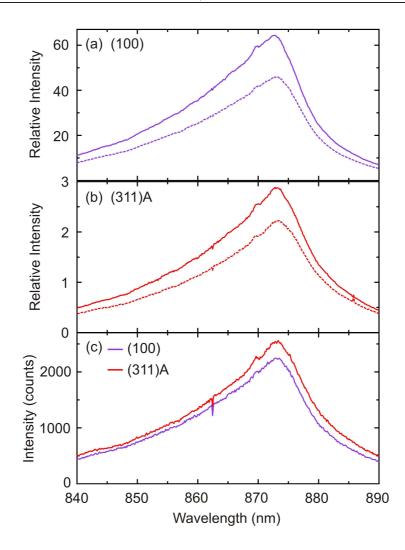


FIGURE 3.4: (a/b) Photoluminescence spectra from two different samples for each of the (100) and (311)A surfaces after receiving the weak  $(NH_4)_2S_x$  treatment. (c) PL spectra for the untreated (100)/(311)A reference samples (purple/red line respectively). Peak intensity of each sample in (a/b) was normalised to the associated reference sample with the same orientation. The intensity in (c) from the reference samples is given in raw counts detected by the CCD camera. The  $40 - 60 \times$  increase in PL peak intensity for treated (100) GaAs is indicative of surface state passivation and the associated reduction in nonradiative recombination pathways. The comparatively small increase for (311)A GaAs suggests  $(NH_4)_2S_x$  does not effectively passivate this surface.

samples in shown in Fig. 3.4(c). The PL peak intensity for  $(NH_4)_2S_x$  treated (100) GaAs surfaces was increased  $40 - 60 \times$  compared to the untreated reference sample. The enhanced PL signal arises from the reduction of non-radiative recombination pathways due to effective surface state passivation.<sup>44,53,252</sup> By contrast, the same treatment produced only a  $2 - 3 \times$  increase in PL intensity for (311)A GaAs surfaces. Such a small PL intensity increase suggests the band-gap surface

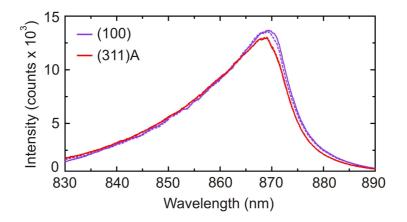


FIGURE 3.5: Photoluminescence spectra for treated/untreated (100) (purple solid/dashed lines) and treated/untreated (311)A (red solid/dashed lines) Al-GaAs/GaAs heterostructures.  $(NH_4)_2S_x$  treatment did not alter the PL signal as recombination was not limited by the surface in these samples (see text).

state density was not appreciably reduced by  $(NH_4)_2S_x$  treatment and subsequent sulfidisation. This corroborates the conclusions drawn from the gate-dependent electrical measurement of devices presented in Sec. 3.1.2.

The shift from electrical/XPS measurements to PL measurements entailed a change of wafer. Initial PL measurements were conducted with pieces from a matched pair of (100) oriented *n*-type and (311)A oriented *p*-type heterostructure wafers used for the devices in Sec. 3.1.2 and Ref. 24. As shown in Fig. 3.5, no PL enhancement was observed for either  $(NH_4)_2S_x$  treated wafer. To explain this, consider that the laser penetration depth in GaAs/AlGaAs is approximately 80 nm at a wavelength  $\lambda = 488$  nm. This excitation would have generated electron/hole pairs well into the bulk of the modulation doped wafer. The band bending of the heterostructure may then have provided an effective barrier for surface recombination. In this case, no recombination would have taken place via surface states for even the untreated samples, and surface state reduction would have no appreciable affect on PL intensity. Another possible alternative for the lack of PL intensity increase could be that the high density of impurities in the AlGaAs doping layer dominates non-radiative recombination. However, this is a less likely scenario as significant PL gain has previously been observed for bulk samples with similar doping density  $N_{Si} \sim 10^{16} \text{ cm}^{-3}$ .<sup>43,44,252</sup> In any case, the PL signal of modulation

doped wafers was not sensitive to changes in surface state distribution. Accurately probing comparative surface dynamics between the (100) and (311)A surfaces by PL necessitates only that other properties of the wafer are roughly the same. This justifies the use of the bulk materials in this case.

### 3.1.5 Possible causes of ineffective passivation

The above results show that  $(NH_4)_2S_x$  alters the bare (311)A GaAs surface by removing the oxide layer and generated Ga-S and As-S bonds, as for (100) GaAs. However, it did not produce a similarly substantial increase in PL intensity and no reliable improvement in the performance of *p*-AlGaAs/GaAs HHMTs. This was characterised by sweep-rate dependent hysteresis and an  $I_d$  plateau where the filling of surface states suppressed hole depletion and changed the pinch-off voltage  $V_{po}$ . The changing length and slope of the  $I_d$  plateau and the corresponding shifts in  $V_{po}$  pointed to a modification of the surface state distribution, but no consistent reduction in the overall density. It is apparent that  $(NH_4)_2S_x$  does not passivate the (311)A surface in the same way as it does for (100) GaAs, and results in surface states with energies that fall within the band gap.

We postulate that the nature of (311)A GaAs surface bonds can explain the ineffective passivation of this surface. Given the scarcity of previous work on the (311)A surface, we turn to the known properties of lower index planes. Higher Miller index planes can be considered linear combinations of lower-index Miller planes; specifically, the bare (311)A GaAs surface consists of (111)A-like single dangling bonds for Ga and (100)-like double dangling bonds for As.<sup>105</sup> I now use previous work on (111)A and (100) sulfidisation to provide possible explanations for our data.

Looking first at Ga bonding, the success of sulfur passivation for (100) arises in part from the Ga-S-Ga bridge bonds covalently satisfying the Ga double-dangling bonds at this surface.<sup>21,44,59,60,262,263</sup> By contrast, surface Ga atoms only have a single dangling bond as three of the four Ga bonds are satisfied by sub-surface Ga-As bonding.<sup>264</sup> Adsorbed S atoms either bond directly on top of Ga surface atoms, or directly substitute at the Ga site.<sup>264–267</sup> Calculations suggest that both situations do not covalently satisfy S bonds; this leaves S-related dangling bonds and associated band gap surface states.<sup>264</sup> This is supported by experimental evidence for ineffective passivation of (111)A GaAs.<sup>267</sup> The similar Ga-S bonding state for (311)A is likely to have contributed to the ineffective passivation observed here.

Turning to As, XPS results in Fig. 3.3 showed that As-S bonds on (311)A mirror those on the (100) surface, as expected. The stability of As-related bonds is reduced compared to Ga-S bonds. While initial H<sub>2</sub>O rinsing removes excess elemental As-As dimers and the associated band gap surface states,<sup>261,268,269</sup> prolonged water exposure breaks the weak As-S bonds and leads to a re-accumulation of surface As and As-O bonds.<sup>253,261</sup> The As-related surface states are below the surface Fermi level for (100) GaAs.<sup>40</sup> If the situation is similar for (311)A, a positive bias applied to the gate in Fig. 3.2 sweeps the surface Fermi level through these states. The variable  $I_d$  plateau length and  $V_{po}$  may well indicate this instability and unreliablility of As-S bond formation, since a process as simple as natural variation of rinse time could have contributed to the difference between device characteristics. Similarly, post-treatment annealing to 360°C is likely to have broken As-S bonds<sup>59–61</sup> and lead to the re-emergence of limited low  $V_G$  depletion and the  $I_d$  plateau in Fig. 3.2(f). The combination of poor As-S bond strength combined with the likely unsatisfied Ga-S dangling bonds explains the insufficient and variable passivation of the (311)A GaAs surface.

The question then is: Is there a solution that can facilitate effective passivation of (311)A GaAs surfaces? The difficulty is that both single and double dangling bonds need to be satisfied. One possibility for satisfying the Ga-dangling bonds is a monovalent adsorbate such as Cl; positive results have been obtained on the (111)A GaAs surface.<sup>270</sup> Producing a solution that incorporated both S and Cl, e.g.,  $S_2Cl_2^{271,272}$  may hold interest as a general approach to passivating (311)A GaAs. In terms of this thesis, investigating alternative passivation solutions was unlikely to completely remove instability and hysteresis in modulation doped devices, due to charge mobility in the dopant layer.<sup>24</sup> I turn to this influence now, using quantum interference effects in a hole quantum billiard.

# 3.2 Using magnetoconductance fluctuations to detect changes in dopant configuration

### **3.2.1** Background, sample and methods

Recent work highlighted in Sec. 1.4.4 has illustrated the impact of small-angle scattering on 'ballistic' transport in modulation doped heterostructures.<sup>84,90,91</sup> The tails of the potential generated by ionised dopant impurities in the AlGaAs layer was shown to cause deviations of electrons even as they travel between large-angle scattering sites.<sup>84,88–91</sup> The sensitivity of magnetoconductance fluctuations (MCF) in semiconductor billiards to electron trajectory and phase means that they are altered by changes in the background potential due to dopants in the AlGaAs layer. The strongest evidence for this was obtained by comparing G(B) traces before and after thermally cycling billiards fabricated from modulation doped heterostructures. MCF became non-reproducible for thermally cycling to temperatures  $T \ge 165$  K.<sup>91</sup> This matches the activation temperature for deep donor trapping sites known as DX centres  $^{93,94,112}$  and gives strong evidence that the decorrelation of MCF traces occurs as a result of spontaneous reorganisation of the donor occupation distribution. This random process changed the form of the background disorder potential crossing the 2DEG and - at least in part - determined a different G(B) signature via small-angle scattering of electrons.<sup>84,90,91</sup>

The sensitivity of MCF to small-angle scattering by dopant impurities opens up its use as a characterisation technique to detect changes in dopant configuration. As mentioned in relation to Fig. 1.19 in Sec. 1.4.5, comparison of the temperature dependence of electron and hole devices suggested that charge migration amongst dopant sites in response to  $V_G$  contributed to the hysteresis of modulation doped *p*-type AlGaAs/GaAs heterostructures.<sup>24</sup> In this section, I present results from a

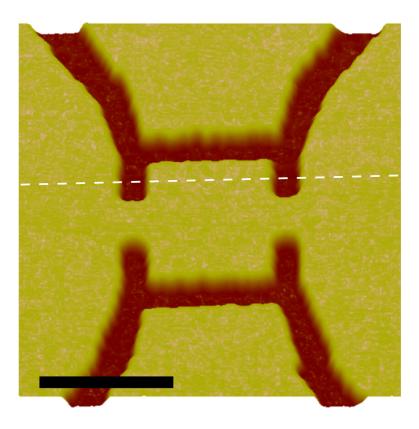


FIGURE 3.6: Atomic force micrograph of the etched square hole billiard prior to polyimide and gate deposition. The 120 nm deep trenches (red regions) define the billiard and entrance/exit QPCs that connected the billiard to the 2DHG. Lateral etching resulted in a billiard size slightly smaller than the 1  $\mu$ m<sup>2</sup> pattern defined by EBL. The white dashed line shows the cut through the billiard used for the cross-sections in Figs 3.9 and 3.11 below. The scale bar represents 1  $\mu$ m.

hole billiard fabricated using the hysteretic wafer. We measured MCF changes in response to applied  $V_G$  in a billiard made from this wafer as a way of confirming and characterising the dopant layer contribution to device instability. Fabrication of the Hall bar and ohmic contacts proceeded as per Sec. 2.2. We etched trenches in the AlGaAs layer to define the billiard rather than define it electrostatically due to gate instability presenting difficulties in confining the 2DHG.<sup>24,106,108–110</sup> The pattern was defined by EBL using PMMA-A3 and wet etching with a 1:8:259 H<sub>2</sub>SO<sub>4</sub>:H<sub>2</sub>O<sub>2</sub>:H<sub>2</sub>O solution to a trench depth of 120 nm. The resulting billiard is shown in Fig. 3.6. The EBL pattern defined a 1  $\mu$ m ×1  $\mu$ m square with QPCs at the entrance and exit. Lateral etching resulted in slightly smaller billiard dimensions. The distance between billiard walls was much smaller than the mean free

path, estimated to be  $\ell = 6.8 \ \mu\text{m}$ . The billiard was covered in a 140 nm thick polyimide layer to prevent gate leakage from the trenches into the 2DHG. Polyimide acts as a negative photoresist and was therefore patterned directly. Polyimide has previously been used as a gate insulator for highly stable ambipolar GaAs/Al-GaAs heterostructure devices.<sup>273,274</sup> The thin film was spun on at 6000 rpm for 60 s and baked at 65°C for 90 s. The regions that were intended to remain after developement were exposed to UV radiation at 10 mW/cm<sup>2</sup> for 70 s. Development in a dedicated developer removed the surrounding polyimide. The sample was then hard baked in a furnace at 250°C for 60 mins to render the remaining polyimide insoluble in alcoholic solvents. The top-gate pattern was then defined

polyimide insoluble in alcoholic solvents. The top-gate pattern was then defined using photolithography, and 20/80 nm layers of Ti/Au were deposited by thermal evaporation.

Conductance G was measured using the 4-terminal techniques outlined in Sec. 2.4.1 using two lock-in amplifiers measuring  $I_d$  and  $V_{4T}$  on either side of the billiard in response to an ac excitation of  $V_{sd}^{ac} = 100 \ \mu\text{V}$  at 13 Hz.  $V_G$  was supplied by a Keithley K2400 SMU, which was also used to monitor gate leakage  $I_G$ . The leakage current remained below  $I_G < 0.5$  nA throughout the experiment. A magnetic field  $|B| \leq 10$  T was supplied by a superconducting solenoid and applied perpendicular to the 2DHG. The experiment was performed in the Kelvinox K100 dilution refrigerator outlined in Sec. 2.4.4, with a mixing chamber base temperature  $T_B = 40$  mK. Heating the mixing chamber allowed continuous temperature variation up to T = 900 mK, with measurements also possible at thermal equilbrium with the <sup>4</sup>He bath, T = 4.2 K. We used a dilution refrigerator rather than the <sup>3</sup>He cryostat used by Scannell *et al.*<sup>91</sup> as lower temperatures T < 100 mK were required to observe hole MCF. This is because the higher hole effective mass generates stronger carrier-carrier interactions that reduce the phase coherence time compared to electrons.<sup>275</sup>

The majority of this experiment deals with differences in MCF for successive G(B) traces. Often clearly different features will be apparent in the raw data. However, quantifying the magnitude of changes is necessary to draw conclusions about trends in the data. I now briefly outline the methods used in the following experiments.

Subtracting successive G(B) traces is the simplest way to obtain the *B*-dependent difference in G, i.e.

$$\delta G(B) = G_2(B) - G_1(B) \tag{3.1}$$

Ideally  $\delta G(B) = 0$  for all B where  $G_2(B)$  and  $G_1(B)$  were identical. In practice, differences in G(B = 0) were hard to avoid, since device instability meant that G would sometimes drift very slightly even at constant  $V_G$ . As a result,  $\delta G(B)$ would develop a constant offset. To deal with this, we defined the difference function  $\Delta(B)$ 

$$\Delta(B) = \delta G(B) - \langle \delta G(B) \rangle \tag{3.2}$$

where the subtraction of the average difference  $\langle \delta G \rangle$  accounts for any offset. The function  $\Delta(B)$  is zero for all B if two G(B) traces are identical, and develops oscillations around 0 when differences in the MCF emerge. These oscillations are larger for greater MCF differences and this can be quantified by the root-meansquare conductance difference:

$$RMS_{\Delta} = \langle \Delta(B)^2 \rangle^{\frac{1}{2}} \tag{3.3}$$

The  $RMS_{\Delta}$  will be used to quantify the degree of MCF difference between traces. This can be normalised to a number between 0 (no correlation) and 1 (complete correlation) following Scannell *et al.*<sup>91</sup> using the autocorrelation function F:

$$F = \sqrt{1 - \frac{\Delta(B)^2}{N}} \tag{3.4}$$

where N is a normalisation function, equal to the average  $\Delta(B)^2$  of completely de-correlated traces.<sup>91,92</sup> N was obtained by randomly generating 100 different G(B) traces and taking the average of the conductance difference  $N = \langle (G_x(B) - G_y(B))^2 \rangle$ . Each of the random traces  $G_x(B)$  and  $G_y(B)$  were constrained to have the same minima and maxima as the measured traces  $G_1(B)$ ,  $G_2(B)$ . Thus F automatically takes a value between 0 and 1. Since MCF are dependent on density, we compared only traces where the conductance at zero-magnetic field G(B = 0)matched to within a very small value,  $0.05 \times 2e^2/h$ . Analysis was performed only

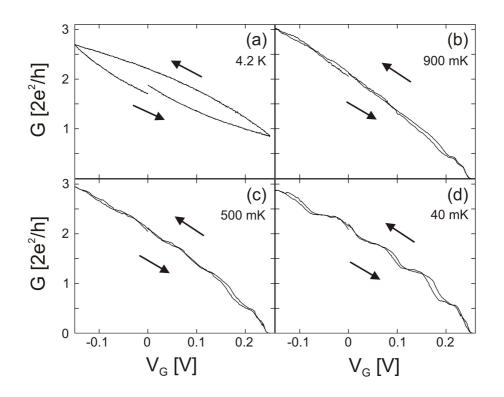


FIGURE 3.7: G modulation for  $V_G$  applied to the top gate covering the billiard at decreasing T.  $V_G$  was incremented at 1 mV/s. Hysteresis in (a) is consistent with that in Fig. 3.1.2 and reduces with decreasing T because the lower thermal energy suppresses charge hopping between acceptor sites. Conductance steps at T = 40 mK arise due to the entrance/exit QPCs.

for |B| < 0.07 T, where the cyclotron radius was larger than the billiard diameter. Fluctuations in G(B) at higher B are due to tighter skipping orbits along the billiard walls, not MCF (see Fig. 1.14(d) in Sec. 1.4.3).

### 3.2.2 Gate characterisation

We first characterised the G dependence on  $V_G$  at B = 0 T for different temperatures, shown in Fig. 3.7. The hysteresis observed at T = 4.2 K is consistent with that shown in Fig. 3.2. The magnitude of the hysteresis decreases as the temperature is lowered because  $k_B T$  becomes smaller relative to the trap depth, but the hysteresis is still present at T = 40 mK. The conductance steps at T = 40 mK are due to the QPCs at the entrance/exit of the billiard. The non-additivity of QPCs in series with separation less than the mean free path is known to produce steps that do not align with integer values of the conductance quantum  $G_Q = 2e^2/h$ .<sup>276,277</sup> The persistence of hysteresis at low T means that these characteristics provide only a guide to device behaviour. Throughout the experiment the characteristics in Fig. 3.7(d) were displaced horizontally in response to shifts and drifts in Gat constant  $V_G$  – this is explored in the next section. A consequence of the nonmonotonic  $G(V_G)$  relationship is that moving away from and returning to the same  $V_G$  results in different  $G(B = 0) \equiv G_0$ . The need to compare traces with similar  $G_0$  meant that traces were taken at whichever  $V_G$  gave the desired  $G_0$ . Typically this was on a conductance plateau, e.g.,  $G_0 = 1.8 \ G_Q$  where the  $G_0$  dependence on  $V_G$  is weak. Here, small differences in  $V_G$  settings would result in minimal variation between  $G_0$  values.

#### **3.2.3** Temporal stability of MCF at constant $V_G$

The presence of hysteresis indicates that charge in the dopant layer may still be active at T = 40 mK. We examined the temporal stability of MCF at constant  $V_G$  with two objectives. Firstly, to ensure that stability was maintained during the ~ 20 min G(B) traces and that G(B) provided a reliable 'fingerprint' of the background disorder potential. The second objective was to observe any changes in the background potential. Depending on the relationship between trap depth and thermal energy, these changes may have two sources. The first is from spontaneous reorganisation due to random thermal fluctuations, and the second is charge migration driven by a static  $V_G$ . To collect this data, G(B) was measured at times t = 0, 8 and 15 hours with constant  $V_G$ . The traces taken at t = 8 and 15 hours were compared with the trace at t = 0 to look for changes in MCF. These changes could also be clearly identified using the conductance difference  $\Delta(B)$ with  $\delta G(B) = G(B, t) - G(B, 0)$  where t = 8, 15 hours.

Figures 3.8(a/b) show this experiment for two  $V_G = -58.5$  mV and -156 mV, which were set to give  $G_0 = 1.8$   $G_Q$  and 2.5  $G_Q$ , respectively. Each MCF trace was symmetrical, G(B) = G(-B), showing that at these  $V_G$ , the disorder potential did not evolve appreciably during each trace. Further, the MCF at  $V_G = -58.5$  mV

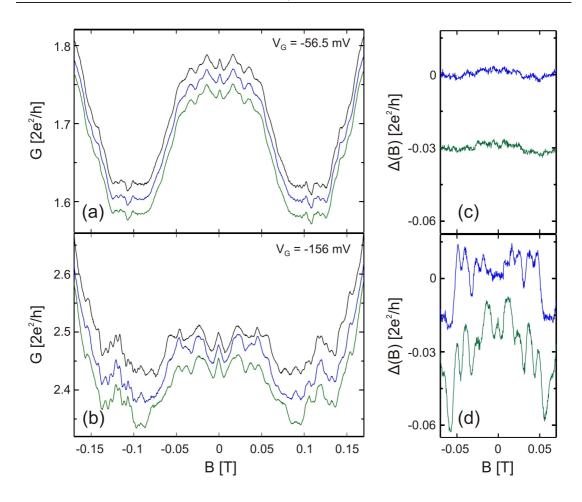


FIGURE 3.8: (a/b) G(B) at constant  $V_G = -58.5$  mV and -156 mV, with  $G_0$  set to (a) 1.8  $G_Q$  and (b) 2.5  $G_Q$  at time t = 0, 8 and 15 hours (black, blue, green traces respectively). Traces were offset vertically for clarity by 0, 0 and  $-0.02 \ G_Q$  in (a) and 0,  $-0.04 \ G_Q$  and  $-0.07 \ G_Q$  in (b), from top to bottom. The inconsistent offsets were due to random jumps and drifts in G at constant  $V_G$  due to the commonly observed gate instability of p-AlGaAs/GaAs devices. (c/d) Conductance difference  $\Delta(B) = \delta G(B) - \langle \delta G(B) \rangle$  corresponding to data in (a/b) for  $\delta G(B) = G(B,t) - G(B,0)$  with t = 8 hours (blue traces) and 15 hours (green traces). The green traces were offset vertically by  $-0.03 \ G_Q$ .  $\Delta(B) \sim 0$  for  $G_0 = 1.8 \ G_Q$ ,  $V_G = -58.5 \ mV$  but develops increasing fluctuations over time for  $G_0 = 2.5 \ G_Q$ ,  $V_G = -156 \ mV$  as more negative  $V_G$  drives charge migration between Si-acceptor sites in the AlGaAs layer.

did not evolve over the 15 hour period; the three traces in Fig. 3.8(a) were practically identical. This suggests that the system was stable at  $V_G = -58.5$  mV on a long time scale. There were no spontaneous reorganisation events in the dot vicinity, and the configuration was not changed by  $V_G$ . Conversely, changes emerged between the three traces at the more negative gate voltage,  $V_G = -156$  mV in Fig. 3.8(b), indicative of a temporal evolution of the underlying disorder potential.

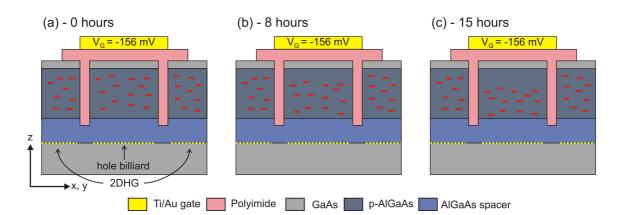


FIGURE 3.9: (a/b/c) Cross-section of the hole billiard along the white dashed line in Fig. 3.6 at t = 0, 8 and 15 hours. The billiard was defined by etching the heterostructure, which caused selective depletion of the 2DHG in the x, y plane. (a) At t = 0 hours negatively charged impurity sites were randomly located. (b/c) Negative  $V_G$  caused a slow migration of holes towards the gate, leading to a higher density of negative impurities closer to the billiard. The configuration was driven further from the initial configuration over longer times. This was sensed as MCF changes in Figs 3.8(b/d).

The different behaviour at different  $V_G$  is clearly demonstrated in the respective  $\Delta(B)$  vs B plots shown in Figs 3.8(c/d). The conductance difference  $\Delta(B)$  was close to zero for all B at  $V_G = -58.5$  mV (Fig. 3.8(c)), confirming that negligible changes in MCF occurred even after 15 hours at this gate voltage. For  $V_G = -156$  mV in Fig. 3.8(d) however,  $\Delta(B)$  developed oscillations around 0. The increasing magnitude of  $\Delta(B)$  oscillations shows that the MCF had changed more after 15 hours than they had after 8 hours. This is consistent with a continual evolution of the background potential over time as it was driven further away from the initial state at t = 0.

The MCF evolution in Figs 3.8(b/d) must be due to an underlying change in disorder potential since neither the hole density or billiard geometry changed during this time.<sup>84,90,91</sup> The likely cause of this behaviour is that  $V_G$  drives charge hopping between acceptor sites in the AlGaAs layer. This process is illustrated in Fig. 3.9. Charge migration occurs because the equilibrium impurity charge distribution at each  $V_G$  is not the equilibrium distribution at  $V_G = 0$ ; more negative  $V_G$  means that a higher positive charge density towards the gate is a more favourable configuration. As such, negative  $V_G$  will drive hopping of holes between impurity sites towards this new equilibrium configuration, leaving a high density of negatively charged ionised impurities in proximity to the 2DHG. More positive  $V_G$  will cause migration in the opposite direction. This process will be slow for thermal energy similar to or less than the acceptor trap depth,  $k_BT \leq E_A$ , and contributes to the hysteresis in  $I_d$  vs  $V_G$ . By reducing the available thermal energy further at T = 40 mK the time scale for the migration was dramatically increased, i.e., 8 to 15 hours. This slow migration towards a new equilibrium configuration was sensed by the MCF changes in Figs 3.8(b/d).

Related to this, the question may be asked: What about other small-angle scattering centres? In particular, could surface states be responsible for the observed MCF changes? We suggest that in the modulation doped sample used here, the much higher density of acceptor impurities separated from the 2DHG by  $\sim 40 - 120$  nm are extremely likely to dominate the background potential and small-angle scattering effects, compared to surface state charges located 120 nm away from the 2DHG. While small-angle scattering from surface charges has been suggested to occur for shallow 2DEGs in heterostructures without modulation doping, it was shown to be strongly suppressed for 2DEG depths of > 100 nm.<sup>25,26</sup> This highlights the importance of the proximity of the scattering centres to the 2DEG/2DHG. The two major differences between this work and that of Mak et $al.^{25}$  and Wang *et al.*<sup>26</sup> is that the 2DHG is 120 nm below the surface in the sample here, with a high density of ionised dopants in the modulation doped layer. As such, charge in the dopant layer is likely to dominate the background potential in this device. A separate argument towards this is that (100) surface states are also active in n-AlGaAs/GaAs heterostructures, but are comparatively shallow traps with different trapping time scales.<sup>24</sup> The time scales meant that surface charges did not produce hysteresis, and instead caused non-linearities in  $I_d$  vs  $V_G$  (see Ref. 24, in particular the discussion relating to Figure 3 of that paper). If true, these surface charges were active in the modulation doped electron billiards used in Scannell et al.<sup>91</sup>, See et al.<sup>90</sup> and all other billiard studies.<sup>65,82,83</sup> Yet gate-induced MCF changes were not observed, likely due to the domination of small-angle scattering by charges in the modulation doping layer. I will therefore assume hereafter

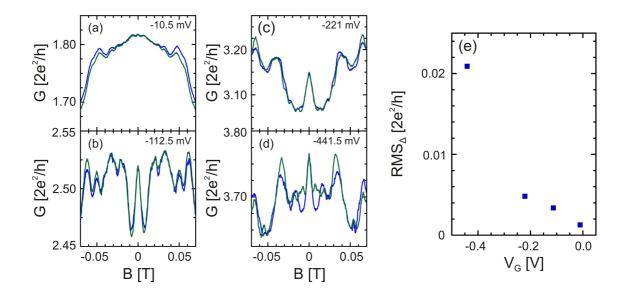


FIGURE 3.10: (a-d) G(B,t) for t = 0 and 1 hr (blue and green traces respectively) at  $V_G = -10.5, -112.5, -221$  and -441.5 mV. The green traces were offset vertically by (a) 0.015  $G_Q$ , (b) 0, (c) 0.015  $G_Q$  and (d) 0.05  $G_Q$  to align  $G_0$ to facilitate easy identification of changes in G(B). Greater differences between traces are apparent at more negative  $V_G$ . (e) Root-mean-square conductance difference  $RMS_{\Delta} = \langle \Delta(B)^2 \rangle^{\frac{1}{2}}$  between the two traces in each panel plotted against  $V_G$ . Higher  $V_G$  caused greater de-correlation between G(B) traces due to increased charge migration between Si-acceptor sites in the AlGaAs layer.

that MCF changes arose due to changes in the charge configuration amongst Siacceptors in the AlGaAs layer.

#### 3.2.4 Evolution of MCF with gate modulation

The data in Fig. 3.8 showed larger conductance differences at the more negative gate voltage. This suggests that the stronger electric field drove more pronounced charge migration. To confirm this effect, we took pairs of G(B) traces one hour apart at increasingly negative  $V_G$ , shown in Figs 3.10(a-d). The traces at higher  $V_G$  are noticeably altered after 1 hour. The  $RMS_{\Delta}$  for each  $\Delta(B)$  between the two traces at constant  $V_G$  is plotted in Fig. 3.10(e).  $RMS_{\Delta}$  increased with more negative  $V_G$ , supporting the notion that an increasing electric field strength drives charge in the dopant layer further from its initial configuration over time.

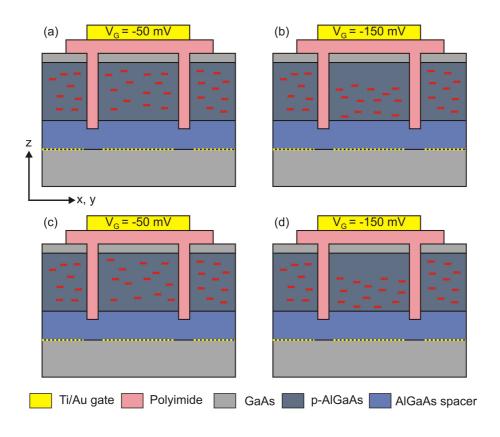


FIGURE 3.11: Device cross-sections along the cut in Fig. 3.6 illustrating a possible scenario for the effect of gate sweeps on the charge configuration in the dopant layer. From (a) to (b),  $V_G$  was swept to more negative  $V_G$  and then from (b) to (c) it was swept back to the original value. This perturbed the charge distribution in the dopant layer. Upon returning to the original  $V_G$  in (c), the density distribution was similar to that in (a), however the charges were not returned to their exact initial configuration. This scenario is likely to occur if there are a large number of trap sites compared to the number of holes. The process would also occur for an experiment starting at (b), sweeping to more positive  $V_G$  in (c) and returning to the more negative  $V_G$  in (d).

The data in Figs 3.8 and 3.10 show that  $V_G$  drives a change in the underlying disorder potential, and that this change was detected by the MCF in the hole billiard. An interesting question is whether particular impurity charge configurations are repeatable. For example, consider collecting MCF for  $G_0 = 1.8 G_Q$ , then sweeping  $V_G$  away from this initial value to induce charge migration amongst impurities. If  $V_G$  is returned to where  $G_0 = 1.8 G_Q$  and a final MCF trace is collected, will the initial and final G(B) traces match? Or is the impurity configuration such that charges are highly unlikely to be returned to the initial state, causing a de-correlation of MCF traces? The latter scenario for this experiment is shown in Fig. 3.11. It would occur if the number of charges is small compared

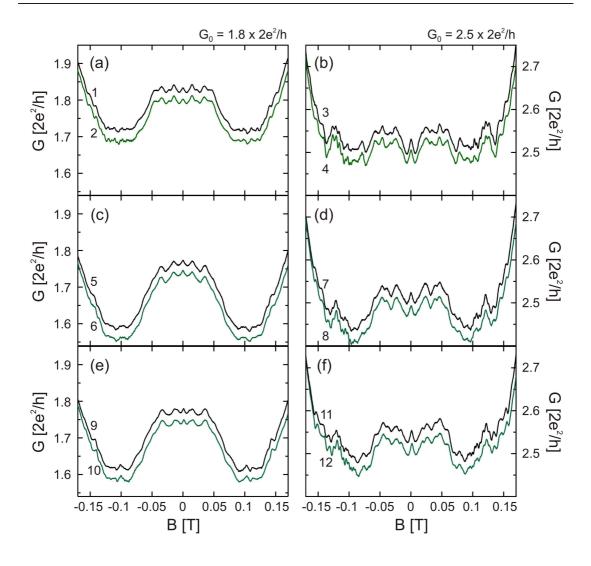


FIGURE 3.12: G(B,t) at t = 0 hours (black traces) and 1 hour (green traces) for  $V_G$  such that  $G_0$  was set to (a/c/e) 1.8  $G_Q$  and (b/d/f) 2.5  $G_Q$ . After collecting the green trace in each panel,  $V_G$  was cycled to the higher or lower setting in the next panel following the order of the numbers 1 - 12. Cycling  $V_G$  produced more marked changes in G(B) between panels compared to the temporal evolution observed within panels.

to the number of impurity sites where the charges may be located.

Data from this  $V_G$  cycling experiment are shown in Fig. 3.12. It proceeded as follows:

- i.  $V_G$  was set to give  $G_0 = 1.8 G_Q$ .
- ii. Two G(B) traces were obtained, one hour apart. This was to check for the degree of temporal evolution during the experiment.

- iii.  $V_G$  was set to give  $G_0 = 2.5 G_Q$ .
- iv. Two G(B) traces were obtained, one hour apart, at this more negative  $V_G$ .

These steps were repeated three times to produce the data in Fig. 3.12. The numbers 1 - 12 indicate the order that the traces were collected. The traces in the left-hand (right-hand) column were obtained at Step ii (iv) of each iteration. Therefore comparing G(B) differences in the left-hand (right-hand) column allowed us to evaluate the effect of cycling to more (less) negative  $V_G$ .

Looking first at the temporal evolution, the black and green traces within each panel were very similar to each other. This shows that MCF did not markedly evolve within each one hour period at Steps ii and iv, similar to the low  $V_G$  data in Fig. 3.10. By comparison, strong differences emerged when comparing traces between different panels in the same column. That is, the MCF were appreciably altered between the black traces (a) and (c) and (e) in the left-hand column, and between the black traces in (b) and (d) and (f) in the right-hand column. This suggests that charges were unlikely to return to the same configuration once  $V_G$ has caused a perturbation of the charge impurity distribution. This is consistent with there being a high density of trap sites in the *p*-AlGaAs layer compared to the hole density. Figure 3.12 also suggests that disturbing the charge configuration by  $V_G$  cycling caused a greater de-correlation than the background temporal evolution. The latter is not surprising as  $V_G$  cycling is likely to represent a stronger perturbation than a constant  $V_G$ .

#### 3.2.5 Thermal cycling and Si-acceptor activation energy

The data presented so far shows that the charge configuration in the AlGaAs layer does not spontaneously re-organise at T = 40 mK, but it can be driven out of equilibrium by the application of an external electric field. This suggests a metastable state at T = 40 mK, with  $k_BT \leq E_A$ . By increasing  $k_BT > E_A$ , it is possible to induce spontaneous reorganisation of charge trap occupation in the dopant layer and determine the activation energy for dopant impurities. The following method was used by Scannell  $et \ al.$ <sup>91</sup>

- 1. Measure initial  $G_1(B)$  at base temperature  $T_B$
- 2. Raise device T to some intermediate temperature  $T_i$  and hold there for 30 mins before returning to  $T_B$
- 3. Measure a second  $G_2(B)$  trace and compare  $G_1(B)$  and  $G_2(B)$  using the autocorrelation function F (Equation 3.4)<sup>91,92</sup>

Scannell *et al.*<sup>91</sup> found significant decorrelation,  $F \leq 0.5$ , for  $T_i \geq 165$  K in *n*-type AlGaAs/GaAs billiards, corresponding to the activation energy of DX centres in n-type AlGaAs.<sup>93</sup> The accuracy of this technique was facilitated by the particular Heliox <sup>3</sup>He cryostat used in Ref. 91, which featured continuous temperature variation between base temperature  $T_B = 240$  mK and room temperature. The suppression of MCF in GaAs hole billiards above  $T=100~{\rm mK^{275}}$  necessitated using the dilution refrigerator, which has a reduced range of controllable intermediate temperatures  $T_i = 40 - 900$  mK and 4.2 K. Fortunately, we expected the activation temperture to be less than 4.2 K from the temperature-dependent gate characteristics in Fig. 3.7. This meant the thermal cycling experiment could be done, although it was slightly more difficult in practice. Temperatures  $T_i \leq 900$  mK were obtainable by carefully heating the mixing chamber, with a constant  $V_G = 0$ . However, obtaining  $T_i = 4.2$  K required completely extracting the <sup>3</sup>He/<sup>4</sup>He mixture to bring the sample into equilibrium with the <sup>4</sup>He bath, and then re-condensing the mixture to obtain  $T_B = 40$  mK. The significant change in device properties due to this thermal cycle necessitated taking measurements at  $V_G = -156$  mV and -270 mV before and after cycling.

Figure 3.13 shows G(B) before and after thermal cycling to  $T_i = 300$  mK, 500 mK, 700 mK, 900 mK and 4.2 K. The before and after cycling traces at  $T_i = 300$  mK exhibit no clear differences. Small changes between traces emerged at higher  $T_i$ ; some examples are indicated by the arrows. Nevertheless, the two before and after traces were still largely the same up to  $T_i = 900$  mK. Conversely, thermal cycling to

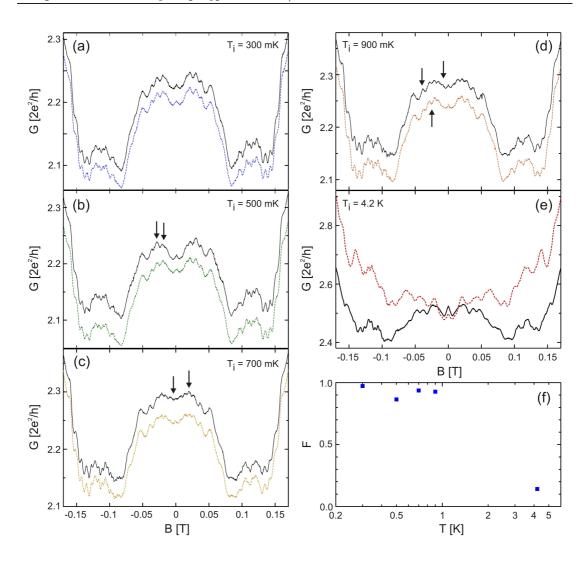


FIGURE 3.13: (a-e) G(B) at T = 40 mK before (black traces) and after (coloured traces) thermal cycling to intermediate temperatures  $T_i$  of (a) 300 mK, (b) 500 mK, (c) 700 mK, (d) 900 mK and (e) 4.2 K.  $T_i$  was held for 30 mins in (a-d), and data was collected with  $V_G = 0$ . Obtaining the data in (e) was more involved, as per the main text. Changes in MCF between traces for  $T_i \leq 900$  mK are highlighted by the arrows. The G(B) traces obtained before and after  $T_i = 4.2$  K have low correlation. (f) Autocorrelation function F vs  $T_i$ for the data in Figs (a-e). De-correlation occurred in the range  $T_i = 0.9 - 4.2$  K.

 $T_i = 4.2$  K produced almost complete de-correlation of the before and after G(B) traces. This interpretation was borne out by the autocorrelation function F values in Fig. 3.13(f). Cycling to  $T_i \leq 900$  mK caused only small spontaneous changes in charge configuration, and the corresponding autocorrelation values were high, F = 0.9-1. The  $T_i = 4.2$  K thermal cycle gave an autocorrelation value F = 0.14. This indicates that significant de-correlation occurs in the  $T_i$  range 0.9 - 4.2 K. Activation of DX centres in *n*-type billiards with trap depth  $\sim 250$  meV<sup>93</sup> caused

 $F \leq 0.5$  for  $T_i \geq 165$  K. From Fig. 3.13(f), F fell below 0.5 in the range  $T_i = 0.9 - 4.2$  K. The activation energy for Si-acceptors in AlGaAs is  $40 - 180 \times$  smaller than for DX centres. This gives an acceptor depth of  $E_A = 1.4 - 6.25$  meV. Given the extra time and gate sweeps involved in obtaining the 4.2 K data, it is likely that this F value was artificially reduced. In this case we expect  $E_A$  to be towards higher end of this range,  $E_A = 3 - 6$  meV.

There is little literature available on *p*-AlGaAs with which to compare this result. One work places Si-acceptor depth in AlGaAs around 60 meV,<sup>113</sup> which is consistent with the known Si-acceptor depth in GaAs of  $E_A = 34.8 \text{ meV}$ ,<sup>278</sup> This is an order of magnitude larger than the  $E_A$  measured here. The shallow traps causing the behaviour in Fig. 3.13 may therefore not be the hydrogenic Si-acceptor. One alternative may be the Si-X defect<sup>279</sup> which is present in p-GaAs at very high doping densities  $N_A \sim 10^{16} \text{ cm}^{-3}$  and is expected to have low activation energy compared to the hydrogenic Si-acceptor.<sup>280</sup> However, little is known about p-Al<sub>0.33</sub>Ga<sub>0.67</sub>As and so conclusively assigning the source of the shallow level will require further work. Note that the bandstructure precludes the measured activation energy having arisen from the shallow hydrogenic Si-donor, despite a coincidentally similar small trap depth  $E_D \sim 6 - 20$  meV.<sup>281</sup> Looking at the *p*-type modulation doped bandstructure in Fig. 1.10(d) (Sec 1.4.1), donor levels will be positioned just below the conduction band. This places them well above the Fermi level and means they are not populated with electrons. If electrons did somehow populate and hop between hydrogenic donor sites, they would follow the band bending and lose energy by leaving the AlGaAs layer on either side. There the electrons would re-combine with holes in the 2DHG or pass into the gate. Thus the Si-donor cannot explain the observed data.

### 3.3 Discussion

In this Chapter I presented results from two investigations into the causes of gate hysteresis and instability in p-AlGaAs/GaAs heterostructures. We explored the

use of  $(NH_4)_2S_x$  surface treatments on (311)A GaAs,<sup>32</sup> and postulated that the ineffective surface state passivation arose from the the need to satisfy both singleand double-dangling bonds. The variable results likely arose from the low stability of As-S bonds. We then used MCF in a hole billiard to detect changes in charge distribution amongst acceptor impurities in the AlGaAs layer.<sup>33</sup> These changes could be induced by using an external electric field to drive charge migration, or thermal cycling to cause spontaneous reorganisation. Thermal cycling was also used to estimate the activation energy of Si-acceptor traps  $E_A \sim 3-6$  meV from the corresponding  $T_i = 2 - 4.2$  K data. These experiments illustrate the potential for MCF as a diagnostic tool to sense background potential fluctuations towards optimisation of doping and impurity incorporation.<sup>33,282</sup> Looking at these results in the context of (311)A p-AlGaAs/GaAs heterostructure-based quantum device development, they provide further evidence for the role of both surface states and dopants in gate instability at low T. The changes in  $I_d$  vs  $V_G$  induced by the  $(NH_4)_2S_x$  solution – particularly in  $I_d$  plateau length and  $V_{po}$  – provide strong evidence for the role of surface states. Similarly, the small-angle scattering from impurities observed using MCF in a billiard highlights the impact of charge migration between acceptor sites.

The data in Fig. 3.13 showed that the activation temperature for Si-acceptors was 2 - 4.2 K. In Ref. 24 we suggested that surface states froze out around 1 - 2 K based on the reduced prominence of the  $I_d$  plateau in HHMTs (see Fig. 1.19 in Sec. 1.4.5 or Fig. 6 in Ref. 24). A second explanation for the temperature dependent behaviour of HHMTs is that charge migration in the dopant layer froze out at 2 K. This would have facilitated enhanced low  $V_G$  depletion and potentially allowed pinch-off to occur before the onset of the  $I_d$  plateau, i.e., before the surface Fermi level reached the highest density of surface states. This may have resulted in an apparent reduction in  $I_d$  plateau with reduced T in Fig. 1.19, rather than an actual reduction in surface state electrical activity. Obtaining further information and teasing apart the potentially coincidental energetics would likely require additional wafer growth and experiments capable of probing only surface states. Of particular interest would be mapping out the (311)A surface

state density spectrum using, e.g., deep level transient spectroscopy  $(DLTS)^{43}$  or C-V measurements.<sup>39,41</sup> Also of interest would be corroboration of the estimated activation energy for charge traps in p-Al<sub>0.33</sub>Ga<sub>0.66</sub>As *via*, e.g., low *T* PL or DLTS.

From a practical perspective, obtaining stable, noise-free hole quantum devices will require addressing both the contributions from the surface and the dopant layer. Indeed, the most stable control of hole quantum systems has come from wafers where both dopants and surface states are negated. This is accomplished by inducing the 2DHG using a degenerately doped cap as a gate.<sup>103,283–287</sup> The cap can be selectively etched to generate sets of independent gates that induce and/or confine the 2DHG. The stability arises as these 'induced' devices abandon both modulation doping and the need for a metal/semiconductor interface. Here, the surface states are above the gate, not between the gate and the 2DHG. This means slow charge trapping in response to  $V_G$  either does not occur or cannot influence the 2DHG as it is effectively screened by the gate. The difficulty for induced devices is that the gate must overlap the ohmic contacts in order to ensure the 2DHG connects to the ohmics. The application of a large voltage between the closely separated gates and ohmics limits the measurement range and device yield. Gate feature size is also limited as surface Fermi level pinning generates depletion regions at the feature edges that reduces conductivity.<sup>25</sup> Relatively stable low-dimensional hole systems have also been induced using metal gates on (100)-oriented heterostructures without doping.<sup>98,273,274</sup> This suggests that doping is therefore the biggest road-block for (100)-oriented modulation doped p-heterostructures.<sup>108</sup> The best solution may be to find a suitably stable p-type doping scheme for (100)-oriented growth. The stability of n-type devices is likely to be due to the deep DX centres trapping any free electrons in the AlGaAs layer at T < 165 K. As previously suggested in Ref. 24, this encourages the intentional introduction of deep traps that could mirror the role of DX centres.

### Chapter 4

# Results: Advanced gating methods for nanowire transistors

## 4.1 Gating nanowires using multiple metal/oxide wrap-gates

The pursuit of greater electrostatic control over electron transport in nanostructures has pointed to using transistors where the gate wraps concentrically around the channel. Moving from traditional planar structures towards structures where the gate is folded or wrapped around the channel is a major motivation in transistor development for high speed logic applications,<sup>3,4,12,288,289</sup> and offers improved control over quantum systems.<sup>290</sup> The morphology of self-assembled semiconductor nanowires makes them excellent candidates to develop wrap-gate technology.<sup>3</sup> Additionally, the cost- and materials-effective synthesis of III-V nanowires compared to planar growth has driven considerable interest in the development of III-V wrap-gated nanowires towards efficient deployment in transistor systems.<sup>3,4,12,135</sup> Strong electrostatic control has also facilitated concepts such as 'external doping', where a base-line charge carrier density is set by a gate voltage rather than doping. This has been used to, for example, fabricate a tunable p - n junction in an InP nanowire,<sup>20</sup> and lock-in charge configurations for quantum measurements at low T.<sup>30</sup> The latter is the subject of Chapter 5 in this thesis. Aside from the new functionality of improved tunability, excluding dopant atoms holds the potential for a range of nanoscale devices. This may consist of, e.g., reducing large-angle scattering to improve mobility, similar to modulation doped<sup>71,72</sup> or electrostatically induced<sup>16,17,283,284</sup> structures. At an even smaller scale, random placement of dopants can cause a large variability in behaviour for device sizes  $\leq 50$  nm.<sup>13,14,36</sup> Using devices where transport is induced solely by strong electric fields supplied by effective gating could address some of these issues.

Returning to nanowires specifically, development will need to go beyond simply the performance advantages brought by III-V materials for nanowire technologies to provide a truly viable alternative to planar Si technology.<sup>12</sup> Scalability is a major factor due to the desire to fit an increasing number of transistors in the same chip area. Nanowires also offer potential advantages here in the form of 3D architectures, where multiple layers of transistors facilitate a greater number of transistors in the same substrate surface area.<sup>3,291</sup> One simple way of achieving this is multiple wrap-gates on vertical structures; this opens the potential for two, three, four etc transistors in the substrate area previously occupied by a single transistor. Additionally, multiple gates on the same transistor opens routes towards simpler logic functionality. For example, the AND operation can be realised with a single two-gate device,<sup>292</sup> rather than the four separate transistors required in CMOS. Devices with more than two gates are also interesting for novel logic processes.<sup>293</sup> While fabrication of a vertical Si nanowire transistor with two wrap-gates has been demonstrated<sup>292</sup> there is a significant time cost in the fabrication, as each individual wrap-gate must be defined consecutively. This entails repeating a large number of processing steps for each gate with a significant time and yield cost.

An alternative 3D architecture would be multiple layers of *lateral* nanowire transistors, each with multiple wrap-gates.<sup>29</sup> The lateral orientation is also favourable for fundamental studies of both basic materials properties and quantum phenomena, as outlined in Sec. 1.5.3. One example of the latter would be using a device with three wrap-gates to define and probe highly tunable quantum dot systems

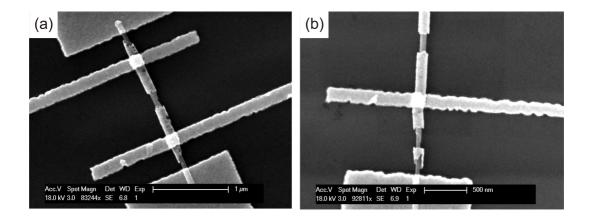


FIGURE 4.1: Scanning electron micrographs of double wrap-gated NWFETs fabricated by first depositing the source, drain and gate contacts. This process resulted in (a) gradual bends (b) sharp kinks in the nanowires. Images courtesy of Adam Micolich.

in nanowires. Here, the two outer gates would be used to define tunnel barriers surrounding a central plunger gate. The first step in any of these programmes naturally is a realisation of a single lateral nanowire transistor with multiple wrapgates. I contributed to the development of a fabrication method for such devices during my research.<sup>29</sup> We developed a method where the same number of fabrication steps were used for devices with a number of gates  $N \ge 2$ .<sup>29</sup> The fact that the number of processing steps does not increase with the number of gates provides a significant scaling advantage over the vertical orientation.

#### 4.1.1 Fabrication process

The main challenge in realising multiple-gate nanowire transistors was in making the initial step from a single gate to two gates. As explained in Sec. 2.3.2, an important aspect of fabrication for single wrap-gated NWFETs developed by Storm *et al.*<sup>145</sup> is that a single resist layer was used for the entire process. However, a single-resist approach to fabricating multiple wrap-gates would necessarily result in ohmic contacts between each gate. This would severely limit a number of applications. As such, we focussed on the development of a more general method that used two separate resist layers. The devices were based on 70 nm diameter InAs nanowires pre-coated with a 12 nm insulating Al<sub>2</sub>O<sub>3</sub> layer, a 16.5 nm W layer and 11.5 nm Au layer that acted as the wrap-gate, and an outer  $12 \text{ nm Al}_2\text{O}_3$  layer to facilitate nanowire transfer as described in Sec. 2.3.

The first fabrication method we investigated was to use the first resist to define the source, drain and gate contacts. The process was similar to that presented for single wrap-gates in Sec. 2.3.2. A second resist was then used to define the multiple gate segments. The process was:

- 1. Define the source/drain contact regions in the first resist via EBL.
- 2. Etch away the outer oxide, Au and W in the source/drain region.
- 3. Expose the regions for multiple gate contacts in the same resist layer *via* EBL.
- 4. Etch the outer oxide in the gate contact region and inner oxide in source/drain region.
- 5. Deposit source, drain and gate contacts. So far this process is identical to single wrap-gates, but with multiple gate contacts to what is at this point still a single wrap-gate.
- 6. Lift-off resist and spin second resist.
- 7. Define etch regions to separate gate segments via EBL.
- 8. Etch away outer oxide, Au and W layers.
- 9. Lift-off second resist for completed device.

Depositing the contacts using the first resist ensured that the nanowire remained fixed in the same position during lift-off at Step 6, and during subsequent gate definition. Using this method we realised NWFETs with multiple wrap-gates, as shown in Fig. 4.1. However, performing subsequent fabrication with contact metal deposited appeared detrimental to nanowire quality. The nanowires in Figs. 4.1(a) and (b) sustained a gradual bend and sharper kinks, respectively. This may have arisen from the forces applied during metal lift-off or resist spinning if unrestrained

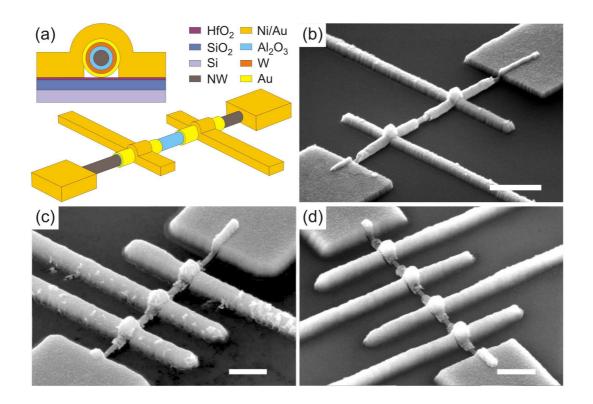


FIGURE 4.2: (a) Schematic and (inset) cross-section of wrap-gated NWFET with two independent segments. The cross-section shows the inner  $Al_2O_3$ , W and Au wrap-gate layers along with the Ni/Au gate contact. The underlying substrate consisted of highly doped Si that was used as a back-gate, insulated from the NWFET by 100/10 nm layers of SiO<sub>2</sub>/HfO<sub>2</sub>. (b/c/d) Scanning electron micrographs of double, triple and four wrap-gated NWFETs. Fabrication of each device entailed the same number of processing steps. SEM images courtesy of Adam Burke.

sections of the nanowire moved when the restrained sections underneath the contacts could not.

In the interests of minimising nanowire damage, we moved to an alternative method. This second process had the additional advantage of less processing steps; it only required two EBL sessions, rather than the three used in the previous method. The process was:

- Etch away outer Al<sub>2</sub>O<sub>3</sub> layer on all nanowires immediately after transfer, before spinning the first resist.
- 2. Use the first resist to define source/drain contacts as well as etch regions to separate wrap-gate segments *via* EBL.

- 3. Etch Au and W layers.
- 4. Lift-off first resist and spin second resist.
- 5. Define source, drain and gate contacts via EBL.
- 6. Deposit contact metal.
- 7. Lift-off second resist for completed device.

Examples of NWFETs with 2, 3 and 4 wrap-gates in series produced using this method are shown in Fig. 4.2. The success of the fabrication relied on the unrestrained nanowires maintaining their position through lift-off and secondary resist spinning in Step 4, as no contacts were deposited using the first resist. Fortunately this held for  $\sim 90\%$  of nanowires, enough to facilitate the proof-of-principle experiments we conducted. The scalability power of this method is that the total number of fabrication steps did not depend on the number of wrap gates. Defining and contacting additional wrap-gates only required changing the EBL patterns in Steps 2 and 5.

As in fabrication for single wrap-gated devices, Au, W and Al<sub>2</sub>O<sub>3</sub> removal was accomplished using KI/I<sub>2</sub>, H<sub>2</sub>O<sub>2</sub> and HF etches, respectively (see detail in Sec. 2.3.2). The outer Al<sub>2</sub>O<sub>3</sub> layer was removed prior to resist spinning at Step 1 rather than immediately prior to gate metal etches at Step 3 since it was only required to facilitate nanowire transfer. We found that etching the Al<sub>2</sub>O<sub>3</sub> at Step 3 resulted in less accurate gate definition. The source/drain contacts were passivated by immersion in the weak (NH<sub>4</sub>)<sub>2</sub>S<sub>x</sub> solution at 40°C for 2 minutes.<sup>240</sup> The contacts consisted of 45/205 nm of Ni/Au deposited in the Lesker thermal evaporator. The thick contacts were required to obtain continuous metal coverage over the wrap-gated segments of the nanowire, which were ~ 170 nm in diameter. The total 420 nm thickness of the contacts plus the nanowire was greater than a single PMMA layer, ~ 300 nm. Thus, we used a double-layer of PMMA for the second resist spun at Step 4. This was accomplished by consecutively spinning and baking two layers of PMMA A5, for a total thickness of ~ 600 nm. EBL was performed using the Raith 150-Two EBL system with a 20 kV accelerating voltage and 20  $\mu$ m aperture.

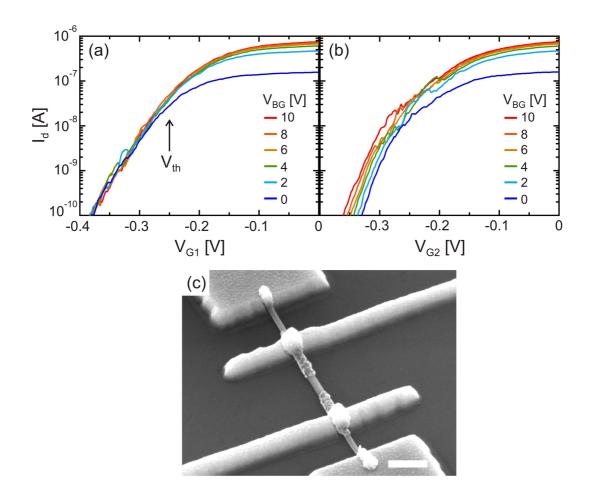


FIGURE 4.3: (a/b) Electrical characteristics and (c) scanning electron micrograph of a double wrap-gated NWFET. (a)  $I_d$  vs  $V_{G1}$  at varying  $V_{BG}$ . The electrical properties of G1 were independent of  $V_{BG}$  below threshold voltage  $V_{th} \sim -0.25$  V due to the wrap-gate providing effective screening of the nanowire from the back-gate potential. This suggests that the W layer of this gate was intact. (b)  $I_d$  vs $V_{G2}$  at varying  $V_{BG}$ . The  $V_{BG}$  dependence of  $I_d$  vs  $V_{G2}$  indicates insufficient screening, likely due to holes in the W layer. SEM image courtesy of Adam Burke.

#### 4.1.2 Electrical characterisation of the two-gate device

We evaluated the electrical performance of the gates by focussing on the transfer characteristics in the subthreshold region, where the gate exercises the strongest control over the channel. Ideally,  $I_d$  vs  $V_G$  in the subthreshold region depends only on temperature and the materials properties of the gate insulator and semiconductor.<sup>1,2</sup> A SR830 lock-in amplifier supplied an ac excitation  $V_{sd} = 30 \ \mu\text{V}$  at 73 Hz to the source and measured the resulting  $I_d$  at the drain. A Keithley K2400 SMU was used to check that each gate did not exhibit significant leakage currents within the operating range. For measurements, gate voltage  $V_{GN}$  was applied to gate number N = 1, 2, 3, 4 by the voltage supplies included in the SR830, with RCR filters used on each output to reduce high frequency noise.  $V_{GN}$  was incremented at 4 mV/s. A back gate voltage  $V_{BG}$  was supplied to the  $n^+$ -Si substrate using a Keithley K2400 SMU. Measurements were performed in liquid nitrogen at T = 77 K using the cryogenic dewar and dipstick outlined in Sec. 2.4.2. This was to freeze-out charge traps at the NW/Al<sub>2</sub>O<sub>3</sub> interface and thereby reduce drift and hysteresis that arises at room temperature.<sup>22,23,28</sup>

The  $I_d$  vs  $V_{G1}/V_{G2}$  transfer characteristics are shown in Fig. 4.3(a/b) for G1 and G2 of the device in Fig. 4.3(c) at varying  $V_{BG}$ . In this measurement, G2 was grounded while G1 was swept, and vice versa. Both gates turn off the NWFET within a 0.4 V range, with subthreshold swings of SS = 40 and 30 mV/dec and threshold voltages  $V_{th} = -0.247$  and -0.236 V, for G1 and G2 respectively. Scaling the subthreshold swing to room temperature gives expected values of SS(T = 300 K) = 117 and 168 mV/dec. Typically, room temperature subthreshold swings of 100 - 750 mV/dec are observed for InAs wrap-gated NWFETs with a Al<sub>2</sub>O<sub>3</sub> dielectric.<sup>22,23,139,140,145,294,295</sup> This places the device in Fig. 4.13 amongst the better performing NWFETs. The comparatively good performance of this device may be partially due to the fact that charge trapping in surface states will have degraded the subthreshold properties of NWFETs measured at room temperature in these previous works.<sup>1,2,12,48</sup>

Looking closely at the SEM in Fig. 4.3(c) reveals various imperfections in the gates. In particular, there are large patches of unintentionally removed material. This was most likely due to small quantities of etchant attacking the Au layer by wicking between the resist and device. The uneven nature of the patches can be explained by the large grain size of sputtered Au films resulting in large voids between grains.<sup>145</sup> Such imperfections are normal in nanofabrication as a result of small changes in materials and reagents between fabrication batches. In particular, recall the much more continuous gates fabricated using a nominally identical process shown in Fig. 4.2. For device applications, the most important aspect is the extent to which these imperfections affect the electrical performance.

The major concern with devices like that in Fig. 4.3 is that the integrity and electrical quality of the gates was compromised. Practically, material removal was only likely to affect device performance if the W layer was also compromised; an intact W layer would provide a strong radial confinement potential irrespective of the properties of the Au layer. Since this was not easily determined from SEM images, we used a simple electrical method of testing the extent of W removal, which I explain now.

The technique we used relies on the capacity of the wrap-gate to screen external electric fields. Significant W removal would noticeably decrease screening capabilities as external electric fields penetrate directly to the nanowire through holes in the metal. To test the screening, we used  $V_{BG}$  to apply an external electric field in parallel to the wrap-gate and observe how this affected the subthreshold region when sweeping  $V_{GN}$ . In an ideal device, current in the subthreshold region depends only on the potential barrier generated by the gate voltage (see Sec. 1.2).<sup>1,2,145</sup> This ideal behaviour was seen for G1, where  $I_d$  vs  $V_{G1}$  fell on the same path below  $V_{th}$ , i.e.  $V_{G1} < -0.25$  V (Fig. 4.3(a)). A back-gate dependence of  $I_d$  vs  $V_{G1}$  emerged above  $V_{th}$  due to positive  $V_{BG}$  inducing a higher carrier concentration in the regions of the nanowire not covered by the wrap-gate. This was noted by Storm *et al.*<sup>145</sup> for single wrap-gate NWFETs also. By contrast,  $I_d$  vs  $V_{G2}$  shows that G2 performance was influenced by  $V_{BG}$  even in the subthreshold region. This suggests that G2 did not effectively screen the nanowire channel from the back-gate potential due to holes in the wrap-gate extending into the W layer.

This simple electrical technique allowed us to determine the suitability of a particular device for applications without recourse to more time consuming imaging methods. Additionally, many of the most probing imaging methods, e.g., transmission electron microscopy, are destructive and cannot be used on devices intended for electrical measurement. This means information is only gained about devices that can never be used anyway. It is also worth noting that G1 of the device in Fig. 4.3 exhibited ideal behaviour despite the poor appearance compared to, e.g., the devices in Fig. 4.2 and in Storm *et al.*<sup>19</sup> This illustrates that effectively

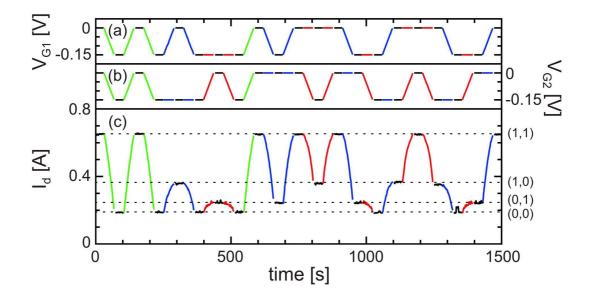


FIGURE 4.4: (a)  $V_{G1}$  and (b)  $V_{G2}$  vs time as they were swept between 0 V (logical 1) and -0.15 V (logical 0). Green traces represent where G1 and G2 were swept together, blue traces where G1 was swept with G2 fixed, and red traces where G2 was swept with G1 fixed. The black traces are 30 s wait times to establish device stability. The program switched each two-input logic state between every other logic state. (c)  $I_d$  output in response to the program in (a/b).  $I_d$  took four distinct values corresponding to each input logic state.

utilising devices rests on ascertaining the actual device performance rather than relying on appearance alone.

We then evaluated the independence and electrical balance between gates. Fabricating gates that reproducibly apply the same potential is important for logic applications or the generation of quantum systems. We investigated this by treating the NWFET as a logic system, using the two gates as inputs and reading  $I_d$  as the output. The power of this was that we could set well defined  $V_{GN}$  inputs and predict the  $I_d$  outputs. Deviations in the measured  $I_d$  from the expected output states could then be traced back to imbalance at the inputs. For instance, take the simplest case of a single gate FET switching between two pre-defined  $V_G$  values. The high  $V_G$  is notated as logical 1 and the lower  $V_G$  as logical 0. Switching between these two values should consistently generate only two  $I_d$  values: a high  $I_d$  for logical input 1, and a low  $I_d$  for logical input 0. Stepping this up to two independent gates generates four logical input states: (1,1), (1,0), (0,1) and (0,0). The expected  $I_d$  output behaviour is four constant values that correspond to the four input states in the following way: Applying (1,1) at the input will generate the highest  $I_d$  output; applying (0,0) and the input generates a low  $I_d$  output; applying inputs (1,0) and (0,1) generates an  $I_d$  output somewhere in between the highest and lowest  $I_d$ . A stable device should return to these four  $I_d$  output values no matter the order of operation.

To test this on the device pictured in Fig. 4.3(c) we formulated a program that swept the gates separately or together between  $V_{GN} = 0$  V (logical 1) and  $V_{GN} =$ -0.15 V (logical 0) such that each of the four logical states switched to each of the other logical states at least once. Figures 4.4(a/b) follows the action of  $V_{G1}$ and  $V_{G2}$  inputs over time as the program ran. The green traces correspond to when the gates were swept together and the blue (red) traces show when G1 (G2) was swept with G2 (G1) constant. A 30 s wait time at constant  $V_{GN}$  was used between sweeps to examine device stability. The output  $I_d$  took four distinct values corresponding to the four input states during each 30 s wait time (Fig. 4.4(c)). These four  $I_d$  values did not depend on the order of operations, indicating excellent device stability and gate independence.

In terms of electrical balance, G1 in this device had a stronger coupling to the nanowire channel than G2. This is apparent by comparing the blue and red traces (G1/G2 only sweeps, respectively) in the range t = 200 - 500 s. Going from (0,0) to (1,0) by sweeping  $V_{G1}$  generated a greater change at the output than sweeping  $V_{G2}$  to go from (0,0) to (0,1). Related behaviour was observed in the range t = 600 - 900 s, where the transition from (1,1) to (0,1) using  $V_{G1}$  entailed a larger  $I_d$  change than that generated by the (1,1) to (1,0) transition using  $V_{G2}$ . Some imbalance was observed in the basic transfer characteristics in Fig. 4.3, however the logical operations here make this imbalance much clearer. An imbalance in this device is not unexpected given the noticeable damage to G2 visible in Fig. 4.3(c).

The interest in two-gate devices for logic applications is the embodiment of logical AND.<sup>292</sup> This functionality occurs when the input logical 0 is set to a  $V_{GN} < V_{th}$ . In the subthreshold region,  $V_{GN}$  generates a potential barrier that blocks

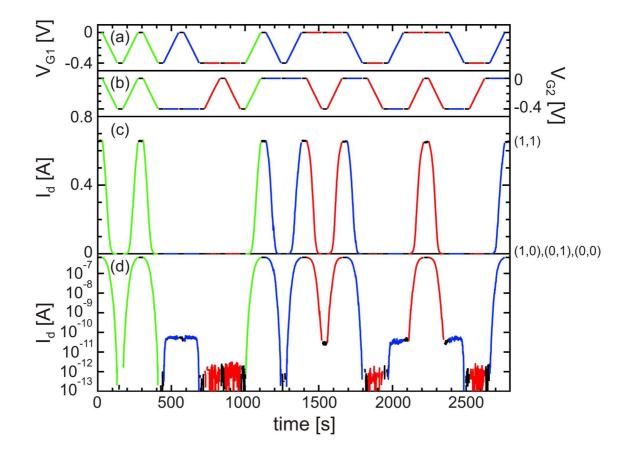


FIGURE 4.5: (a)  $V_{G1}$  and (b)  $V_{G2}$  executing the same program as Fig. 4.4, but logical 0 was set to -0.4 V. (c) Sweeping  $V_{G1}$  or  $V_{G2}$  is sufficient to obtain  $I_d \sim 0$ , i.e., for logic inputs (1,0), (0,1) and (0,0). The device embodies the AND operation as  $I_d$  flows only when both G1 and G2 are above threshold, at input logic state (1,1). (d)  $I_d$  data in (c) plotted on a log<sub>10</sub> scale reveals that gate imbalance affected device performance. Nevertheless,  $I_d$  was reduced by a factor of  $10^{-4} - 10^{-7}$ , which enabled AND operation.

 $I_d$ , independent of the voltage on the other gate. This means that logical inputs (1,0), (0,1) and (0,0) all cause  $I_d \sim 0$  at the output. Non-zero  $I_d$  is obtained only when both gates are above threshold, i.e., the input is (1,1). Figs 4.5(a/b) show the logic program applied to the same device, but with input logical 0 set to  $V_{GN} = -0.4$  V, which was below  $V_{th} = -0.25$  V. The input logical 1 was again set to  $V_{GN} = 0$  V. Fig. 4.5(c) shows that the  $I_d$  output was zero for all inputs aside from when  $V_{G1} = V_{G2} = 0$  V. That is, logical state (1,1) on the inputs generated 1 at the output, while (1,0), (0,1), (0,0) at the inputs all generated logical 0 at the output, effectively embodying the AND operation. Table 4.1 shows the truth table for AND, and the values used to embody the operation using the double wrap-gated NWFET. The program previously used in Fig. 4.4 generated a four

level output  $I_d$  – rather than the two output AND – because the logical 0 of  $V_{GN} = -0.15$  V set there was well above  $V_{th}$ .

TABLE 4.1: (left) Truth table for logical AND. (right) Device inputs/outputs from Fig. 4.5.

AND			Device input		Device output
Input 1	Input 2	Output	$V_{G1}$ [V]	$V_{G2}$ [V]	$I_d$ [A]
1	1	1	0	0	$6.5 \times 10^{-7}$
1	0	0	0	-0.4	$\sim 10^{-10}$
0	1	0	-0.4	0	$\sim 10^{-12}$
0	0	0	-0.4	-0.4	$\sim 10^{-13}$

To examine the effect of gate imbalance on AND realisation,  $I_d$  was plotted on a  $\log_{10}$  scale in Fig. 4.5(d). Similarly to the data in Fig. 4.4, the stronger coupling of G1 caused the (1,0) state to generate a higher output than the (0,1) state. However,  $I_d$  in both states was  $10^4$  to  $10^7$  times lower than the  $I_d$  when (1,1) was applied to the input, which lead to an effective  $I_d = 0$  in Fig. 4.5(c). This reinforces the point that even devices with a poor appearance often have adequate electrical performance. The advantage of the multiple gate approach to AND realisation is that it requires only a single transistor with two gates. In CMOS technology, AND requires two *n*-type and two *p*-type transistors. Note that the device could likely be operated much faster than the slower sweeps conducted here; vertical InAs wrap-gated transistors with GHz operation have been demonstrated, by undertaking optimisation of various parasitic effects.<sup>141,142,296</sup> The slow sweep rate was used here to accurately capture the steady state device properties.

#### 4.1.3 Electrical characterisation of the four-gate device

As mentioned previously, the multiple wrap-gate fabrication process facilitated realisation of NWFETs with four gates using the same number of processing steps as for NWFETs with two gates. Figure 4.6 shows the transfer characteristics for each gate of the four-gate device in Fig. 4.2(d). There was a high degree of similarity in the electrical properties of each gate, with average subthreshold swing  $SS = 34 \pm 9$  mV/dec and median threshold voltage  $V_{th} = -0.46 \pm 0.04$  V. The

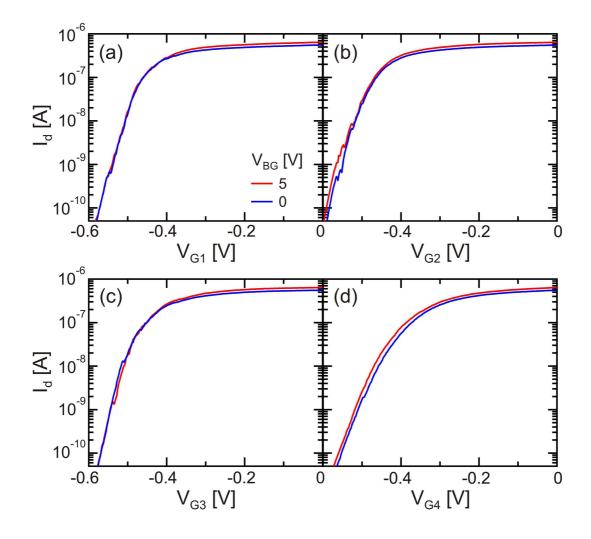


FIGURE 4.6:  $I_d$  through the four wrap-gated NWFET in response to (a)  $V_{G1}$ , (b)  $V_{G2}$ , (c)  $V_{G3}$  or (d)  $V_{G4}$  at  $V_{BG} = 0$  V (blue traces) or 5 V (red traces). The comparative lack of dependence on  $V_{BG}$  and strong similarity between the characteristics of each gate are indicative of a high quality device.

 $V_{BG}$  dependence of wrap-gate sweeps was much reduced compared to the two-gate device studied in Fig. 4.3, indicating excellent screening and therefore a high degree of material integrity. The comparative lack of  $V_{BG}$  dependence above threshold is likely due to the additional gate coverage. This would increase the associated screening and reduce  $V_{BG}$  influence over electron density in the nanowire. The effective screening in this device illustrates the capacity to fabricate high quality gates with lengths ~ 400 nm and separations of 200 nm using our fabrication process.

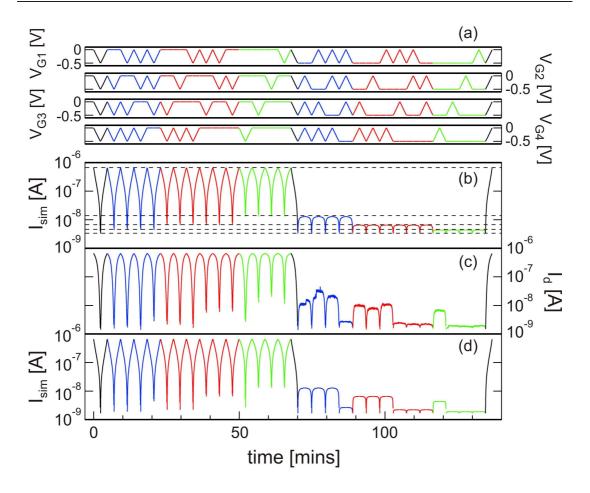


FIGURE 4.7: (a) Logic program applied to  $V_{GN}$ , N = 1 - 4 for the four-gate NWFET with logical input 1 set to  $V_{GN} = 0$  V and logical input 0 set to  $V_{GN} = -0.5$  V. The program switched from/to logic input (1,1,1,1) to/from all other input logic states outlined in Table 4.2, then from/to (0,0,0,0) to/from all other logic states. Black, blue, red and green traces correspond to one, two, three or four gates swept, respectively. All other gates were held constant. (b) Simulated current  $I_{sim}$  for an ideal device with equal gate-nanowire coupling for all gates. The 18 input logic states result in 5 ouput logic states. (c) Measured  $I_d$  for the four-gate device in Fig. 4.2(d). The device deviates from expected behaviour due to different coupling of G4. (d)  $I_{sim}$  taking into account the different G4 gate coupling (see text).

Evaluating the temporal stability and gate balance of this device was more complex than for the device with two wrap-gates. The logic programs in Figs. 4.4 and 4.5 consisted of switching each of the four logic states between every other logic state. This becomes impractical when dealing with four independent gates; four inputs generate a total of 18 possible input logic states. This means a prohibitively large number of sweeps would be needed to map out the four-gate device in the same way as the two-gate device. Instead, we focussed on the essential details to illuminate gate imbalance. First, we considered ideal device behaviour, where the nanowiregate coupling is identical for each gate. In this case, changing G1 from logical 0 to logical 1 would result in the same  $I_d$  output as changing G2, G3 or G4 from logical 0 to logical 1. That is, four gates with identical coupling to the nanowire should generate the same output for the input permutations (1,1,1,0), (1,1,0,1), (1,0,1,1) and (0,1,1,1). Considering the remaining logical input states reveals that 5 output states arise from the 18 possible input states in an ideal four-gate device. These are summarised in Table 4.2.

TABLE 4.2: Input and outputs for the four-gate device where the gates couple equally to the channel. Ordered from highest to lowest  $I_d$  output

Input logic states	Output logic state
(1, 1, 1, 1)	5
(1,1,1,0),(1,1,0,1),(1,0,1,1),(0,1,1,1)	4
(1, 1, 0, 0), (1, 0, 0, 1), (0, 0, 1, 1), (0, 1, 1, 0), (1, 0, 1, 0), (0, 1, 0, 1)	3
(1, 0, 0, 0), (0, 1, 0, 0), (0, 0, 1, 0), (0, 0, 0, 1)	2
(0,0,0,0)	1

Recognising gate imbalance thus consisted of simply evaluating the degeneracy of  $I_d$  output states. Fig. 4.7(a) shows the protocol we used. We started with all gates at  $V_{GN} = 0$  V, i.e., logic input state (1,1,1,1), output 5. We then swept all gates to  $V_{GN} = -0.5$  V and back to establish logic state (0,0,0,0), output 1. We then systematically ran through each permutation of each of the logic output states 2, 3 and 4. This entailed sweeping a different number of gates depending on the initial and final logic states. The blue traces correspond to three gates being swept with one held constant, red traces correspond to two gates being swept with two held constant, green traces correspond to individual gates being swept with the other three held constant, and the black traces correspond to all four gates swept between (1,1,1,1) and (0,0,0,0).

Before looking at the device result, we needed to understand ideal device behaviour. We did this by simulating the current output  $I_{sim}$  of an ideal four-gate device, shown in Fig. 4.7(b). The simulation used a simple model of the nanowire resistance,  $R = R_0 + \Sigma R_N(V_{GN})$ , where  $R_0$  is the nanowire resistance when all gates were set to 0 V, and  $R_N(V_{GN})$  was the gate voltage-dependent resistance of the nanowire segment beneath each gate, N = 1 - 4. For the data in Fig. 4.7(b), each function  $R_N(V_{GN})$  was an identical linear function of  $V_{GN}$  to simulate equal gate-nanowire coupling. As expected, the  $I_{sim}$  output was driven between the five logic states outlined in Table 4.2. These output states 5 - 1 (top to bottom) are highlighted with the dashed lines.

The measured response from the actual device is given in Fig. 4.7(c). The  $I_d$ output did not consist of five clearly defined output states. This immediately indicates the presence of gate imbalance. The major deviations from ideal behaviour arose from G4. This is most apparent in traces that involved only altering  $V_{G4}$ ; these are the first set of green traces in each series. Sweeping  $V_{G4}$  alone caused a greater deviation from either the (1,1,1,1) state or (0,0,0,0) state than when any of the other gates were swept alone. A related effect was observed when three gates were swept simultaneously (blue traces). The last set in each blue series shows that when  $V_{G4}$  was not swept, the resulting change in current was smaller than the other three sets when  $V_{G4}$  was swept. This suggests that the function  $R_4(V_{G4})$ differs from  $R_N(V_{GN})$  for N = 1, 2, 3. To show this clearly, the simulation was repeated with  $R_1(V_{G1}) = R_2(V_{G2}) = R_3(V_{G3}) = 0.984 \times R_4(V_{G4})$ , and the resulting simulated  $I_{sim}$  plotted in Fig. 4.7(d). This simulation matched actual device performance more closely, suggesting the gate coupling for G4 deviated from the coupling for G1, G2 and G3 by a simple scaling factor 0.984. Firstly this indicates that G1, G2 and G3 were all very well matched gates, and that G4 differed only slightly. This positive result suggests that even better results could be obtained with some optimisation of the fabrication process.<sup>23,296</sup> The technique used here also illustrates the potential to compensate for gate imbalance. The different coupling of G4 could be accounted for by applying the evaluated scaling factor to the relevant gate voltages sourced in subsequent experiments/applications.

#### 4.1.4 Multiple transistors on the same nanowire

The fabrication platform we developed also enabled realisation of two independent wrap-gate transistors on the same nanowire. This device is shown in Fig. 4.8(a).

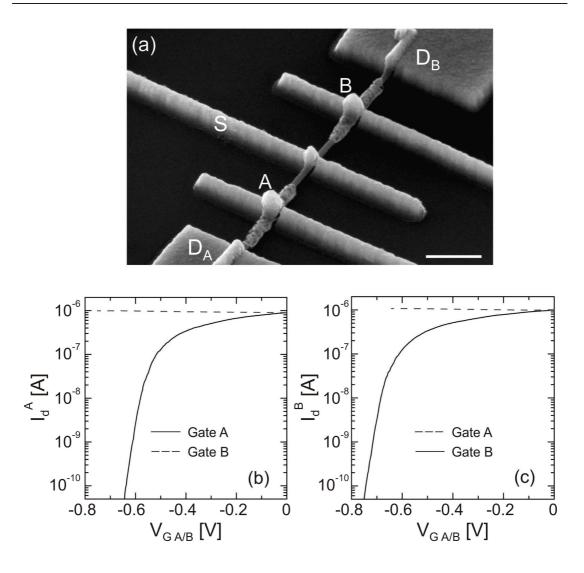


FIGURE 4.8: (a) Scanning electron micrograph of two transistors A and B on the same nanowire with common source contact S and respective drain contacts  $D_A$  and  $D_B$ . (b)  $I_d^A$  for transistor A and (c)  $I_d^B$  for transistor B measured at the drain contact of each transistor in response to applied  $V_{GA}$  or  $V_{GB}$  with  $V_{BG} = 0$  V. Solid traces in each frame represents the current response from sweeping the gate of that particular transistor. Dashed traces show the current obtained when sweeping the gate of the other transistor.

Fabrication differed from previous devices only by the inclusion of a common source contact in the middle of the nanowire. This contact was defined alongside the two drain and gate contacts during the second EBL step. The electrical data in Fig. 4.8(b/c) demonstrates independent operation of the two transistors A and B with  $V_{sd}$  applied to the common source.  $I_d^A$  and  $I_d^B$  were measured at their respective drain contacts using two lock-in amplifiers coupled to the  $V_{sd}$  reference signal. This facilitated simultaneous measurement of  $I_d^A$  and  $I_d^B$  in response to The transfer characteristics of each transistor – solid lines in Fig. 4.8(b/c)) – were very similar to that seen in multiple wrap-gate devices above. The transistors also match each other well, with identical subthreshold swings and a difference in  $V_{th}$ of only ~ 0.15 V. The dashed lines show the  $I_d$  response in each transistor when the gate of the other transistor was swept to deplete that side of the device. In both devices,  $I_d$  rose as the other device was depleted. This was not because the gate was directly influencing the opposing channel. Rather, it is because the two transistors passed current to separate grounded drain contacts in a parallel circuit. Since there is a constant  $V_{sd}$  applied to the source, reducing  $I_d^A$  by application of  $V_{GA}$  necessarily caused more current to flow to ground through Transistor B, and vice versa.

#### 4.1.5 Summary

Multiple wrap-gate NWFETs offer a versatile route to many novel devices and functionalities.<sup>29</sup> The devices can implement two-input AND logic on a single transistor as opposed to the four required by CMOS. The simple fabrication method facilitated three and four wrap-gated devices with the same number of processing steps as two wrap-gates. This contrasts with vertical wrap-gated nanowires where the number of steps scales with the number of gates.<sup>292</sup> Gate control and balance was already satisfactory for most applications, although work could be done in improving the oxide/nanowire interface<sup>23,296</sup> and the quality of sputtered Au. The latter may reduce the extent of gate material removal seen in Fig. 4.3(c). Nevertheless, even devices with poor appearance often provided suitable electrical performance. We presented a method to characterise the performance of potentially damaged gates using screening of an external potential, and a second method to evaluate the electrical balance between gates and compensate for this in future experiments. From here, the multi-gate devices could be utilised for strong electric field control in the generation and manipulation of quantum systems, as well as novel many-input logic systems.<sup>293</sup> Finally, we demonstrated two transistors with a common source on the same nanowire using the same process, which could also be used to facilitate device scaling in 3D architectures.<sup>3,291</sup> The multiple wrapgates could be combined with more complex heterostructure nanowire growth to produce devices with unique functionality that further enhances scalability. For example, similar structures – albeit without wrap-gates – have been fabricated to realise inverters on single nanowires.<sup>297</sup> The stronger confinement and versatility supplied by our multiple wrap-gate fabrication technique has the potential to enhance nanowire development in virtually any electrical application.

### 4.2 Gating nanowires using nanoscale patterned polymer electrolytes

As outlined in Sec. 1.6.1, polymer electrolyte (PE) dielectrics offer an interesting alternative to metal/oxide gates to generate strong electric fields.<sup>8</sup> The strong gating arises from the fact that the mobile ions in the polymer matrix are driven to  $\sim 1$  nm away from the semiconductor channel.<sup>8,203</sup> The mobile ions offer a particular advantage to nanomaterials such as nanowires, as the ions follow the surface morphology of the semiconductor material. When used to gate nanowires, the ions are expected to wrap concentrically around the nanowire. As such, PEs provide an organic, spin cast route to wrap-gating.<sup>28,31,196,219</sup> Furthering the use of PEs with nanomaterials requires patterning them on the nanoscale to facilitate individual device contact and control. This would also open up the possibility of using multiple independent PE gates on the same nanowire. Patterning the PE is desirable in general as it eliminates overlap with the source and drain contact, which reduces parasitic capacitance, leakage currents and contact erosion.<sup>8</sup> Patterning via photolithography,<sup>211</sup> ink-jet printing<sup>193,197</sup> and injection into microfluidic channels<sup>217</sup> has enhanced the use of PE electrolytes in microelectronics. However, the resolution of these methods is insufficient to fit the PE between source/drain contacts on a nanowire, which typically have a 200 nm  $-3 \mu m$  separation. In this Section I

obtained electrical performance comparable to metal/oxide wrap-gate transistors. We also demonstrated two independent gates on a single nanowire using patterned PE dielectrics, which exhibited excellent electrical balance.

#### 4.2.1 Electron beam patterning of PEO/LiClO<sub>4</sub>

Electron beam lithography (EBL) operates by high energy electrons selectively causing the scission or cross-linking of chains in a polymeric resist layer. This alters the solubility of the patterned regions when immersed in developer solutions and causes either the patterned regions to be removed, or the surrounding resist to be removed (see Sec. 2.1.2). PEO – a common polymer backbone for many PEs – functions as a negative-tone EBL resist,<sup>298</sup> where the patterned area remains after development. The polymer chains in PEO are cross-linked on exposure to high energy electrons, making the exposed regions comparatively insoluble in a range of developers including methanol, tetrahydrofuran and  $H_2O$ .<sup>298</sup> PEO has not found extensive application as a resist outside niche biological applications,<sup>299,300</sup> despite the ability to pattern PEO features with dimensions below  $\sim 200$  nm.<sup>299–302</sup> Further, EBL patterning of a salt-doped PE has not previously been investigated. Expanding EBL patterning to salt doped PEO required addressing the concern that the added ions may capture incident electrons, which could potentially prevent cross-linking. This may inhibit pattern formation and/or lower the ionic conductivity, which would be detrimental for device performance. There is some research to suggest EBL should at least not be detrimental to device performance; unpatterned PE films used in battery applications have seen a conductivity enhancement after irradiation with high energy electrons.<sup>303,304</sup>

My initial work towards realising PE-gated NWFETs was to evaluate the viability of EBL patterning and optimise the process. Thin films were produced using the following method.<sup>196,204,217</sup> First, PEO (MW 100 k) and LiClO<sub>4</sub>  $\cdot$  3H<sub>2</sub>O with

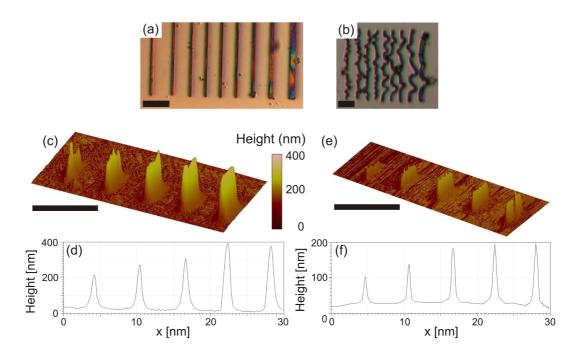


FIGURE 4.9: (a/b) Optical micrographs of 50  $\mu$ m long patterned PEO/LiClO<sub>4</sub> strips with  $w_d = 50, 100, 200, 300, 400, 500, 750, 1000$  and 2000 nm from left to right, with doses (a) 400  $\mu$ C/cm<sup>2</sup> and (b) 100  $\mu$ C/cm<sup>2</sup>. The high dose in (a) resulted in  $w_m$  saturation for  $w_d < 500$  nm. The pattern distortions in (b) were sometimes observed when low doses were used. (c/e) 3D rendered atomic force micrographs and (d/f) cross-sections of 4  $\mu$ m long PEO/LiClO<sub>4</sub> strips with  $w_d = 100$  nm patterned using doses of 50, 100, 200, 300 and 400  $\mu$ C/cm<sup>2</sup> from left to right. The polymer:salt ratios were 10:1 in (c/d) and 2.4:1 in (d/f). Reducing dose at constant  $w_d$  decreased  $w_m$  and feature height.

polymer:salt ratios 10:1, 8:1 or 2.4:1 were dissolved in methanol by sonication for 10 - 15 minutes. The mixture was left standing overnight to precipitate out any large undissolved particulates. The supernatant was spun onto the sample at 4000 rpm for 60 s and the sample was baked at 90°C on a hot-plate for 30 minutes to evaporate the remaining solvent. Preliminary patterning tests were done using the Sirion SEM/EBL tool, with a 5 kV accelerating voltage and typical beam currents ~ 20 pA. Development consisted of a 25 s rinse in H<sub>2</sub>O to remove the unexposed regions of the film, and the sample was dried with N<sub>2</sub> gas.

Figure 4.9 shows optical microscope and atomic force microscope (AFM) images of patterned PEO/LiClO<sub>4</sub> films on Si substrates. These patterns provided initial confirmation that this PE could be patterned by EBL. They were then used to gauge the resolution limits for the PE strips that were ultimately used on the NWFETs. The biggest limit to resolution for negative-tone EBL resists is the proximity effect, whereby secondary electron emission at high accelerating voltages and doses causes cross-linking outside the intended regions. The resultant pattern broadening meant that two line-widths were important for EBL-defined PEO strips. The first is the line-width  $w_d$  defined by the EBL pattern, and the second is the measured PEO line-width,  $w_m$ . We took  $w_m$  as the full-width at half maximum as measured by AFM. Having already chosen a low accelerating voltage, 5 kV, to reduce the proximity effect,<sup>238</sup> we turned to optimising dose within other constraints of the materials system.

Figure 4.9(a) shows one example of how dose dominates pattern definition for a 10:1 PEO/LiClO<sub>4</sub> film. The pattern traced by the electron beam was a series of lines of differing  $w_d = 100 - 2000$  nm, but the resulting  $w_m$  saturated for  $w_d \leq 500$  nm when a dose of 400  $\mu$ C/cm<sup>2</sup> was used. Since  $w_d$  no longer uniquely determined  $w_m$ , further reducing  $w_m$  entailed setting a constant line-width 100 nm and reducing the dose to various values from 400 to 50  $\mu$ C/cm<sup>2</sup>. The resulting pattern is shown in Fig. 4.9(c/d), with the obtained  $w_m = 1.2 \ \mu$ m - 820 nm from right to left.

Two materials limitations prevented further dose reduction. Firstly, the feature height on this sample also fell from 350 nm to 150 nm with decreasing dose. This aspect was somewhat variable, with heights < 100 nm sometimes observed at doses of 100  $\mu$ C/cm<sup>2</sup>. It was important that feature height did not fall further than this to ensure the PEO continuously covered the nanowire. Secondly, low doses < 100  $\mu$ C/cm<sup>2</sup> provided insufficient cross-linking to generate suitable surface adhesion. This occasionally resulted in unrestrained swelling for some sections of the PEO/LiClO<sub>4</sub> during development. The result was pattern distortions like those shown in Fig. 4.9(b). Pattern distortions frequently occurred for patterns with defined widths  $w_d < 100$  nm also. These results pointed towards using  $w_d =$ 100 nm and a dose of 100  $\mu$ C/cm<sup>2</sup> for the 10:1 PEO/LiClO<sub>4</sub> films. When applying these patterns to NWFETs on the  $n^+$ -Si/SiO<sub>2</sub>/HfO<sub>2</sub> substrates, we found that  $w_m$  was further reduced to values as low as ~ 600 nm. This suggested a reduced interaction volume and proximity effect, which has been previously observed for bilayer dielectrics.<sup>305</sup> Further reduction in line widths are likely to be possible if smaller feature heights are viable for a particular device architecture.<sup>301,302</sup>

The final aspect of patterning was the influence of increased salt content. The height and  $w_m$  for features patterned with 8:1 PEO/LiClO<sub>4</sub> films were similar to that for the 10:1 films. However, the increased salt content for the 2.4:1 films resulted in a reduction of feature dimensions, as seen in Fig. 4.9(e/f). This suggests that some electron capture did occur at the expense of cross-linking. The dose needed to be increased by a factor of 8 to obtain the same feature size from 2.4:1 films as for 10:1 films. The concern still was that electron capture may adversely affect device performance through neutralising Li<sup>+</sup> ions. As with the multiple wrap-gate NWFETs, the best way to evaluate electrical properties was with electrical methods from incorporating the PE into NWFETs. The PE line widths  $w_m \sim 600 - 800$  nm were sufficient to fit strips of the dielectric between the contacts of our  $3 - 6 \ \mu m$  long nanowires.

#### 4.2.2 Single PE-gated nanowire transistors

A schematic and AFM image of a NWFET featuring a nanoscale patterned  $PEO/LiClO_4$  dielectric are shown in Fig. 4.10(a/b). The fabrication method proceeded as follows:

- 1. As-grown InAs nanowires were deposited onto the measurement substrate.
- 2. Source, drain and gate contacts were defined in PMMA by EBL.
- The contact metal was 25/75 nm Ni/Au deposited by thermal evaporation. Lift-off in NMP removed all unwanted metal.
- 4.  $PEO/LiClO_4$  thin films were prepared and spun as outlined above.
- 5. Strips were patterned in the PE film to link the gate electrodes and NW using the Raith150-Two. The strips were defined with  $w_d = 100$  nm and dose of 100  $\mu$ C/cm<sup>2</sup>.

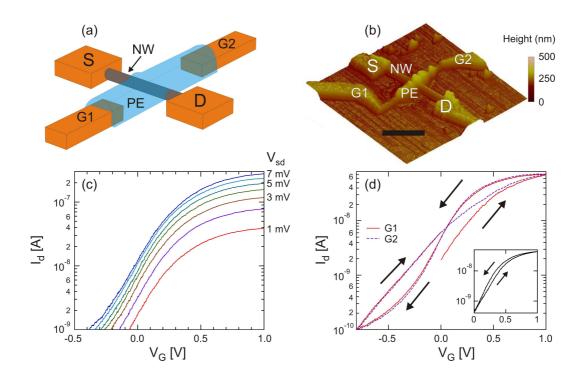


FIGURE 4.10: (a) Schematic and (b) atomic force micrograph of a NWFET featuring a nanoscale patterned PE gate dielectric, showing Ni/Au source (S), drain and gate contacts (G1 and G2). The polymer electrolyte (PE) spans the gap between the gate electrodes and the nanowire. (c)  $I_d$  vs  $V_{G1}$  at increasing  $V_{sd}$  for a PE-gated NWFET. The low subthreshold swing is indicative of effective gating and compares well with metal/oxide wrap-gated NWFETs. (d) Hysteresis for a device with  $s_{G1} = 1 \ \mu m$  and  $s_{G2} = 4 \ \mu m$ . The two time-limited behaviours of surface state trapping and EDL formation generate a complex figure-of-eight loop shape when devices were swept to the full off state. Despite the factor of 4 difference  $s_{G1}$  and  $s_{G2}$ , the electrical characteristics of G1 and G2 were almost identical apart from the 'virgin' behaviour of the initial trace. This is because properties of the EDL do not depend on the thickness of the dielectric (see Fig. 4.11 below). (Inset) Restricting  $V_G$  sweep range avoided the figure-of-eight hysteresis loop.

#### 6. Devices were completed by development of the PE films in $H_2O$ for 25 s.

It is worth noting the low number of fabrication steps compared to metal-oxide wrap-gate fabrication. This process does not rely on a series of specialised etches, and devices are completed simply by developing the PE resist. This makes the above process a powerful method for achieving strong, local electrostatic gating of NWFETs with high yield.

The device in Figs 4.10(a/b) featured two gate electrodes, G1 and G2, connected by a single PE dielectric. The two separate gate electrodes were used to test the dependence of device properties on electrode/NW separation  $s_{GN}$ . The G1/NW separation on all devices with a single PE dielectric was held constant at  $s_{G1} = 1 \ \mu$ m, while  $s_{G2}$  was varied from  $1 - 4 \ \mu$ m. Devices were again studied using the SR830 to supply ac excitation  $V_{sd}$  at 73 Hz at the source and detect  $I_d$  at the drain. Gates were tested to ensure there were negligible leakage currents in the measurement range using a Keithley K2400 SMU, and low-noise Yokogawa GS210 voltage sources were used to increment  $V_{G1}$  or  $V_{G2}$  at 5 mV/s for data collection. Each gate was swept separately with the opposite electrode grounded. This was a matter of convenience; floating the opposing gate electrode had no influence on device performance. All experiments on PE-gated NWFETs in this section were conducted at room temperature in ambient.

Figure 4.10(c) shows the transfer characteristics  $I_d$  vs  $V_{G1}$  for a device featuring a PEO/LiClO<sub>4</sub> dielectric with 10:1 polymer:salt ratio at increasing  $V_{sd} = 1 - 7$  mV. In this figure the gate was swept towards negative  $V_{G1}$ . The strong electrostatic gating supplied by the PE dielectric turned off  $I_d$  within a 1.5 V range, aided by the low SS = 271 mV/dec. Twelve devices featuring dielectrics with 10:1 salt ratio were studied, which gave an average  $SS = 307 \pm 33$  mV/dec and  $V_{th} = 0.16\pm0.06$  V at  $V_{sd} = 2$  mV across the 22 working gates. The average SS compares very favourably to previously studied InAs NWFETs; room temperature SS for metal/oxide wrap-gates is typically 100-750 mV/dec, <sup>22,23,139,140,145,294,295</sup> while the less effective substrate gates typically achieve only SS = 1 - 4 mV/dec. <sup>22,30,145,146</sup> The major advantage of substrate gating is simplicity of fabrication compared to metal/oxide wrap-gates. The patterned PE presented here offers an attractive compromise between the two common architectures; PEs enable strong, local gate control using simple fabrication methods.

A commonly encountered feature in most research-level FETs is gate hysteresis. These devices are no different, with an example of the hysteresis observed shown in Fig. 4.10(d). Interestingly, the hysteresis took a figure-of-eight form rather than a simple clockwise or anti-clockwise loop. This can be explained by the presence of two competing time-limited behaviours. The first contribution was likely to be charge trapping in InAs surface states,  $^{22,23,48}$  similar to the effect of GaAs

surface states on p-AlGaAs/GaAs devices in Chapter 3. For the electron devices here, surface states should cause a clockwise hysteresis loop, as traced out in the discussion surrounding Fig. 1.7 in Sec. 1.3.2. Indeed, this is commonly observed in metal/oxide wrap-gated NWFETs.<sup>22,23</sup>

The second contribution to hysteresis is likely to have arisen from the limited ionic conductivity in the PE, which causes electric double layer (EDL) formation to take a finite amount of time (milliseconds to seconds).<sup>8,203,204</sup> In contrast to surface states, time-limited EDL formation should cause an anti-clockwise hysteresis loop. The rationale for this is as follows. Consider starting at  $V_G = 0$ , with a random equilibrium distribution of ions in the PE dielectric, and initial electron density  $n_i$  contributing to initial current  $I_{d0}$ . Applying negative  $V_G$  drives negative ions towards the NW channel, depleting electrons. Upon sweeping  $V_G$  back to 0 V, the ions in the PE take a finite time to reach the initial equilibrium configuration. This results in a net negative ionic charge in proximity to the channel, with a final electron density and current lower than the initial values, i.e.,  $n_f < n_i$  and  $I_{df} < I_{di}$ . This traces an anti-clockwise hysteresis loop.

The different loop directions and time scales for these two contributions likely resulted in the complex figure-of-eight pattern when the device was swept to the full 'off' state. The figure-of-eight could be negated by limiting sweep range, as demonstrated in the inset to Fig. 4.10(d). The exact parameters varied between devices, but halting the sweep towards depletion in the range  $I_d = 10^{-9} - 10^{-10}$  A was usually sufficient to prevent the figure-of-eight hysteresis. The dynamics of ion transport and surface state charge trapping will be examined in detail as part of Chapter 6.

PE dielectrics possess a number of interesting features that distinguish them from conventional dielectrics. These arise from the differing nature of EDL formation *via* ion transport in PEs and direct field effects from the polarisation of molecular dipoles in conventional dielectrics (recall Fig. 1.25 in Sec. 1.6.1).<sup>8</sup> The NWFETs developed here facilitated demonstration of some of these effects, which also provided confirmation that the gating was due to ionic transport and EDL formation.

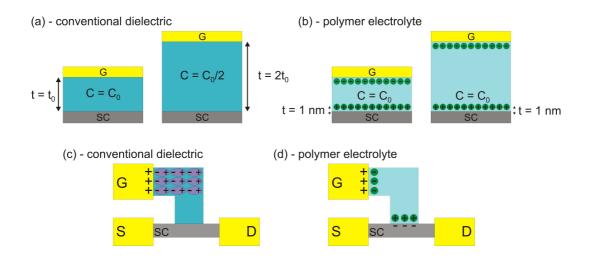


FIGURE 4.11: Comparison between conventional dielectrics and polymer electrolytes when used between a gate G and semiconductor channel, SC. (a) In a conventional dielectric, doubling the layer thickness t halves the dielectric capacitance C. (b) In a polymer electrolyte, the capacitance is a property of the EDL, since this is where the voltage drop occurs. Increasing the separation between the gate and semiconductor does not influence the dielectric capacitance. (c) Gating using a conventional dielectrics is facilitated by molecular polarisation. Gating does not occur if the gate electrode does not align with the semicondcutor channel. (d) Ion transport in polymer electrolytes facilitates very different behaviour. The positive voltage on G drives ion transport and EDL formation even for geometries where the gate electrode is offset from the channel and/or screened by the source/drain contacts.<sup>197,205</sup>

Investigating the interesting features was partially to allay concerns that electron capture at the EBL patterning stage caused neutralisation of ions and that any gating effect was due to direct field coupling through a PEO layer functioning as a conventional dielectric. Closely positioned metal electrodes have been shown to exert electrostatic control over nanowires,<sup>306,307</sup> so it was worth ruling out direct field effects as the major contribution to gate control in these devices. There are four major differences that we used to determine that the dielectrics operated as PEs.

- Effective gating: If EBL neutralised Li<sup>+</sup> ions and gating occurred by direct field effect, the performance of the gates could be expected to be similar to devices with no dielectric at all.
- LiClO<sub>4</sub> concentration: This should strongly influence device performance if EDL formation dominates the gating behaviour.

- 3. NW/gate separation dependence: The capacitance of conventional dielectrics scales with thickness. This means that performance is reduced for increasing NW/gate separation. By contrast, the voltage in a PE drops only across the EDL at the NW surface. This means the capacitance depends only on the properties of the EDL, which are not affected by material thickness (see Fig. 4.11(a/b)).<sup>8,308</sup>
- 4. Novel gate geometries: Conventional dielectrics work only if gate electrode is in the 'line-of-sight' of the transistor. Ion transport in PEs enable a unique 'non-local' gating effect, where the gate potential drives charge to the channel surface even if the electrode and channel are offset (see Fig. 4.11(c/d)).<sup>197,205</sup>

I will now demonstrate the first three features listed here using single PE-gated NWFETs, with novel gating geometries (feature 4) investigated in the next section.

Figure 4.12(a) presents  $I_d$  vs  $V_{G1}$  for representative devices featuring no PE dielectric, and PE dielectrics with polymer:salt ratios 10:1, 8:1 and 2.4:1. The gating effect for the device with no patterned PE dielectric was dramatically diminished. There was a small residual direct field effect, but this only reduced  $I_d$  by a factor of 3 in the same  $V_{G1}$  range where the dielectric with a 10:1 ratio reduced  $I_d$  by a factor of almost  $10^3$ . As for the salt concentration, there was little difference in performance between the dielectrics with ratios 10:1 and 8:1. However, performance was reduced for the device with ratio 2.4:1. These observations were supported by the SS values across a number of devices. For 10:1,  $SS = 307 \pm 33 \text{ mV/dec}$  (as above), 8:1 devices gave  $SS = 286 \pm 45 \text{ mV/dec}$  and devices with 2.4:1 ratios gave  $SS = 431 \pm 53 \text{ mV/dec.}$  These SS values are consistent with previously determined trends for ionic conductivity versus concentration for Li<sup>+</sup> in PEO.<sup>202,308–310</sup> Conductivity tends to increase with salt concentration until it peaks for ratios around 8:1. Further increases in salt concentration causes crystallisation of the host polymer.<sup>8,202</sup> Crystallisation impairs conductivity because ion transport takes place in amorphous regions of the material.<sup>202,303,311</sup>

A practical consideration was that patterning PE films for NWFETs using the low accelerating voltage was sometimes problematic. Obtaining accurate alignment

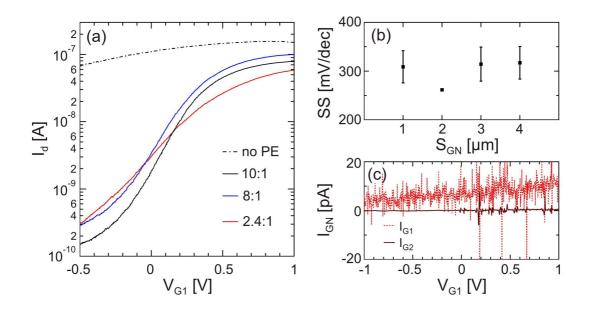


FIGURE 4.12: (a)  $I_d$  vs  $V_G$  for four NWFETs: One with no dielectric between G1 and the NW (dot dash black trace), and one each featuring PE dielectrics with polymer:salt ratios 10:1 (black solid trace), 8:1 (blue solid trace) and 2.4:1 (red solid trace). The much more effective gating offered by devices with PE dielectrics and the dependence of gate characteristics on salt concentration strongly suggests that ion transport dominates over any direct field effects. (b) Subthreshold swing SS as a function of gate/NW separation  $s_{GN}$  for devices featuring PE dielectrics with 10:1 polymer:salt ratio. Error bars represent the standard deviation of each data set. The effectiveness of the gate did not depend on  $s_{GN}$ ; this is because the gate capacitance in PEs depends on properties of the EDL. The value for  $s_{GN} = 2 \ \mu m$  deviated from the trend and had a low associated standard deviation because only two devices with this  $s_{GN}$  worked, and they coincidentally had very similar SS. (c) Current measured at G1 (dotted trace) and G2 (solid trace) for applied voltage  $V_{G1}$ . Both  $I_{G1}$  and  $I_{G2}$  were essentially zero, as no electrical current passes through polymer electrolytes. The clean and accurate signal for  $I_{G2}$  is due to the extremely sensitive measurement capabilities of the K6517 electrometer compared to the K2400 SMU used to measure  $I_{G1}$ .

was achieved by viewing pre-patterned alignment markers underneath the resist layer prior to exposure. This relies on the electron beam penetrating the resist layer, and the backscattered electrons reaching the detector. However, the low accelerating voltage means the electron beam penetration depth and number of backscattered electrons was low. As a result, alignment marker visibility was often impaired, and sometimes the films were entirely opaque to the electron beam. This prevented accurate alignment and significantly reduced device yield. Films with 8:1 polymer:salt ratio were more susceptible to this than films with ratio 10:1, perhaps due to the increased ion concentration. As such, most of this work focussed on dielectrics with salt concentration 10:1, which provided the best compromise between patternability and device performance.

Turning to how the gate/NW separation  $s_{GN}$  affected device performance, an isolated example was actually already given in Fig. 4.10(d). G2 for this device had  $s_{G2} = 4 \ \mu$ m, compared to the  $s_{G1} = 1 \ \mu$ m. Despite the factor of four increase in separation, the two traces were identical. Although not all devices had the same remarkable similarity between gates, on average  $s_{G2}$  did not affect device performance. This is illustrated by plotting SS against  $s_{GN}$  in Fig. 4.12(b). The lack of SS dependence despite gate/NW separations up to 4  $\mu$ m clearly demonstrates that gate capacitance was not affected by the material thickness. This gives strong support to the notion that the gating effect is due to EDL formation by ionic charge transport.<sup>8</sup>

One feature that PEs do have in common with conventional dielectrics is that no electrical current passes through the material under gating.<sup>8</sup> The charge motion through the polymer consists only of ions drifting under the influence of an external electric field, and no charge passes into or out of the metal contacts. This is demonstrated in Fig. 4.12(c), where a Keithley K2400 SMU was used to apply  $V_{G1}$  and measure any leakage current  $I_{G1}$ . A Keithley K6517 electrometer was used to simultaneously measure the current  $I_{G2}$  at G2, on the opposite side of the PE dielectric. Both currents  $I_{G1}$  and  $I_{G2}$  were on the order of pA for the entire  $V_{G1}$  range, with the fluctuations arising due to noise.

#### 4.2.3 Dual PE-gated nanowire transistors

EBL patterning of the PE opened up the possibility of NWFETs with multiple independent gates, linked to the nanowire using separate PE strips. The schematic and AFM image for one of these devices are shown in Fig. 4.13(a/b). Fabrication of this device differed from the single dielectric devices only in the EBL pattern

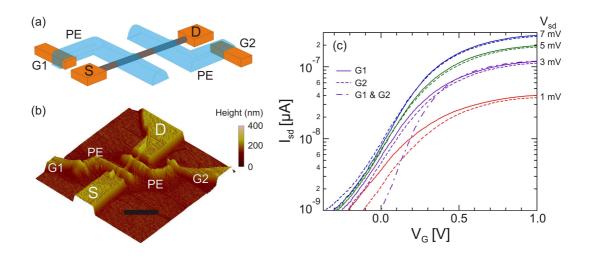


FIGURE 4.13: (a) Schematic and (b) atomic force micrograph NWFET featuring two independent nanoscale patterned PE gate dielectrics, showing Ni/Au source (S), drain and gate contacts (G1 and G2). G1 and G2 were offset to preclude direct field effects influencing the channel; gating in this device operated solely through EDL formation. (c) Transfer characteristics of a dual PE-gated NWFET with  $V_{G1}$  swept and  $V_{G2}$  grounded (solid lines),  $V_{G2}$  swept and  $V_{G1}$  grounded (dashed lines) and  $V_{G1}$  and  $V_{G2}$  swept simultaneously. The two gates are very well matched, with remarkably similar subthreshold properties. The increased depletion when both gates are swept together is indicative of independent operation.

used, with all other steps remaining the same. This device also featured a 'nonlocal' gating geometry, where the metal electrode was physically offset from the channel.<sup>197,205</sup> This geometry eliminated direct field coupling between the gate and NW, and ensured that gating occurred solely *via* ionic transport and subsequent EDL formation.

Transfer chacteristics for G1 and G2 of a dual PE-gated NWFET are shown in Fig. 4.13(c). The electrical properties closely resemble that of the single PE-gated NWFET in Fig. 4.10, supported by the similar SS = 332 mV/dec and 321 mV/decat  $V_{sd} = 2 \text{ mV}$  for G1 and G2, respectively. This strongly highlights the unique properties of PEs – despite a  $s_{GN} \sim 4 \mu \text{m}$  and a non-local geometry, SS for these devices was within experimental error of SS for single PE-gate devices with  $s_{GN} = 1 \mu \text{m}$ . This makes it clear that EDL formation is the major contribution to gating in these structures. Simultaneously sweeping both G1 and G2 improved depletion, with a resulting SS = 192 mV/dec (dot-dash line in Fig. 4.13(c)).

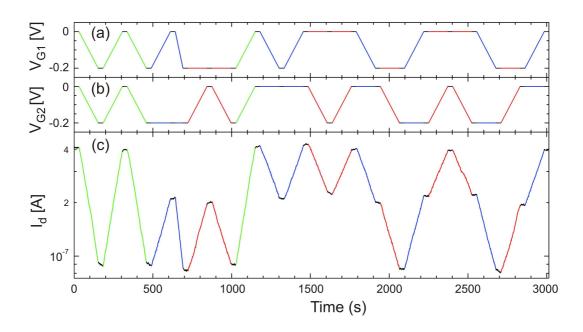


FIGURE 4.14: (a)  $V_{G1}$ , (b)  $V_{G2}$  and (c)  $I_d$  vs time with constant  $V_{sd} = 42 \text{ mV}$ for a logic program with logical 1 set as  $V_{GN} = 0$  V and logical 0 set as  $V_{GN} =$ -0.2 V. Gates were incremented at 2 mV/s together (green traces) or separately, with either  $V_{G1}$  swept and  $V_{G2}$  held constant (blue traces) or  $V_{G2}$  swept and  $V_{G1}$  held constant (red traces). A 30 s wait time was used between sweeps to examine the stability of  $I_d$  output states. The equality of the  $I_d$  output states for input states (1,0) and (0,1) highlights the excellent electrical balance of the gates in this device.

The data in Fig. 4.13(d) and the SS values show that G1 and G2 on this device had remarkably good electrical balance. To further illustrate the gate independence and electrical balance between the gates, we applied the two-gate logic program used previously for metal/oxide wrap-gates in Figs. 4.4 and 4.5. The resulting data for the dual PE-gated NWFET is shown in Fig. 4.14. Each gate voltage  $V_{GN}$  was swept between two pre-defined values – in this case 0 V and -0.2 V – in order to switch each input logic state between every other input logic state (Figs. 4.14(a/b)). Similarly for the metal/oxide wrap-gated device, the  $I_d$  output in Fig. 4.14(c) took distinct values corresponding to the input logic states. The long term stability of the PE-gated device was reduced compared to the wrapgated devices; this is likely due to the increased surface state influence here, which was eliminated for wrap-gated devices by measuring at T = 77 K.

An impressive feature of the data in Fig. 4.14(c) is that the  $I_d$  output was equal for the (1,0) and (0,1) logic states. The degeneracy of these outputs is indicative of identical coupling between the gate and NW for both G1 and G2. This excellent electrical balance was actually observed in the majority of dual PE-gate NWFETs, indicating a high degree of reproducibility. Reproducibility is an important topic in nanoscale devices, because small defects can have a large influence. The reproducibility of the PE-gates far exceeds that of the metal/oxide wrap gates we studied, and is likely a product of the EDL. Since the EDL consists of ions 1 nm away from the semiconductor,<sup>8,203</sup> only defects at the nanowire surface could influence device properties. This contrasts with the metal/oxide wrap-gates where defects in the oxide layer could also contribute to irreproducibility.<sup>13</sup>

#### 4.3 Discussion

In this Chapter I presented two routes to realising strong local electrostatic control of NWFETs with multiple gates. We advanced the use of metal/oxide wrap-gates by developing a highly scalable method to produce NWFETs with 2, 3 and 4 wrap-gates in series along the channel and developed electrical techniques to characterise gate performance. Materials optimisation of these structures would likely lead to GHz operation<sup>296</sup> and deployment in novel logic circuits.<sup>293</sup> I then presented a method to nanoscale pattern the  $PEO/LiClO_4$  polymer electrolyte using EBL, and fabricated NWFETs gated using two independent PE dielectrics. These devices had excellent electrical reproducibility and balance. The major limitation for polymer electrolytes in general remains switching speed. The limited ionic mobility means that switching over large gate ranges is limited to  $f \sim 1 - 100$  Hz.<sup>8</sup> An interesting class of polymer electrolytes is ion gels, <sup>197,212,214,312</sup> which consist of an ionic liquid that has been gelated by a relatively small fraction of polymer (typically < 10% polymer).<sup>214</sup> The comparatively high ionic conductivity that results from such a high ion density facilitates much faster transistor switching.<sup>197,210,219</sup> Investigating the nanoscale patternability of these would be an interesting project, since both PEO<sup>215,313,314</sup> and PMMA<sup>212,315</sup> have been used in different ion gels to provide the polymer backbone. Nevertheless, the patterned PE presented here is

still an excellent choice for the characterisation of nanodevices, due to the ability to generate strong local electric fields with a simple fabrication method.

Over the next two chapters I focus on experiments that illuminate two further applications for PE-gated NWFETs, and how these relate to dopant and surface impurities in InAs nanowires. In the next chapter I present results on using a PE to 'freeze-in' charge states at cryogenic temperatures to enhance the study of quantum systems. This technique allows a range of base-line charge densities to be set in a single device for separate cool-downs, as if the same device had different doping densities for different cool-downs. This is an interesting approach for quantum systems where a low impurity density is required to, e.g., achieve high mobilities. I then present results showing how surface and background impurity charges can dominate device behaviour, and tune the background potential generated by these impurities using the PE dielectric. In Chapter 6 I show that PEO without salt doping can act as a proton-conducting polymer electrolyte, and investigate the competing dynamics of ionic transport and surface state trapping as part of an effort to evaluate the use of III-V NWFETs as proton-to-electron transducers.

## Chapter 5

# Results: Using polymer electrolytes for external doping and setting background disorder potential

#### 5.1 Setting and freezing ionic potential

In this Chapter I show how patterned PE gates can be used to enhance the study of quantum devices at cryogenic temperatures. Transport of Li<sup>+</sup> and ClO<sub>4</sub><sup>-</sup> ions in poly(ethylene oxide) is strongly temperature dependent. The polymer chain relaxations that transport ions are hindered at lower temperatures and ions are immobilised for temperatures  $T \leq 220$  K.<sup>208</sup> This results in a corresponding decrease in dielectric performance, illustrated in Fig. 5.1. The subthreshold swing of this PEO/LiClO<sub>4</sub>-gated NWFET steadily degraded with decreasing temperature, until at T = 220 K the gate characteristics resembled the device with no dielectric in Fig. 4.12(a) of Sec. 4.2.2.

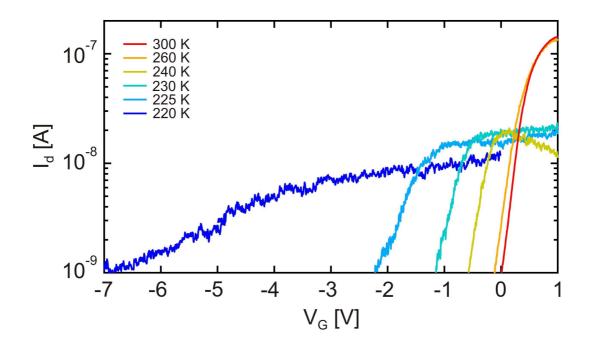


FIGURE 5.1: Current  $I_d$  through a PE-gated NWFET vs gate voltage  $V_G$  applied to the PE gate electrode with decreasing temperature  $T = 300 \rightarrow 220$  K. Lower temperatures inhibit ionic motion and degrade transistor performance.

Using a 'frozen' PE gate in parallel with a second, conventional gate, e.g., a substrate/back-gate, brings added functionality to low temperature measurements in the following way. Cooling below T = 220 K with a non-zero  $V_G$  on the PE gate electrode sets and freezes in a particular non-equilibrium configuration of  $\text{Li}^+/\text{ClO}_4^-$  ions. That is, cooling with positive (negative)  $V_G$  fixes more  $\text{Li}^+$  ( $\text{ClO}_4^-$ ) ions close to the nanowire. Below T = 220 K, the ion configuration constitutes a potential  $V_{PE}$  that is independent of the voltage on the gate electrode  $V_G$ ,<sup>208</sup> the back-gate voltage  $V_{BG}$ , or any other gate in parallel.<sup>30,31</sup> This means that the potential  $V_{PE}$  can be set at room temperature, frozen in place by cooling to cryogenic temperatures, and the resulting system properties probed using  $V_{BG}$ .

Figure 5.2 outlines the basics of this methodology. The device consisted of a NWFET with a PE gate and  $n^+$ -Si substrate as back gate, similar to that used in the previous chapter. Utilising a substrate/back gate becomes a more viable option at low T compared to at room temperature, since conventional metal and degenerately doped semiconductor gates are able to more effectively influence  $I_d$  with reduced T.<sup>1,2</sup> Fig. 5.2(a) illustrates that at T = 300 K, a back gate voltage

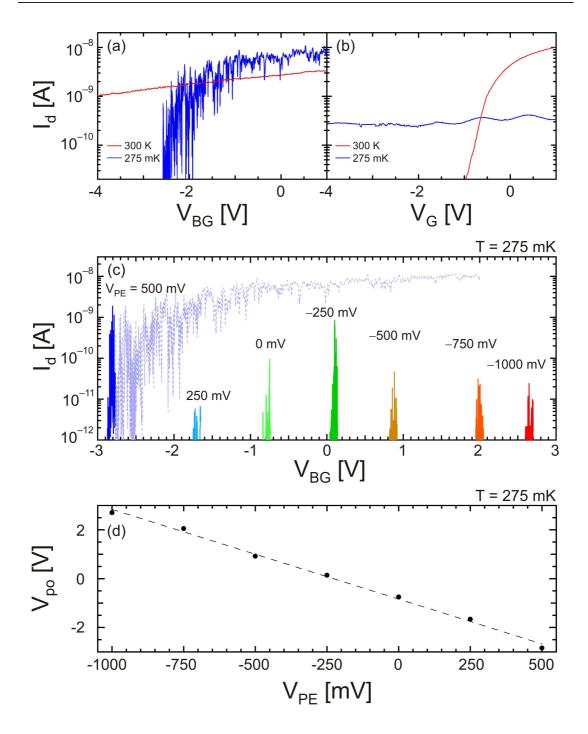


FIGURE 5.2: (a)  $I_d$  vs.  $V_{BG}$  applied to a n<sup>+</sup>-Si backgate and (b)  $I_d$  vs.  $V_G$  applied to a gate electrode with a PE dielectric at T = 300 K and 275 mK. Low temperatures improved the performance of the back gate, but froze ionic motion in the PE. For the T = 275 mK trace in (a), we used  $V_{PE} = 500$  mV to show that low-voltage operation of the back-gate was maintained even at the highest carrier density used in this work. For all other traces in (a/b) each gate was grounded when not in use. (c)  $I_d$  vs  $V_{BG}$  for the range  $V_{po} \leq V_{BG} \leq V_{po}+100$  mV with different frozen in  $V_{PE}$ . The pinch-off voltage  $V_{po}$  is the  $V_{BG}$  where  $I_d = 0$ . Each trace was collected on a separate cooldown, with  $V_{PE}$  set at T = 300 K and held during cooldown. The dashed blue trace shows the full  $V_{BG}$  range; a similar tail was present for all other  $V_{PE}$ , but excluded from the figure for clarity. (d) Setting  $V_{PE}$  within a 1.5 V range at T = 300 K tuned  $V_{po}$  roughly linearly through a 6 V range at T = 275 mK. Using  $V_{PE}$  to alter the  $V_{BG} = 0$  V carrier density constitutes a novel 'external doping' technique.

range  $-4 \text{ V} \leq V_{BG} \leq 1 \text{ V}$  tunes  $I_d$  by only a factor of 3 and is insufficient to pinch off the channel. By contrast, the nanowire can be pinched off within the same range at low T = 275 mK.  $I_d$  exhibits quantum interference effects typically observed in NWFETs and quantum dot resonance signatures near pinchoff,  $I_d = 0 \text{ A}$ .<sup>171,241</sup> The latter is discussed in detail throughout this chapter. The temperature dependence of the PE performance is the reverse of the back gate (Fig. 5.2(b)). The strong gating effect at T = 300 K vanishes at T = 275 mK; the potential  $V_{PE}$  is frozen-in and independent of  $V_G$ . The weak quantum interference oscillations reflect residual direct field coupling between the gate electrode and the nanowire<sup>306,307</sup> across the 1  $\mu$ m of PEO.

The simplest consequence of setting and freezing  $V_{PE}$  is a shift in the back gate threshold voltage and pinch-off point. This is because a positive (negative)  $V_{PE}$ increases (decreases) the electron density at zero back gate voltage  $V_{BG} = 0$  V. The data in Fig. 5.2(c) demonstrates this effect, which was collected using the following methodology. The ion configuration was set by sweeping  $V_G$  at room temperature to the desired  $V_{PE}$ . This voltage was held constant while the device was cooled to T = 275 mK, which locked-in the base-line charge configuration in the nanowire.  $V_G$  was then swept to 0 and the voltage source was disconnected to fully demonstrate the charge freeze-in effect. Transfer characteristics were then collected by sweeping  $V_{BG}$ . Setting the next desired  $V_{PE}$  was accomplished by first warming the cryostat and device to room temperature to allow the ion configuration in the PE to be reset. The device was cooled down with this different  $V_{PE}$ , and a new set of measurements collected at T = 275 mK.

Setting positive voltage  $V_{PE} \ge 0$  V accumulated electrons in the channel, and an increasingly negative  $V_{BG}$  was required to pinch off the nanowire. For negative  $V_{PE} \le -250$  mV the channel was already depleted at low T, requiring positive  $V_{BG}$  to accumulate electrons in the NW. In this way we could use  $V_{PE}$  to tune the back gate operating point linearly across a 6 V range (Fig. 5.2(d)). Using  $V_{PE}$ to set the  $V_{BG} = 0$  V electron density can be thought of as an 'external doping' technique, since  $V_{PE}$  essentially serves the same purpose as dopant atoms. There are a number of advantages to using an external, field-effect technique over dopant impurity atoms. Firstly, different effective doping densities can be set within the same device by simply cooling down with a different  $V_{PE}$ . Secondly, excluding dopants from the channel can be expected to increase the charge carrier mobility, as in modulation doping.<sup>5,71,72</sup> Thirdly, external doping may be useful in situations where doping is difficult or inappropriate. In particular, having greater control over base-line carrier density is useful for InAs nanowires, since they are typically undoped and the electron density is set by properties of the surface accumulation layer.<sup>48</sup>

#### 5.2 Setting the background disorder potential

Our interest in tuning carrier density arose from the work by Heiner Linke's group at Lund University on enhancement of thermoelectric power factor  $S^2\sigma$ .<sup>171</sup> As mentioned in Sec. 1.5.5, they found that background disorder potential in nanowires leads to quantum dot-like states that couple with propagating modes to enhance the thermoelectric power factor over a large carrier density range.<sup>171,176–178</sup> Our aim in this project was to evaluate the suitability of polymer electrolytes for thermoelectric measurements since they offer the dual advantages of strong carrier density tuning and low thermal conductivity. The high thermal conductivity of a metal wrap-gate along the length of the nanowire may 'short-out' an applied thermal gradient. This would dramatically reduce the thermovoltage  $V_T$ , and mean the measurement was dominated by the metal gate, rather than the nanowire itself.

To study the disorder properties and thermoelectric response of PE-gated NWFETs we used the device in Fig. 5.3. It consisted of an InAs nanowire contacted with three ohmic contacts and a polymer electrolyte gate dielectric. The resistive heating strips provided a temperature gradient to generate open circuit thermovoltage  $V_T$ . The  $n^+$ -Si substrate was used as the back gate. Two different measurement set-ups were required to obtain the closed circuit electric current  $I_d$  and open circuit thermovoltage  $V_T$ , as described in Figs 5.3(b/c). Typically we performed  $I_d$ vs  $V_{BG}$  and source-drain bias spectroscopy measurements before switching to the

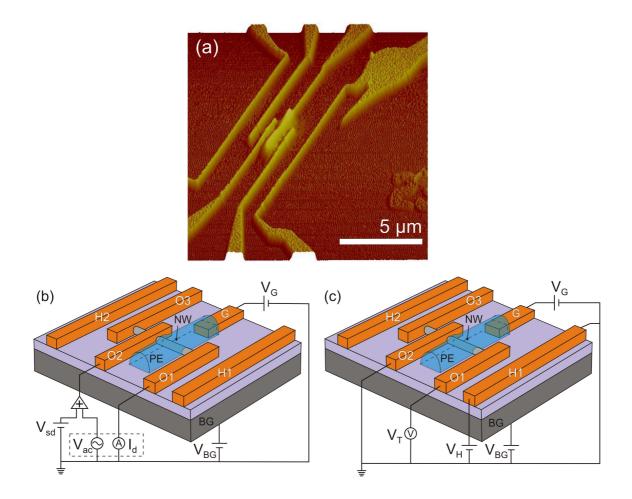


FIGURE 5.3: (a) Atomic force micrograph and (b/c) schematics of the device used in this chapter. Schematics show nanowire (NW), ohmic contacts (O1, O2, O3), heater strips (H1, H2), gate electrode (G), polymer electrolyte (PE) and  $n^+$ -Si substrate back gate (BG). (b)  $I_d$  measurement set-up. A Stanford SR830 lock-in amplifier (represented by the dashed box) sourced an ac excitation  $V_{sd}^{ac} = 100 \ \mu\text{V}$  at 13 Hz and measured  $I_d$ . A dc voltage  $V_{sd}^{dc}$  was supplied by a Yokogawa 7651 voltage source and added to  $V_{ac}$  for source-drain bias spectroscopy measurements using passive circuitry. (c) Open-circuit thermovoltage  $V_T$  measurement set-up. A dc voltage  $V_H$  was applied to heater strip H1, and the associated current flow generated a thermal gradient  $\Delta T$  across the nanowire via Joule heating. The resulting open circuit thermovoltage  $V_T$  was measured between O1 and O2.  $V_T$  was amplified using a Stanford SR560 voltage preamplifier and then read out using a Keithley 2000 multimeter. In both set-ups  $V_G$ was applied to the polymer electrolyte gate at room temperature, and  $V_{BG}$  to the back gate at low T using a Keithley K2400 SMU.

 $V_T$  circuit. After  $V_T$  vs  $V_{BG}$  was obtained for each desired heating voltage  $V_H$ , a final  $I_d$  vs  $V_{BG}$  trace was taken to confirm measurement consistency. Measurements were performed in the Heliox <sup>3</sup>He cryostat described in Sec. 2.4.3. The base temperature was  $T_B \sim 275$  mK throughout, and higher temperatures up to

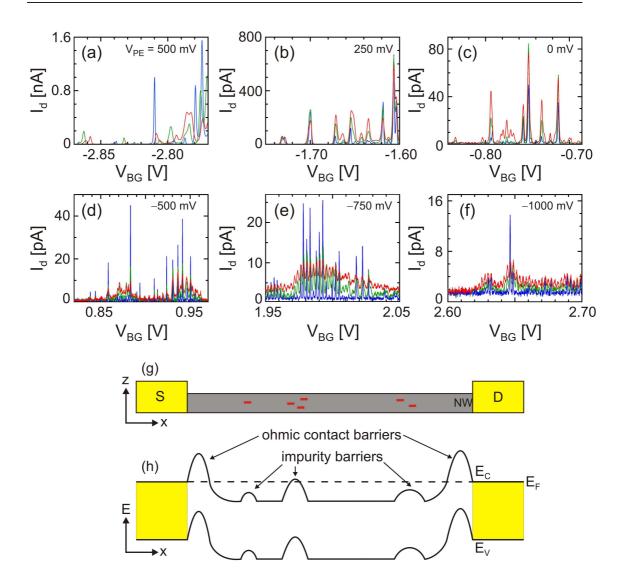


FIGURE 5.4: (a-f)  $I_d$  vs  $V_{BG}$  near pinch-off for increasingly negative  $V_{PE}$  settings with T = 0.275, 1.5, 2.5 K (blue, green and red traces respectively). Coulomb blockade resonances are apparent at all  $V_{PE}$ . (g) Physical schematic and (h) band diagram of a nanowire (NW) with ohmic contacts at the source (S) and drain (D). Strongly confining barriers in the vicinity of the ohmic contacts can arise from distortions of the gate potential. Weaker barriers along the nanowire length arise due to randomly placed ionised impurities, represented as red negative charges in (g).

~ 4.5 K were obtained by carefully controlling the <sup>3</sup>He pot heater and/or the amount of gas between the pot and charcoal sorb.  $V_H$  generated a thermal gradient  $\Delta T$  that produced the open circuit  $V_T$ . It also locally raised the average temperature of the device, even while the cryostat maintained base temperature  $T_B = 275$  mK. We determined the approximate average device temperature due to  $V_H$  by measuring  $I_d$  at each  $V_H$  and comparing the peak heights of quantum dot  $I_d$  resonances with those measured when heating the cryostat. We found that applying  $V_H = 20,60$  and 100 mV corresponded approximately to average device temperatures  $T_D = 0.7, 1.3$  and 2.0 K for the device, respectively.

Figure 5.4 shows  $I_d$  vs  $V_{BG}$  near pinch-off for six  $V_{PE}$  values between +500 and -1000 mV. Traces were taken at T = 275 mK, 1.5 K and 2.5 K obtained by heating the cryostat.  $I_d$  resonances were observed at all  $V_{PE}$ , consistent with previous observations of impurity-related transport barriers that define quantum dots along the nanowire length.  $^{171,175,178}$  The most striking aspect of the data in Fig. 5.4 is that the regularity and robustness of the peaks increased with negative  $V_{PE}$ . The quantum dot signatures for  $V_{PE} \ge 250$  mV were unstable, and often not reproducible for consecutive traces. This is apparent from Fig. 5.4(a), where the system at  $V_{PE} = 500 \text{ mV}$  was not robust enough to produce a reproducible signal across the traces at each T. By contrast, the systems for  $V_{PE} < 0$  mV were more stable; reproducible behaviour was obtained for days at a time while  $V_{BG}$  remained in the measurement range and a low T was maintained. Thermal excitation and electric fields altered the impurity distribution, and therefore the quantum dot resonances, in a similar fashion to the MCF changes observed in the hole billiard presented in Chapter 3. Changing  $V_{BG}$  by more than approx. 1 V and returning to the region of interest and/or separate cool-downs with the same  $V_{PE}$  gave similar overall peak heights and spacing, but with different specific peak positions and heights. This indicates a qualitative reproducibility of each system at set  $V_{PE}$ , but suggests the location and/or strength of the barriers can be altered by external fields and temperature changes.

With no 0D confinement introduced during nanowire synthesis or device fabrication, the question is: what is the origin of the quantum dot system? The Li<sup>+</sup>/ClO<sub>4</sub><sup>-</sup> ions are unlikely to directly provide the disorder potential; a high density of ions closely following the nanowire surface would provide a relatively uniform potential. Rather, we suggest  $V_{PE}$  accentuates the effect of pre-existing potential barriers. In nanowires, potential barriers have been observed to arise both due to the ohmic contacts at either end of the device,<sup>241</sup> and from charged impurities at the surface and in the bulk of the nanowire<sup>171,175,178</sup> (see illustration in Fig. 5.4(g/h)). The barriers near the ohmic contacts arise as the gate potential becomes nonhomogenous in the vicinity of the source/drain contacts, and produces a local electron depletion.<sup>241</sup> Depending on the interplay between the gate/contact potentials, the generated barriers can constitute a quantum dot with well defined confinement dimensions. By contrast, barriers generated by surface charges or ionised impurities in the bulk tend to provide weak, random confinement that changes with the location and charging state of the impurities.<sup>171,175</sup> Combined, Figs 5.4(g/h) show that nanowires at low T provide a highly non-uniform potential landscape for electrons. This example highlights very clearly how the behaviour of nanoscale devices can be dominated by surface and bulk impurities.

The changing nature of  $I_d$  resonances in Fig. 5.4 allows us to hypothesise the origin of each system. Increasingly negative  $V_{PE}$  produced narrower, more closely spaced peaks with maxima that decreased by over two orders of magnitude. The reduced peak height and greater regularity with more negative  $V_{PE}$  points to better defined barriers, while the reduced peak spacing is indicative of a larger dot size.<sup>63</sup> These features suggest the disorder at positive  $V_{PE}$  defines a series of small, weakly defined quantum dots, likely due to random impurity charges at the surface or in the bulk. At negative  $V_{PE}$ , the smaller peak spacing and reduced peak height indicate at least one large dot with well defined barriers. The most likely cause for this is barriers located at the ohmic contacts. The impurity defined barriers between the ohmic contacts then define small, weakly coupled quantum dots that contribute to the complex signatures in Figs. 5.4(d/e/f). Overall, these results show that in addition to setting a base-line carrier density,  $V_{PE}$  can be used to tune the background disorder potential for a low temperature quantum system.

We turn now to a focussed study of the system at  $V_{PE} = -500$  mV, which proved the simplest system to understand. In addition to  $I_d$  measurements, we use sourcedrain bias spectroscopy measurements, a powerful tool for mapping quantum dot energetics (recall Sec. 1.4.3). This allowed us to examine our hypotheses regarding the source of quantum dot-like states in our device. We then looked at the thermovoltage signal. Investigating thermoelectric applications for PEs was doubly interesting since the PE could be used to tune the background potential as well as the carrier density.<sup>171</sup> This is a strong example of how 'non-ideal' properties, i.e., surface charges and bulk impurities can sometimes be harnessed to generate new functionalities.

#### 5.3 Probing the disorder potential

Figure 5.5 shows current, source-drain bias spectroscopy (SDBS) and thermovoltage data for  $V_{PE} = -500$  mV. The  $I_d$  data in Fig. 5.5(a) is the same as in Fig. 5.4(d). SDBS involved the addition of two voltages at the source; an ac component  $V_{sd}^{ac} = 100 \ \mu\text{V}$  at 13 Hz to generate the  $I_d$  signal detected at the drain using the phase sensitive detection of a lock-in amplifier, and a dc bias  $V_{sd}^{dc}$  to split the source/drain chemical potentials.<sup>80</sup> The experiment proceeded by stepping  $V_{BG}$ and sweeping  $V_{sd}^{dc}$ , with the resulting  $I_d$  plotted on a colour scale as a function of both  $V_{BG}$  and  $V_{sd}^{dc}$ . The purple diamonds where  $I_d = 0$  are the transportprohibited regions outlined in Sec. 1.4.3. The high  $I_d$  outside the diamonds (green to red colours) arises since a dot energy level is always between the source and drain potentials. Moving  $V_{BG}$  along the line  $V_{sd}^{dc} \sim 0$  reproduces the resonances in  $I_d$  when the diamonds close.  $V_T$  was obtained in response to applied heater voltages  $V_H = 20, 60, 100 \text{ mV}$ , which provided a small thermal gradient and also raised the average temperature of the device.

The temperature dependence of  $I_d$  peaks in Fig. 5.5(a) gives initial clues to the energetics of the system.<sup>78,316,317</sup> Peaks with the greatest amplitude at T = 275 mK became wider and shorter with increasing T, while the amplitude of suppressed peaks increased with increasing T. The former is observed in the quantum limit where the addition energy is large compared to  $k_B T$ , and transport can only take place via the ground state.<sup>316</sup> The latter behaviour is typically a product of thermally activated transport through excited states where the excited level spacing is small.<sup>316</sup> While both behaviours can be present in a single quantum dot where different levels have different couplings to the leads,<sup>78,316</sup> the observed temperature dependence in Fig. 5.5(a) is also a feature of multiple, serially coupled

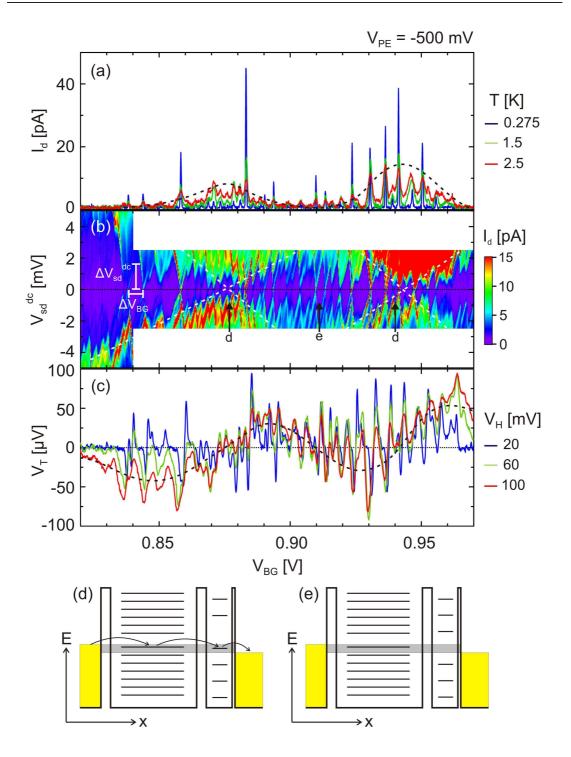


FIGURE 5.5: (a)  $I_d$  vs  $V_{BG}$  at T = 0.275, 1.5, 2.5 K, (b) source-drain bias spectroscopy (SDBS) and (c) thermovoltage  $V_T$  vs  $V_{BG}$  for heater voltages  $V_H = 20,60$  and 100 mV near pinch-off with  $V_{PE} = -500$  mV. The temperature dependence in (a) and the double diamond structure in (b) suggests the nanowire is broken into two serially coupled quantum dots according to the explanation in the text. The background modulation of  $V_T$  at higher  $V_H$  is consistent with this explanation. The underlying  $I_d$  resonances, diamonds and  $V_T$ modulations are indicated by the dashed lines in (a/b/c). Diamond dimensions  $\Delta V_{sd}^{dc}$  and  $\Delta V_{BG}$  are used calculate dot charging energy, capacitance and length. (d/e) Conduction band diagram along the nanowire length x for (d) matched and (e) mismatched dot levels. Electrons traverse the system only when both dots have an energy level within the source/drain window shown in light grey. The data corresponding to the situations (d) and (e) are noted in (b).

quantum dots.<sup>316,317</sup> The latter seems the most likely origin for the data here, given previous evidence for disorder-defined quantum dots in InAs nanowires.<sup>171,175</sup>

SDBS was vital in determining the specific system defined by the background potential at  $V_{PE} = -500$  mV; the data in Fig. 5.5(b) strongly suggests two serially coupled quantum dots. The most striking clue that there were two dots is the background of larger diamonds superimposed on the smaller diamonds, highlighted by the white dashed lines. The size of SDBS diamonds is proportional to the quantum dot charging energy, which is inversely proportional to dot size. Thus, a dot with large (small) physical size generated the small (large) SDBS diamonds. The observed pattern arose because transport through the entire double-dot system is possible only when energy levels from both dots are between the potential of the source and drain. This is illustrated in Figs. 5.5(d/e). When the levels of the two dots are aligned, only a small  $V^{dc}_{sd}$  is required to induce charge transport through the dot and a high  $I_d$  (Fig. 5.5(d)). If the levels are mismatched, a much higher  $V_{sd}^{dc}$  is required to induce high  $I_d$  (Fig. 5.5(e)). Looking back at the data in Fig. 5.5(b), the first case was seen at the points marked 'd'. At these  $V_{BG}$  values, the larger diamonds close, and a low  $V_{sd}^{dc}$  was sufficient to induce high  $I_d$  (red on the colour scale). The situation corresponding to Fig. 5.5(e) is marked with an 'e' in Fig. 5.5(b). At this  $V_{BG}$  setting there was a large mismatch in the dot energy levels, meaning that a significantly higher  $V_{sd}^{dc}$  was required for onset of high  $I_d$ . Between these two limits, the green and red  $I_d$  bands follow a diamond-like shape, highlighted by the white dashed lines.

Using this interpretation, we can return to Fig. 5.5(a) briefly and understand the temperature dependent data in more detail. We previously noted regions of suppressed  $I_d$  peaks that grew with increasing T, and regions of large peaks whose intensity fell with increasing T. This can be understood using the models presented in Figs. 5.5(d/e) in the following way. The regions where large peak heights were observed correspond to the situation in Fig. 5.5(d), where the energy levels are well matched. Increasing T caused the peaks to become shorter and wider, which is the expected behaviour for transport through the ground state.<sup>316</sup> Looking now at the suppressed peaks at T = 275 mK, these occurred in the region corresponding to the situation in Fig. 5.5(e). No current could pass through the device at  $V_{sd}^{dc} = 0$  because the dot levels were misaligned. However, increasing T effectively widens the  $V_{sd}^{dc}$  bias window. This facilitated temperature activated transport through the system, and produced a non-zero  $I_d$ . Similar to the SDBS data, the  $I_d$  peak height was modulated by a background signal highlighted by the black dashed line. It consisted of broad  $I_d$  peaks that align with the vertices for the large diamonds in Fig. 5.5(b). This strongly suggests that the background signal represents the resonances of the smaller quantum dot.

The SDBS data facilitated an estimation of the capacitance  $C_{dot}$ , addition energy  $\Delta E_a$ , and length L of both dots. I start with the larger dot, which gave rise to the smaller diamonds. The varying size of these diamonds is due to orbital effects from the degenerate energy levels.<sup>76</sup> In general, the addition energy  $\Delta E_a = \frac{e^2}{C_{dot}} + \Delta E_{0D}$ . The capacitive contribution  $\frac{e^2}{C_{dot}}$  related to Coulomb blockade is always present due to the repulsive potential of electrons already in the dot. Conversely, the spacing between 0D energy levels  $\Delta E_{0D}$  is only included when the added electron must also start filling the next 0D level. To obtain  $C_{dot}$  we looked at smaller diamonds, which were likely to involve adding electrons at a constant 0D energy level  $E_{0D}$ . Here, the addition energy is simply  $\Delta E_a = e^2/C_{dot}$ . The addition energy is related to the spacing between resonance peaks  $\Delta V_{BG}$  by  $\Delta E_a = \alpha e \Delta V_{BG}$  where the 'lever arm'  $\alpha = C_{BG}/C_{dot}$  scales the gate-dot coupling. Obtaining a measured value of  $\alpha$  is done using the ratio between the  $V_{sd}^{dc}$  and  $V_{BG}$  required to overcome Coulomb blockade  $\alpha = \Delta V_{sd}^{dc}/\Delta V_{BG}$ . This can be read directly from SDBS, illustrated in Fig. 5.5(b).<sup>63,76,316</sup>

Taking an average  $\Delta V_{BG}$  and  $\Delta V_{sd}^{dc}$  from the smallest diamonds gave  $\alpha = 0.37$ ,  $\Delta E_a = 1 \text{ meV}$  and  $C_{dot} = 0.16 \text{ fF}$ . Dot length L was obtained from the coupling between the gate and the dot using  $\alpha = C_{BG}/C_{dot}$  and a cylinder on a plane model  $C_{BG} = \frac{2\pi\epsilon_r\epsilon_0 L}{\cosh^{-1}(1+t/\varrho)}$ .<sup>318,319</sup> Solving for L with effective dielectric constant  $\epsilon_r = 4.2$  and thickness t = 110 nm of the SiO<sub>2</sub>/HfO<sub>2</sub> dual layer and nanowire radius  $\rho = 25$  nm gave a dot length  $L \sim 600$  nm. The same was done for the smaller dot using the single diamond defined by the dashed lines in Fig. 5.5(b). For this we found  $\alpha = 0.05$ ,  $\Delta E_a = 4$  meV,  $C_{dot} = 0.02$  fF and dot length  $L \sim 30$  nm.

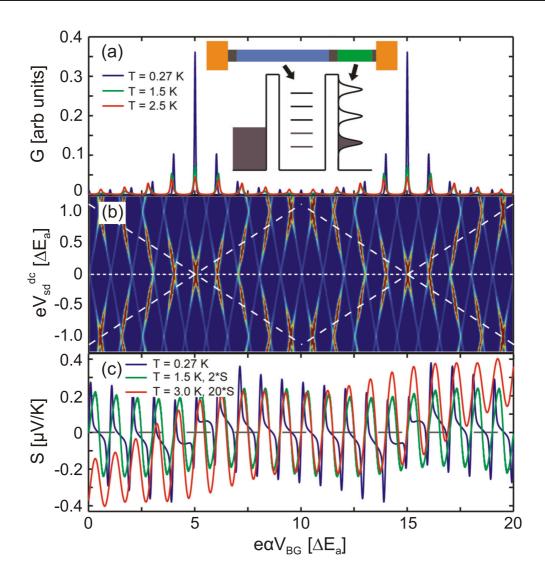


FIGURE 5.6: Theoretical simulations of (a)  $G = I_d/V_{sd}^{ac}$  at T = 0.27, 1.5 and 2.5 K, (b) SDBS with differential conductance on the colour axis where blue (red) is low (high)  $\partial I_d/\partial V_{sd}^{ac}$ , and (c) Seebeck coefficient  $S = -V_T/\Delta T$  at average T = 0.27, 1.5 and 3 K. The voltages were scaled by electron charge e and lever arm  $\alpha$  to be in units of addition energy  $\Delta E_a$ . The model consisted of a nanowire containing two serially coupled quantum dots, shown in the inset to (a). The small dot has a stronger lead coupling, which was included in the description of the lead for the purposes of the simulation. This simple model reproduced the major features of the data in Fig. 5.5. Simulations were done by Martin Leijnse.

However, there is considerable uncertainty in the values for the smaller dot due to the determination of  $\alpha$  from the slope of the dashed lines. The uncertainty arises both from possible error in the placement of the lines and the limited measurement range where quantum dot-like states were observed.

To confirm the double-dot interpretation, theoretical simulations were carried out by Martin Leijnse (Lund University) using the Landauer-Büttiker formula to calculate the conductance G through the system. The data is shown in Fig. 5.6, with full details on the method given in Ref. 30 and the supporting information for that paper. Briefly, the model consisted of two serially coupled quantum dots of different sizes. An equal energy level spacing was used within each dot; these spacings differed between the two dots according to their differing size. The data in Fig. 5.5 also suggested that the smaller dot had a strong coupling to the adjacent lead. The rationale for this is that the effect of increased coupling to the leads is a broadening of the energy levels within the dot.<sup>63</sup> The broad nature of the  $I_d$ resonances associated with the small dot therefore strongly suggest a strong lead coupling for this dot. In Figs 5.5(d/e) the strong coupling was illustrated by the thin barrier between the small dot and the adjacent lead. To keep the theoretical calculation simple, the small dot was included as a modification to the drain lead. The inset of Fig. 5.6(a) shows this resulted in a drain lead consisting of broad energy levels. This encapsulates the same physics as a dot that strongly couples to

the leads. The modelling reproduced all the major features outlined in the data, including the superimposed backgrounds on the  $I_d$  vs  $V_{BG}$  and SDBS data. The simulations were also used to calculate the Seebeck coefficient  $S = -V_T/\Delta T$  and provide insight into our thermovoltage measurements, which I turn to now.

#### 5.4 Understanding the thermovoltage signal

The  $V_T$  vs  $V_{BG}$  signal in Fig. 5.5(c) consists of a series of resonances produced in response to the changing energy levels within the quantum dot system.<sup>179–181,183,185</sup> In a quantum dot, the relative contributions of sequential tunnelling and higher order processes determines the  $V_T$  lineshape (recall Fig. 1.24). Sequential tunnelling typically dominates at higher T, producing a continuous, sawtooth dependence of  $V_T$  on  $V_G$ .<sup>179,181</sup> As T is lowered, higher order tunnelling processes begin to dominate, driving a transition towards a lineshape proportional to the derivative of conductance,  $V_T \sim \frac{1}{G} \frac{\partial G}{\partial V_G}$ .<sup>180,183,185</sup> These consist of single, discrete  $V_T$  oscillations around  $V_T = 0$  that correspond to the  $V_G$  position of  $I_d$  resonances. The derivative-like lineshape dominated the  $V_H = 20$  mV trace in Fig. 5.5(c). It consisted of sharp oscillations, the zeros of which align with the peaks of  $I_d$  resonances in in Fig. 5.5(a). This behaviour is most clear around, e.g.,  $V_{BG} = 0.83 - 0.86$  V. The lineshape of the oscillations changes as increased  $V_H = 60,100$  mV raises the average device temperature. At higher  $V_H$ ,  $V_T$  varies more smoothly with  $V_{BG}$  and takes on the sawtooth lineshape expected at higher T.<sup>181,320</sup>

A second consequence of the higher  $V_H$  and associated average T increase was that  $V_T$  developed a slowly varying vertical offset. At  $V_H = 100$  mV in particular,  $V_T$  oscillations in some regions were either entirely positive or entirely negative. Rather than oscillating about  $V_T = 0$ ,  $V_T$  oscillated approximately about the dashed line in Fig. 5.5(c). The lineshape traced by the dashed line is consistent with the  $V_T$  signal for a smaller quantum dot with strong lead coupling. The positive slope zero crossings at  $V_G = 0.875$  and 0.945 V line up with the inferred  $I_d$  peaks of the smaller dot and the closing of the larger diamonds in Figs 5.5(a) and (b), respectively. The absence of a clearly defined sawtooth can be explained by the stronger lead coupling of the smaller dot washing out the signal, similar to the broad dashed  $I_d$  resonances in Fig. 5.5(a). Again, the major features of the data were reproduced in the simulations shown in Fig. 5.6(c). Importantly, the theoretical calculations assumed the same temperature gradient  $\Delta T$  for all traces. This shows that the  $V_T$  behaviour was due to an increase in average temperature rather than an increased temperature gradient. In all, it is apparent that the  $V_T$ signal for serially coupled quantum dots is simply the addition of the two signals, as for the conductance and SDBS results.

#### 5.5 Disorder potential at other $V_{PE}$

I now touch briefly on measurements made at two other  $V_{PE}$  settings shown in Fig. 5.7. The  $I_d$  data at various  $V_{PE}$  in Fig. 5.4 suggested that for positive  $V_{PE}$ ,

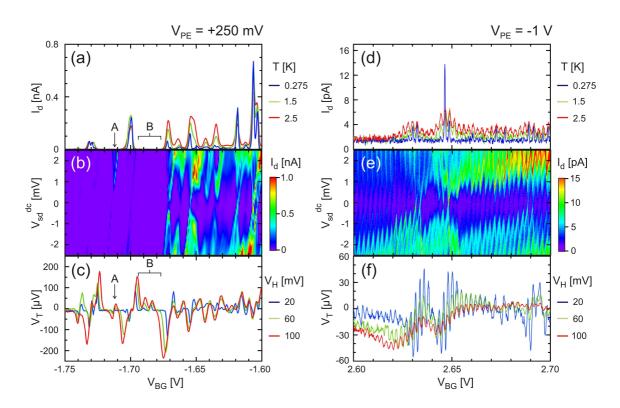


FIGURE 5.7: (a/d)  $I_d$  at T = 0.275, 1.5, 2.5 K, (b/e) source-drain bias spectroscopy (SDBS) and (c/f) thermovoltage  $V_T$  dependence on  $V_{BG}$  near pinchoff for (a/b/c)  $V_{PE} = 250$  mV and (d/e/f)  $V_{PE} = -1$  V. At positive  $V_{PE}$  the nanowire consisted of a series of weakly coupled quantum dots. The system at  $V_{PE} = -1$  V was similar to that observed at  $V_{PE} = -500$  mV, although it is unclear from the data how many dots were present. The markers A and B in (a) and (c) highlight where resonances in  $V_T$  did not have a corresponding clear signal in  $I_d$  or SDBS data. This encourages further work on  $V_T$  as a characterisation tool.

the nanowire was broken into a series of small, weakly defined quantum dots; likely more than two. This is supported by the SDBS and  $V_T$  data for  $V_{PE} = 250$  mV. The irregularly spaced, overlapping diamonds in Fig. 5.7(b) are consistent with multiple dots in series with poorly defined barriers. Unlike the data for  $V_{PE} =$ -500 mV, it is not clear how many quantum dots constitute this system. Similarly, the complex  $V_T$  oscillations in Fig. 5.7(c) are likely to have arisen from the addition of signatures from multiple quantum dots.<sup>321</sup> The  $V_T$  temperature evolution is clearer than it was for  $V_{PE} = -500$  mV, due to the increased peak spacing here. At  $V_H = 20$  mV, the  $V_T$  oscillations follows the derivative of the corresponding  $I_d$  resonance in Fig. 5.7(a). With increasing  $V_H$ , the  $V_T$  lineshape in Fig. 5.7(c) evolves towards a continuous, sawtooth shape as sequential tunnelling dominates transport.

The data in Fig. 5.7(c) also shows an interesting aspect of using  $V_T$  as a characterisation tool. At  $V_H = 100$  mV, strong  $V_T$  signals emerge that do not have clearly visible corresponding  $I_d$  resonances; two examples are at  $V_{BG} = -1.71$  V and -1.685 V, indicated by the letters A and B in Figs 5.7(a/c). Crucially, there is no clear signal in the SDBS data around B either. Even so, the presence of a  $V_T$  signal shows that a quantum dot level passed through the Fermi level at these  $V_{BG}$ . Ordinarily this would result in an  $I_d$  resonance. The absence of a resonance suggests that a mis-alignment of the energy levels between multiple dots prevented electrons from traversing the dot. A large  $V_T$  signal at low G often arises due to the  $\frac{1}{G}$  term in the relationship  $V_T \sim \frac{1}{G} \frac{\partial G}{\partial V_G}$ .<sup>189</sup> However, it is interesting that the large  $V_T$  was prominent at higher temperatures, where the derivative-like line shape may not be applicable.<sup>179,181</sup> A second possibility is that the signals arose because  $V_T$  traces the evolution of dot energy level(s).<sup>179,181,183</sup> It is possible that this may occur even if transport was blocked due to a misalignment between consecutive dot energy levels. Unfortunately, from the available data it is only possible to speculate about the source. Nevertheless, the fact that the  $V_T$  signal remains clear even when there is no  $I_d$  signal or clear SDBS data highlights the potential for thermovoltage as a characterisation technique.<sup>189</sup> Here,  $V_T$  has illuminated the presence and evolution of dot energy states that would have gone undetected using only  $I_d$  and SDBS. Further work using well controlled double-dot systems would be extremely valuable to properly examine this effect.

Going to more negative  $V_{PE}$ , the system at  $V_{PE} = -1$  V in Figs 5.7(d-f) resembles that seen with  $V_{PE} = -500$  mV. The small diamonds and  $\Delta V_{BG}$  oscillations in  $I_d$  and  $V_T$  oscillations arise from a larger dot. This is modulated by the signature of at least one smaller dot. The absence of clear larger diamonds in the SDBS data means it is likely there is more than one smaller dot, but again it is not clear how many are present. The larger dot was approximately 700 nm long, with  $\Delta E_a \sim 0.75$  mV and  $C_{dot} = 0.07$  fF. A similar period of  $\Delta V_{BG}$  oscillation and therefore similar dot size was observed at  $V_{PE} = -750$  mV also (see Fig. 5.4(e) for  $I_d$  data and Ref. 31 for SDBS and  $V_T$ ). The consistency in the larger dot lengths, and the similarity of these values to the source/drain contact separation ~ 850 nm suggests that these dots were defined by the barriers generated in the vicinity of the ohmic contacts. The smaller dots at each  $V_{PE}$  can be attributed to random surface or bulk impurities that shifted with thermal cycling or large applied  $V_{BG}$ . The disorder-induced quantum dots produced the strongest signatures when  $V_{BG}$  was swept to positive voltages. The positive back-gate potential would have induced an electron density close to the nanowire surface like that in Fig. 1.21(d). The fact that the influence of disorder was strongest when the electron density was closest to the surface supports previous suggestions that surface impurities are the dominant source of disorder in these devices.<sup>241,322</sup>

#### 5.6 Discussion

In terms of thermoelectrics, there are two areas of significance for these measurements. Firstly, Fig. 5.5(c) confirms that the PE gate is compatible with thermoelectrics; the  $V_T$  signal took the expected form and had a magnitude on the same order as previous work on quantum dots in nanowires under similar heating powers.<sup>171,180</sup> This ensures the compatibility of PEO for applications involving tuning the disorder potential and setting base-line charge density for thermoelectric power factor enhancement. Secondly, the pronounced background signal in  $V_T$  was much clearer than the background in conductance, providing stronger evidence for the existence of a double-dot system. Thermovoltage is related to the average energy of carriers in the system, and therefore often carries more information than the conductance alone. In principle the amplitude of the Seebeck coefficient in quantum dots gives a measure of charging energy  $\Delta E_a$  and capacitance  $C_{dot}$ .<sup>179,181</sup> Therefore it may be possible to fully characterise quantum dot systems using the Seebeck coefficient  $S = -V_T/\Delta T$ ,<sup>320</sup> since  $C_G$ , and thereby dot size can be obtained from  $I_d$  resonance spacing  $\Delta V_G = e/C_G$ . However, higher order tunnelling effects reduce the S amplitude from the theoretical value as the lineshape moves towards the derivative-like form.<sup>180,182,183,185</sup> This introduces difficulties in a reliably determining  $S.^{323}$  Fully developing a thermovoltage-based characterisation

technique for double quantum dots using a rigorous theoretical understanding of well controlled experimental systems would be an interesting project. Extracting dot energetics from  $V_T$  rather than SDBS would be useful for situations where driving the device out of equilibrium with a large electric field would disrupt the system energetics<sup>306</sup> and affect the ability to correctly probe the system. The other advantage is that  $V_T$  measurements are significantly less time consuming than SDBS measurements; collecting the data in Fig. 5.5(b) took over 10 hours, while collecting a single  $V_T$  trace with the resolution in Fig. 5.5(c) took less than 30 mins. The measurement technique by showing that the thermovoltage remains the straightforward addition of signatures even for multiple quantum dots with different sizes and lead couplings.

More broadly, the device studied here provided an interesting case study along the thematic lines of this thesis. Firstly, we intended to examine the set-and-freeze functionality of the PE gate dielectric as means to exclude dopant atoms from quantum devices. Indeed, the data in Fig. 5.2 showed that this technique could be used to linearly tune the back-gate operating point as if the same device had different doping densities on different cool-downs. While showing this, we also collected data that illustrated the dominant role that surface and bulk impurity charges play in influencing electron transport through InAs nanowires at low temperature. These charges form a background disorder potential that contributes to nanowires presenting as a series of quantum dots states near  $V_{po}$ . The clear, robust quantum dot signatures obtained at many of the  $V_{PE}$  settings highlights the important role of impurities in nanoscale devices. Nevertheless, these 'non-ideal' features can be used to understand new physics and work towards possible thermoelectrics applications.

I now turn to quite different work involving PEO-based dielectrics, yet one where I again encountered the importance of surface states in determining device behaviour.

## Chapter 6

# Results: Protons in poly(ethylene oxide)

#### 6.1 Motivation; a control experiment

Section 4.2.2 presented experiments on PEO/LiClO<sub>4</sub>-gated NWFETs aimed at ensuring the strong gating effect was due to EDL formation *via* ion transport. This included measuring devices with no gate dielectric, dielectrics with differing salt concentration, devices with differing electrode/NW separation, and devices with a non-local gate architecture where direct field effects were excluded. Data from these experiments showed conclusively that ionic motion was responsible for the effective gating in these NWFETs. Data on the temperature dependence of PE gating in Fig. 5.1 supports this conclusion; device performance dropped dramatically when the ions were immobilised at T = 220 K. We did one more experiment not yet mentioned, which was the fabrication of NWFETs featuring patterned PEO dielectrics *without* LiClO<sub>4</sub>. Fabrication of devices proceeded as usual, except for the exclusion of LiClO<sub>4</sub> when mixing the polymer electrolyte solution. The same electron beam parameters were used to pattern the dielectrics - e.g., accelerating voltage, electron dose - since the morphology/dimensions of

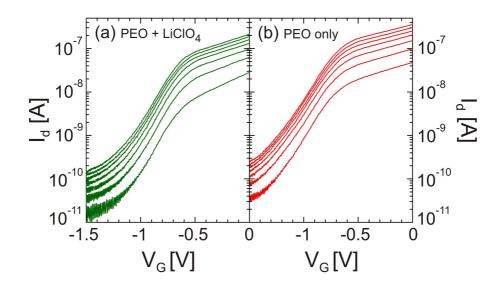


FIGURE 6.1: Transfer characteristics for InAs transistors featuring patterned PEO dielectrics (a) with and (b) without LiClO<sub>4</sub>. Traces are for  $V_{sd} = 1-7 \text{ mV}$  from bottom to top. The characteristics are almost identical; the difference in overall  $I_d$  is within device to device variation.

patterned dielectrics without Li was comparable to those with a polymer:salt ratio of 10:1. Without ions to facilitate EDL formation, device behaviour was expected to resemble that of devices without dielectrics (Fig. 4.12(a) in Sec. 4.2.2), or devices when the ionic motion was frozen out at T = 220 K (Fig. 5.1 in Sec. 5.1). By contrast, experiments showed that the performance of undoped PEO dielectrics was comparable to the performance of devices with  $LiClO_4$ . The data in Fig. 6.1 highlights this equality. The performance of both devices is almost identical, despite the dielectric of one device not containing the  $Li^+/ClO_4^-$  ions that are normally assumed to facilitate such effective gating. Analysis of 16 devices (25 working gates) across 3 separate fabrication batches featuring PEO gates without  $LiClO_4$  gave an average subthreshold swing  $SS = 321 \pm 50 \text{ mV/dec}$  at source-drain voltage  $V_{sd} = 2$  mV in ambient. This is well within experimental error of the value of  $SS = 307 \pm 33$  mV/dec given in Sec. 4.2.2 for devices with LiClO<sub>4</sub>. Naturally, device to device variation produced some differences in e.g. SS, on/off ratio,  $V_{th}$ , hysteresis area, etc, between any two randomly chosen devices, but the average SS values show that gate performance was not hampered by removing the LiClO<sub>4</sub>. Most remarkably, PEO-only dielectrics with a non-local configuration gated the NWFETs. This precluded a direct field effect and gave very strong evidence that

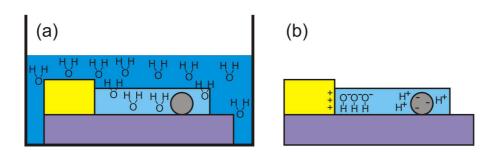


FIGURE 6.2: Illustration of process for H<sub>2</sub>O uptake and transport in PEO films. (a) Cross-section of PEO-gated NWFET during development of patterned PEO film, with grey nanowire, light blue PEO film, gold electrode and purple substrate. The PEO film absorbs H<sub>2</sub>O molecules.<sup>298</sup> A similar process is likely to occur under hydrated atmospheres. (b) Dissociation and of H<sub>2</sub>O and subsequent  $\rm H^+/OH^-$  ion transport under a gate bias alters the electron concentration in the nanowire.

a mobile ionic species was present in PEO films without LiClO<sub>4</sub> doping.

We postulated that  $H_2O$  uptake and subsequent  $H^+/OH^-$  transport was the most likely explanation for the data in Fig. 6.1. Patterned PEO films readily adsorb  $H_2O$ ,<sup>298</sup> and PEO intentionally doped with  $H^+$  exhibits proton conduction.<sup>324–326</sup> There is also small body of evidence for  $H^+$  transport in undoped PEO,  $^{309,325,327,328}$ with transport arising from the same polymer chain relaxation process as for polymer electrolytes.<sup>327–329</sup> These studies suggest native H<sup>+</sup> conductivity in PEO on the order of  $\sigma = 10^{-10} - 10^{-9}$  S/cm.<sup>309,325,327</sup> This is low compared to Li<sup>+</sup> conductivity in PEO at room temperature,  $\sigma = 10^{-7} - 10^{-6}$  S/cm.<sup>303,309–311,330</sup> In our case, the PEO films were exposed to H<sub>2</sub>O during development, and could also have adsorbed  $H_2O$  directly from the atmosphere during measurement (see Fig. 6.2). The departure from previous work is that the equality of transistor characteristics here suggests an equality of ionic conductivity for PEO films with and without  $LiClO_4$  doping. Through the rest of the chapter I focus on reconciling the results presented here with previous work on H<sup>+</sup> transport in PEO. We suggest that  $H_2O$  exposure produced PEO-based polymer electrolytes with an ionic conductivity that matches that of salt-doped PEO-based polymer electrolytes. We then used this feature to develop devices and prototype circuits featuring proton-gated NWFETs that could be used in a platform for biological interfacing.

Before undertaking this investigation there was an obvious question: Is the high conductivity of undoped PEO films due to Li contamination during processing? This was answered by measuring the Li content of representative PEO films using laser ablation inductively coupled plasma mass spectrometry (ICPMS). The ICPMS was performed by Helen Rutlidge at the Mark Wainwright Analytical Centre at UNSW. Laser ablation causes fine particles to be expelled from the surface, which are swept into the analysis chamber. A plasma ionises the particles, whose mass and concentration can be analysed by the mass spectrometer. For this measurement I fabricated two sets of representative films spun onto Si substrates. The films were unpatterned as ICPMS required larger coverage to make an accurate determination. The results gave 2580  $\mu g/g$  Li for the intentionally Li-doped film and 12.6  $\mu$ g/g Li for the film without PEO. The limit of detection was 1.4  $\mu$ g/g. The Li concentration differed significantly; by a factor of over 200. Looking at previous literature, such a large difference means that contamination cannot explain the electrical behaviour. The ICPMS results show the undoped films have an effective PEO:LiClO<sub>4</sub> ratio of 2000:1. Even films with ratio of 100:1 have previously been observed to have four orders of magnitude lower ionic conductivity compared to films with a 10:1 ratio<sup>309</sup> and reducing ion concentration significantly reduces dielectric capacitance.<sup>308</sup> This makes it extremely unlikely that Li contamination can explain the equality of transistor characteristics in Fig. 6.1, and the results that I present throughout the rest of the Chapter.

#### 6.2 Dependence on $H_2O$ concentration

Our hypothesis was that  $H_2O$  adsorption essentially dopes PEO films with  $H^+/OH^$ ions. In this case, additional  $H_2O$  adsorption should increase the ionic concentration in the PEO and improve the ionic conductivity. This in turn should result in more effective EDL formation and improved transistor action. To examine this, we studied the hydration dependence of PEO-gated NWFETs using the humidity controlled measurement chamber outlined in Sec. 2.4.5. The measurements were

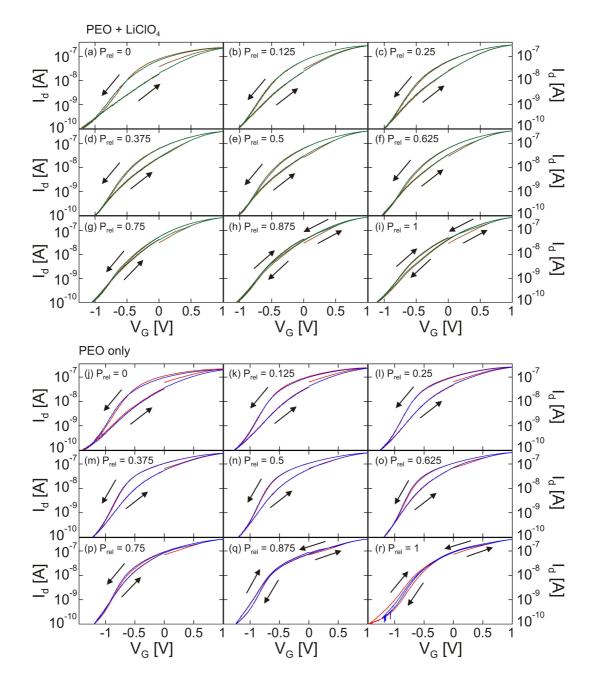


FIGURE 6.3: Hydration dependent transfer characteristics for PEO-gated InAs NWFETs (a-i) with and (j-r) without LiClO<sub>4</sub> doping. Two consecutive traces were taken at  $V_{sd} = 2 \text{ mV}$  (green/brown for doped PEO, red/blue for undoped PEO). Water vapour pressure was increased from P = 0 to 24 mbar in 3 mbar increments, increasing relative vapour pressure  $P_{rel}$  from 0 to 1. Added H<sub>2</sub>O was adsorbed into the PEO dielectric, increasing on current  $I_{on}$ , reducing hysteresis and altering the subthreshold swing SS and threshold voltage  $V_{th}$  (see text).

done at the University of Queensland by myself and Bernard Mostert, who constructed the specialised system. Each panel in Fig. 6.3 shows two consecutive  $I_d$  vs  $V_G$  traces for one device (a-i) with and (j-r) another without LiClO<sub>4</sub>. Note these are not the same devices as those used for Fig. 6.1.

To obtain the data in Fig. 6.3, the following methodology was used:

- 1. The vacuum chamber was pumped overnight using a rotary pump prior to obtaining the data in Fig. 6.3(a/j) at P = 0 mbar, which we defined as P < 0.1 mbar on the 50 mbar gauge.
- Water vapour was introduced by opening the bleed valve between the vacuum chamber and a reservoir containing the degassed, deionised water until the desired P was reached.
- 3. The PEO was allowed to fully absorb the additional water and come to equilibrium. This typically took 20 - 30 minutes and was judged by monitoring the evolution of  $I_d$  during that time.
- 4. Electrical measurements in Fig. 6.3 were done with  $V_{sd} = 2 \text{ mV}$  and  $V_G$  incremented at 5 mV/s.  $V_G$  ran from 0 V to 1 V and then towards negative voltages until  $I_d$  dropped to 0.1 nA before finally returning to  $V_G = 0 \text{ V}$ . The arrows show the direction of each trace. The sweep was stopped at  $I_d = 0.1 \text{ nA}$  to avoid the complex figure-of-eight hysteresis seen in PEO gated devices when swept to their full off state (see Fig. 4.10(d) in Sec. 4.2.2).

Steps 2 - 4 were repeated, incrementing P by 3 mbar each time until saturation pressure  $P_{sat} = 24$  mbar was reached. The presented relative pressure values are normalised to the saturation pressure,  $P_{rel} = P/P_{sat}$ .

As P increases, there are clear changes in four of the transistor properties: subthreshold swing SS, the threshold voltage  $V_{th}$ , on current  $I_{on}$  and, the hysteresis loop area. To properly examine the changes in transistor performance, and the underlying trends in ionic transport I present extracted values for the four transistor parameters in Fig. 6.4. The common scales for y-axes in the left and right

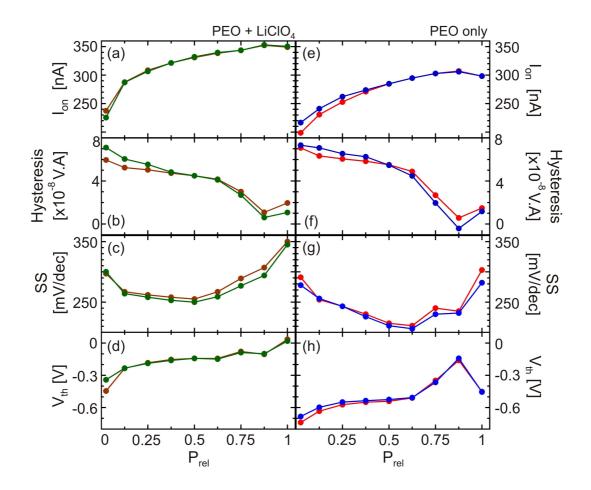


FIGURE 6.4: Extracted transistor characteristics from each of the traces in Fig. 6.3. The colour corresponds to the colour of the trace in Fig. 6.3. Increasing  $P_{rel}$  donated H<sup>+</sup>/OH<sup>-</sup> to PEO films, altering (a/e) on current,  $I_{on}$ , (b/f) hysteresis loop area, (c/g) subthreshold swing SS and (d/h) threshold voltage  $V_{th}$  according to discussion in the text. The data at  $P_{rel} = 1$  tends to counter the trends up to that point. This is likely due to a thin layer of condensed water forming on the surface at saturation vapour pressure affecting device operation.

columns indicates that the transistor properties for both devices fall within a similar range. This confirms the equality of transistor behaviour seen in Fig. 6.3.  $V_{th}$ differed most between these two devices, reflective of a large  $V_{th}$  variation across all devices; a range of up to 0.5 V was generally observed even within fabrication batches. A relatively large  $V_{th}$  variation occurs for metal/oxide wrap gated InAs nanowires also.<sup>145</sup>

Looking first at  $I_{on} = I_d(V_G = 1 \text{ V})$  in Figs 6.4(a/e)), Sweeping  $V_G$  positive caused migration of positive ions to the NW/PEO interface, increasing the electron density within the nanowire and thereby  $I_d$ . This process is limited by the ionic concentration since the electron density is set by the number of ions at the semiconductor/PEO interface. The increase in  $I_{on}$  with increasing P for both devices is therefore consistent with increased H<sup>+</sup>/OH<sup>-</sup> concentration due to water adsorption. The  $V_{th}$  response to increased hydration (Figs 6.4(d/h)) is also consistent with higher ionic concentration.  $V_{th}$  tended to less negative values at higher  $P_{rel}$  as less charge on  $V_G$  was required to produce the same H<sup>+</sup> density at the PEO/NW interface. This resulted in  $I_d$  turning off at lower  $V_G$ .

Similarly, the hysteresis in  $I_d$  vs  $V_G$  is an indicator of ionic mobility in the PEO (Figs 6.4(b/f)). This is because the time taken to establish the EDL is dependent on ionic mobility. The hysteresis loop area falls with increasing  $P_{rel}$ , suggesting a corresponding improvement in ionic mobility. However, using hysteresis to gain quantitative information about conductivity/mobility is problematic due to the contribution to the hysteresis from charge trapping at the InAs surface.<sup>22,23,28,331</sup> The charge trap dynamics depend on the chemistry of the insulator/semiconductor interface and therefore tend to vary for different insulator/semiconductor combinations. Since little is known about the InAs/PEO interface, accurate quantitative information about H<sup>+</sup>/OH<sup>-</sup> mobility cannot be drawn from these measurements. Nevertheless, we determined the ionic conductivity using time dependent relaxation/pulsing measurements discussed in the next section.

The SS response to PEO hydration was more complicated. For low  $P_{rel} < 0.5$ , SS decreased, consistent with the notion that increased ion concentration improves gate efficiency. For  $P_{rel} \ge 0.5$ , SS increased, implying less effective gating. This opposes the trends in the three other device parameters; an increase in  $I_{on}$  and decrease in  $V_{th}$  and hysteresis are all indicators of improved device performance, likely due to added H<sub>2</sub>O. One possible cause of the increased SS is surface state occupation; high surface state charge density is a common source of SS degradation.<sup>1,4</sup> This could arise in two possible ways. Firstly, H<sub>2</sub>O adsorbed on the nanowire surface may directly induce a higher surface state density. I suggest this is *not* the case for InAs devices, as this process should increase hysteresis with increasing  $P_{rel}$ ; the opposite is observed. Secondly, the increased surface charge density could be induced indirectly. The increased ionic density at the NW/PEO

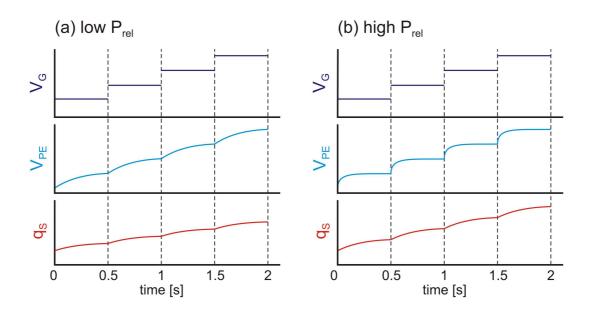


FIGURE 6.5: Illustration of how the measurement protocol affects surface charge  $q_S$  at each  $V_G$  for (a) low  $P_{rel}$  and (b) high  $P_{rel}$ . Measurements were taken every 0.5 s, before  $V_G$  was incremented by 2.5 mV. In (a), the slow ionic response limits the amount of surface charge filling. In (b), the higher ionic conductivity gives a faster  $V_{PE}$  response. This means surface states have longer to fill at a higher  $V_{PE}$ . This increases the surface charge present at each measurement point.

interface may increase the surface state occupation at each measurement point. To understand this, consider the measurement protocol outlined in Fig. 6.5. Each measurement consisted of incrementing  $V_G$  by 2.5 mV, waiting 0.5 s, then collecting  $I_d$  before incrementing  $V_G$  by another 2.5 mV. During this 0.5 s waiting time, the voltage  $V_{PE}$  at the NW/PEO interface builds up as the ions are transferred to the interface. At low  $P_{rel}$ , this occurs slowly, and takes most of the 0.5 s wait time. At high  $P_{rel}$ ,  $V_{PE}$  is established quickly and is stable for most of the 0.5 s wait. Data presented in the next section confirms this. Meanwhile, the amount of charge trapped in surface states is dependent on  $V_{PE}$ ; higher  $V_{PE}$  induces a higher surface charge  $q_s$ . Since surface state filling is a time limited process, holding a particular  $V_{PE}$  for a longer time also increases  $q_S$  as surface states continue to slowly fill at constant  $V_{PE}$ . In all, holding a higher  $V_{PE}$  for a longer time results in a higher  $q_S$ . Looking at the high  $P_{rel}$  case in Fig. 6.5(b),  $V_{PE}$  spends longer at higher values during each measurement cycle than in the low  $P_{rel}$  case (Fig. 6.5(a)). This means a higher surface charge will be induced at higher  $P_{rel}$ . A higher surface charge gives a higher surface state capacitance, which in turn degrades  $SS^{2}$ 

This process would explain why hysteresis continued to improve for  $P_{rel} > 0.5$ while SS degraded. A higher ionic conductivity reduced EDL formation time, and thereby hysteresis, but caused more charge to be trapped in surface states, thereby increasing SS.

The above discussion highlights how the differing temporal dynamics of surface states and ionic migration in these devices can generate complex behaviour for the gate sweeps. The competing dynamics of these two influences also resulted in the complex figure-of-eight hysteresis observed for PE-gated NWFETs when swept to more negative voltages (Fig. 4.10(d) in Sec. 4.2.2). The next section presents results from transient measurements which illuminate and to some extent separate the effects of surface states and ionic transport. These measurements also allowed determination of the ionic conductivity.

#### 6.3 Transient behaviour and ionic conductivity

The dependence on  $P_{rel}$  of  $I_{on}$ ,  $V_{th}$  and hysteresis loop area suggested that that ion concentration n and mobility  $\mu$  improved with added water content. Thus the conductivity  $\sigma = qn\mu$  should also improve. An accurate determination of  $\sigma$ is a vital part of understanding ion dynamics. It is also important for dielectric applications since  $\sigma$  determines how fast the transistor can switch between on and off states.<sup>8</sup> To measure  $\sigma$ , we turned to transient measurements. EDL formation via ionic motion in a PE can be modelled as a resistor and capacitor in series. The effective resistance R accounts for the resistance experienced by the ions as they set up the double-layer capacitance,  $C.^{8,219}$  Applying a voltage to the gate electrode causes a time-dependent build-up of the voltage  $V_{PE}$  at the NW/PEO interface according to  $V_{PE} = V_G e^{-t/\tau_{ion}}$ . The time constant  $\tau_{ion} = RC$  quantifies how quickly the EDL is established. In the region where  $I_d$  is linear in  $V_{PE}$ , we have  $I_d(t) \sim e^{-t/\tau_{ion}}$ . The RC time constant can thus be measured by applying a  $V_G$  pulse in the region  $V_G > V_{th}$  and observing the  $I_d$  response in the nanowire. However, surface states will also influence the transient response, since they also

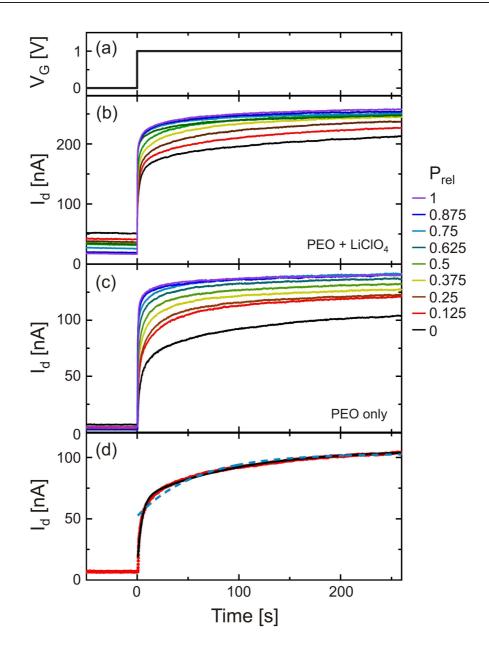


FIGURE 6.6: (a)  $V_G$  pulse to measure  $I_d$  transient response for devices (b) with and (c) without LiClO<sub>4</sub> under increasing  $P_{rel}$ . (d) Data (red dots) for the PEO only device at  $P_{rel} = 0$  with best fits of  $I_d(t) = I_0 + Ae^{-t/\tau_{ion}}$  (blue dotted line) and  $I_d(t) = I_0 + Ae^{-t/\tau_{ion}} + Be^{-t/\tau_{SS}}$  (black solid line).

act as an RC circuit with time constant  $\tau_{SS} = R_{SS}C_{SS}^{39}$  that can be as long as 30 mins for InAs nanowires.<sup>331</sup> In this case, the transient will take the form  $I_d(t) = I_0 + Ae^{-t/\tau_{ion}} + Be^{-t/\tau_{SS}}$ . Finally, a direct field may add some small component to the  $I_d$  transient. To eliminate this aspect only devices with a nonlocal geometry were used for all transient measurements. This ensured that the experiments accurately probed ion/surface dynamics.

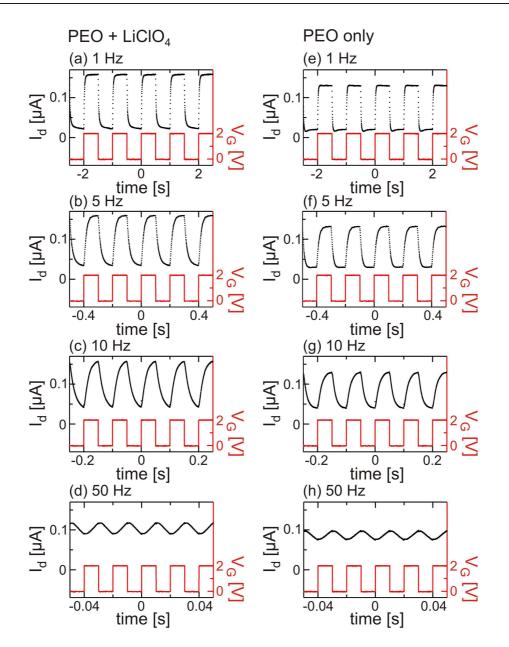


FIGURE 6.7:  $I_d$  transient response (black/left axes) to square pulse trains at different frequency f applied to  $V_G$  (red/right axes) of devices (a-d) with and (e-h) without LiClO<sub>4</sub> at  $P_{rel} = 1$ . Switching the gate faster than the surface state trapping time results in stable operation. For high f > 10 Hz the switching time exceeds the time required for complete EDL formation.

Figure 6.6 shows the result of applying a 1 V pulse to the gate electrodes of NWFETs with non-local PEO dielectrics. Figure 6.6(a) is an illustration of the applied pulse, with the  $I_d$  response for devices with and without LiClO<sub>4</sub> at each  $P_{rel}$  shown in Figs 6.6(b/c). Both devices exhibit a sharp rise in  $I_d$  followed by a longer, slow relaxation. The slow response is on the order of seconds and does not saturate even after 4 minutes, as expected if  $\tau_{SS}$  is on the order of minutes.<sup>331</sup> As

 $P_{rel}$  increases, the overall  $I_d$  increases, consistent with the data for  $I_{on}$  in Figs 6.3 and 6.4. The initial transient response becomes notably faster with increasing  $P_{rel}$ , consistent with increased ionic conductivity. Fits were performed for each device at each  $P_{rel}$ ; an example of the fitted data for the case  $P_{rel} = 0$  is shown in Fig. 6.6(d). As expected, the data (red dots) does not fit a simple exponential  $I_d(t) = I_0 + Ae^{-t/\tau_{ion}}$  (blue dotted line). The fit of  $I_d(t) = I_0 + Ae^{-t/\tau_{ion}} + Be^{-t/\tau_{SS}}$ (black solid line) follows the data more closely. This makes a strong case for the presence of two time limiting behaviours – i.e. ion migration and surface states. However close inspection reveals that the fit is not as close as one might like to reliably extract  $\sigma$  values. This may be due to the fact that assigning a single  $\tau$  for each ionic/surface state component is not always appropriate. Surface states in particular tend to consist of numerous different traps each with different trapping time.<sup>39</sup> Most importantly, fitting multiple exponential terms to a single trace with this many free parameters is bound to lead to errors in the extracted values.<sup>332,333</sup>

To improve the accuracy of  $\sigma$  determination we sought to exclude the influence of surface states altogether. This was accomplished by pulsing  $V_G$  faster than the typical surface state filling time. In this way  $V_G$  would switch before the surface states had time to trap/de-trap charge. We used a function generator to apply square pulse trains to  $V_G$  at varying frequency f.  $I_d$  was amplified by a Femto current pre-amplifier and monitored on a digital signal oscilloscope (DSO). The DSO was also used to directly monitor the  $V_G$  pulse train. The results at  $P_{rel} = 1$ are shown in Fig. 6.7. The  $I_d$  transient response is much improved;  $I_d$  saturates and becomes stable within 0.1 s for positive  $V_G$  pulses. The  $V_G$  pulses to zero produce a less ideal response, perhaps due to the filling of surface states with a short time constant. By increasing the switching frequency to f = 5 Hz, the  $I_d$ transients contain only a single exponential contribution due to ionic motion, and  $I_d$  saturates once the EDL is established. As f is increased beyond 10 Hz, the time between pulses drops below the time required for complete ion separation across the  $\sim 4 \ \mu$ m thick dielectric, cutting off the  $I_d$  response.

Linearising the data proved the best way to extract  $\sigma$ . The  $I_d$  response to pulses of  $V_G$  to 1 V follow the relationship  $I_d(t) = I_0 + I_{sat}(1 - e^{t/\tau_{ion}})$  where  $I_0 = I_d(t = 0)$ 

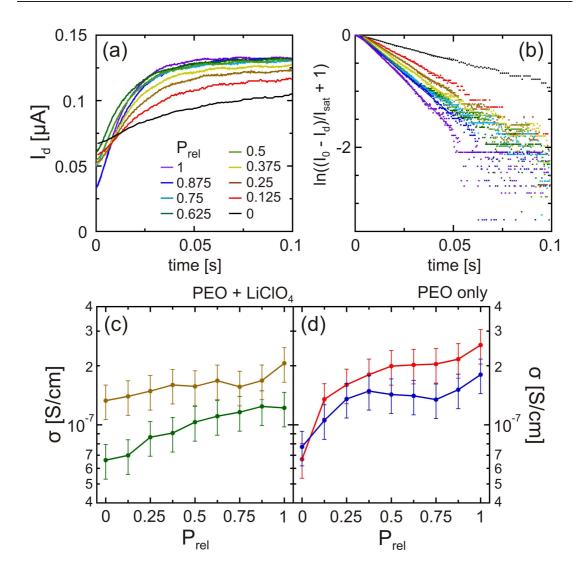


FIGURE 6.8: (a)  $I_d$  transient response from 5 Hz data for a PEO-only gated device at each  $P_{rel}$ . Transient form is close to an exponential that saturates at long t, but deviates from ideal behaviour around t = 0 s when the device switches. (b) Plotting the data (dots) as  $\ln(\frac{I_0-I_d}{I_{sat}}+1)$  vs time t gives a linear trend with slope of  $-1/\tau_{ion}$ ; a steeper slope indicates a shorter time constant. Only the regions that followed a linear trend were fitted (solid lines). (c/d) Ionic conductivity  $\sigma$  extracted from two devices each with and without LiClO<sub>4</sub>. The error bars are largely due to uncertainty in device dimensions. The  $\sigma$  extracted from both sets of devices are within device-to-device variation, and match previously reported values for LiClO<sub>4</sub> doped PEO. Devices without LiClO<sub>4</sub> have a slightly stronger hydration dependence.

and  $I_d$  saturates for long t,  $I_d \rightarrow I_0 + I_{sat}$  (see data in Fig. 6.8(a)). The two constants  $I_0$  and  $I_{sat}$  were collected from the data, and linear plots of  $\ln(\frac{I_0-I_d}{I_{sat}}+1)$ vs t were used to extract the slope  $-1/\tau_{ion}$ . An example is shown in Fig. 6.8(b). This method was more accurate compared to performing an exponential fit to raw data as the curves did not always conform to an ideal exponential at the start and end of each pulse. This is likely due to non-ideal behaviour at the point where ions change direction. At long times,  $I_d$  saturation and the possible influence of surface states causes deviations from ideal behaviour. Linearising the data made it much clearer where the fit should be performed to obtain the most accurate results. Pulses at 5 Hz were used as they typically provided the best compromise between the desire to exclude surface state influence at long times, but also capture an entire  $I_d$  transient response. Once  $\tau_{ion}$  was given from the linear fits,  $\sigma$  could be calculated by  $\tau_{ion} = RC = \frac{l}{\sigma A}C$ . The PEO dielectric length l and cross-sectional area A were measured with atomic force microscopy. C is the capacitance of the electrical double layer at the PEO/NW interface, given by the capacitance for concentric cylinders  $C = \frac{2\pi\epsilon_r\epsilon_0 L}{ln(1+t/\varrho)}$  where  $\epsilon_0$  is the permittivity of free space,  $\epsilon_r = 10$  and t = 1 nm are the dielectric constant and thickness of the EDL, respectively,<sup>8,196,203</sup> L is the length of the nanowire covered by PEO, and  $\varrho$  the nanowire radius.

Figures 6.8(c/d) give the extracted  $\sigma$  as a function of  $P_{rel}$  for two devices with LiClO<sub>4</sub> doped PEO and two devices with undoped PEO dielectrics. The ionic conductivity extracted from both sets of devices is of order  $\sigma = 10^{-7}$  S/cm, consistent with previously reported values for LiClO<sub>4</sub>/PEO polymer electrolyte films at room temperature.<sup>303,309–311,330</sup> This value is much higher than previous measurements of proton conductivity in PEO,  $\sigma = 10^{-9} - 10^{-10}$  S/cm.<sup>309,325,327</sup> The question is: What is the difference in our devices? The mobility  $\mu$  for Li<sup>+</sup> and H<sup>+</sup> in PEO is likely to be very similar, since both species progress by segmental motions of polymer chains.<sup>329</sup> The key to the high H<sup>+</sup> conductivity  $\sigma = qn\mu$  is therefore likely to be a high ionic concentration *n*. There are two aspects that could contribute to the difference between our results and previous work.<sup>309,325,327</sup> First, care was taken in previous work on proton conduction in PEO to actively exclude water during fabrication. Secondly, the parallel plate capacitor geometry typically used to evaluate conductivity has recently been shown to strongly inhibit water uptake in organic materials.<sup>250,251</sup> By contrast, the device was immersed in  $H_2O$  to develop the pattern and the uncovered dielectrics were exposed to a hydrated atmosphere in the measurement chamber. As a result a much increased  $H^+/OH^-$  concentration can be expected compared to previous work.

As for the hydration dependence of  $\sigma$ , there was a general trend to higher  $\sigma$  with increasing  $P_{rel}$ , which was an expected consequence of water adsorption from the atmosphere. Conductivity extracted from devices without LiClO<sub>4</sub> had a slightly stronger hydration dependence than those with Li-doped dielectrics, consistent with Li<sup>+</sup> transport contributing to EDL formation. However, the variation in conductivity is less than an order of magnitude for each device, without a clear, consistent trend. In addition,  $\sigma$  is not zero at vacuum, even when the chamber had been pumped for more than 15 hours. Together these results suggest that the majority of  $H^+/OH^-$  carrier were introduced during processing and were not extracted by vacuum.  $H^+/OH^-$  incorporation could have been during thin film preparation, or at the development stage. Any additional water adsorption in the measurement chamber likely added only a little to the pre-existing  $H^+/OH^$ concentration. The conclusion is that by simply not attempting to exclude water from fabrication it is possible to produce a PEO polymer electrolytes with the same conductivity as intentionally doped films. This may be useful in cases where the high capacitance of polymer electrolytes is required, but salt doping is undesirable.

An interesting feature of the data in Fig. 6.6 is that  $I_d$  increased as surface states filled. This is the opposite of expected behaviour according to the explanation in Sec. 1.3.2 and the data presented in Chapter 3 on AlGaAs/GaAs heterostructures. For an *n*-type semiconductor channel with surface traps, positive  $V_G$  should trap negative charge and compensate the gate voltage. This would give a *reduction* in  $I_d$  over time; this is not observed for these InAs surface traps. The behaviour of InAs nanowire surface traps may be due to the high density of states in the conduction band and the related surface Fermi level pinning. The latter causes a surface accumulation layer which means electron transport is expected to take place close to the surface, rather than in the bulk (see Fig. 6.9(a)).<sup>48</sup> The energetic and spatial proximity of surface states to the conduction channel is likely to result in fundamentally different behaviour to surface states in the band gap. For

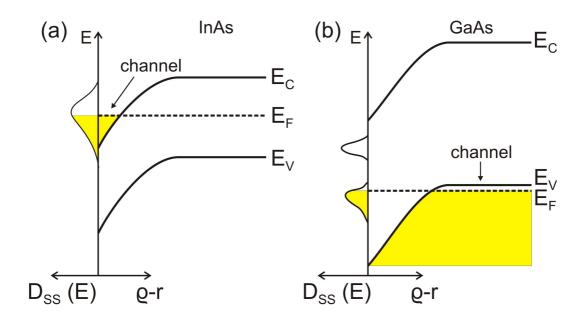


FIGURE 6.9: Surface state density  $D_{SS}(E)$  and radial band diagram for (a) InAs and (b) *p*-doped GaAs nanowires with radius  $\rho$ . The high density of surface states in the conduction band pins the InAs surface Fermi level in the conduction band. This makes undoped InAs nanowires *n*-type, with a conduction channel at the surface of the nanowire. Charge pulled into InAs surface states may further enhance to conduction. The band-gap surface states in GaAs generate a surface barrier, confining the hole conduction channel to the centre of the nanowire. Charge trapping in GaAs surface states opposes an applied gate voltage.

In As nanowires, charge pulled into the surface states may in fact be non-localised and contribute to conduction in the surface accumulation layer. Alternatively, de-trapped charge may fall directly into the conduction band and contribute to conduction here. Either situation would result in increasing  $I_d$  for positive  $V_G$ pulses as observed.

To test this hypothesis, we turned to *p*-GaAs nanowires, which we expected to possess band-gap surface states and a conduction channel that extends to the centre of the nanowire, away from the surface (see Fig. 6.9(b)). In addition to testing the influence of the GaAs surface on device behaviour, utilising a different nanowire material could give an independent measurement of ionic conductivity  $\sigma$ . Using a *p*-type channel also gave us scope to investigate the potential use of III-V nanowires as proton-to-electron transducers and the opportunity to explore complementary amplification/inverting applications.

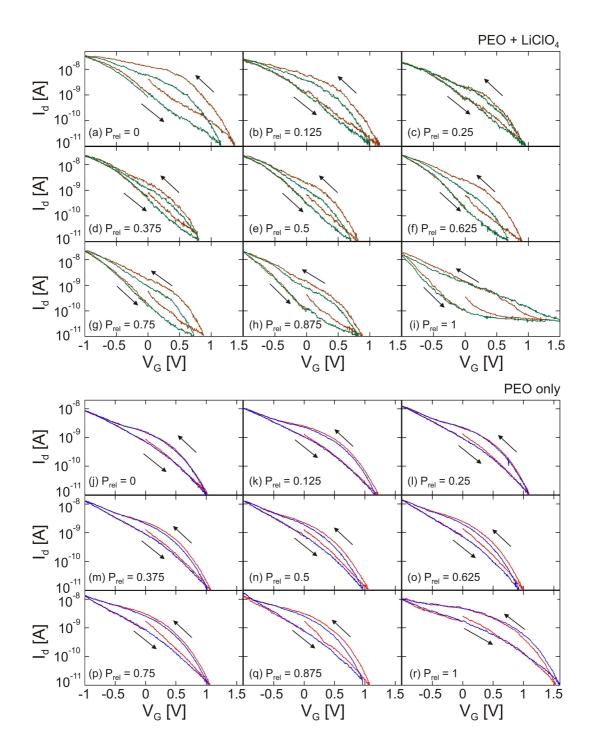


FIGURE 6.10: Hydration dependent transfer characteristics for PEO-gated p-GaAs NWFETs (a-i) with and (j-r) without LiClO<sub>4</sub> doping. Two consecutive traces were taken at  $V_{sd} = 1$  V. The p-GaAs devices have similar operating range and on/off ratios to the *n*-InAs devices. The increasing hysteresis at high  $P_{rel}$  is likely due to an increased contribution from surface states compared to n-InAs devices.

#### 6.4 Complementary *p*-GaAs devices

The basic performance of p-GaAs devices was ascertained by the transfer characteristics shown in Fig. 6.10. The p-GaAs devices are well matched to the n-InAs devices in terms of  $V_G$  operating range ~ 2 V and on/off ratio of  $10^3 - 10^4$ . The correspondence of these two parameters is essential for complementary switching and amplification applications. The overall current for p-GaAs devices is lower because techniques for ohmic contact formation to p-type GaAs nanowires are less mature than those for InAs nanowires. This results in high contact resistance and necessitated measuring p-GaAs devices with  $V_{sd} = 1$  V. Rifat Ullah from our group at UNSW is currently working to improve the linearity and contact resistance of contacts to p-GaAs nanowires as part of his research.

The hydration dependence of the *p*-GaAs transfer characteristics was not as strong as for *n*-InAs devices. Hysteresis tended to reduce up to  $P_{rel} \sim 0.75$  but rose again towards  $P_{rel} = 1$ . Additionally, some GaAs devices became very unstable/noisy at  $P_{rel} = 1$ . Similar instability was not observed for any InAs devices. The difference points to surface effects, possibly related to the formation of a condensed water layer at  $P_{rel} = 1$ . Water adsorption on the GaAs surface may induce a higher density of surface states that increase hysteresis and instability. Since charge trapping in band-gap surface states directly opposes charge accumulation at the semiconductor/PEO interface, a surface state component would drive larger hysteresis loops against the hysteresis-reducing effect of improved ionic conductivity. This means it was not helpful to draw conclusions from these measurements. Similarly to *n*-InAs devices, transient measurements gave a clearer picture of ion and surface dynamics.

Figure 6.11 shows the outcome of a  $V_G = -1$  V pulse to the gate of a *p*-GaAs NWFET featuring PEO dielectrics with and without LiClO<sub>4</sub>.  $V_G$  was pulsed negative rather than positive to induce charge carriers and operate the device in the linear  $I_d$  vs  $V_G$  region, mirroring the *n*-InAs experiments. In contrast to the InAs devices,  $I_d$  falls after the initial step, consistent with the filling of band-gap

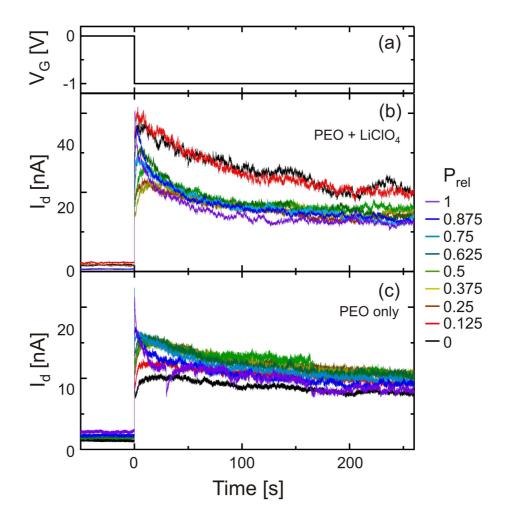


FIGURE 6.11: (a)  $V_G$  pulse to measure  $I_d$  transient response for *p*-GaAs devices featuring PEO gates (b) with and (c) without LiClO<sub>4</sub> under increasing  $P_{rel}$ . Filling of surface states reduces  $I_d$  over time. This component depends on  $P_{rel}$ , and very quickly suppresses  $I_d$  at high  $P_{rel} = 0.875, 1$ . The noise seems to arise from the devices; Fig. 6.10 also has more noise compared to Fig. 6.3.

surface states compensating the gate voltage. The GaAs surface state response is fast compared to InAs surface states, especially for higher hydrations. This supports the suspicion from the  $I_d$  vs  $V_G$  data in Fig. 6.10 that as  $P_{rel}$  approaches 1, a high density of surface states comes to dominate hysteresis. GaAs devices had much higher noise than InAs NWFETs; this was also seen in the transfer characteristics (Figs 6.3 and 6.10). The noise source was not the measurement set-up or a product of the reduced signal, since GaAs and InAs devices measured in the same  $I_d$  range show different noise magnitudes. Rather, it seems specific to GaAs devices and could be due to e.g. spontaneous trapping/de-trapping in surface states.

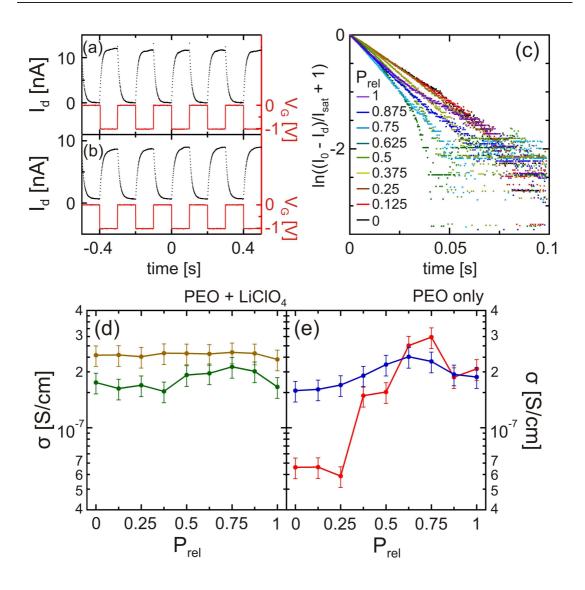


FIGURE 6.12: (a/b)  $I_d$  switching (black traces/left axes) in response to a f = 5 Hz square pulse train applied to  $V_G$  (red traces/right axes) for p-GaAs devices gated by PEO (a) with and (b) without LiClO<sub>4</sub> doping at  $P_{rel} = 1$ . Device stability and noise was greatly improved compared to Fig. 6.11. (c) 5 Hz data from device in (b) at each  $P_{rel}$  plotted as  $\ln(\frac{I_0-I_d}{I_{sat}}+1)$  vs time. Non-linear behaviour at high  $P_{rel}$  is likely due to increased surface state influence. (d/e) Ionic conductivity  $\sigma$  extracted from two devices each with and without LiClO<sub>4</sub>. The error bars are largely due to uncertainty in device dimensions. The extracted  $\sigma$  is on the same order as for *n*-InAs devices, however the trend is likely to be less accurate due to the stronger surface state influence.

We again applied square pulse trains in an attempt to operate faster than the surface traps; f = 5 Hz data is shown in Fig. 6.12(a/b) for dielectrics with and without LiClO<sub>4</sub>. Stability and noise dramatically improved compared to Figs 6.10 and 6.11. The data was taken at  $P_{rel} = 1$ , where the greatest instability was observed. This strongly supports the notion that the noise arose due to trapping/de-trapping of charge in GaAs surface states. Similarly to InAs devices we find that the influence of surface states on device operation can be somewhat mitigated by operating at speeds faster than the trapping/de-trapping times.

The square pulse trains facilitated a second determination of ionic conductivity  $\sigma$ for dielectrics both with and without  $LiClO_4$  doping. However, these efforts were hampered by the comparatively fast acting surface states, especially for higher  $P_{rel}$ . Figure 6.12(c) shows linearised data from a p-GaAs NWFET featuring a PEO dielectric without LiClO<sub>4</sub>. At low  $P_{rel} < 0.75$ , the slope increases with  $P_{rel}$ and the trends are largely linear. For  $P_{rel} > 0.75$ , the data deviates from linear behaviour as rapid filling of surface states suppresses the  $I_d$  transient responses. The curved response in this plot means the original data consists of more than one exponential component. Since the band-gap surface states of GaAs act in opposition to the gate voltage, this resulted in an artificial reduction in  $\tau_{ion}$  and  $\sigma$  for high  $P_{rel}$ . This is apparent on the plots of  $\sigma$  vs  $P_{rel}$  in Figs 6.12(d/e), where  $\sigma$  is suppressed for  $P_{rel} > 0.75$ . However, I stress that this is a measurement artight due to surface states, and not an accurate measurement of  $\sigma$ . Beyond this caveat, the extracted  $\sigma$  values agree with the order of magnitude obtained from In As devices in Fig. 6.8(c/d). Again, the variation for  $\sigma$  extracted from devices with Li-doped dielectrics is smaller than for those with undoped dielectrics. The low dependence of  $\sigma$  on hydration over all prevents meaningful conclusions about ion dynamics, but supports the conclusion drawn from the n-InAs devices that the majority of the  $H^+/OH^-$  ions were introduced during processing.

### 6.5 Circuit/transducer applications

Underlying the above discussion is the conversion of proton signals to electron signals by PEO-gated n-InAs and p-GaAs devices. As explained in Sec. 1.6.2, proton-to-electron transducers are of great interest for biological interfacing, with a number of organic transistors showing promise towards this. Full-scale integration of biological systems with conventional electronics rests on developing biological logic gates and signal amplifiers; ideally using complementary n- and p-type transistors. While organic transistors have the advantages of flexibility and channel permeability, they lack integrability with Si-based microelectronics and have poor electron/hole mobilities. A combined organic/inorganic approach using III-V nanowires has distinct advantages, including growth directly on Si substrates,<sup>129–131</sup> n- and p-type materials with high electron/hole mobilities,<sup>22,334</sup> dimensions that are well matched to cells and proteins,<sup>227</sup> and a high surface-to-volume ratio that lends scope for enhanced detection capabilities compared to thin film devices.<sup>232</sup> With this in mind we investigated the switching capabilities of PEO-based proton-gated n-InAs and p-GaAs NWFETs and their use in a basic inverter circuit.

For operation in a switching circuit, transistors are switched between their full on and off states; i.e. pulses are not limited to the linear  $I_d$  vs  $V_G$  ranges utilised above. Figures 6.3 and 6.10 show both *n*-InAs and *p*-GaAs devices switch between  $I_{on}$  and  $I_{off}$  in the range  $-1 \text{ V} < V_G < +1 \text{ V}$ . To evaluate their performance as switches, voltage pulses between  $V_G = \pm 1$  V were applied, presented in Fig. 6.13. To mimic a wet biological environment the chamber was hydrated to a greater extent than usual. This consisted of leaving the bleed value open for 5-10 mins; far longer than the  $\sim 10 - 20$  s required to reach each  $P_{rel}$  in the preceding experiments. The long hydration time would have ensured condensation on the device substrate and that the PEO films were completely hydrated. The device response in Fig. 6.13 is similar to that in Figs 6.7 and 6.12, with some notable differences. First,  $I_d$  in both devices in Fig. 6.13 drops to  $I_{off} \sim 0$  when turned off. Secondly, the transients do not follow a simple exponential because  $I_d$  vs  $V_G$ is no longer linear in the steady state. At f = 1 Hz, the *p*-GaAs device was very noisy and  $I_d$  in the *n*-InAs device fell after the initial transient, in contrast to the rise in current previously observed. Both effects can likely be attributed to water-induced band-gap surface states. This improved with switching speed, and a stable signal was obtained from both devices up to f = 50 Hz.

Finally, we constructed complementary circuits from n-InAs and p-GaAs PEOgated NWFETs. Figures 6.14(a/b) contain data and the schematic for an inverter

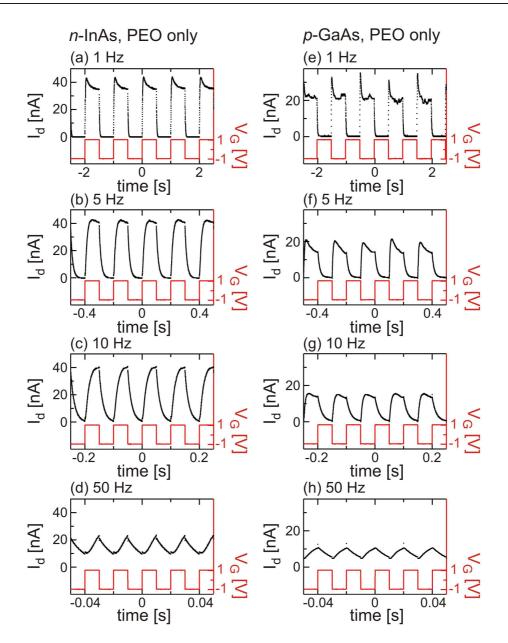


FIGURE 6.13: Square pulse trains  $-1 \text{ V} < V_G < +1 \text{ V}$  at frequency f for (a-d) n-InAs and (e-h) p-GaAs devices with PEO only dielectrics at  $P_{rel} = 1$ . The devices are switched on and off up to 50 Hz.

with the channels of *p*-GaAs and *n*-InAs devices connected in series. The devices were on separate chips in the same chamber at  $P_{rel} = 1$ , connected *via* external circuitry. Ultimately the ideal situation would be to have the devices side-byside on the same chip linked by a common PEO gate. Here, the devices were linked *via* external circuitry as it was the most efficient option for a simple proof of principle given the significant obstacles to placing nanowires side-by-side on the same chip. It is possible to achieve lateral nanowire alignment using trenches

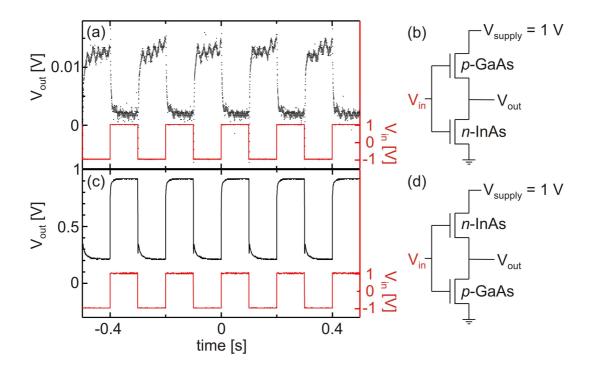


FIGURE 6.14: (a) Data and (b) schematic for an inverter circuit with complementary *n*-InAs and *p*-GaAs devices. The circuit inverts the  $V_{in}$  signal at  $V_{out}$ but gain is severely limited by the poor ohmic contacts to *p*-GaAs. The low signal resulted in visible 50 Hz noise at  $V_{out}$ . (c/d) The circuit was reversed and  $V_{supply}$  instead applied to the source of the *n*-InAs. Gain improved as  $V_{in}$  could swing  $V_{out}$  close to  $V_{supply}$ .

defined in an EBL resist, with a yield of approximately 14% of the trenches filled by a single nanowire.<sup>335,336</sup> Two consecutive steps of this process for the two different materials may result in the desired configuration. However, the low yield for this alignment process means that going through this process for a prototype would not be an efficient use of time. Significantly better alignment yield can be achieved by carefully depositing nanowires from solution onto a substrate with prepatterned electrodes while a sinusoidal voltage is applied across the electrodes.<sup>337</sup> The time cost here would be in establishing a working system to perform this more complex alignment technique. In addition, the poor *p*-GaAs ohmic contacts limit the the inverter performance; optimising these prior to attempting complex device fabrication/application is the more efficient approach.

For the inverter circuit, a dc supply voltage  $V_{supply}$  was applied to the source of the *p*-GaAs. A square pulse train  $V_{in}$  was applied to the gate electrodes of both devices simultaneously and  $V_{out}$  was read out between the *n*-InAs/*p*-GaAs channels by a DSO. Figure 6.14(a) shows that the circuit inverts the  $V_{in}$  signal at  $V_{out}$ ;  $V_{in}$ is high when  $V_{out}$  is low and vice versa. However, the  $V_{out}$  signal is very weak, with a gain of only ~ 0.01. This is due the high *p*-GaAs source/drain contact resistance. Looking back at Fig. 6.10, the applied  $V_{sd} = 1$  V generated current  $I_{on} \sim 10^{-8}$  A and the resistance in the on state is therefore ~ 100 MΩ. The high resistance means that  $V_{out}$  can never reach the supply voltage as much of it will drop over the contacts even when the resistance of the transistor channel is low. To illustrate this,  $V_{supply}$  was instead sourced to the *n*-InAs device drain in Figs 6.14(c/d). This circuit displays excellent fidelity; the  $V_{out}$  signal closely resembles the square wave at  $V_{in}$ . Additionally,  $V_{in} = 1$  V swung  $V_{out}$  close to the supply voltage  $V_{supply} = 1$  V due to the low resistance across the *n*-InAs contacts. For  $V_{in} = -1$  V,  $V_{out} \sim 0.2$  V rather than ground due again to the high contact resistance of the *p*-GaAs nanowires.

In terms of the potential for applications, the data presented here highlights that III-V semiconductor nanowires could be very useful elements in biological interfacing. They can be gated by the field effect from proton/ion signals with operating range  $V_G \pm 1$  V and the high electron/hole mobility means that fidelity and switching speed is only limited by the biological/organic material. Stability over long time-scales may be compromised by surface states, but operation at even f = 1 Hz nullified this influence as the pulse time is insufficient to allow charge trapping/detrapping.

There are some areas for improvement in future work. Most notably, the maximal switching speed remains around 50 Hz, limited by the ionic conductivity and device geometry. These devices have a non-local configuration; the gate electrode/nanowire separation is ~ 4  $\mu$ m, meaning it takes a long time to establish the EDL. Device switching would occur much faster for thinner layers. One application would be sensing electrical signals from biological sources, e.g., cells/neurons. In this case the biological material would sit on top of the PEO, 100 nm away from the nanowire. Given 100 nm of PEO at a proton conductivity of  $10^{-7}$  S/cm, one could expect a response time  $\tau_{ion} \sim 0.6$  ms, and therefore switching speeds exceeding 1 kHz. An alternative would be to utilise materials with higher proton conductivity such as maleic chitosan<sup>220,223</sup> or Nafion.<sup>338</sup> However, it is not clear whether either of these materials could be patterned at the nanoscale to fit between the nanowire source/drain contacts. Nanoscale patternability is a major advantage for PEO here.

Improving inverter gain will largely rest on developing low resistance ohmic contacts to p-GaAs nanowires. Unfortunately this process is not trivial for nanowires; their structure renders them susceptible to damage and destruction under annealing. This places constraints on annealing temperature and time that are absent for traditional planar structures. For example, ohmic contacts to p-type heterostructures with AuBe contacts required annealing to temperatures of 490°C. This temperature completely destroys GaAs nanowires. Increasing the doping in the nanowire itself gives better contact formation, but comes at the expense of gate controllability. A high  $V_G > 10$  V was required to turn off more highly doped p-GaAs nanowires; this is clearly not optimal for low voltage operation. Juggling all these variables and finding the right compromise constitutes a significant project in itself. Also of concern is that GaAs surface states caused increased noise and instability at higher  $P_{rel}$ , although this may not affect high speed operation where switching occurs on a time-scale shorter than that of the trapping/de-trapping time. Development of p-GaAs nanowires is at an earlier stage compared to InAs nanowires, so a concerted optimisation programme may well resolve these issues. If not, one option is to turn to alternative, more developed nanowire materials, e.g.  $InP^{19,144}$  or *p*-InAs.<sup>339</sup>

These problems aside, the data presented in this chapter makes a strong case for development of III-V nanowires towards a biological interfacing platform that could connect the organic, biological world with traditional inorganic microelectronics. Parallel to this, the discovery of relatively high proton conductivity in nanoscale patterned PEO films opens up the opportunity to use this dielectric as a polymer electrolyte without the need for additional salt doping.

## Chapter 7

## Discussion and conclusions

This thesis has centred on the impact of the simple concept of the field-effect transistor, and how devices based on it permeate a wide array of research areas. The drive for new, more efficient technologies has steered research towards III-V semiconductors,  $^{3,4,12}$  quantum devices<sup>5-7</sup> and organic materials.<sup>8-11</sup> I explored each of these areas in this thesis and showed how their overlap can produce novel, exciting devices, such as NWFETs with organic gates. I also showed many of the challenges in developing these new devices towards potential applications. Issues due to dopant impurities and surface states repeatedly arise in nanoscale device development, due to the increased probability of inhomogeneous dopant distributions for smaller device sizes,  $^{13-15}$  and the effective increase in surface-to-volume ratio bringing the surface closer to the channel.<sup>4,21</sup> Understanding the physics behind surfaces and dopants constitutes interesting science in itself, and is also vital to establishing the future utility of emergent devices and realising new functionalities.

The issues arising from surfaces and dopants were most prominent in the study of p-AlGaAs/GaAs heterostructure devices in Chapter 3. Despite the strong similarities to the hugely successful n-AlGaAs/GaAs heterostructures, a simple change in crystal orientation from (100) to (311)A brought about a significant change in electrical behaviour. The differences in surface states and dopant impurity energetics rendered the *p*-type heterostructures prone to hysteresis and instability.<sup>24</sup> Examining the influence of the surface contribution highlighted the importance of the mixed valency of the (311)A GaAs surface and suggests that effective passivation for this surface may prove more difficult than for (100) GaAs surfaces.<sup>32</sup> Similarly, focussing on charge dynamics in the dopant layer revealed that shallow Si-acceptor traps cause instability as charges migrate between trapping sites.<sup>33</sup> By comparison, the stability of *n*-type devices is likely to be due – at least in part – to the deep DX centre charge traps in the AlGaAs layer. This encourages work on finding a stable *p*-type dopant species for (100)-oriented *p*-AlGaAs/GaAs heterostructures, where surface states may not contribute to hysteresis.

Nanowires are an interesting case study on the caveats of increased surface-tovolume ratio. On the one hand, their geometry makes them naturally suited to wrap-gating for the most effective electric field control.<sup>145</sup> The multiple wrap-gated NWFETs developed in Chapter 4 hold promise towards strongly confined quantum systems, e.g., quantum dots, and the scalability power of the fabrication technique encourages their use in dense, 3D architectures.<sup>29</sup> On the other hand, the increased surface-to-volume ratio means that the non-ideal III-V surface necessarily strongly influences carrier transport. This was especially true for the device in Chapter 5, where surface charge likely contributed to potential barriers that defined a series of quantum dots along the nanowire length.<sup>30</sup> However, these impurities, along with the background bulk impurities that also contribute, can be used to realise new functionality in the form of nanoscale thermoelectric elements. In these, the coupling between the quantum dot-like states and propagating modes enhances the thermoelectric efficiency across a wide range of gate voltages.<sup>171</sup> Harnessing this effect reliably and predictably will likely proceed by the study of devices with controlled potential barriers via, e.g., heterostructure barrier growth or multiple wrap-gates.

Surface states also causes instability for the NWFETs measured at room temperature. This included the PE-gated NWFETs studied in Chapters 4 and 6.<sup>28</sup> I showed that while surface states contributed significantly to hysteresis in the steady state transfer characteristics for PE-gated NWFETs, this influence could be negated by switching the gate faster than the typical charge trapping time. The switching speed limit was then determined by the ionic conductivity in the polymer electrolyte. Improving switching speed through geometries where the gate electrode is closer to the nanowire, or exploring nanoscale patterning of ion-gels with a high ionic conductivity<sup>197,210,219</sup> would be the next step for future research. The results in Chapter 6 also constituted a proof-of-principle operation for III-V nanowires in complementary proton-to-electron transducer circuits.<sup>11</sup> Optimisation of the ohmic contacts to *p*-type nanowires will be central to the future success of such a programme, along with the switching speed gains outlined above.

Together, these results illustrate the importance of cutting edge semiconductor research, and how it simultaneously drives and responds to the cycle of development of new technologies through materials study and optimisation, and the subsequent realisation of new device functionalities. This thesis also shows how these devices are at the centre of a huge range of research programmes, from fundamental studies of quantum systems, to interfacing with biological environments. In all, it shows that non-ideal materials behaviour brings both challenges and opportunities in the advancement of our scientific knowledge and the continual quest for a more energy efficient world.

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